

# LECTURE 4: CRYSTALLOGRAPHY BASICS

- From molecules to 3D crystals
- Translation in 3D crystals → new symmetry operations (glide planes & screw axes)
- From point groups to space groups
- Crystal lattice, lattice points & unit cell
- International Tables of Crystallography

# FROM MOLECULES TO **CRYSTALS**

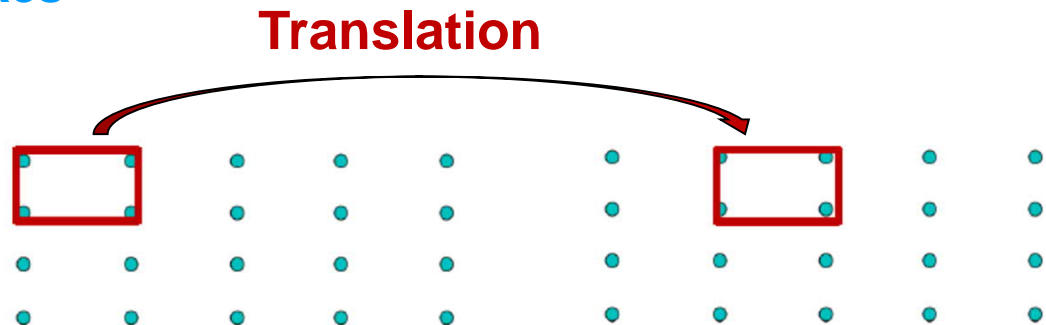
There are two things which make macroscopic (infinite) crystals different from discrete molecules in terms of symmetry: **Space-filling & Translation**

## **Space-filling**

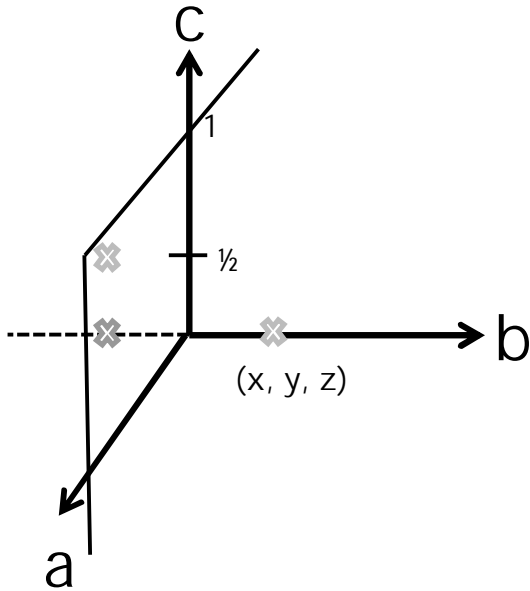
- Macroscopic crystals need to continuously fill the space → 5-cubes can not satisfy this (except in quasicrystals)
- For molecules 5-fold rotation is possible, but not for crystals

## **Translation in crystals**

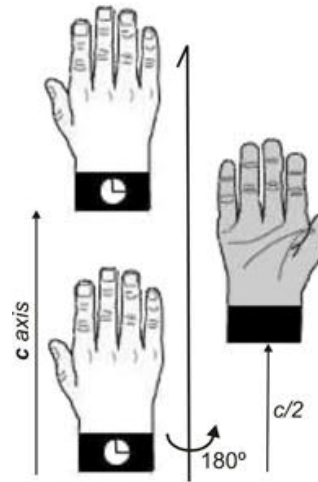
- Translation: move from one point to another (the entire object)
- This does not exist in molecules, but is the essence of macroscopic crystals exhibiting long-range order
- Combining translation with other symmetry operations/elements → new symmetry operations/elements (not included in point groups): **glide planes** & **screw axes**



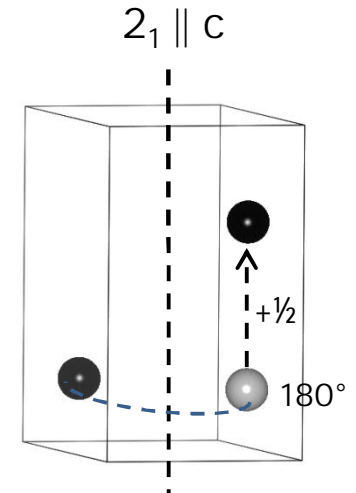
# GLIDE PLANE $c (\perp b)$



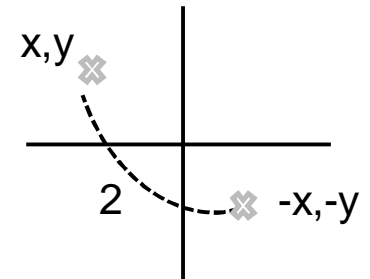
Rotation (c) followed by translation (t)



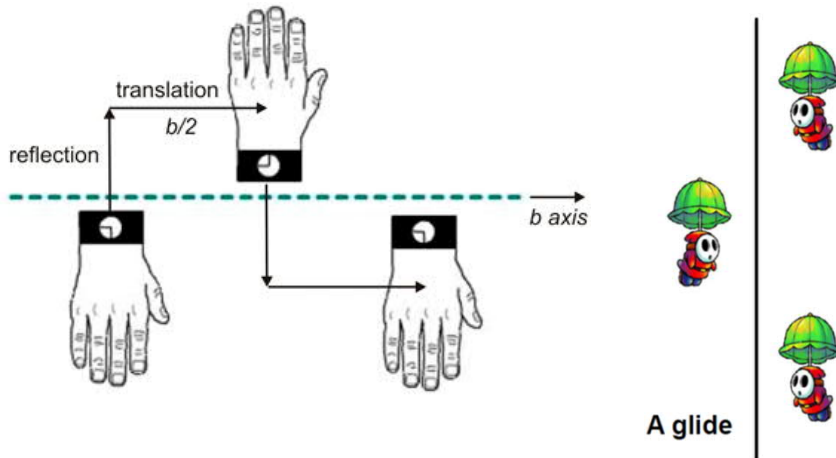
# SCREW AXIS



$$x, y, z \rightarrow -x, -y, z + \frac{1}{2}$$



Reflection (m) through ac-plane, followed by (half) translation (t) along c-axis



A glide

## In infinite lattice there are additional translation symmetry elements:

### Translation

Translation (move) from one point to another point

### Screw axis

Rotation plus translation

$n_m$  ( $2_1, 3_1, 3_2, 4_1, 4_2, 4_3, 6_1, 6_2, 6_3, 6_4, 6_5$ )

