1. Derive an expression for the Kapitza resistance between a solid and a liquid using the acoustic mismatch model. (We're looking for a proper derivation. Just presenting the lecture notes again is not enough.)

Hint: As in the lecture notes, start with Snell's law.

- 2. One cubic centimeter of non-pressurized ³He liquid is cooled in a cube-shaped cell. Determine the time constant with which the liquid cools at 10 mK. Use the values $c_3 = 23 T \text{ J/mol } \text{K}^2$ for the heat capacity of ³He and $R_K = 0.04 A^{-1}T^{-3} \text{ m}^2 \text{K}^4/\text{W}$ for the contact resistance.
- 3. How thick layer of silver powder should be sintered on the walls of the copper cell in the previous problem to achieve a ³He cooling time constant of one minute at 10 mK? Use the value of contact resistance given in the previous problem and assume a filling ratio of 0.5 for the sinter and a surface area of $2 \text{ m}^2/\text{g}$.
- 4. A helium chamber has a sintered silver heat exchanger with a total surface area of $50 \,\mathrm{m}^2$. Estimate the lowest attainable temperature
 - a) in pure liquid ${}^{3}\text{He}$
 - b) in liquid ${}^{3}\text{He} {}^{4}\text{He}$ mixture

when the heat leak into the sample is 0.1 nW. At a temperature below 10 mK, one can use a value of $1000 A^{-1}T^{-1} \text{ m}^2\text{K}^2/\text{W}$ for the Kapitza resistance of pure ³He and $16 A^{-1}T^{-2} \text{ m}^2\text{K}^3/\text{W}$ for the mixture.

5. Most materials contract when their temperature is decreased. Explain why metals typically contract less than plastics and dielectrics. Why some materials can have a negative coefficient of contraction, i.e. they expand while cooled down?