

# Functional Inorganic Materials

## Fall 2021

Tuesdays: 14.15 - 16.00  
Thursdays: 12.15 - 14.00  
Remote Zoom lectures

#	Date	Who	Topic
1	Tue 02.11.	Maarit	Introduction + Materials design
2	Thu 04.11.	Antti	Computational materials design
<b>3</b>	<b>Tue 09.11.</b>	<b>Maarit</b>	<b>Superconductivity: High-<math>T_c</math> superconducting Cu oxides</b>
4	Thu 11.11.	Maarit	Ionic conductivity (Oxygen): Oxygen storage and SOFC
5	Tue 16.11.	Maarit	Ionic conductivity (Lithium): Li-ion battery
6	Thu 18.11.	Antti	Thermal conductivity
7	Tue 23.11.	Antti	Thermoelectricity
8	Thu 25.11.	Maarit	Hybrid materials
9	Tue 30.11.	Maarit	Luminescence and optically active materials
10	Thu 02.12.	Antti	Piezoelectricity
11	Tue 07.12.	Antti	Pyroelectricity and ferroelectricity
12	Thu 09.12.	Antti	Magnetic and multiferroic oxides

# LECTURE 3: (High- $T_c$ ) Superconductivity

- ❖ Zero resistance, Meissner effect, Cooper pair, Josephson junction
- ❖ History & Impact
- ❖ Physics – Materials/Chemistry – Technology
- ❖ New-material discoveries: Design principles & Good luck
- ❖ Perovskite & Ruddlesden-Popper
- ❖ Multi-layered crystal structure & Homologous series
- ❖ Aliovalent substitution / Isovalent substitution (= Chemical pressure)
- ❖ Mixed-valency & Oxygen nonstoichiometry
- ❖ p-type & n-type: Importance of Cu coordination number/sphere

## LECTURE EXERCISE 3

1. The following copper oxide compounds are high- $T_c$  superconductors:  
 $\text{YBa}_2\text{Cu}_3\text{O}_{7\pm\delta}$ ,  $\text{Bi}_2\text{Sr}_2\text{CuO}_{6\pm\delta}$ ,  $\text{Bi}_2\text{Sr}_2\text{CaCu}_2\text{O}_{8\pm\delta}$  and  $\text{Bi}_2\text{Sr}_2\text{Ca}_2\text{Cu}_3\text{O}_{10\pm\delta}$

a) Give the systematic name (= chemical formula abbreviation) for each compound.

b) Which of the compounds should have the highest  $T_c$  (when optimized); give the reason for your choice!

c) Explain the importance of oxygen nonstoichiometry parameter  $\delta$  for these compounds.

d) Explain why a multilayered crystal structure is useful for the high- $T_c$  copper-oxide superconductors.

2.  $\text{Nb}_3\text{Sn}$  and  $\text{Sn}_3\text{Ge}$  are important "low-temperature" superconductors and members of the so-called "A15 family" of intermetallic compounds. Please make a quick/small literature search (3-7 references) to discuss few aspects (history, different characteristics, etc.) which you find most interesting related to these materials.

## Superconductivity of Nb<sub>3</sub>Sn

B. T. MATTHIAS, T. H. GEBALLE, S. GELLER, AND E. CORENZWIT  
*Bell Telephone Laboratories, Murray Hill, New Jersey*  
 (Received June 10, 1954)

Intermetallic compounds of niobium and tantalum with tin have been found. The superconducting transition temperature of Nb<sub>3</sub>Sn at 18°K is the highest one known.

SOME intermetallic compounds crystallizing with the  $\beta$ -wolfram structure become superconducting, as was first pointed out by Hardy and Hulm.<sup>1</sup> In particular one of these, V<sub>3</sub>Si, showed a remarkably high transition temperature between 16.9°K and 17.1°K. These authors made various attempts to raise this temperature by introducing a third component but were not successful.

The  $\beta$ -wolfram structure is a very peculiar structure with rather varying interatomic distances,<sup>2</sup> a fact which may render the addition of a third component rather difficult. It seemed therefore more favorable to look for another  $\beta$ -W compound with a large volume and a favorable electron/atom ratio<sup>3</sup> in order to raise the superconducting transition temperature. There is very little known about the systematic occurrence of intermetallic compounds in this  $\beta$ -W structure. The fact that thus far no niobium compounds have been reported seemed therefore not significant.

It was expected that in the Nb-Sn and Ta-Sn this crystal form would be found, an assumption which was verified. We have determined that Nb<sub>3</sub>Sn and Ta<sub>3</sub>Sn both crystallize in a  $\beta$ -W structure with a lattice constant of about 5.3Å. The Ta<sub>3</sub>Sn was measured in the apparatus previously described,<sup>4</sup> and became superconducting near 6°K. The transition temperature of the Nb<sub>3</sub>Sn was determined by immersing the sample surrounded by a copper coil in liquid hydrogen. The self-inductance of the coil was measured on a General Radio Model 650A Bridge at 1 kc/sec as the sample was slowly cooled. Figure 1 shows the results for two different samples made under somewhat different conditions which were cooled from 18.5°K to 17.5°K during a period of about 30 minutes. The sharpness of the transition together with the reproducibility between samples indicates that these samples are indeed well-defined compounds. The onset of superconductivity at

18.05°K $\pm$ 0.1° is determined by extrapolating the line of steepest slope to the high temperature line. Temperatures were measured by a copper constantan thermocouple secured to the measuring coil and independently checked with the vapor pressure of hydrogen.

### APPENDIX

While the synthesis of an intermetallic compound is generally a rather straightforward process, it may be necessary to describe briefly the formation of these

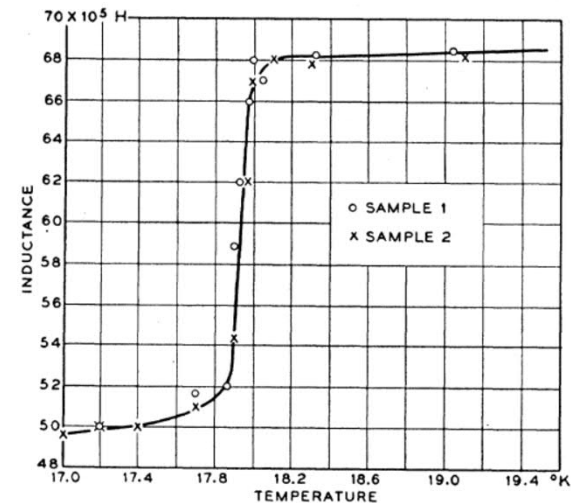


Fig. 1. Variation of susceptibility with temperature of Nb<sub>3</sub>Sn.

compounds. No reference to Nb-Sn or Ta-Sn was found in the literature. The melting point of niobium is nearly 400° above the boiling point of tin, and an arc furnace is therefore out of place. A complete reaction can, however, easily be obtained by having molten tin run over Nb or Ta powder in a closed-off quartz tube at 1200°C. Nb<sub>3</sub>Sn and Ta<sub>3</sub>Sn seem to be formed by a peritectic reaction between 1200°C and 1550°C.

<sup>1</sup> G. Hardy and J. K. Hulm, *Phys. Rev.* **89**, 884 (1953).

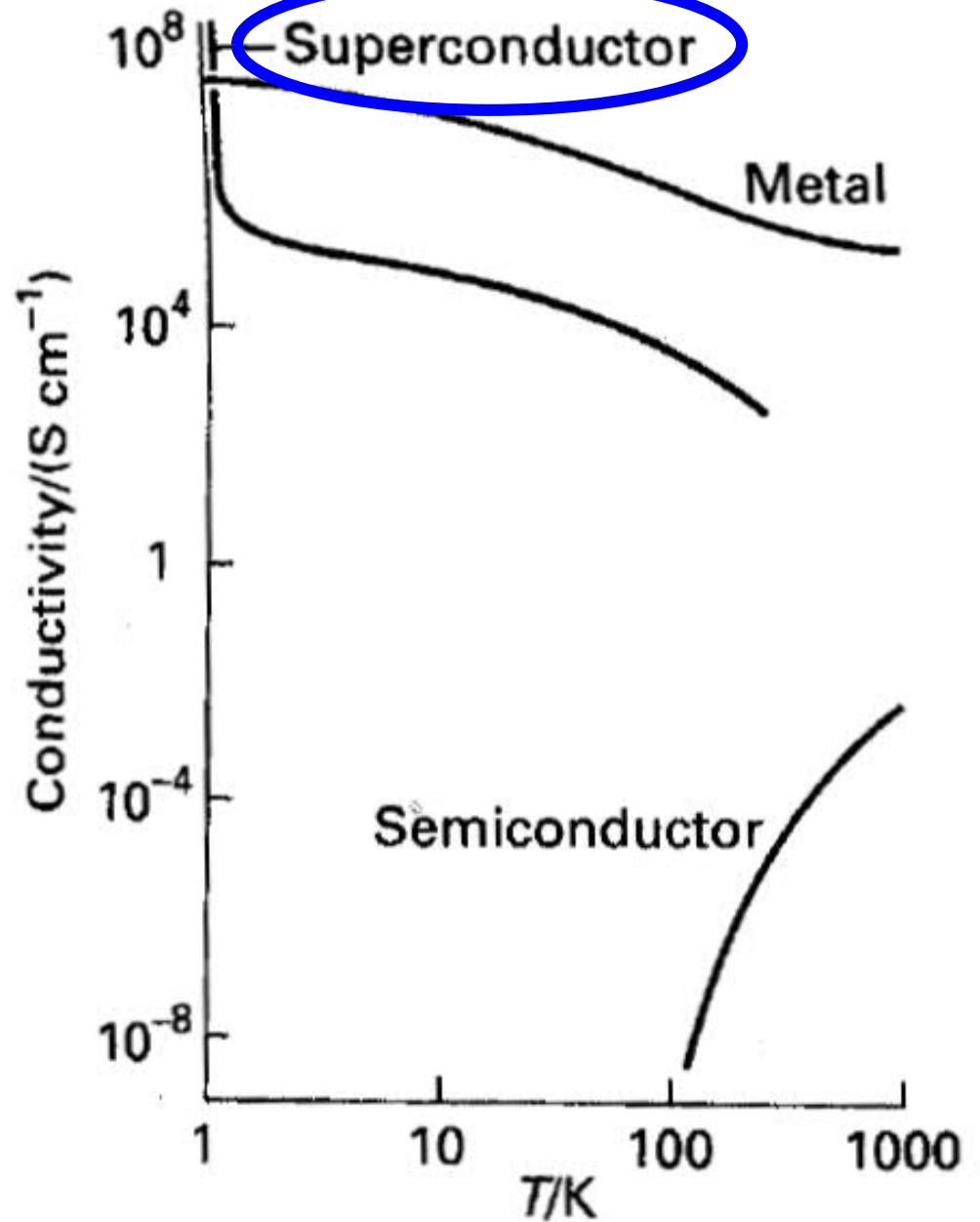
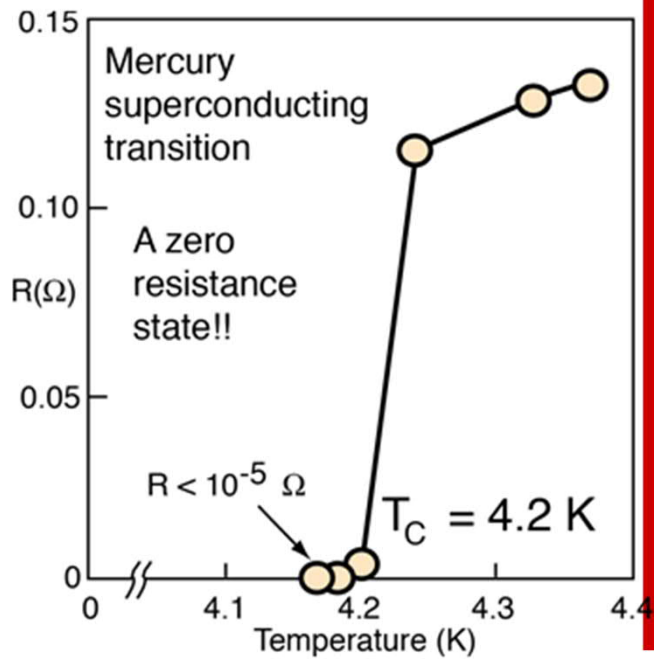
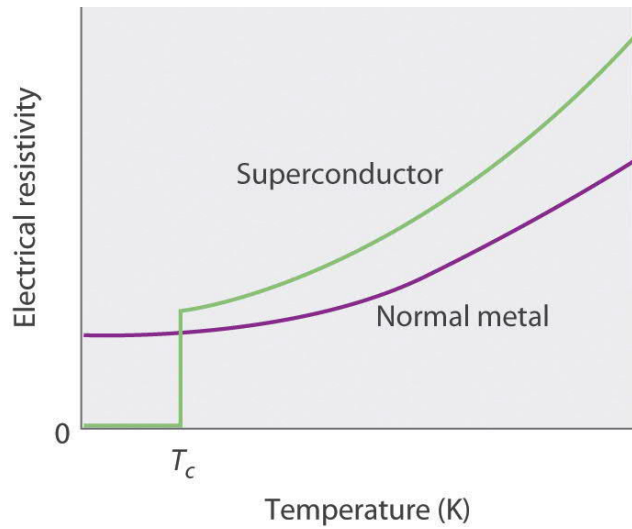
<sup>2</sup> H. I. Wallbaum, *Z. Metallkunde* **31**, 362 (1939).

<sup>3</sup> B. T. Matthias, *Phys. Rev.* **92**, 874 (1953).

<sup>4</sup> B. T. Matthias and J. K. Hulm, *Phys. Rev.* **87**, 799 (1952).