For example:  $2_1$ : rotation  $180^\circ$  and translation  $\frac{1}{2}(m/n)$

### Glide plane

Reflection against a mirror plane plus (half) translation parallel to the plane

Axis glide plane:  $a, b, c$  (siirrot  $\frac{1}{2}a, \frac{1}{2}b, \frac{1}{2}c$  kuhunkin liukutason suuntaan)

Diagonal glide plane:  $n$  [ $\frac{1}{2}(a+b), \frac{1}{2}(b+c), \frac{1}{2}(c+a)$ ]

$d$  [ $\frac{1}{4}(a+b), \frac{1}{4}(b+c), \frac{1}{4}(c+a)$ ] (so-called diamond glide plane)

(from historical reasons)  
**DIFFERENT**

# SYMMETRY SYMBOLS

## ■ Schoenflies (S) symbols

- were developed first
- in molecular symmetry & spectroscopy

## ■ Hermann-Mauguin (HM) symbols

- in crystallography
- long and short forms

## ■ Graphical symbols

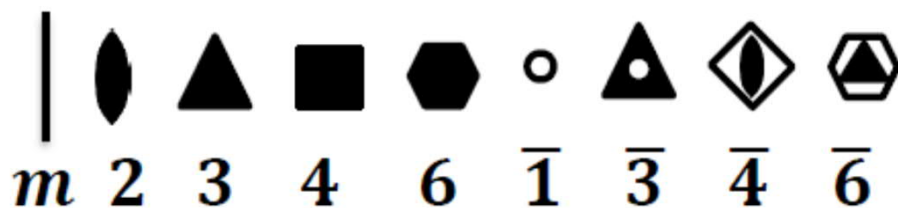


Table 1.1 Symmetry elements of crystal point groups.

System	Point group symmetry		Symmetry elements	Number of operations
	<i>S</i>	<i>H/M</i>		
Triclinic	$C_1$	1	$I = C_1$	1
	$C_1 = S_2$	$\bar{1}$	$I, i (= S_2)$	2
Monoclinic	$C_2$	2	$I, C_2$	2
	$C_2 = C_{1h} = C_{1v}$	<i>m</i>	$I, \sigma$	2
	$C_{2h}$	$2/m$	$I, C_2, \sigma_h, i$	4
Orthorhombic	$C_{2v}$	<i>mm2</i>	$I, C_2, 2\sigma$	4
	$D_2$	222	$I, 3C_2$	4
	$D_{2h}$	<i>mmm</i>	$I, 3C_2, 3\sigma, i$	8
Tetragonal	$C_4$	4	$I, C_4$	4
	$S_4$	$\bar{4}$	$I, S_4 (= C_2)$	4
	$D_4$	422	$I, C_4 (= C_2), 2C_2', 2C_2''$	8
	$C_{4v}$	$4mm$	$I, C_4, 2\sigma_v, 2\sigma_d$	8
	$C_{4h}$	$4/m$	$I, C_4 (= S_4), \sigma_h, i$	8
	$D_{2d}$	$\bar{4}2m$	$I, S_4 (= C_2), 2C_2', 2\sigma_d$	8
	$D_{4h}$	$4/mmm$	$I, C_4 (= S_4), 2C_2', 2C_2'', 2\sigma_v, 2\sigma_d, i$	16
Trigonal	$C_3$	3	$I, C_3$	3
	$C_{3i} = S_6$	$\bar{3}$	$I, S_6 (= C_3), i$	6
	$D_3$	32	$I, C_3, 3C_2$	6
	$C_{3v}$	$3m$	$I, C_3, 3\sigma_v$	6
	$D_{3d}$	$\bar{3}m$	$I, S_6 (= C_3), 3C_2, 3\sigma_d, i$	12
Hexagonal	$C_6$	6	$I, C_6$	6
	$C_{3h}$	$\bar{6}$	$I, S_3 (= C_3), \sigma_h$	6
	$D_6$	622	$I, C_6, 3C_2', 3C_2''$	12
	$D_{3h}$	$\bar{6}m2$	$I, C_3 (= S_3), 3C_2, 3\sigma_v, \sigma_h$	12
	$C_{6h}$	$6/m$	$I, C_6 (= S_6), \sigma_h, i$	12
	$C_{6v}$	$6mm$	$I, C_6, 3\sigma_v, 3\sigma_d$	12
	$D_{6h}$	$6/mmm$	$I, C_6 (= S_6), 3C_2', 3C_2'', 3\sigma_v, 3\sigma_d, \sigma_h, i$	24
Cubic	$T$	23	$I, 3C_2, 4C_3$	12
	$T_h$	$m\bar{3}$	$I, 3C_2, 4C_3 (= S_6), 3\sigma_h, i$	24
	$T_d$	$\bar{4}3m$	$I, 3C_2 (= S_4), 4C_3, 6\sigma_d$	24
	$O$	432	$I, 3C_2, 4C_3, 3C_4$	24
	$O_h$	$m\bar{3}m$	$I, 3C_2, 4C_3 (= S_6), 3C_4 (= S_4), 3\sigma_h, 6\sigma_d, i$	48

# Symmetry elements (Schönflies/Hermann-Mauguin)

Identity  $E$

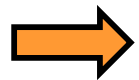
Symmetry/inversion center  $i / \bar{1}$

Rotation axis  $C_n / 1, 2, 3, \dots$

Reflection/mirror plane  $\sigma / m$

Improper rotation axis  $S_n / (\bar{1}, \bar{2}), \bar{3}, \bar{4}, \bar{6}$

One point remains unchanged



Point/Molecular symmetry

NOTE:

- Improper rotation axis  $\bar{1}$  and inversion center  $\bar{1}$  are equivalent
- Mirror plane  $m$  and improper rotation axis  $\bar{2}$  are equivalent

