

Solid State Chemistry

CHEM-E4155 (5 cr)

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Solid state chemistry

- Synthesis, structures, properties, and applications of crystalline inorganic materials
- **Atomic-level structure** of materials is at the very heart of solid state chemistry
- *“If you want to understand function, study structure”*
 - **Francis Crick** (1962 Nobel Prize in Physiology or Medicine – Structure of DNA)
- The classification illustrated below is a rather traditional one: concepts such as **metal-organic frameworks** bring organic/organometallic chemistry into the picture
- What is **your** background? Chemistry, materials science, something else?

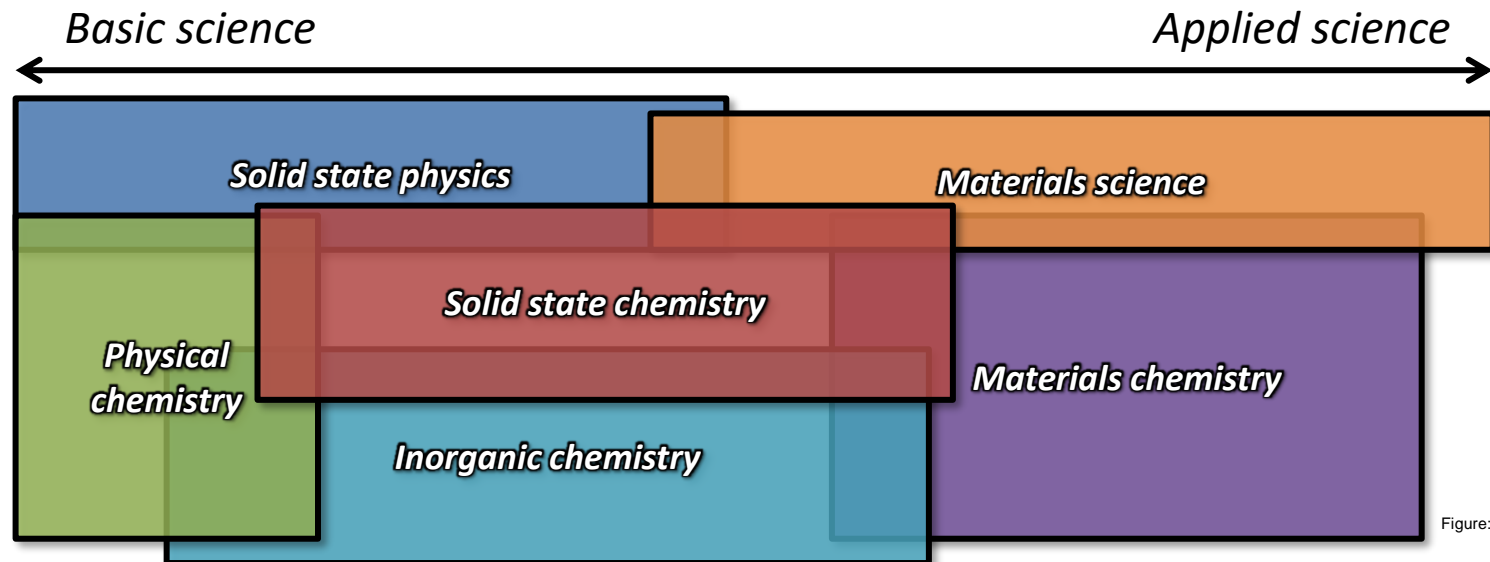


Figure: AJK

Course outline

- **Lectures**
 - 14 lectures (course calendar shown on a later slide)
 - Lectures on **29.1.** and **31.1.** are video lectures (I have to be in a mandatory military refresher exercise that week).
 - Each lecture includes a set of exercises (a MyCourses Quiz)
 - We start the exercises together during the lecture.
 - Additionally: Zulip workspace for discussions and questions about exercises.
- **Wiki Project**
 - We create content in the [Aalto Solid State Chemistry Wiki](#)
 - Includes both independent and collaborative work (peer review)
- **Grading**
 - Exercises 50%, Wiki Project 50%
- **Workload**
 - Lectures, combined with exercises ~28 h
 - Home problem solving ~28 h
 - Independent project work ~40 h

Honor code for exercises

- The purpose of the exercises is to **support your learning**
- Most of the exercises are graded automatically
 - There may also be some manually graded exercises
- It is perfectly fine to discuss the exercises with the other students
 - In fact, I encourage discussion during the teaching sessions
- You are **not allowed** to take answers directly from other students
 - This also means that you are **not allowed** to give answers directly to others
- The exercise answers and timestamps are monitored throughout the course

Course calendar

Lectures in lecture hall Ke2
(A304, Kemistintie 1)
Mon: 15:15-17.00
Wed: 8.30-10.00

Week	Lect.	Date	Topic
Week 1	1	Mon 8.1.	Structure of crystalline materials. Symmetry.
Structure	2	Wed 10.1.	Hall C100. Structural databases, visualization of crystal structures.
Week 2	3	Mon 15.1.	Bonding in solids. Description of crystal structures.
Bonding	4	Wed 17.1.	Band theory. Band structures.
Week 3	5	Mon 22.1.	Phase diagrams, crystal growth.
Synthesis	6	Wed 24.1.	Solid state synthesis.
Week 4	7	Video lect.	XRD, Miller indices. Powder XRD databases. Microscopies.
Characterization	8	Video lect.	Spectroscopies and thermal analysis.
Week 5	9	Mon 5.2.	Abundance of elements, geochemistry, minerals.
Main groups	10	Wed 7.2.	Main group compounds, allotropes, Zintl phases.
Week 6	11	Mon 12.2.	<i>d</i> -block metals, ligand field theory, magnetism.
<i>d</i> -block metals	12	Wed 14.2.	<i>d</i> -block metal oxides and other compounds.
<i>Examination week for study period III (no lectures).</i>			
Week 7	13	Mon 26.2.	12.15-14.00, C100. Defects, non-stoichiometric compounds.
Defects and doping	14	Wed 28.2.	8.30-10.00. Semiconductors, doping.
<i>Final Wiki project deadline after mid-March.</i>			

Literature for the course

- *Solid State Chemistry and its Applications* (2nd ed.), Anthony R. West, **2013**, Wiley.
- *Inorganic Structural Chemistry* (2nd ed.), Ulrich Müller, **2006**, Wiley.
- [Aalto Solid State Chemistry Wiki](#)