**Ted Geballe 100 years  
 Jan. 2020**





# SUPERCONDUCTIVITY

## Superconductivity

- 1911: Kamerlingh-Onnes
- $\rho = 0$
- Hg ( $T_c = 4.2$  K)



Nobel 1913

## Meissner effect

- 1933: Meissner and Ochsenfeld:
- $\chi = B/H < 0 \rightarrow$  levitation



## High- $T_c$ superconductivity

- 1986: Bednorz and Müller
- $(\text{La,Ba})_2\text{CuO}_4$  ( $T_c = 30\sim 40$  K)



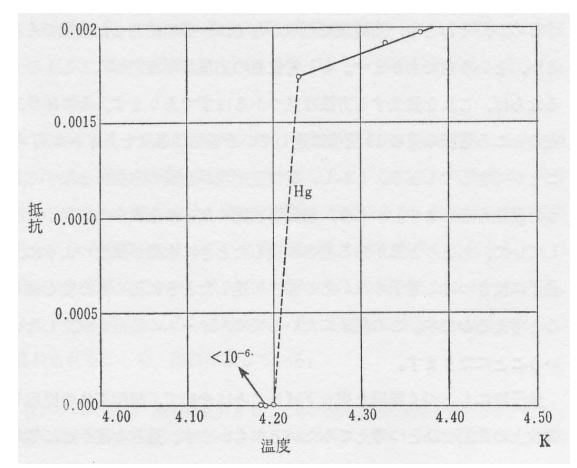
Nobel 1987

## Present record in $T_c$ :

138 K for  $\text{HgBa}_2\text{Ca}_2\text{Cu}_3\text{O}_{8+\delta}$

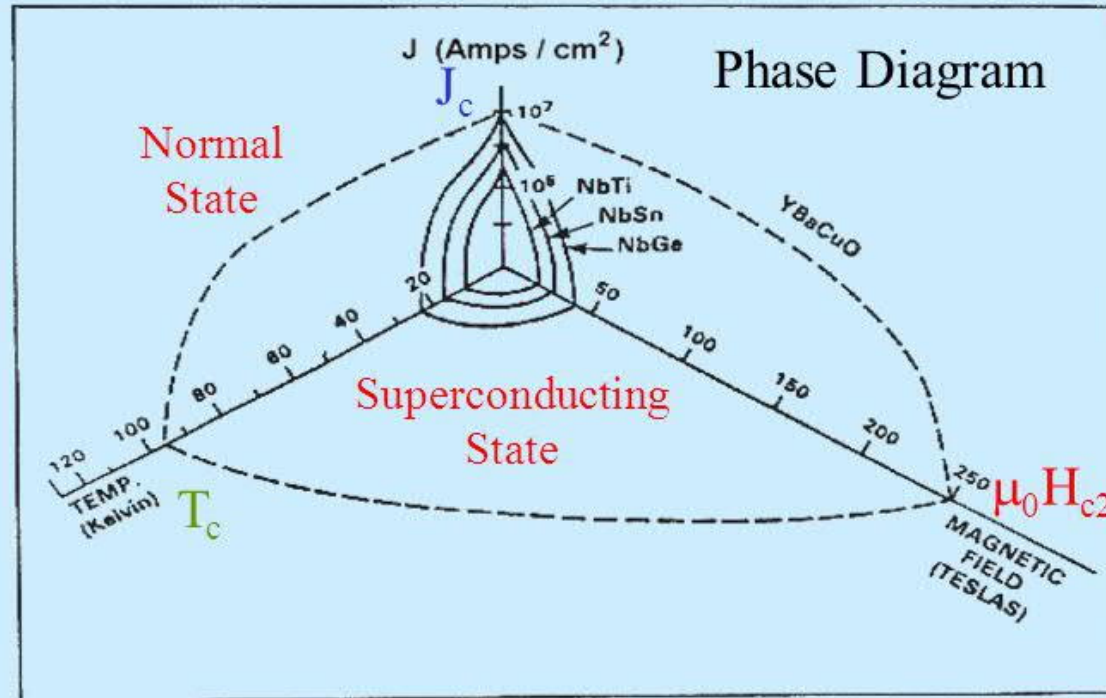


Kamerlingh-Onnes Institute,  
@University of Leiden, the Netherlands



Heike Kamerlingh-Onnes and J.D. van der Waals

# What are the Limits of Superconductivity?



$$f_{\text{super}} = f_{\text{normal}} + \alpha(T)|\psi|^2 + \frac{\beta(T)}{2}|\psi|^4 + \frac{1}{2m^*} \left| \left( \frac{\hbar}{i} \vec{\nabla} - e^* \vec{A} \right) \psi \right|^2 + \frac{\mu_0 h^2}{2}$$

Ginzburg-Landau  
free energy density

Temperature  
dependence

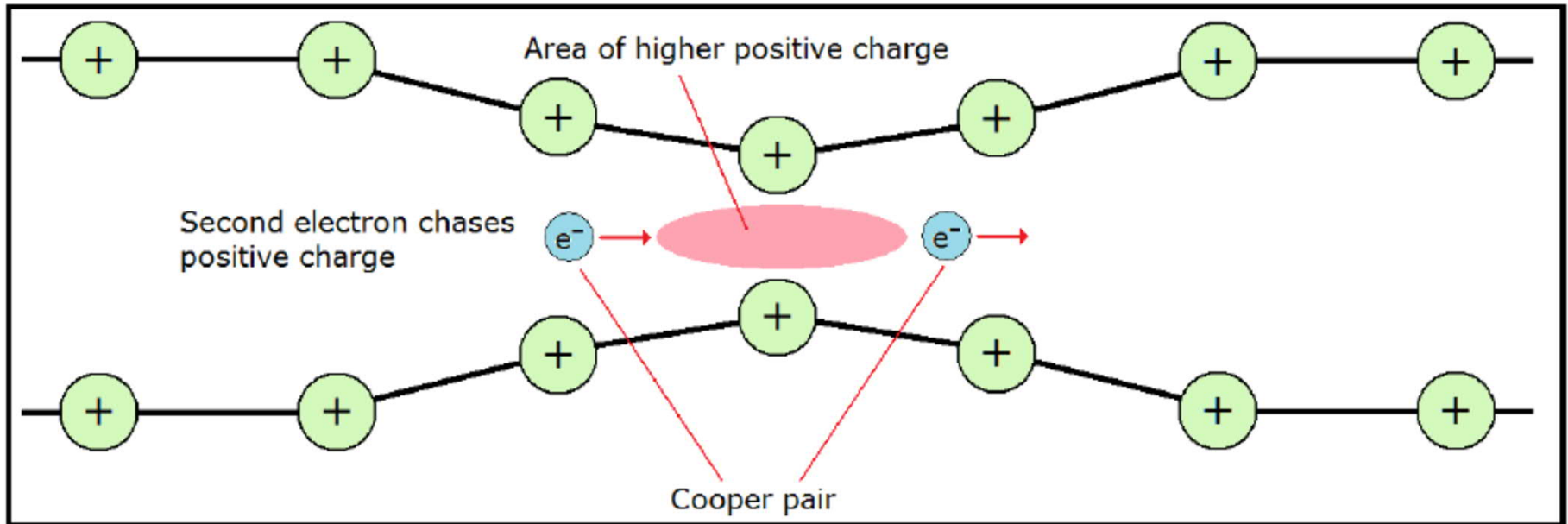
Currents

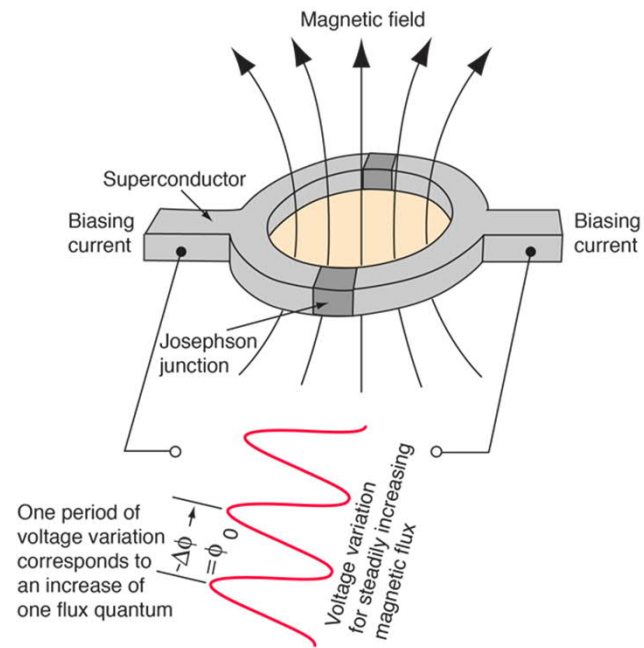
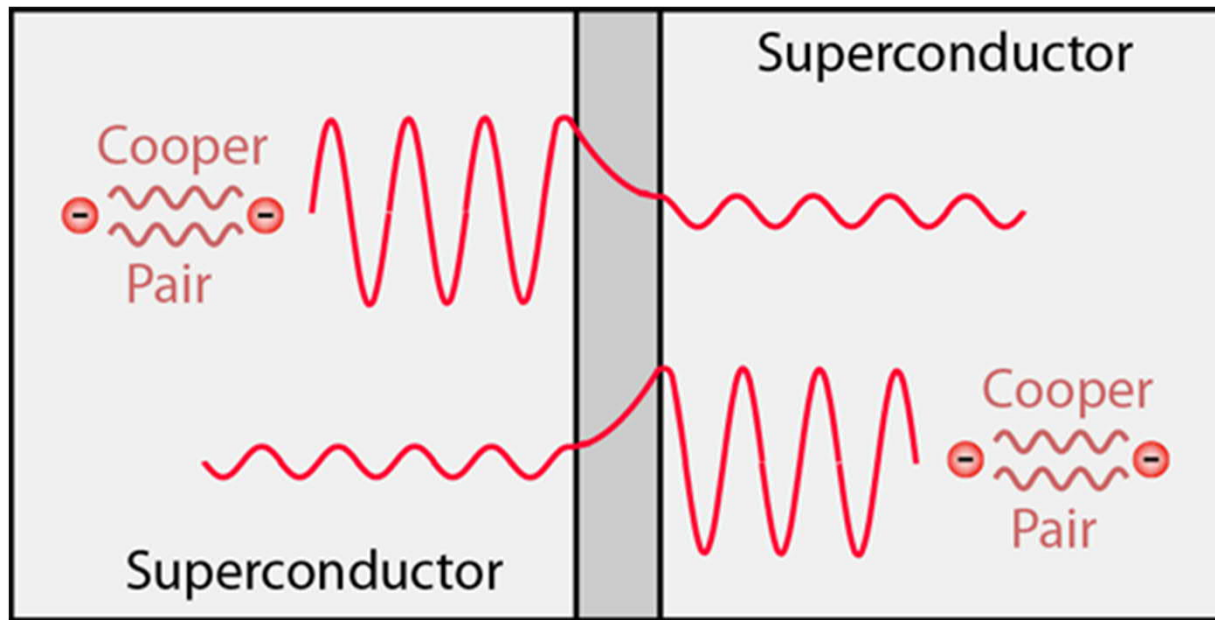
Applied magnetic field



# Bardeen, Cooper & Schrieffer

- BCS theory 1957
- Nobel 1972
- Cooper pairs
- Coupled through Phonons in conventional superconductors



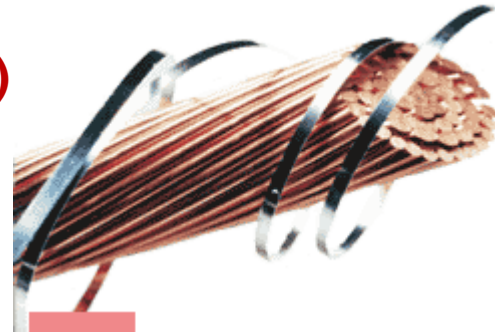


## Josephson Junction & SQUID

- 1962 Brian David Josephson (Nobel 1973)
- Two superconductors separated by a thin insulating layer
- Tunneling of Cooper pairs through the junction
- Macroscopic quantum effect
- Josephson junction device has become the standard measure of voltage
- Superconducting quantum interference device (SQUID) based on Josephson junctions: measurement of extremely weak signals (e.g. subtle changes in the human body's electromagnetic energy field)

# Application Examples of (high- $T_c$ ) Superconductors

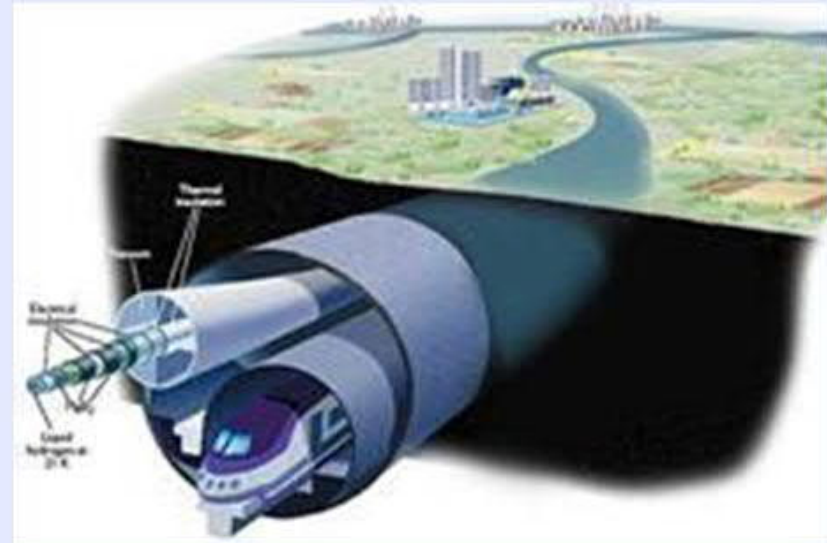
- Cables and wires ( $\rho = 0$ )  
→ public power supply (Copenhagen 2001)
- Strong magnets ( $\rho = 0$ )
- Microwave filters ( $\rho = 0$ )  
→ to improve signal reception  
in wireless phone towers
- Levitation (Meissner)  
→ “True” MAGLEV trains
- Sensitive magnetic probes (Josephson)  
→ SQUID, NMR, MRI
- Supercomputers (Josephson)