## WHAT WE LIKE TO KNOW ABOUT THE CRYSTAL STRUCTURE

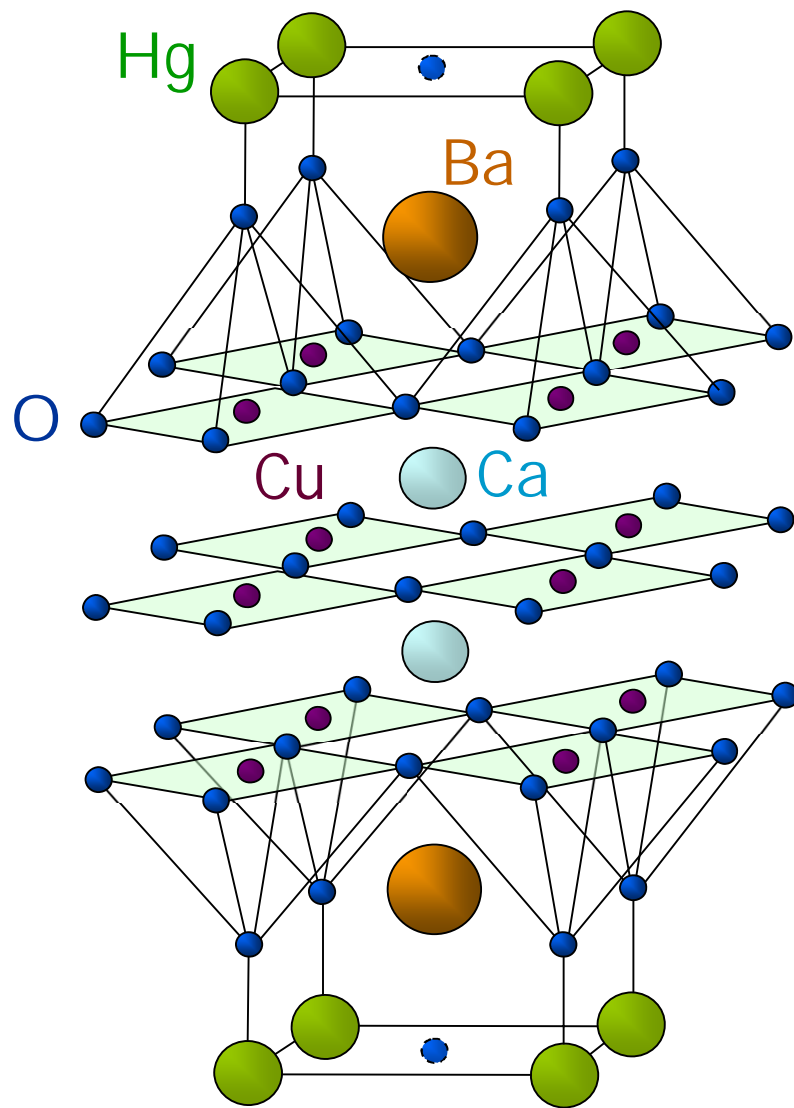
### CRYSTALLOGRAPHY

- symmetry
- unit cell
- lattice parameters
- number of formula units in unit cell
- space group
- etc.

### CRYSTAL CHEMISTRY

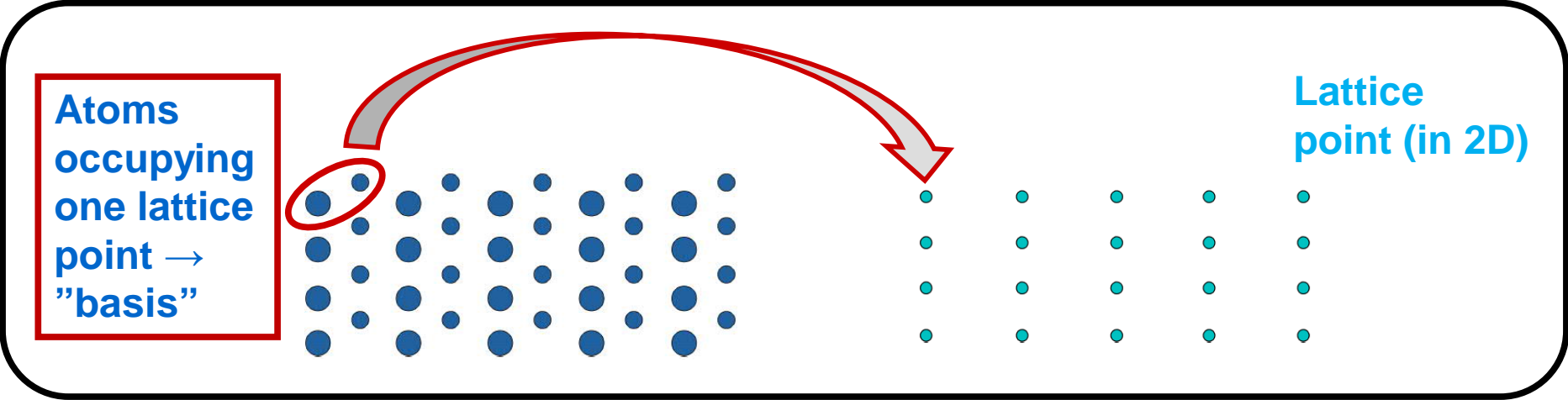
- coordination numbers
- coordination polyhedra
- bond lengths/angles
- occupation factors
- etc.

Discussed in Lecture 2!



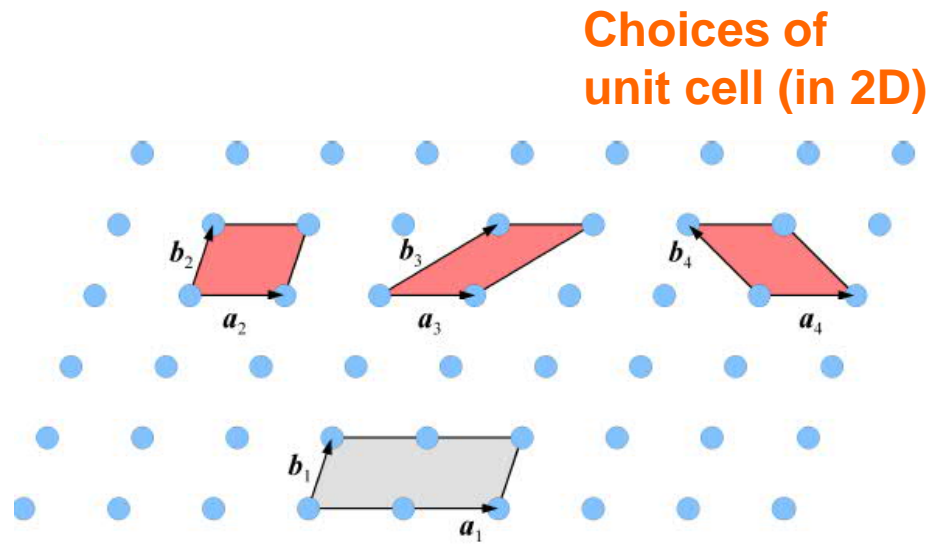
# CRYSTAL LATTICE

- Regular (infinite 3D) arrangement of lattice points
- **Lattice point**: consists of **one or more atoms (= basis)**
- Each lattice point has identical environment + symmetry properties (point group)



