

1. Derive an expression for the conductance of a cylindrical pipe in the case of viscous laminar flow.

Hint: Use the Navier-Stokes equation in a steady state to solve the velocity field inside the cylinder.

2. The  $^3\text{He}$  circulation of a dilution refrigerator is maintained by a booster pump (with a pumping speed  $S = 4 \text{ m}^3/\text{s}$  below 10 Pa pressure). The pumping line consists of cylindrical pipe sections, which have the following lengths and average temperatures:

| part # | length $l$ (m) | $T$ (K) |
|--------|----------------|---------|
| 1      | 0.1            | 1       |
| 2      | 0.3            | 4.2     |
| 3      | 0.6            | 77      |
| 4      | 5              | 293     |

The temperature of the evaporation chamber (still) is kept at 0.6 K, so that only  $^3\text{He}$  is evaporated. At this temperature, the vapor pressure of helium mixture is 4.6 Pa. The desired flow rate is 2 mmol/s. Determine the diameters of the pipe sections so that the pressure difference over each section is the same. Use the experimental temperature dependence  $\eta_3 = 4.22 \cdot 10^{-7} (T/\text{K})^{0.647} \text{ s Pa}$  for the viscosity of  $^3\text{He}$  gas (compare to ideal gas:  $\eta = 9.2 \cdot 10^{-7} (T/\text{K})^{0.5} \text{ s Pa}$ ).

3. A dewar is filled with liquid  $^4\text{He}$  at normal pressure and 4.2 K temperature. The bath is cooled by pumping. What fraction of the liquid is evaporated before it becomes superfluid ( $T = 2.17 \text{ K}$ )? The latent heat is  $L \approx 22 \text{ J/g}$ . The  $^4\text{He}$  enthalpy is  $H(4.2 \text{ K}) = 9.25 \text{ J/g}$  and  $H(2.17 \text{ K}) = 3.5 \text{ J/g}$ .
4. A  $^3\text{He}$  evaporation cryostat, which has separate  $^3\text{He}$  and  $^4\text{He}$  evaporators, is pumped by mechanical pumps, which have a pumping speed  $S = 35 \text{ m}^3/\text{h}$ . See Fig. 1 for a schematic. The conductance of the pumping lines is  $C = 25 \text{ m}^3/\text{h}$ . The heat load of the  $^4\text{He}$  evaporator consists of the  $^4\text{He}$  enthalpy arriving into the evaporator, the thermally anchored  $^3\text{He}$  return capillary, and a 0.1 mW external load. The  $^3\text{He}$  evaporator is loaded by the  $^3\text{He}$  arriving through the  $^4\text{He}$  evaporator and an external load of 0.1 mW. What are the

equilibrium temperatures of the two evaporators? Use  $L_4 = 85 \text{ J/mol}$  and  $L_3 = 25 \text{ J/mol}$  for the heats of evaporation,  $C_3 = (2.1 + 2.2 T/\text{K}) \text{ J}/(\text{mol K})$  for the  $^3\text{He}$  specific heat, and assume that 40% of the cooling power of the  $^4\text{He}$  evaporator goes into cooling the  $^4\text{He}$  liquid itself ( $4.2 \text{ K} \rightarrow T_4$ ). The  $^4\text{He}$  and  $^3\text{He}$  vapor pressures behave in the temperature range in question approximately as  $P_4 = 640e^{-10.6\text{K}/T} \text{ kPa}$  and  $P_3 = 46e^{-3.7\text{K}/T} \text{ kPa}$ , respectively. The parts of the piping marked with flow impedances  $Z_{43}$  and  $Z_{04}$  are considered thin in order to uphold some pressure differences across the system.

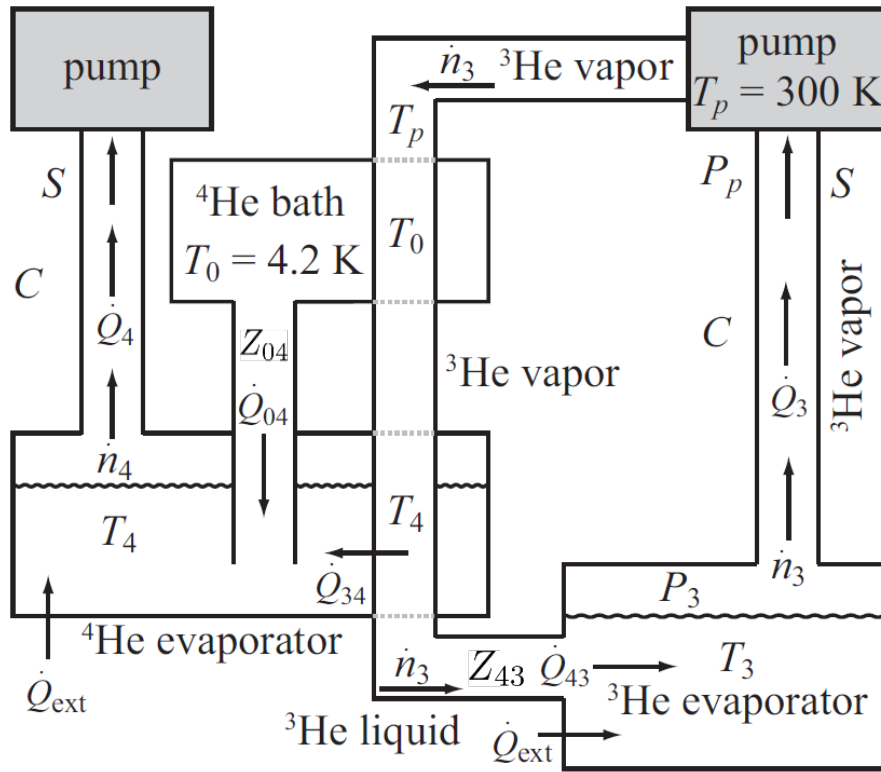


Figure 1: Schematic of a  $^3\text{He}$  evaporation cryostat.