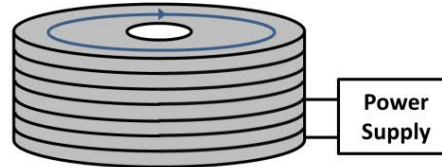
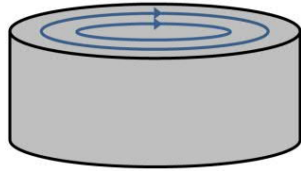
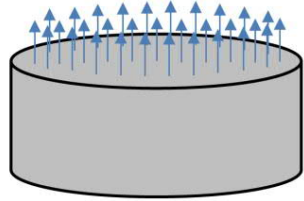
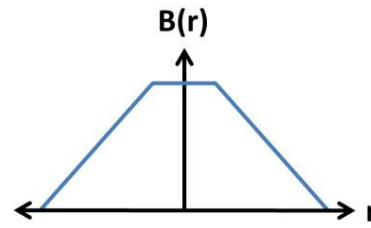
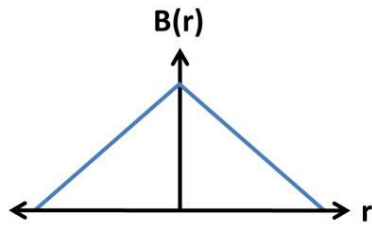
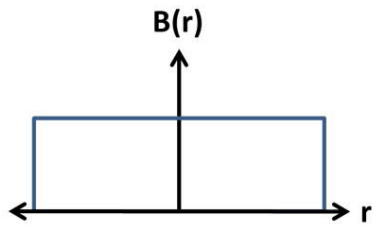


# Applications using Superconductors

## Superconducting power transmission

- currently we waste ~ 20 % of our energy just transporting it around
- potentially the next industrial revolution

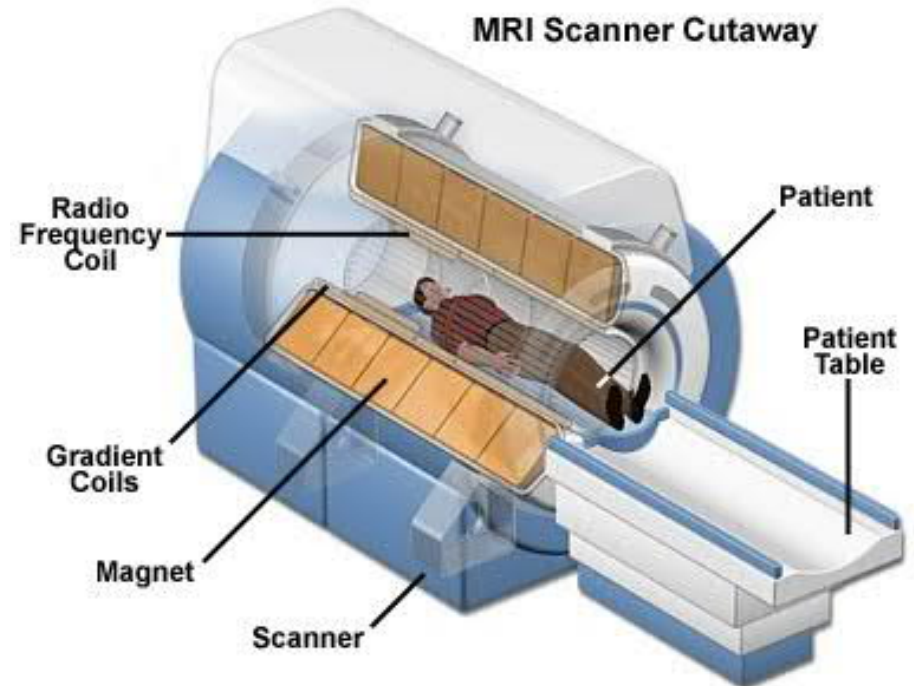




**Permanent ferromagnet**  
*Spin*  
Neodymium N-B-Fe (1.5 T)

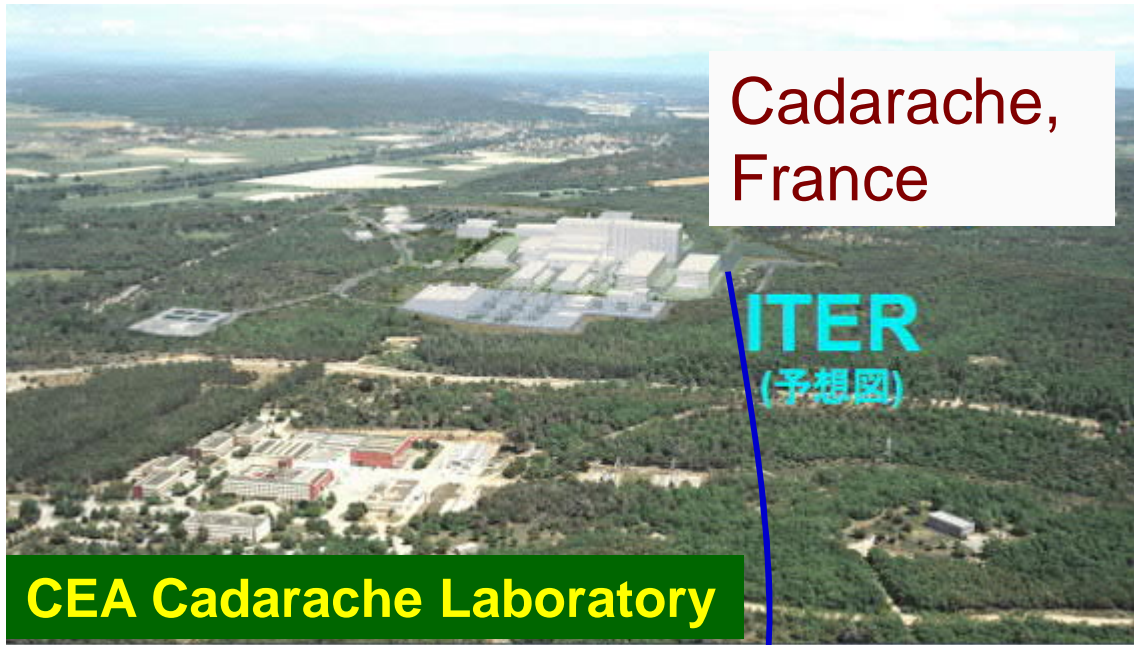
**Bulk superconductor**  
*Induced, superconducting loop current*  
HTS (17 T), MgB<sub>2</sub> (5 T)

**Electromagnet**  
*Supplied loop current*  
Cu (2 T), HTS (> 30 T)

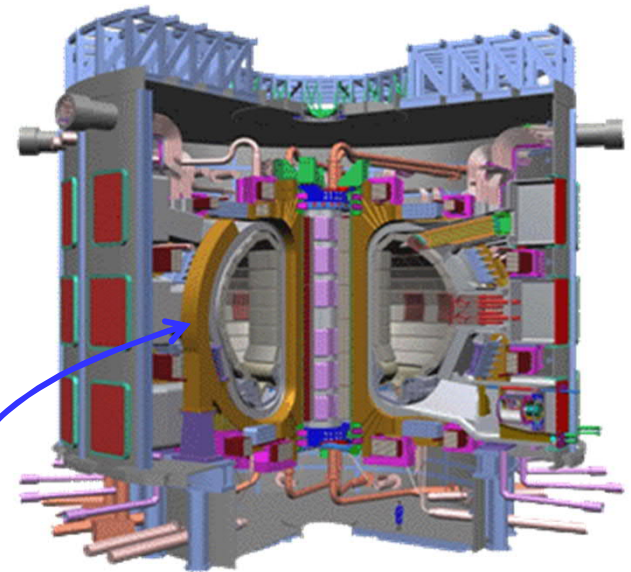


# ITER: International Thermonuclear Experimental Reactor

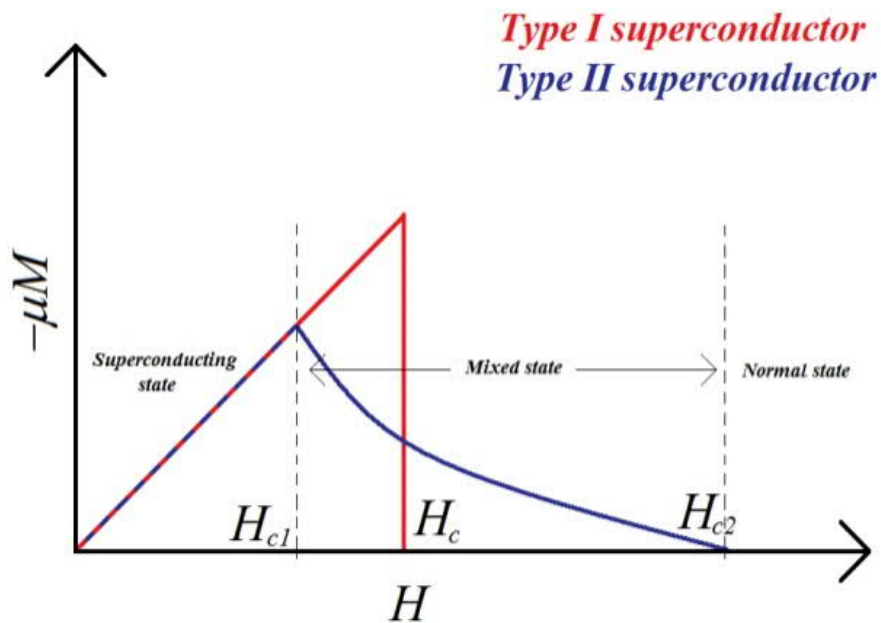
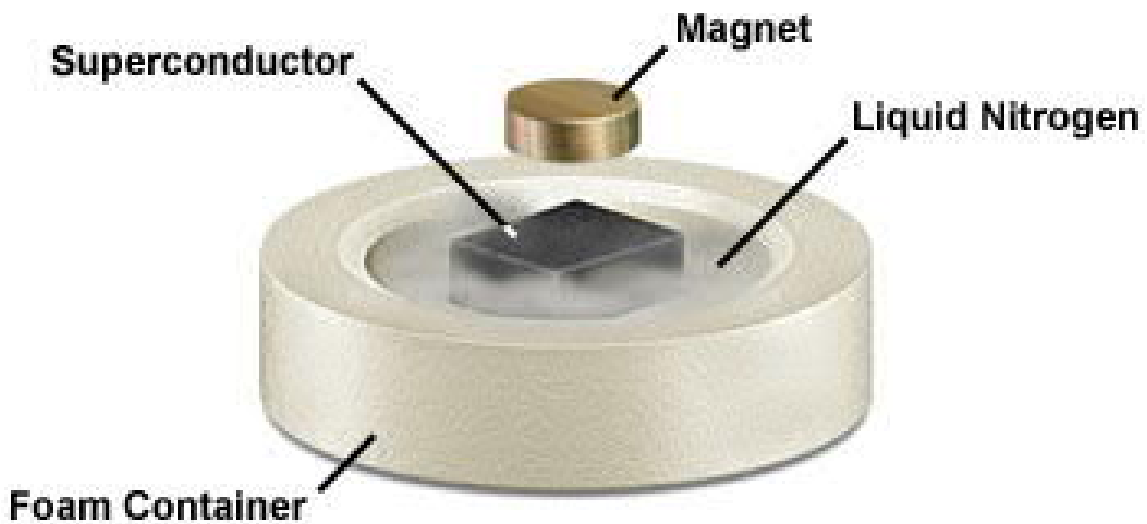
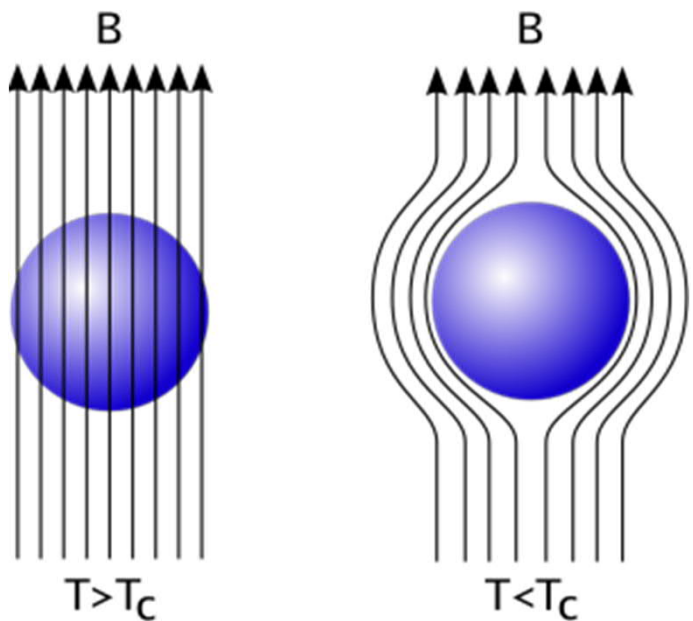
since October, 2007