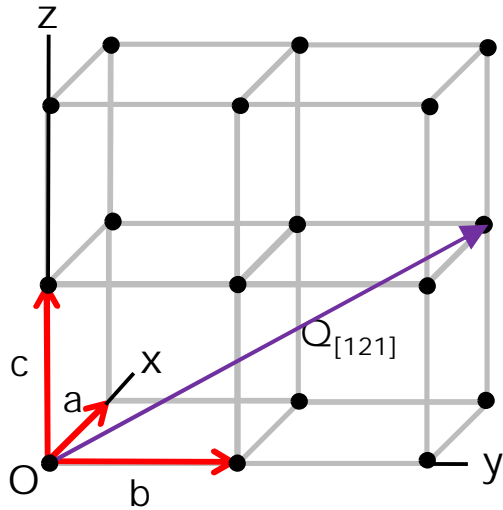
# UNIT CELL

- Smallest possible microscopic 3D part of the crystal lattice that repeats itself periodically and completely fills the lattice volume, and is enough to describe the lattice perfectly
- Choice of the unit cell not always unambiguous: several possible choices

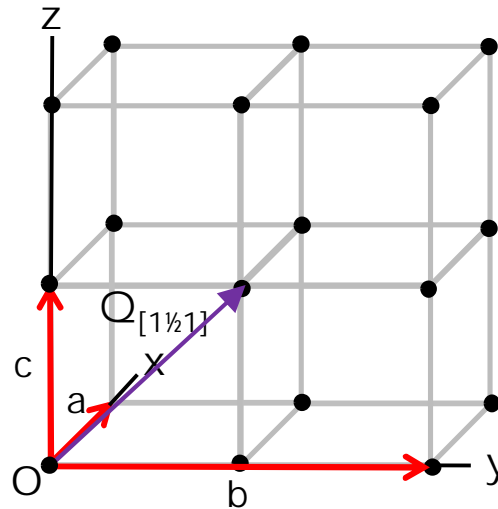




Each lattice point should be able to be described with the vector  $Q_{[uvw]} = ua + vb + wc$ , where  $u$ ,  $v$  and  $w$  all are integer values



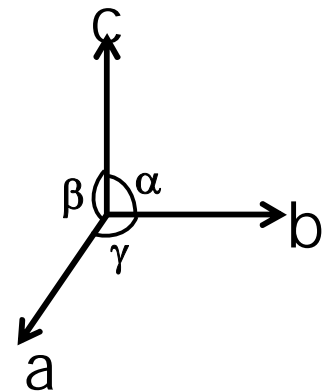
Unit cell: YES



Unit cell: NO

## What we need to tell about the UNIT CELL

- Shape & size of the unit cell *plus* the atomic positions in the unit cell
- Shape & size are given by:
  - Lattice parameters:  $a, b$  ja  $c$
  - Angles between the axes:  $\alpha, \beta$  ja  $\gamma$



# CLASSIFICATIONS

- "Macroscopic shape of the crystal" → Point group for the lattice → **7 CRYSTAL SYSTEMS** (can fill the space without holes; no information of the lattice points/atoms)

## CRYSTAL SYSTEM & LATTICE POINTS (historical importance)

- Combination of crystal system and lattice type → **14 Bravais lattices**
- **Lattice type:** positions of lattice points ( $\neq$  atoms) within the cell considered: primitive (P), body-centered (I), face-centered (F), base-centered (A/B/C), rhombohedral (R)

## CRYSTAL SYSTEM & SYMMETRY (most important in crystallography)

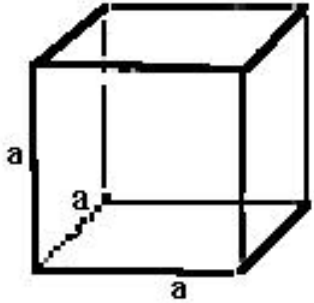
- Possible combinations of point symmetry elements (which leave one point fixed) → **32 (geometric) crystal classes** → describe completely the symmetry of macroscopic crystals

There are an infinite number of **three**-dimensional point groups, but the crystallographic restriction results in there being only **32** crystallographic point groups.

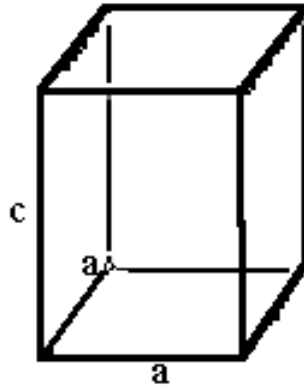
- Considering also the translational symmetry operations in 3D → **230 space groups**

# CRYSTAL SYSTEMS (7)

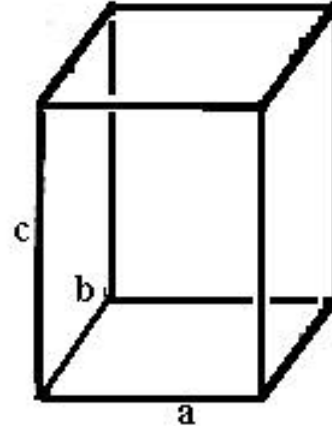
No information of the positions of atoms



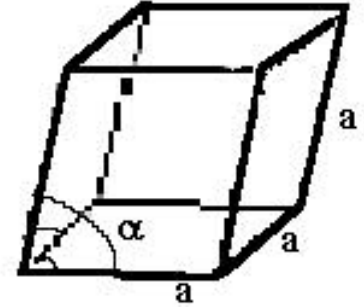
Cubic  
 $a = b = c$   
 $\alpha = \beta = \gamma = 90^\circ$   
NaCl, MgAl<sub>2</sub>O<sub>4</sub>



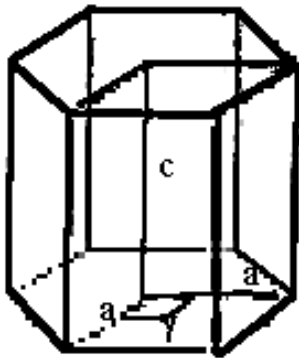
Tetragonal  
 $a = b \neq c$   
 $\alpha = \beta = \gamma = 90^\circ$   
TiO<sub>2</sub>, K<sub>2</sub>NiF<sub>4</sub>



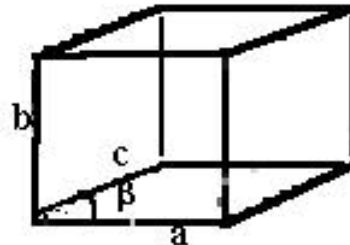
Orthorhombic  
 $a \neq b \neq c$   
 $\alpha = \beta = \gamma = 90^\circ$   
YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7</sub>