**Nuclear Fusion  
Reactor**



# The Meissner Effect





## Super-Maglev Train

- 603 km / hour
- Test line 42.8 km

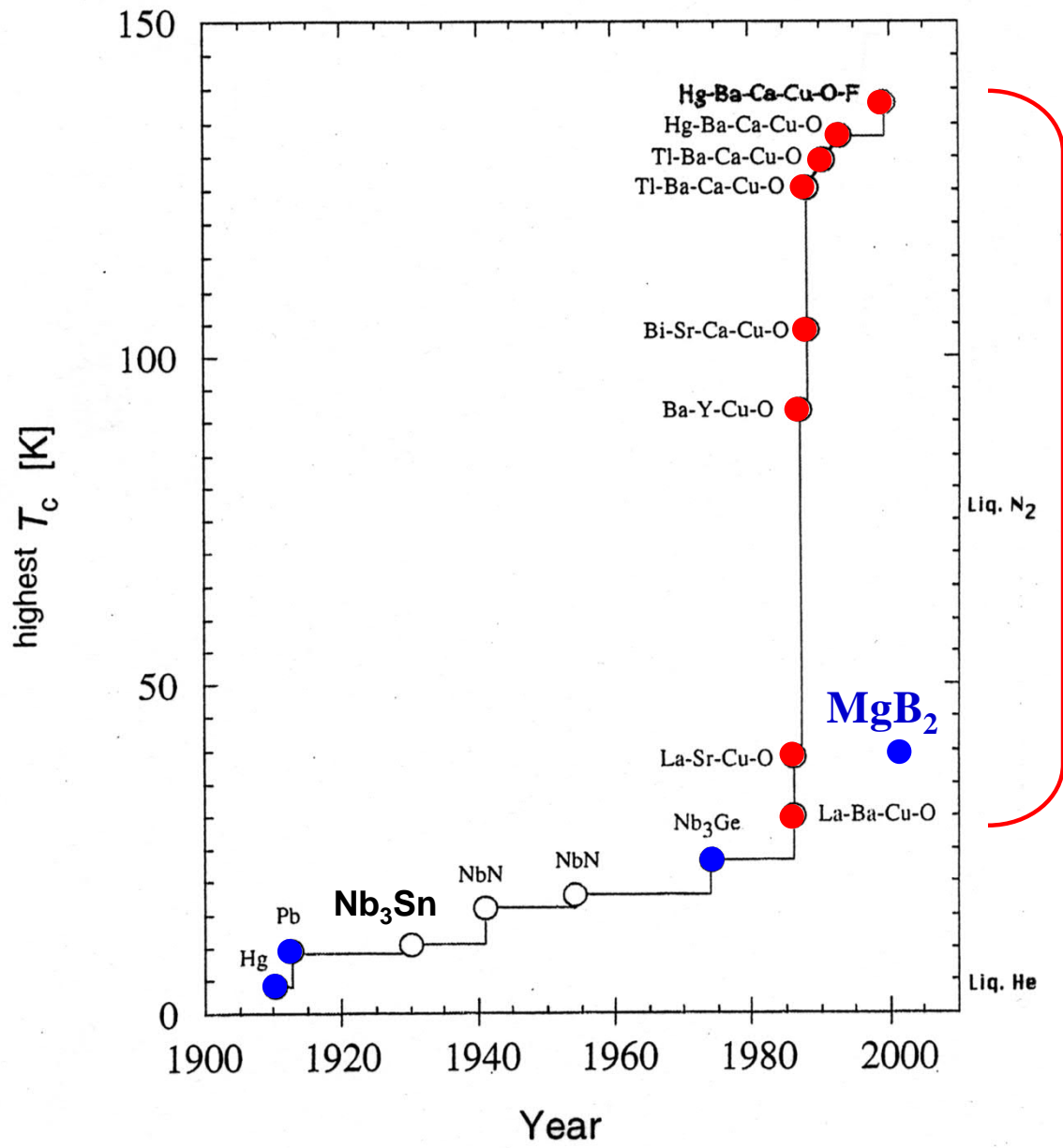


## Superconducting Elements

1	1	H																	2	He																
2	3	Li	4	Be																	5	B	6	C	7	N	8	O	9	F	10	Ne				
3	11	Na	12	Mg																	13	Al	14	Si	15	P	16	S	17	Cl	18	Ar				
4	19	K	20	Ca	21	Sc	22	Ti	23	V	24	Cr	25	Mn	26	Fe	27	Co	28	Ni	29	Cu	30	Zn	31	Ga	32	Ge	33	As	34	Se	35	Br	36	Kr
5	37	Rb	38	Sr	39	Y	40	Zr	41	Nb	42	Mo	43	Tc	44	Ru	45	Rh	46	Pd	47	Ag	48	Cd	49	In	50	Sn	51	Sb	52	Te	53	I	54	Xe
6	55	Cs	56	Ba	57	La	72	Hf	73	Ta	74	W	75	Re	76	Os	77	Ir	78	Pt	79	Au	80	Hg	81	Tl	82	Pb	83	Bi	84	Po	85	At	86	Rn
7	87	Fr	88	Ra	89	Ac	104	Rf	105	Ha	106	Sg	107	Bh	108	Hs	109	Mt	110	Ds	111	Rg	112	Uub												

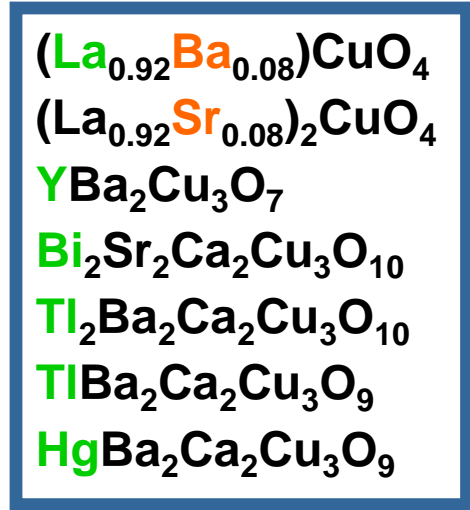
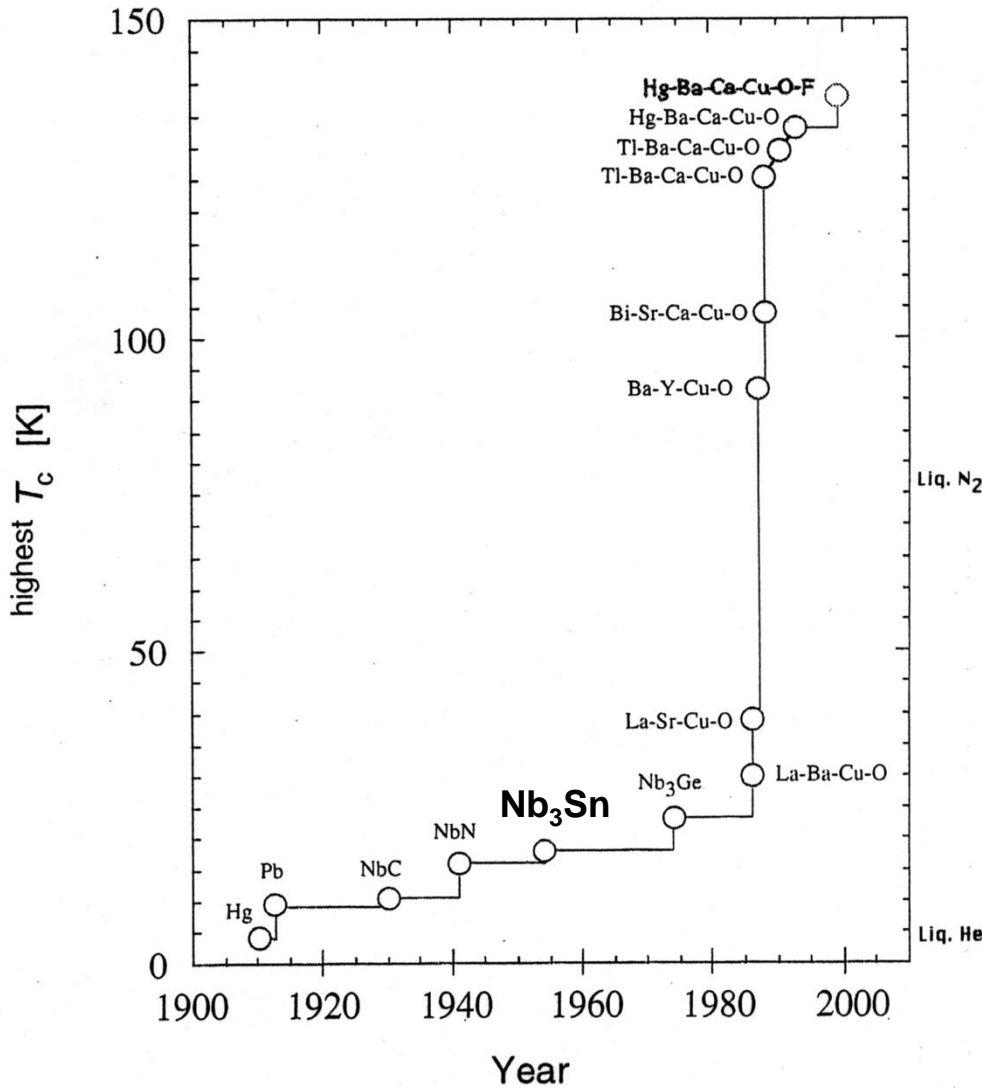
- In Bulk at Ambient Pressure
- At High Pressure
- In Modified Form

58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
90	91	92	93	94	95	96	97	98	99	100	101	102	103
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cr	Es	Fm	Md	No	Lr



"HTSC"

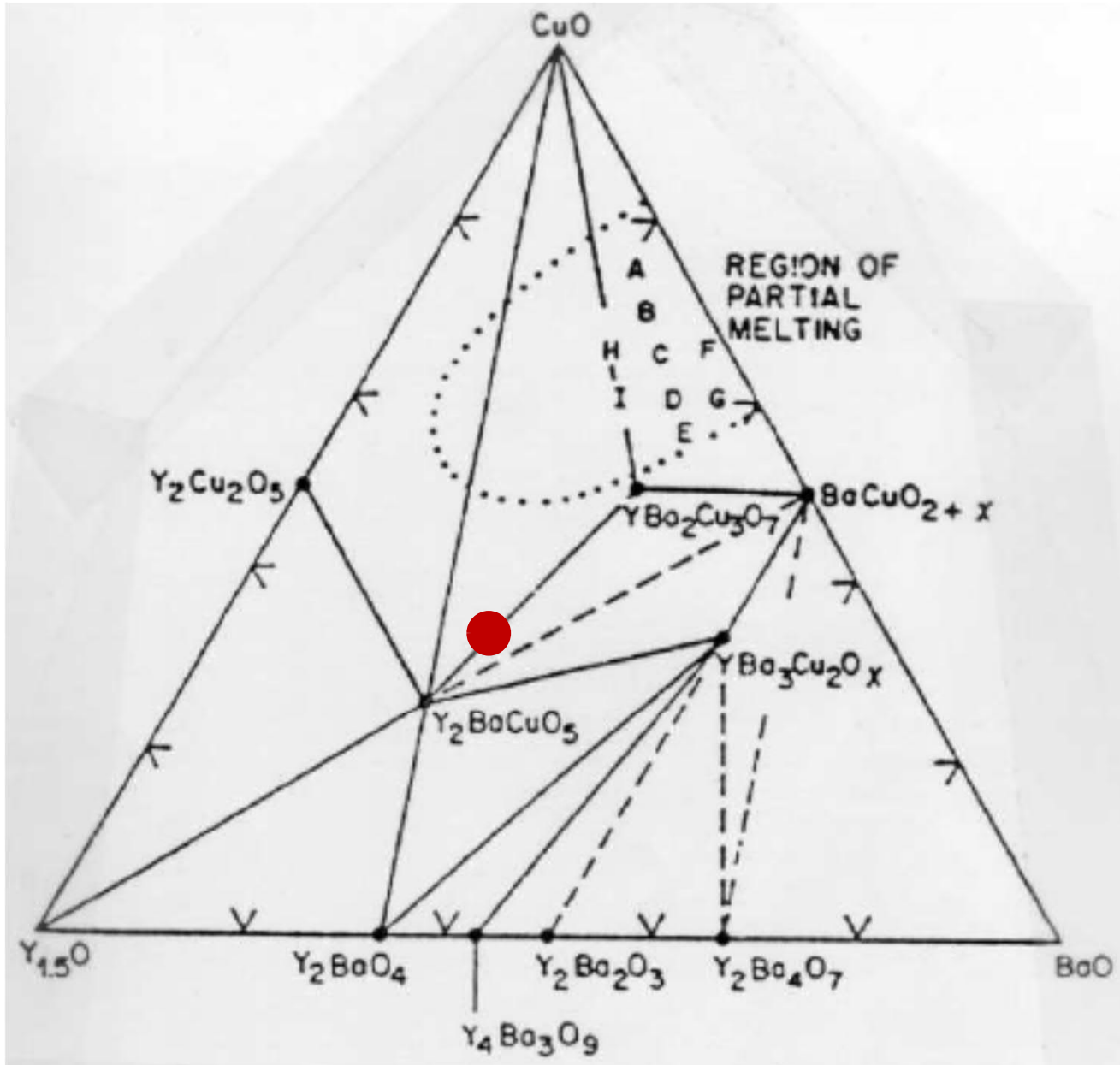
# Search for new high- $T_c$ superconductors



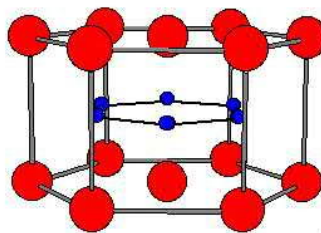
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Period 2	3		4																																																																										10																															
	Li		Be																																																																										Ne																															
Period 3	11		12																																																																										18																															
	Na		Mg														IIIB										IVB										VB										VIB										VIIB										VIIIB										IB										IIB																				Ar	
Period 4	19		20		21		22		23		24		25		26		27		28		29		30		31		32		33		34		35		36																																																																									
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Period 6	55		56		57 to 71		72		73		74		75		76		77		78		79		80		81		82		83		84		85		86																																																																									
	Cs		Ba		La to Lu		Hf		Ta		W		Re		Os		Ir		Pt		Au		Hg		Tl		Pb		Bi		Po		At		Rn																																																																									
Period 7	87		88		89 to 103		104		105		106		107		108		109																																																																																											
	Fr		Ra		Ac to Lr		Rf		Ha		Sg		Ns		Hs		Mt																																																																																											

Lanthanide series →	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71
	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
Actinide series →	89	90	91	92	93	94	95	96	97	98	99	100	101	102	103
	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

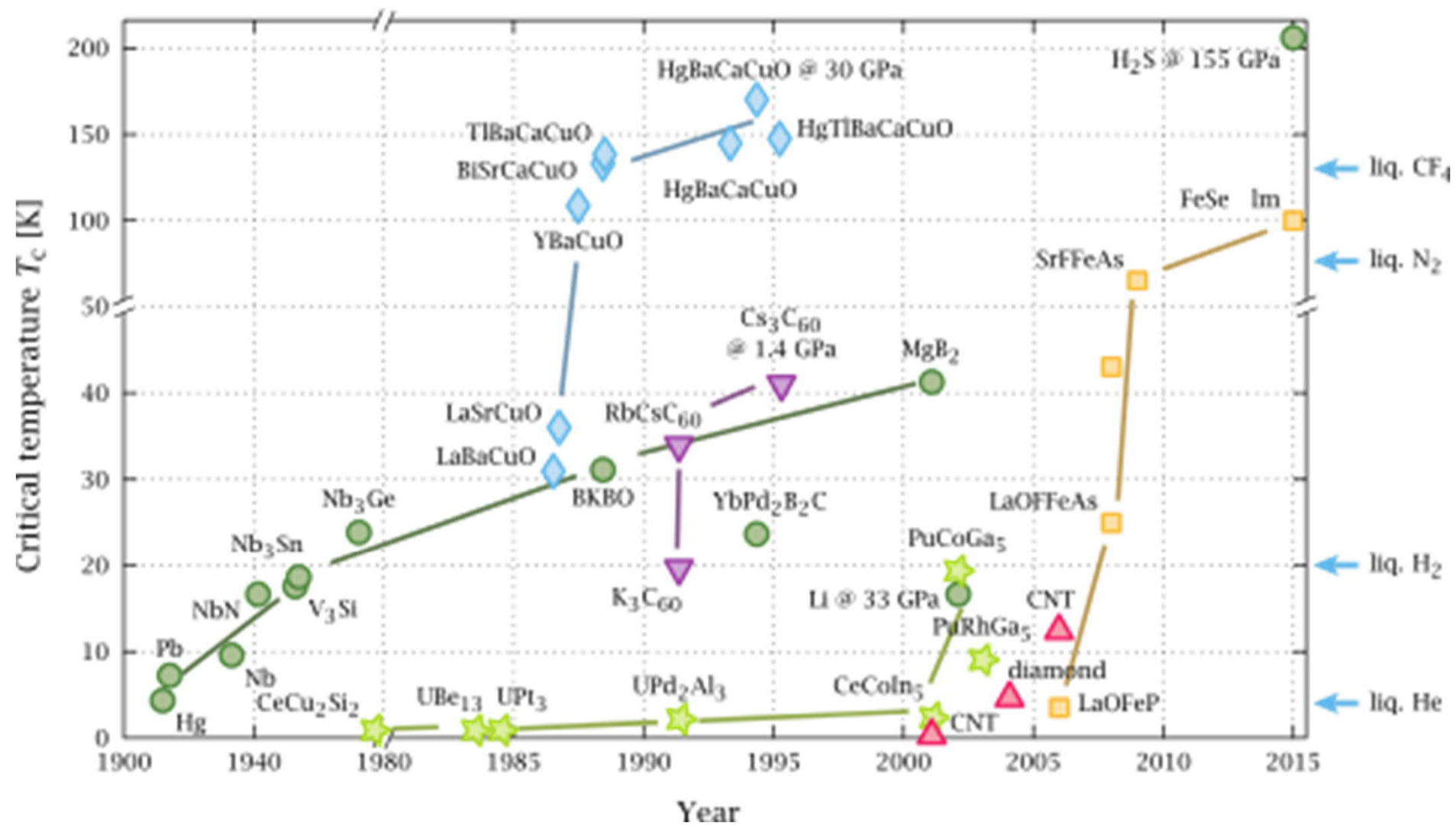
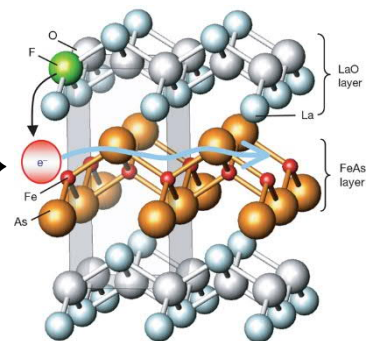
# Pseudoternary Y-Ba-Cu-O Phasediagram (0.5Y<sub>2</sub>O<sub>3</sub>-BaO-CuO): 1000°C, 1 atm O<sub>2</sub>



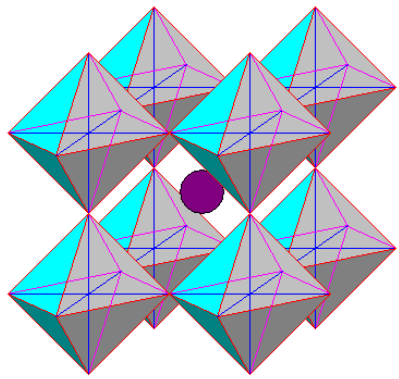
Akimitsu 2001:  
MgB<sub>2</sub>



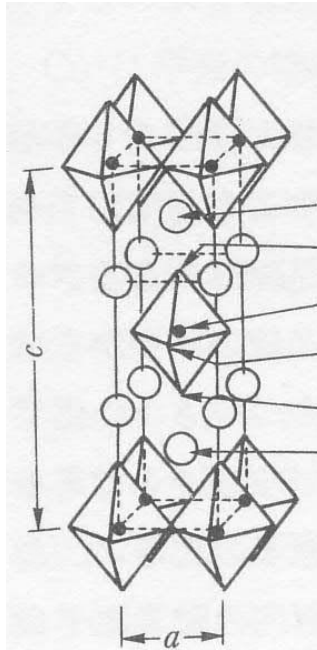
Hosono 2006 →  
[La(O,F)][FeAs]



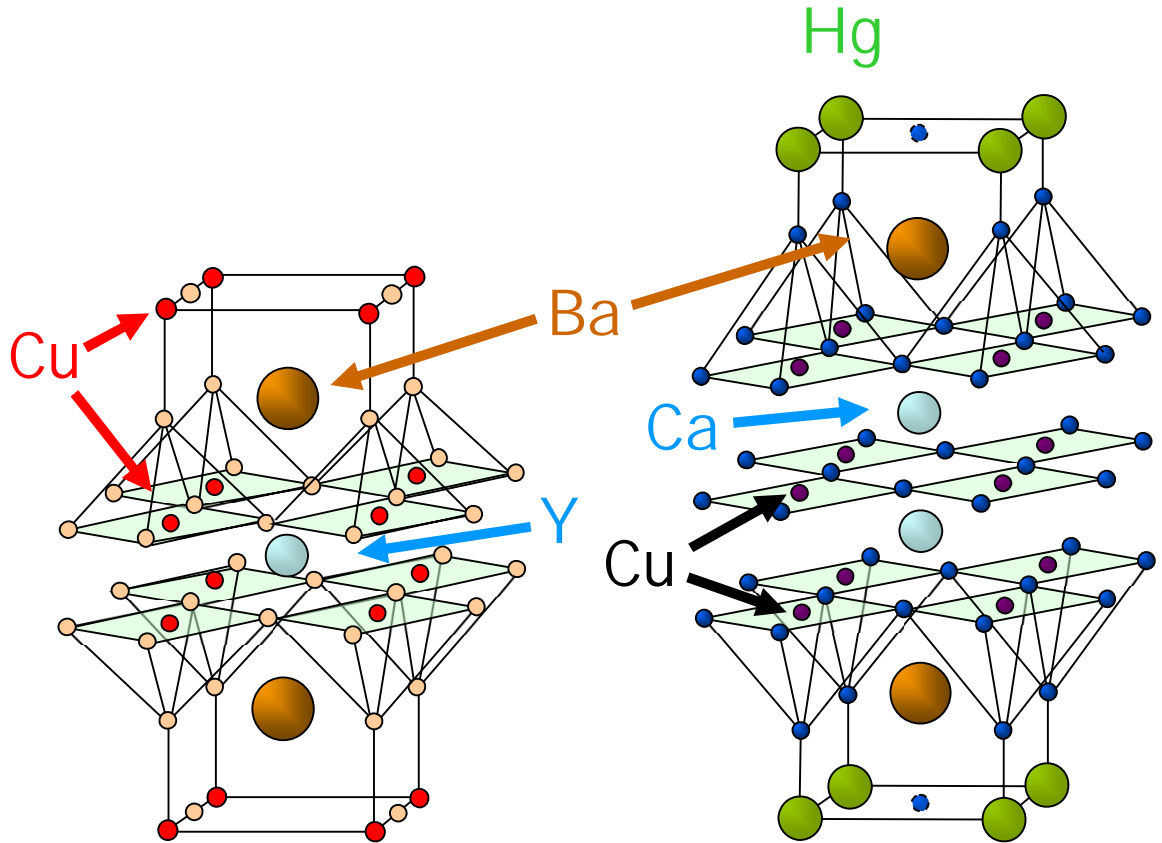
# Crystal Structures of High- $T_c$ Superconductive Copper Oxides



Perovskite  $\text{CaTiO}_3$



$T_c \approx 35 \text{ K}$

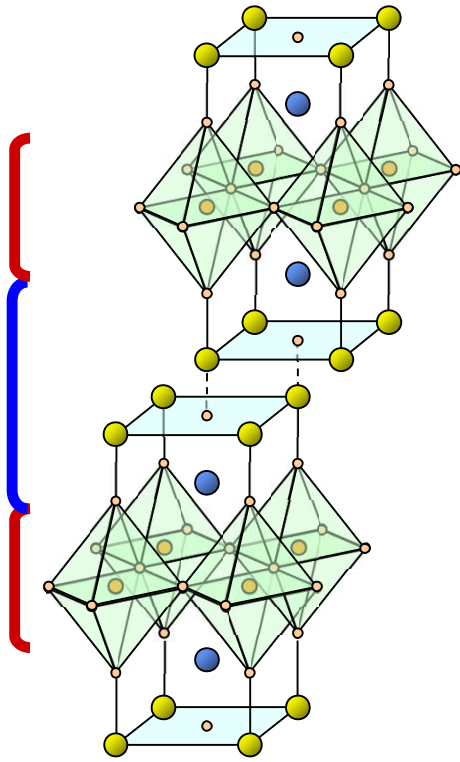


$T_c \approx 92 \text{ K}$

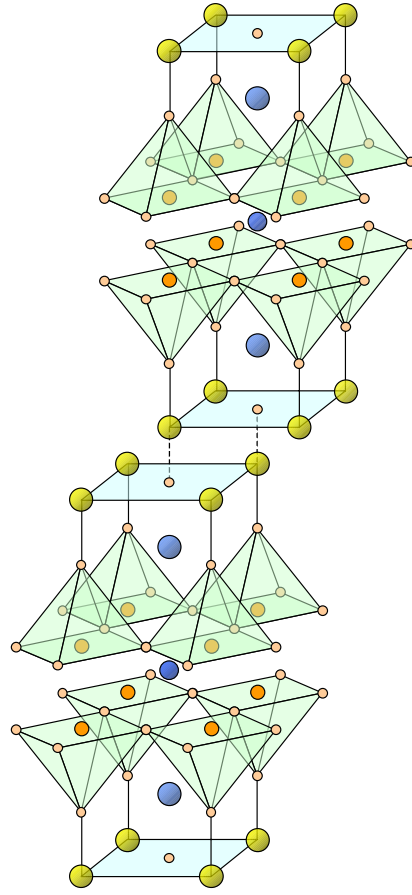


$T_c \approx 135 \text{ K}$

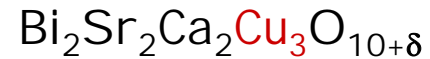
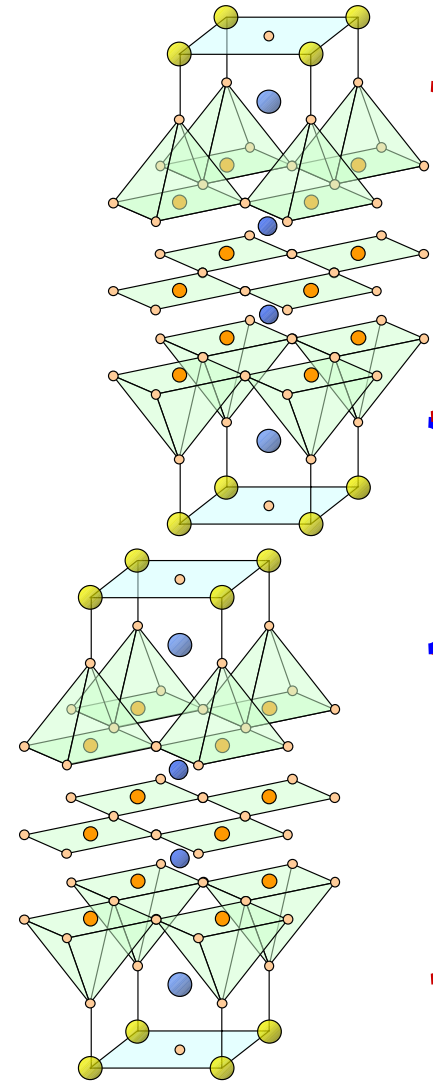
# Homologous Series of Superconducting Copper Oxides