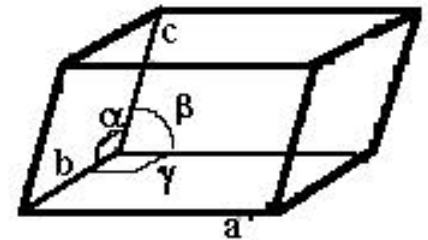
Rhombohedral  
 $a = b = c$   
 $\alpha = \beta = \gamma \neq 90^\circ$   
BaTiO<sub>3</sub> (low-T)



Hexagonal  
 $a = b \neq c$   
 $\alpha = \beta = 90^\circ, \gamma = 120^\circ$   
LiNbO<sub>3</sub>



Monoclinic  
 $a \neq b \neq c$   
 $\alpha = \gamma = 90^\circ, \beta \neq 90^\circ$   
KH<sub>2</sub>PO<sub>4</sub>



Triclinic  
 $a \neq b \neq c$   
 $\alpha \neq \beta \neq \gamma \neq 90^\circ$

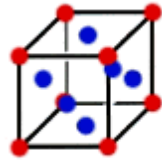
Element	What it does	Possible in crystal system
Identity (1)	-	All
Inversion ( $\bar{1}$ )	Inversion	All
Mirror plane (m)	Mirror	All but triclinic
2- fold rotation (2)	Rotate $180^\circ$	All but triclinic
3-fold rotation (3)	Rotate $120^\circ$	Trigonal, Hexagonal and Cubic
4-fold Rotation (4)	Rotate $90^\circ$	Tetragonal and Cubic
6-fold Rotation (6)	Rotate $60^\circ$	Hexagonal

# BRAVAIS LATTICES (14)

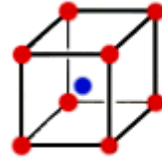
Positions of lattice sites (not atoms) included



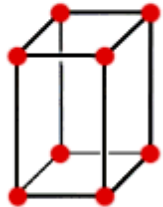
Simple cubic



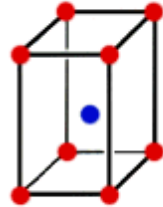
Face-centered cubic



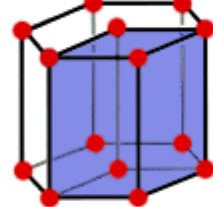
Body-centered cubic



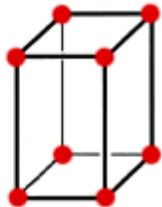
Simple tetragonal



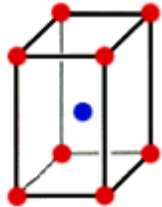
Body-centered tetragonal



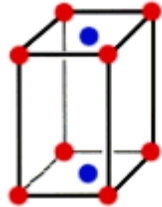
Hexagonal



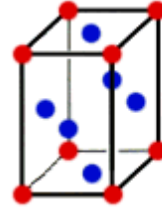
Simple orthorhombic



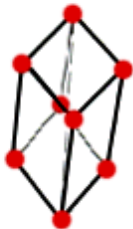
Body-centered orthorhombic



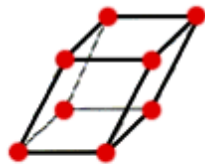
Base-centered orthorhombic



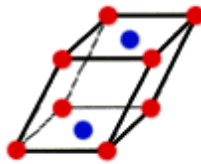
Face-centered orthorhombic



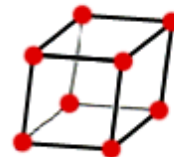
Rhomboidal



Simple Monoclinic



Base-centered monoclinic



Triclinic

Centering	Lattice points/cell	Abbreviation
Primitive	1	P
Base (A,B, or C) centered	2	A,B or C
Body centered	2	I
Hexagonal rhombohedral	3	hR
Face centered	4	F

# EXAMPLE

- What is the Bravais lattice type of NaCl: Cubic F (basis: Na-Cl)

Rock-salt (NaCl) structure

Paolo Fornasini  
Univ. Trento

NaCl	$a=5.64 \text{ \AA}$
KBr	$a=6.60 \text{ \AA}$
CaO	$a=4.81 \text{ \AA}$

conventional unit cell (8 atoms per cell)

Non-Bravais lattice

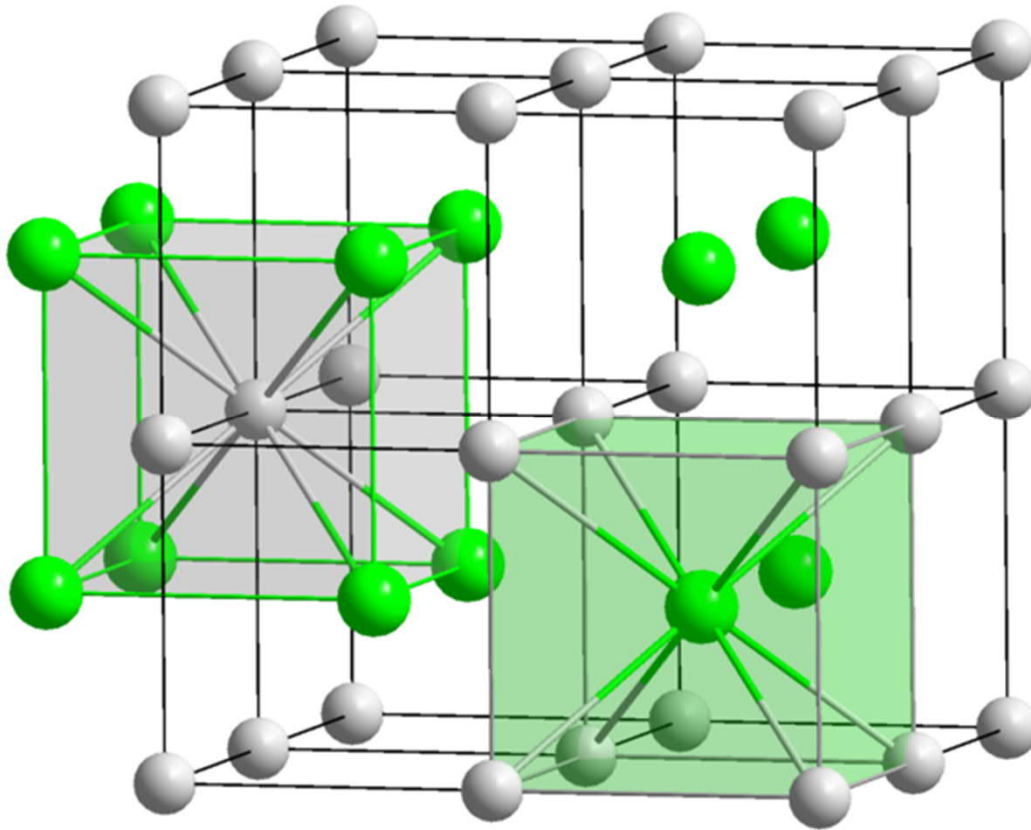
fcc Bravais lattice + 2-atom basis

$(0,0,0)$   $\left(\frac{1}{2}, \frac{1}{2}, \frac{1}{2}\right)$

Cordination number = 6

# EXAMPLE

- What is the Bravais lattice type of CsCl: Cubic P (basis: Cs-Cl)