Bi-2201;  $T_c \approx 20$  K



Bi-2212;  $T_c \approx 80$  K

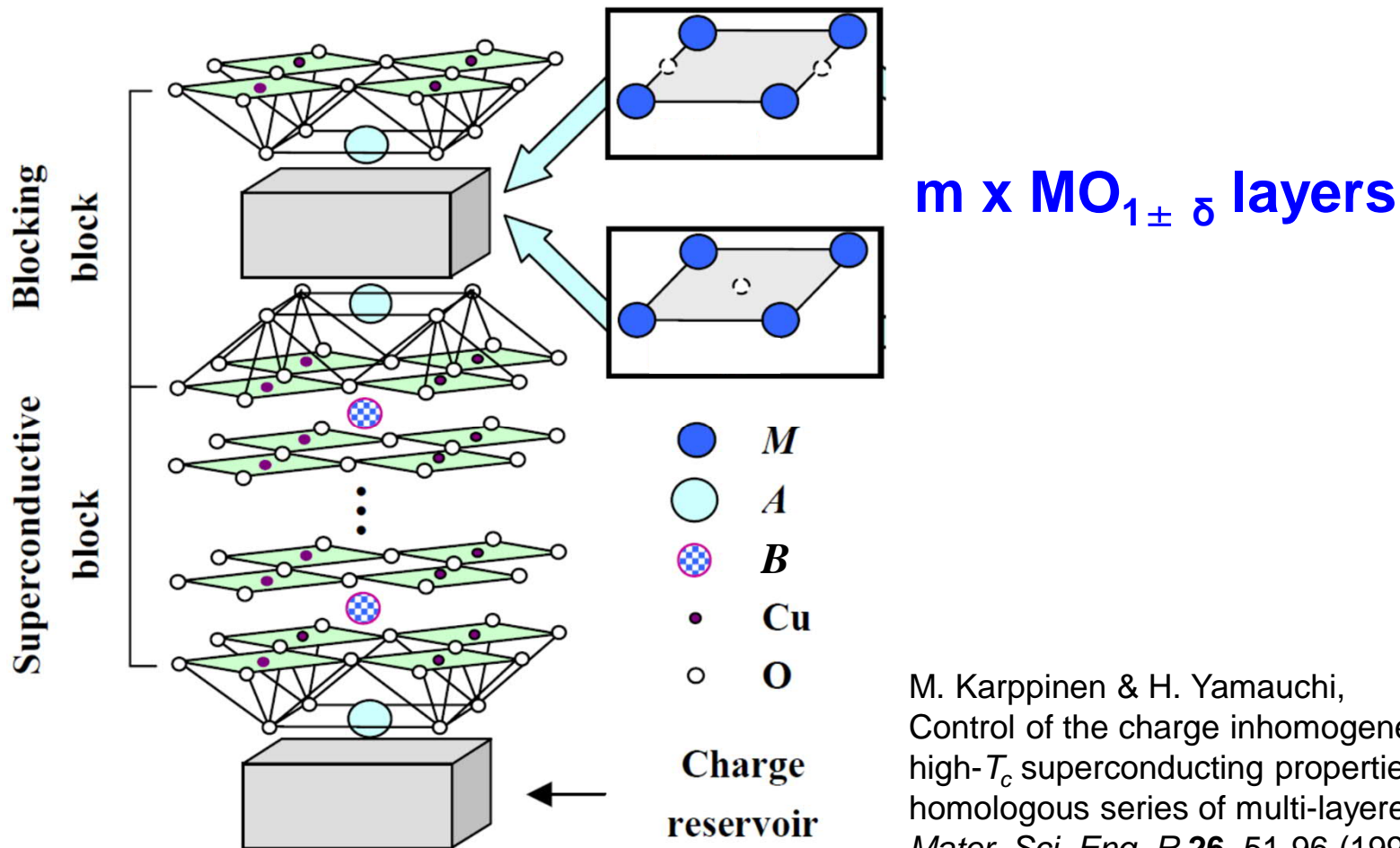


Bi-2223;  $T_c \approx 110$  K

Superconducting Block  
Blocking Block  
Superconducting Block  
Blocking Block  
Superconducting Block

# GENERAL FORMULA

- $M_m A_2 B_{n-1} Cu_n O_{m+2+2n \pm \delta}$
- $M-m2(n-1)n$
- **HOMOLOGOUS SERIES:**  $M$ ,  $m$ ,  $A$  and  $B$  fixed,  $n$  varies

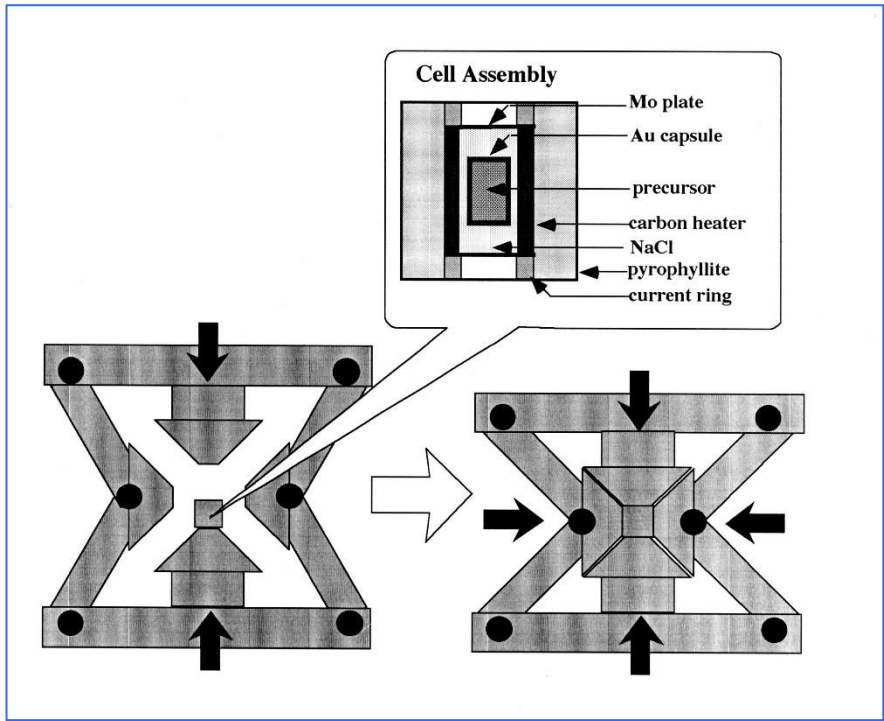
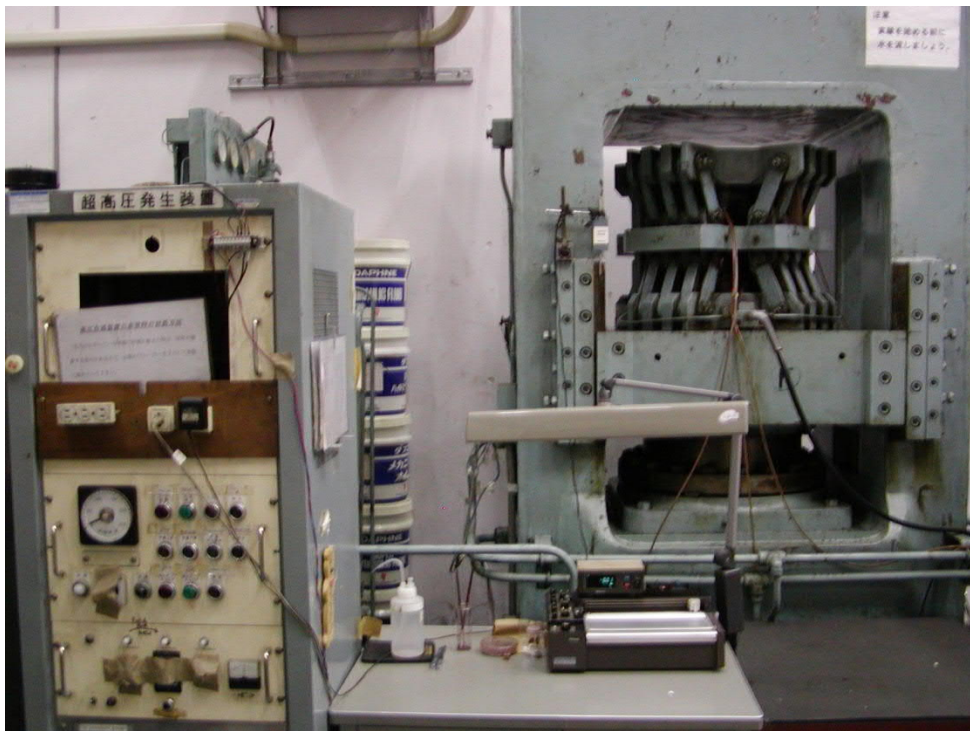
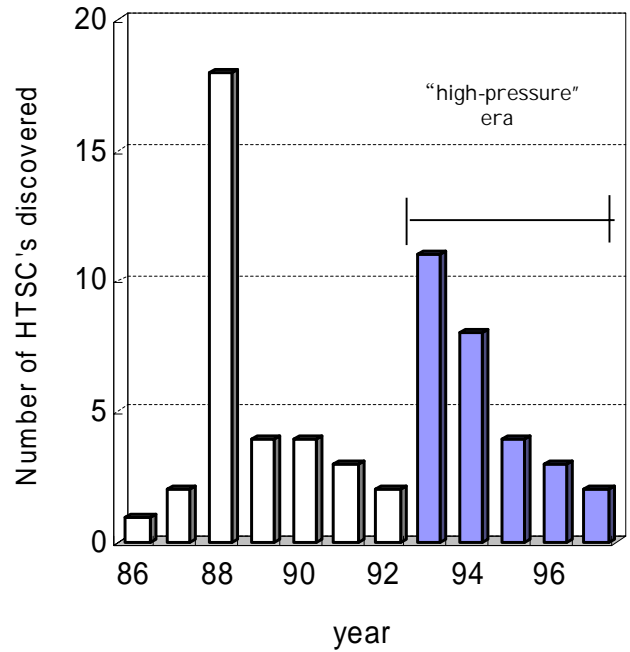


M. Karppinen & H. Yamauchi,  
Control of the charge inhomogeneity and  
high- $T_c$  superconducting properties in  
homologous series of multi-layered copper oxides,  
*Mater. Sci. Eng. R* **26**, 51-96 (1999).

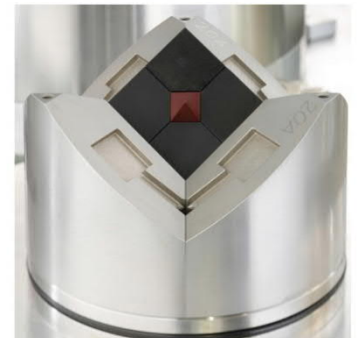
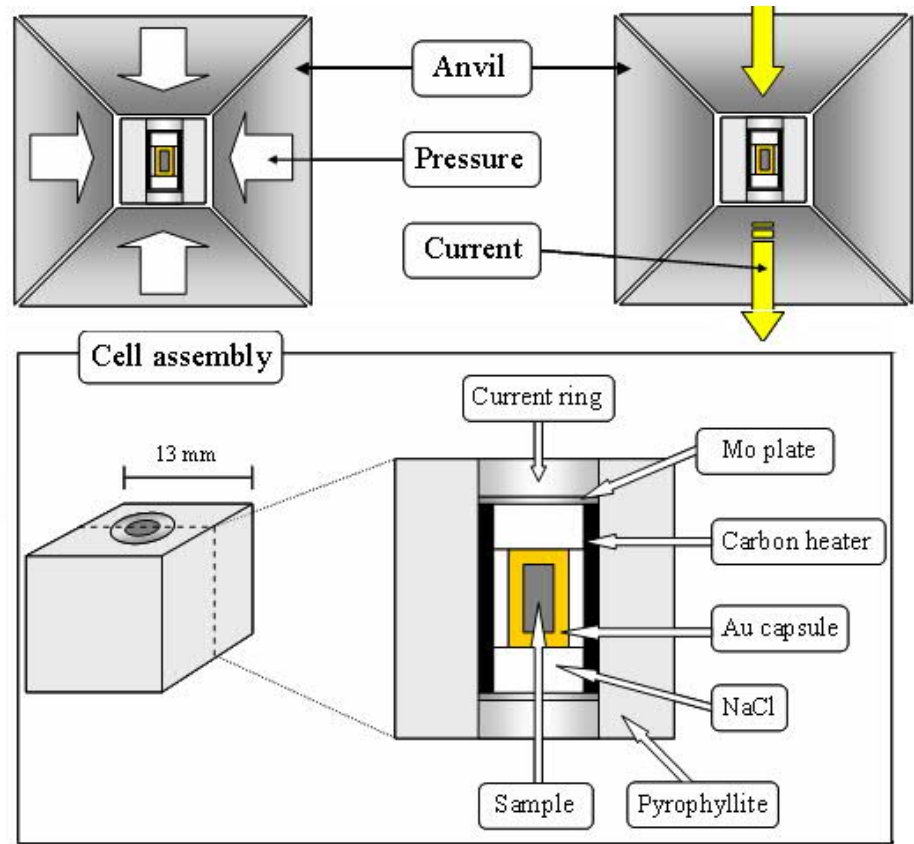
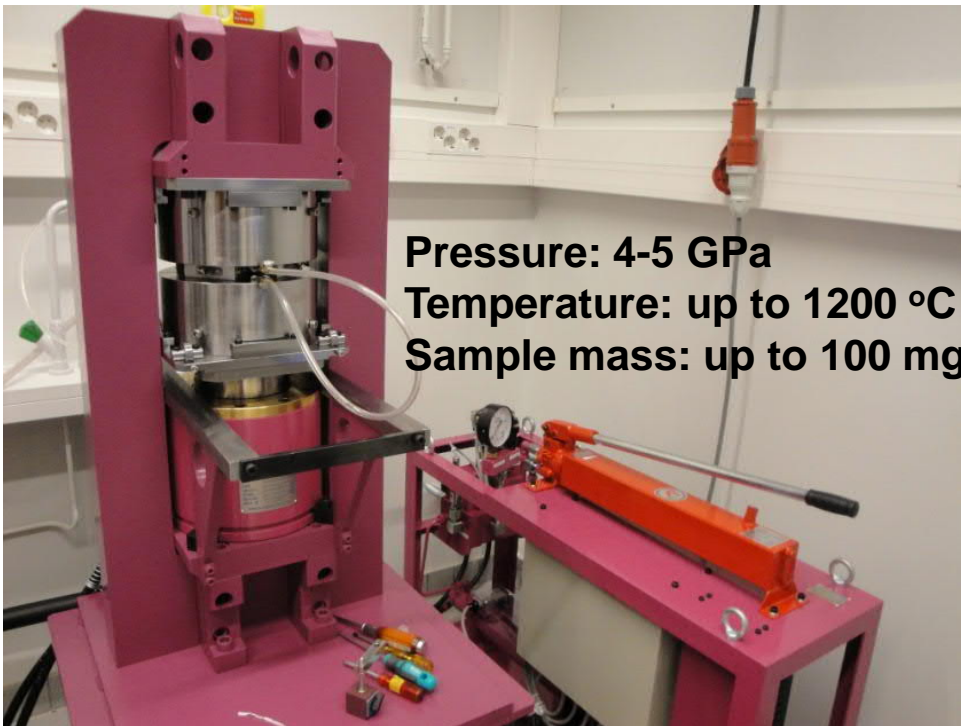


# Ultra-High-Pressure Synthesis

pressure: 2 ~ 8 GPa  
sample: 50 ~ 300 mg



# OUR HIGH-PRESSURE EQUIPMENT



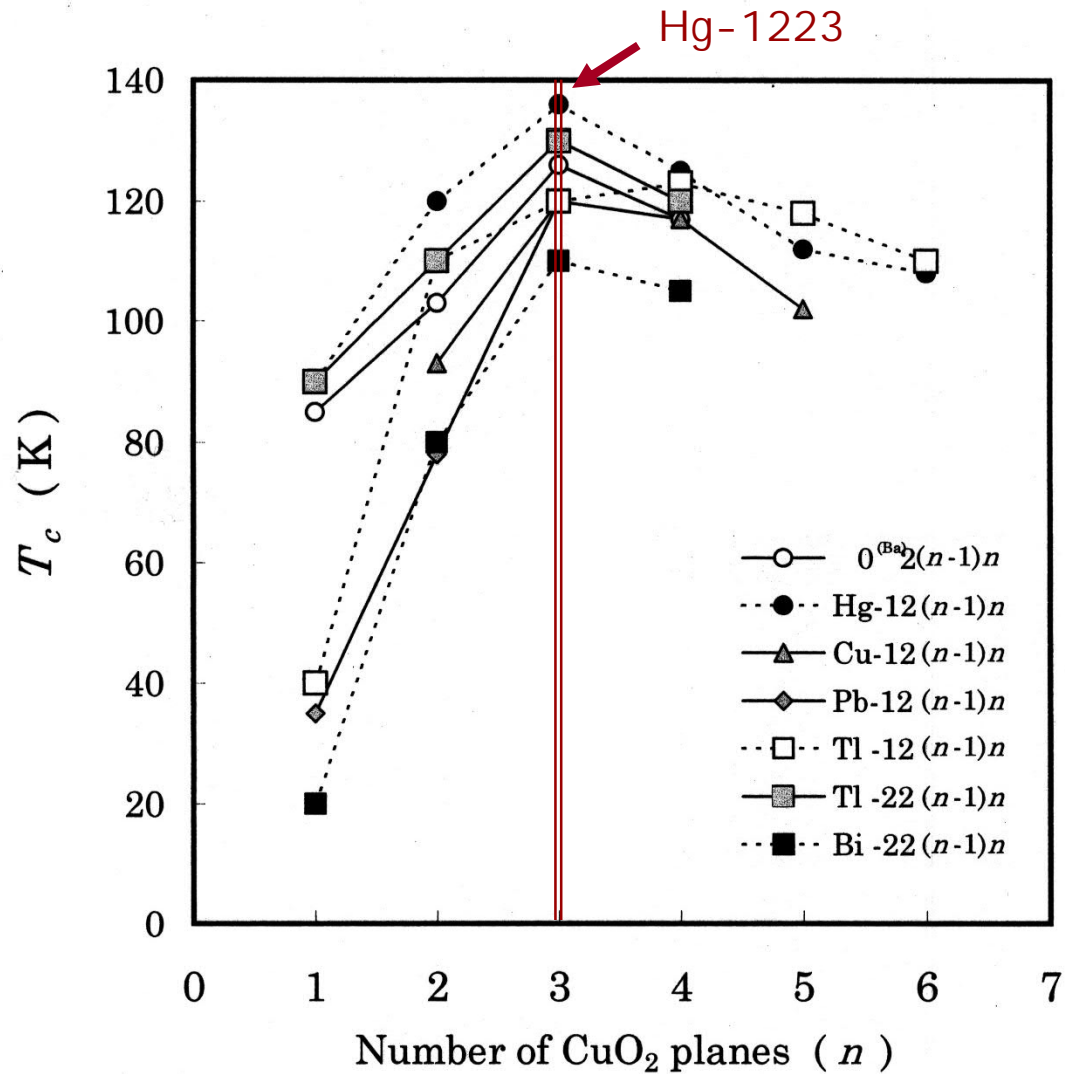


$$M - m2(n-1)n$$

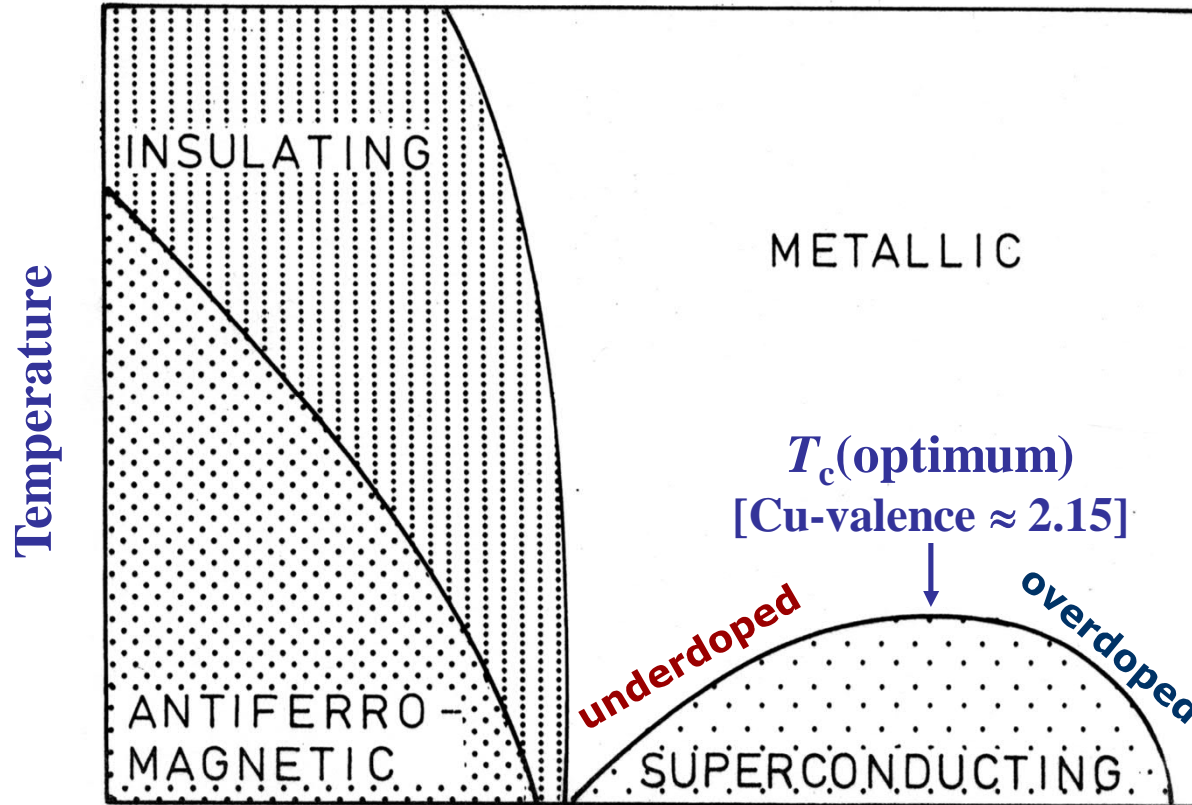
1 H																		2 He
3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne	
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar	
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr	
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe	
55 Cs	56 Ba	57 to 71	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn	
87 Fr	88 Ra	89 to 103	104 Rf	105 Ha	106 Sg	107 Ns	108 Hs	109 Mt										

57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr

$T_c$  versus “number of consecutively stacked  $\text{CuO}_2$  planes”



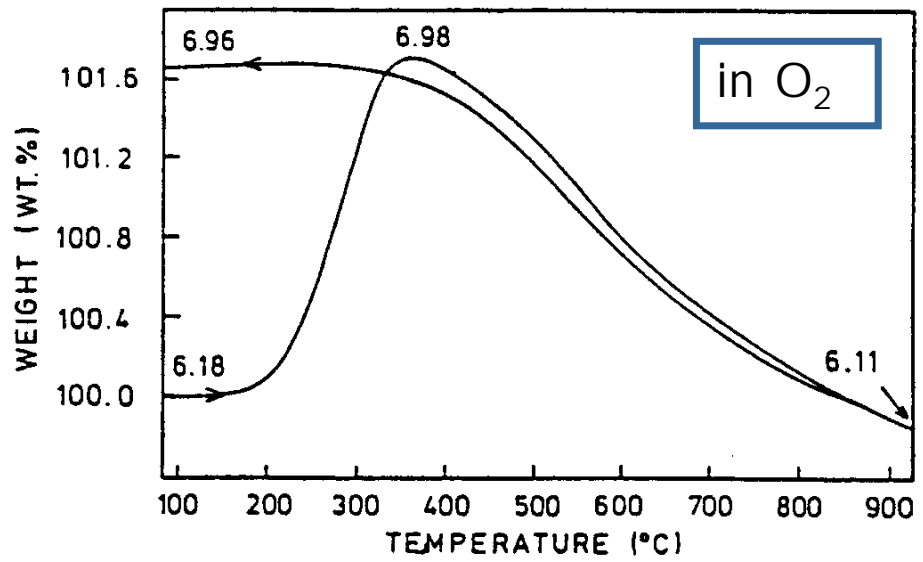
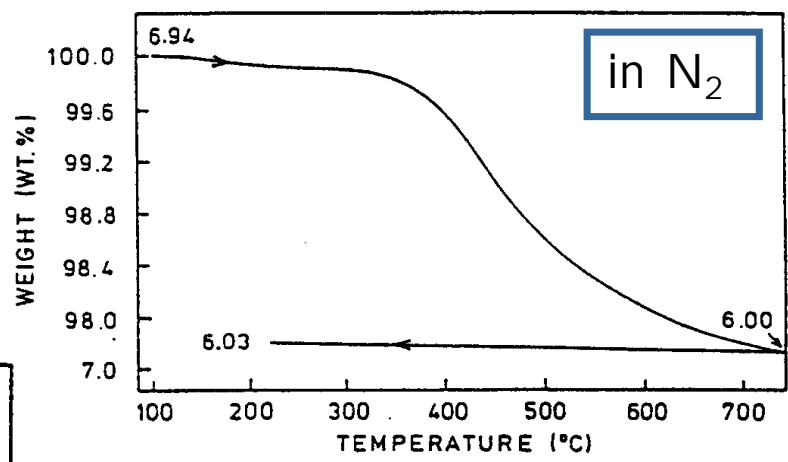
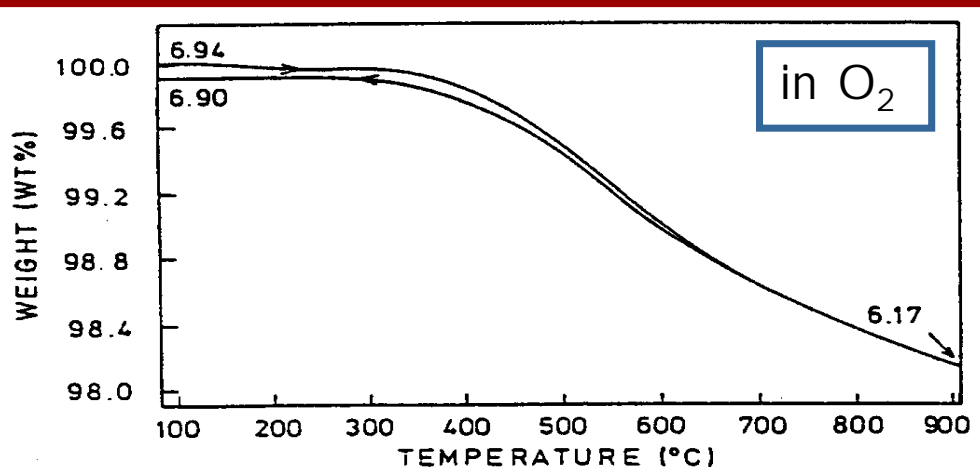
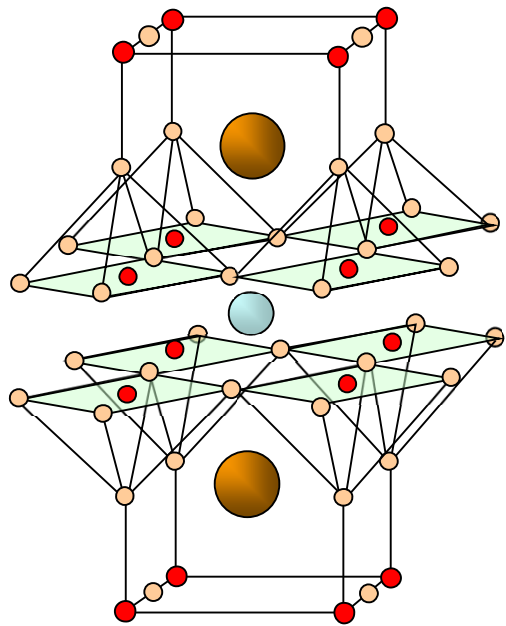
# Phase Diagram of HTSC