# CRYSTAL CLASSES (32)

The 32 Point Groups			
1	4	$\bar{3}$	6mm
$\bar{1}$	$\bar{4}$	32	$\bar{6}m2$
2	4/m	3m	6/mmm
m	422	$\bar{3}m$	23
2/m	4mm	6	$m\bar{3}$
222	$\bar{4}2m$	$\bar{6}$	432
mm2	4/mmm	6/m	$\bar{4}3m$
mmm	3	622	$m\bar{3}m$

The 32 Point Groups (Schoenflies)			
1 ( $C_1$ )	4 ( $C_4$ )	$\bar{3}$ ( $C_{3i}$ )	6mm ( $C_{6\sigma v}$ )
$\bar{1}$ ( $C_i = S_2$ )	$\bar{4}$ ( $S_4$ )	32 ( $D_3$ )	$\bar{6}m2$ ( $D_{3\sigma h}$ )
2 ( $C_2$ )	4/m ( $C_{4\sigma h}$ )	3m ( $C_{3\sigma v}$ )	6/mmm ( $D_{6\sigma h}$ )
m ( $C_\sigma$ )	422 ( $D_4$ )	$\bar{3}m$ ( $D_{3d}$ )	23 (T)
2/m ( $C_{2\sigma h}$ )	4mm ( $C_{4\sigma v}$ )	6 ( $C_6$ )	$m\bar{3}$ ( $T_h$ )
222 ( $D_2$ )	$\bar{4}2m$ ( $D_{2d}$ )	$\bar{6}$ ( $C_{3\sigma h}$ )	432 (O)
2mm ( $C_{2\sigma v}$ )	4/mmm ( $D_{4h}$ )	6/m ( $C_{6\sigma h}$ )	$\bar{4}3m$ ( $T_d$ )
mmm ( $D_{2\sigma h}$ )	3 ( $C_3$ )	622 ( $D_6$ )	$m\bar{3}m$ ( $O_h$ )



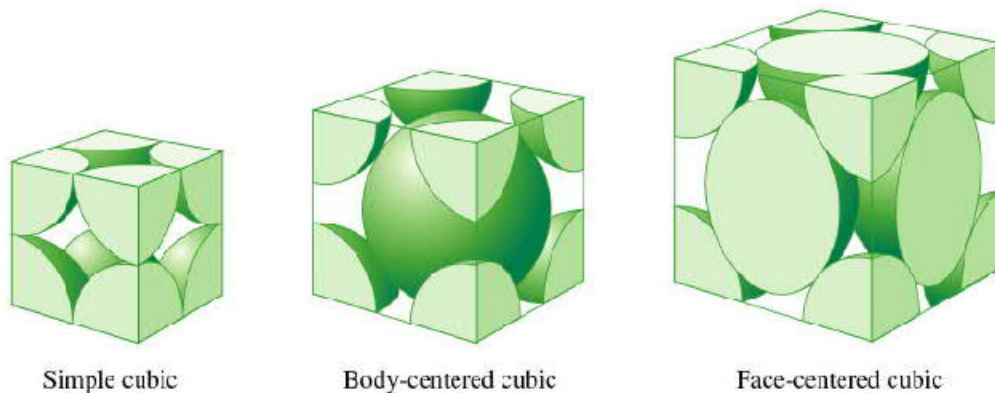
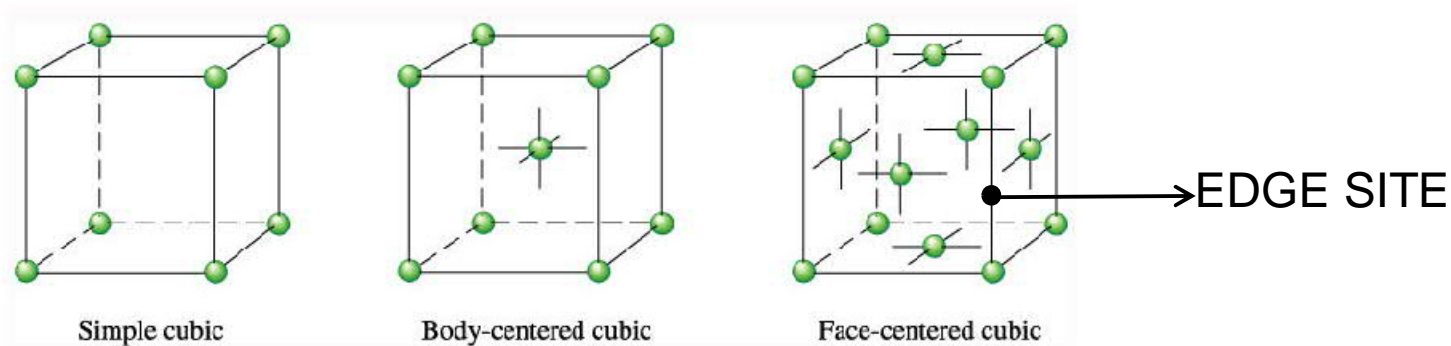
<b>Crystal System</b>	<b># of Point Groups</b>	<b># of Crystal Lattices</b>
<b>Cubic</b>	<b>5</b>	<b>3</b>
<b>Tetragonal</b>	<b>7</b>	<b>2</b>
<b>Orthorhombic</b>	<b>3</b>	<b>4</b>
<b>Monoclinic</b>	<b>3</b>	<b>2</b>
<b>Triclinic</b>	<b>2</b>	<b>1</b>
<b>Hexagonal</b>	<b>7</b>	<b>1</b>
<b>Trigonal</b>	<b>5</b>	<b>1</b>
<b>Totals</b>	<b>32</b>	<b>14</b>