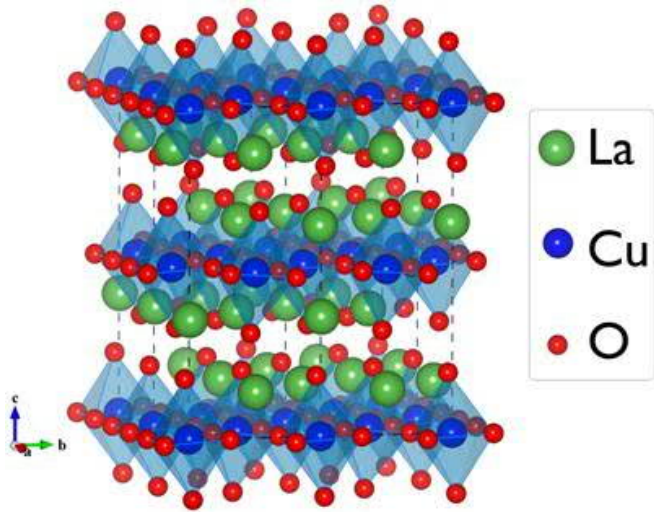


CuO<sub>2</sub>-plane hole concentration [p(CuO<sub>2</sub>)]  
Valence of copper [V(Cu)]

↑  
Chemistry

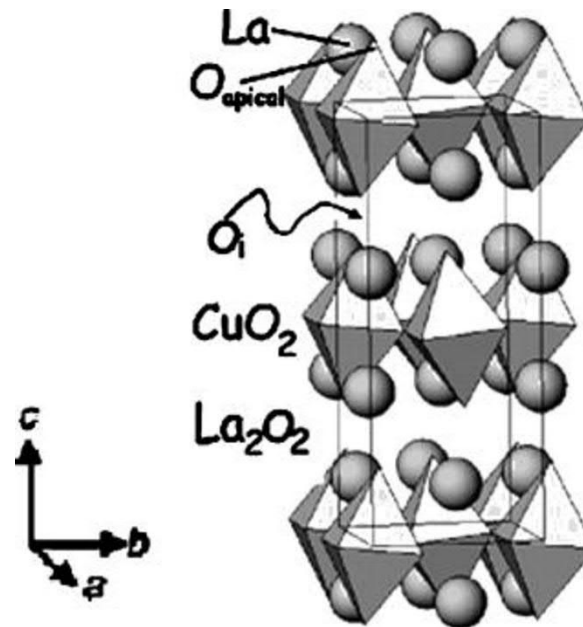
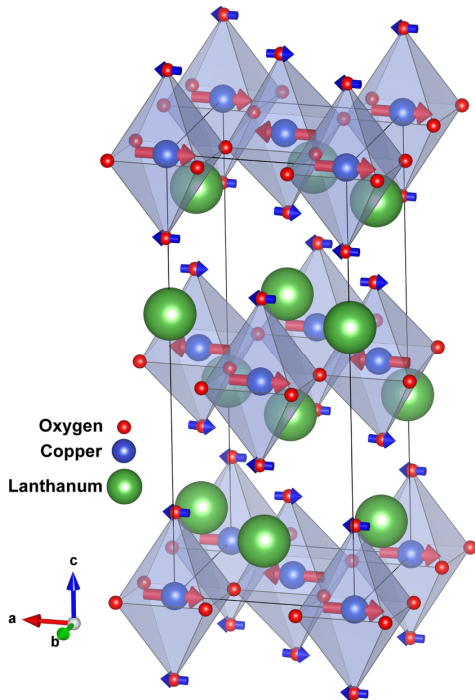
Hole-carrier concentration  
 $p(\text{CuO}_2) = V(\text{Cu}) - 2$

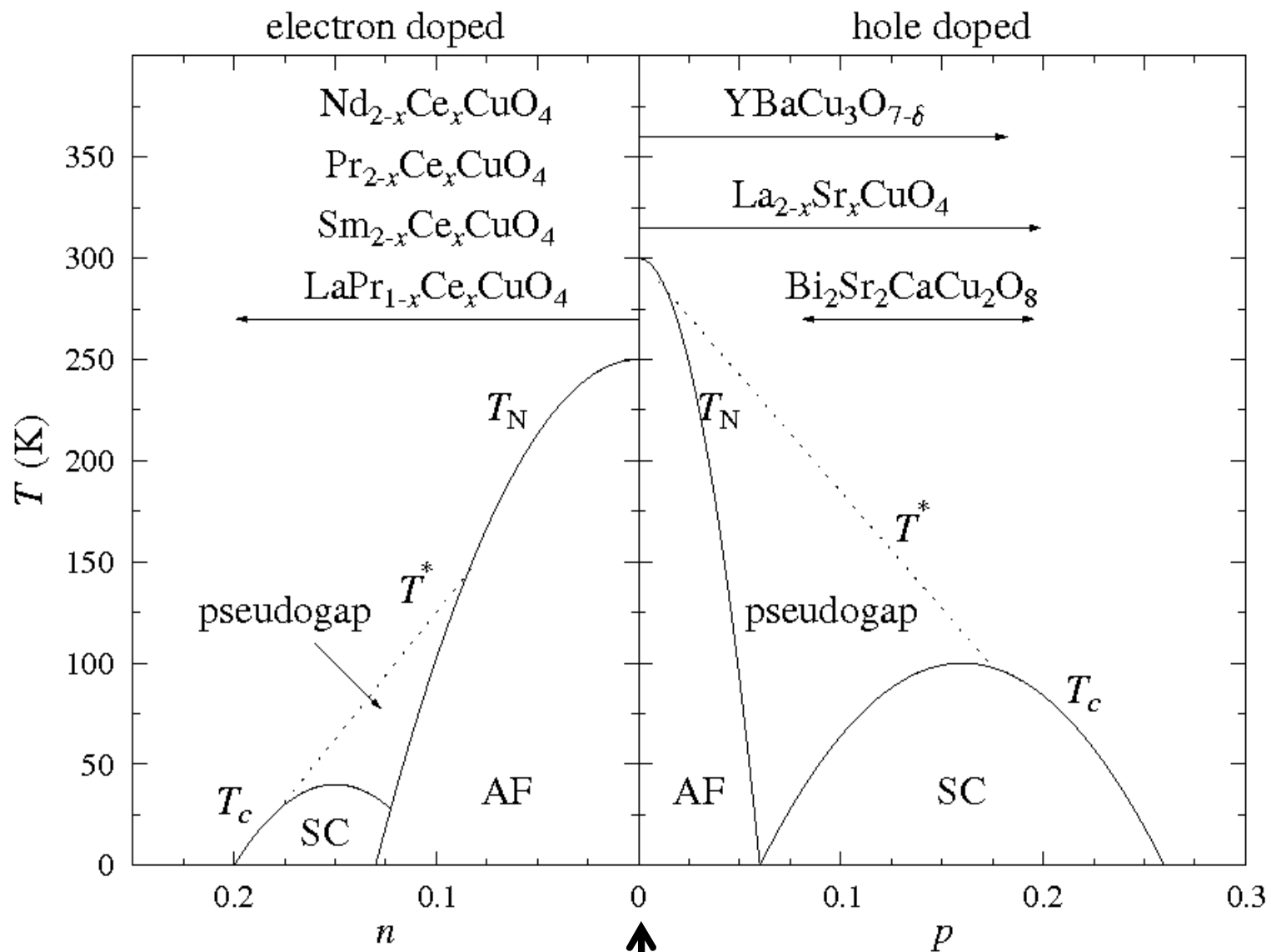




# La<sub>2</sub>CuO<sub>4</sub>

- Is this compound superconducting
- How to increase hole concentration
- What is its M-m2(n-1)n type name
- Does it belong to any general structure





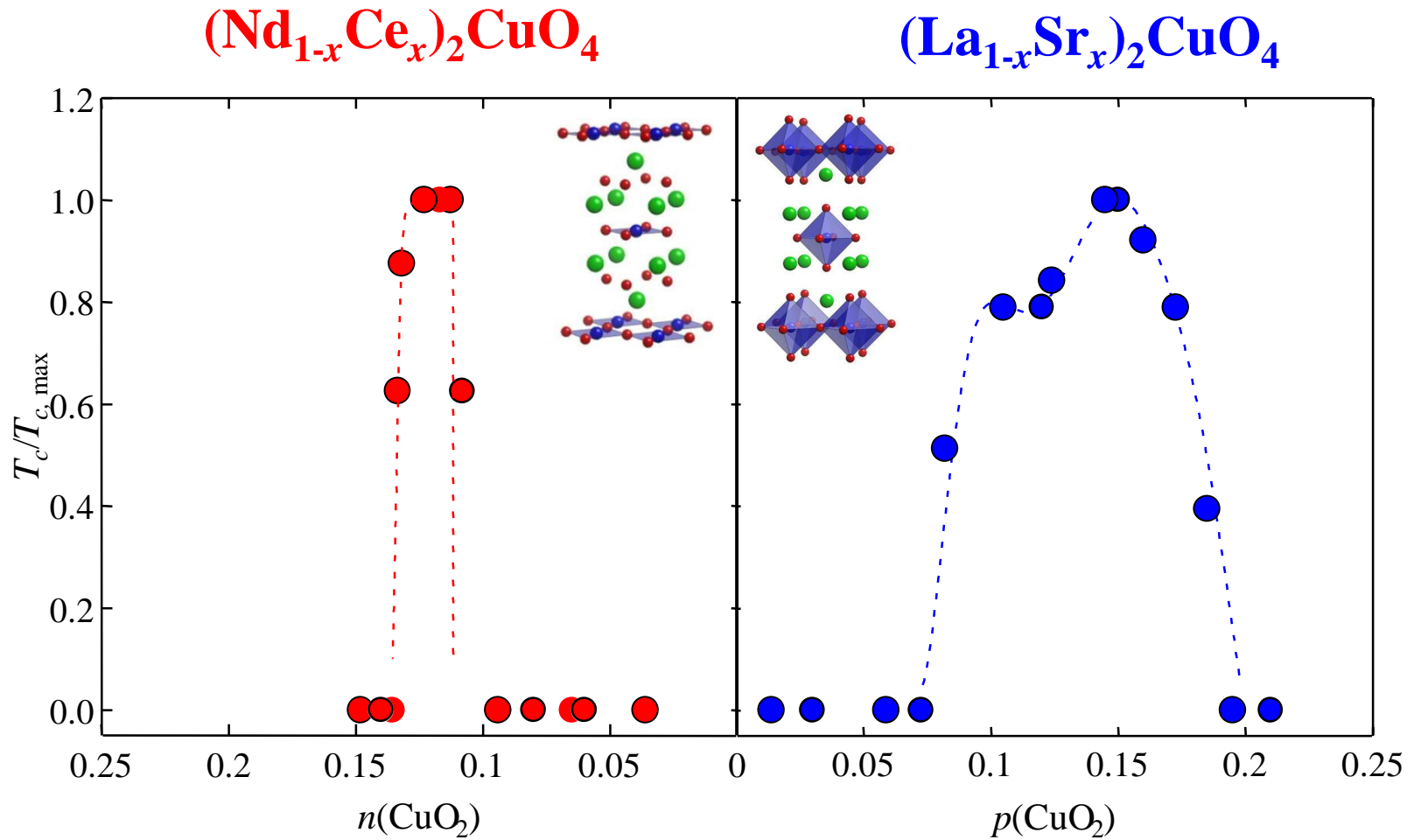
$n = \text{electron density} = 2 - V(\text{Cu})$

$V(\text{Cu}) = 2$

$p = \text{hole density} = V(\text{Cu}) - 2$



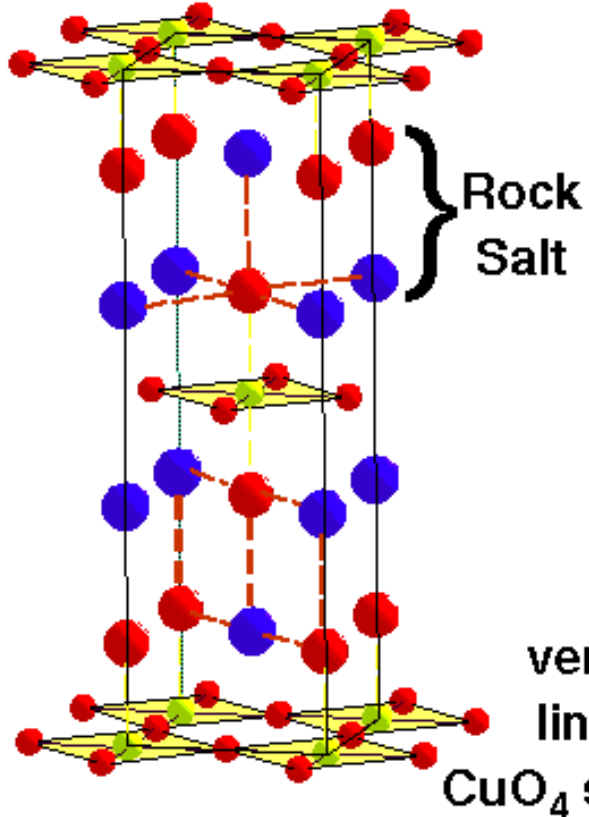
# Superconductivity phase diagrams, *n*-type vs. *p*-type



$$n(\text{CuO}_2) \equiv 2 - V(\text{Cu})$$

$$p(\text{CuO}_2) \equiv V(\text{Cu}) - 2$$

p-type doping



n-type doping