System	Minimum Requirements
Cubic	Four 3-fold rotation axis
Tetragonal	One 4-fold rotation (or R <sub>I</sub> ) axis
Orthorhombic	Three perpendicular 2-fold axis
Rhombohedral	One 3-fold rotation (or R <sub>I</sub> ) axis
Hexagonal	One 6 fold rotation (or R <sub>I</sub> ) axis
Monoclinic	One 2 fold rotation axis or mirror plane
Triclinic	none

System	Point groups
Cubic	23, $m\bar{3}$ , 432, $\bar{4}3m$ , $m\bar{3}m$
Tetragonal	4, $\bar{4}$ , 4/m, 422, 4mm, $\bar{4}2m$ , 4/mmm
Orthorhombic	222, 2mm, mmm
Trigonal	3, $\bar{3}$ , 32, 3m, $\bar{3}m$
Hexagonal	6, $\bar{6}$ , 6/m, 622, 6mm, $\bar{6}m2$ , 6/mmm
Monoclinic	2, m, 2/m
Triclinic	1, $\bar{1}$

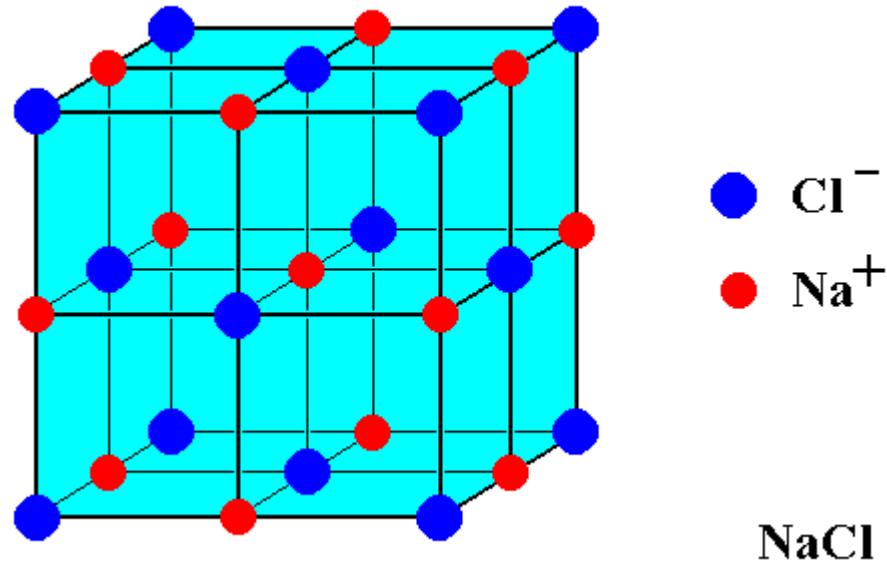
# NUMBER of FORMULA UNITS in UNIT CELL (Z)

- typically 1 – 6, but can be tens or even hundreds
- atom inside unit cell: belongs only to one unit cell
- atom on unit cell face: belongs to two unit cells
- atom on unit cell edge: belongs to four unit cells
- atom on unit cell corner: belongs to eight unit cells



# EXAMPLE

- How many NaCl formula units in unit cell?
- Answer: 4



# SPACE GROUPS

- From finite molecule (or macroscopic crystal) to infinite lattice → translation symmetry must be included
- Possible combinations of symmetry elements (including the translation symmetry elements): **230**  
→ **230 space groups**
- Space groups (and the characteristic information in 2 pages) are listed in **International Tables for Crystallography**  
→ **BIBLE of CRYSTALLOGRAPHY**
- Next-next slide: Space Group P4/mmm as an example

<https://it.iucr.org/>



## Triclinic

(For the enlarged unit cells, click [here](#))

1. [P1](#)      2. [P-1](#)

## Monoclinic

(For a fuller list with alternative unique axes, origins, or enlarged unit cells click [here](#))

3. [P121](#)    4. [P121](#)    5. [C121](#)    6. [P1m1](#)    7. [P1c1](#)  
8. [C1m1](#)    9. [C1c1](#)    10. [P12/m1](#)    11. [P121/m1](#)    12. [C12/m1](#)  
13. [P12/c1](#)    14. [P121/c1](#)    15. [C12/c1](#)

## Orthorhombic

(For a fuller list with alternative axes and origins click [here](#))

16. [P222](#)    17. [P222<sub>1</sub>](#)    18. [P2<sub>1</sub>2<sub>1</sub>2](#)    19. [P2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>](#)    20. [C222<sub>1</sub>](#)  
21. [C222](#)    22. [F222](#)    23. [I222](#)    24. [I2<sub>1</sub>2<sub>1</sub>2<sub>1</sub>](#)    25. [Pmm2](#)  
26. [Pmc2<sub>1</sub>](#)    27. [Pcc2](#)    28. [Pma2](#)    29. [Pca2<sub>1</sub>](#)    30. [Pnc2](#)  
31. [Pmn2<sub>1</sub>](#)    32. [Pba2](#)    33. [Pna2<sub>1</sub>](#)    34. [Pnn2](#)    35. [Cmm2](#)  
36. [Cmc2<sub>1</sub>](#)    37. [Ccc2](#)    38. [Amn2](#)    39. [Abm2](#)    40. [Ama2](#)  
41. [Aba2](#)    42. [Fmm2](#)    43. [Fdd2](#)    44. [Imn2](#)    45. [Iba2](#)  
46. [Ima2](#)    47. [Pmmm](#)    48. [Pnnn](#)    49. [Pccm](#)    50. [Pban](#)  
51. [Pmma](#)    52. [Pnna](#)    53. [Pmna](#)    54. [Pcca](#)    55. [Pbam](#)  
56. [Pccn](#)    57. [Pbcn](#)    58. [Pnnm](#)    59. [Pmnn](#)    60. [Pbcn](#)  
61. [Pbca](#)    62. [Pnma](#)    63. [Cmcm](#)    64. [Cmca](#)    65. [Cmmm](#)  
66. [Cccm](#)    67. [Cnma](#)    68. [Ccca](#)    69. [Fmmm](#)    70. [Fddd](#)  
71. [Immm](#)    72. [Ibam](#)    73. [Ibca](#)    74. [Imma](#)

## Tetragonal

(For the enlarged C- and F-centred unit cells, click [here](#))

75. [P4](#)      76. [P4<sub>1</sub>](#)      77. [P4<sub>2</sub>](#)      78. [P4<sub>3</sub>](#)      79. [I4](#)  
80. [I4<sub>1</sub>](#)      81. [P-4](#)      82. [I-4](#)      83. [P4/m](#)      84. [P4<sub>2</sub>/m](#)  
85. [P4/n](#)      86. [P4<sub>1</sub>/n](#)      87. [I4/m](#)      88. [I4<sub>1</sub>/a](#)      89. [P422](#)      168. [P6](#)      169. [P6<sub>1</sub>](#)      170. [P6<sub>2</sub>](#)      171. [P6<sub>3</sub>](#)      172. [P6<sub>4</sub>](#)  
90. [P42<sub>1</sub>2](#)    91. [P4<sub>1</sub>2<sub>1</sub>2](#)    92. [P4<sub>2</sub>2<sub>1</sub>2](#)    93. [P4<sub>2</sub>2<sub>2</sub>](#)    94. [P4<sub>2</sub>2<sub>1</sub>2](#)    173. [P6<sub>3</sub>](#)      174. [P-6](#)      175. [P6/m](#)      176. [P6<sub>3</sub>/m](#)      177. [P622](#)  
95. [P4<sub>1</sub>2<sub>2</sub>](#)    96. [P4<sub>2</sub>2<sub>1</sub>2](#)    97. [I422](#)      98. [I4<sub>1</sub>2<sub>2</sub>](#)    99. [P4mm](#)      178. [P6<sub>1</sub>2<sub>2</sub>](#)    179. [P6<sub>2</sub>2<sub>2</sub>](#)    180. [P6<sub>2</sub>2<sub>2</sub>](#)    181. [P6<sub>4</sub>2<sub>2</sub>](#)    182. [P6<sub>3</sub>2<sub>2</sub>](#)  
100. [P4bm](#)    101. [P4<sub>2</sub>cm](#)    102. [P4<sub>3</sub>nm](#)    103. [P4cc](#)      104. [P4nc](#)      183. [P6<sub>3</sub>mm](#)    184. [P6cc](#)      185. [P6<sub>3</sub>cm](#)    186. [P6<sub>3</sub>mc](#)    187. [P-6m2](#)  
105. [P4<sub>2</sub>mc](#)    106. [P4<sub>2</sub>bc](#)    107. [I4mm](#)      108. [I4cm](#)      109. [I4md](#)      188. [P-6c2](#)      189. [P-62m](#)      190. [P-62c](#)      191. [P6/mmm](#)    192. [P6/mcc](#)  
110. [I4cd](#)      111. [P-42m](#)    112. [P-42c](#)    113. [P-4<sub>2</sub>1m](#)    114. [P-4<sub>2</sub>1c](#)  
115. [P-4m2](#)    116. [P-4c2](#)    117. [P-4b2](#)    118. [P-4n2](#)    119. [I-4m2](#)  
120. [I-4c2](#)    121. [I-42m](#)    122. [I-42d](#)    123. [P4/mmm](#)    124. [P4/mcc](#)  
125. [P4/nbm](#)    126. [P4/nnc](#)    127. [P4/mbm](#)    128. [P4/mnc](#)    129. [P4/nmm](#)  
130. [P4/ncc](#)    131. [P4<sub>2</sub>/mmc](#)    132. [P4<sub>2</sub>/mcm](#)    133. [P4<sub>2</sub>/nbc](#)    134. [P4<sub>2</sub>/nnm](#)  
135. [P4<sub>2</sub>/mbc](#)    136. [P4<sub>2</sub>/mnm](#)    137. [P4<sub>2</sub>/nmc](#)    138. [P4<sub>2</sub>/ncc](#)    139. [I4/mmm](#)  
140. [I4/mcm](#)    141. [I4<sub>1</sub>/amd](#)    142. [I4<sub>1</sub>/acd](#)

## Trigonal

(For the R-centred cells with hexagonal axes and the larger H-centred trigonal cells, click [here](#))

143. [P3](#)      144. [P3<sub>1</sub>](#)      145. [P3<sub>2</sub>](#)      146. [R3](#)      147. [P-3](#)  
148. [R-3](#)    149. [P312](#)    150. [P321](#)    151. [P3<sub>1</sub>12](#)    152. [P3<sub>2</sub>12](#)  
153. [P3<sub>1</sub>12](#)    154. [P3<sub>2</sub>12](#)    155. [R32](#)    156. [P3m1](#)    157. [P31m](#)  
158. [P3c1](#)    159. [P31c](#)    160. [R3m](#)    161. [R3c](#)    162. [P-31m](#)  
163. [P-31c](#)    164. [P-3m1](#)    165. [P-3c1](#)    166. [R-3m](#)    167. [R-3c](#)

## Hexagonal

168. [P6](#)      169. [P6<sub>1</sub>](#)      170. [P6<sub>2</sub>](#)      171. [P6<sub>3</sub>](#)      172. [P6<sub>4</sub>](#)  
173. [P6<sub>3</sub>](#)      174. [P-6](#)      175. [P6/m](#)      176. [P6<sub>3</sub>/m](#)      177. [P622](#)  
178. [P6<sub>1</sub>2<sub>2</sub>](#)    179. [P6<sub>2</sub>2<sub>2</sub>](#)    180. [P6<sub>2</sub>2<sub>2</sub>](#)    181. [P6<sub>4</sub>2<sub>2</sub>](#)    182. [P6<sub>3</sub>2<sub>2</sub>](#)  
183. [P6<sub>3</sub>mm](#)    184. [P6cc](#)      185. [P6<sub>3</sub>cm](#)    186. [P6<sub>3</sub>mc](#)    187. [P-6m2](#)  
188. [P-6c2](#)      189. [P-62m](#)      190. [P-62c](#)      191. [P6/mmm](#)    192. [P6/mcc](#)  
193. [P6<sub>3</sub>/mcm](#)    194. [P6<sub>3</sub>/mmc](#)

## Cubic

195. [P23](#)      196. [F23](#)      197. [I23](#)      198. [P2<sub>1</sub>3](#)      199. [I2<sub>1</sub>3](#)  
200. [Pm-3](#)    201. [Pn-3](#)    202. [Fm-3](#)    203. [Fd-3](#)    204. [Im-3](#)  
205. [Pa-3](#)    206. [Ia-3](#)    207. [P432](#)    208. [P4<sub>3</sub>2](#)    209. [F432](#)  
210. [F4<sub>3</sub>2](#)    211. [I432](#)    212. [P4<sub>3</sub>2](#)    213. [P4<sub>1</sub>2](#)    214. [I4<sub>1</sub>2](#)  
215. [P-43m](#)    216. [F-43m](#)    217. [I-43m](#)    218. [P-43n](#)    219. [F-43c](#)  
220. [I-43d](#)    221. [Pm-3m](#)    222. [Pn-3n](#)    223. [Fm-3n](#)    224. [Pn-3m](#)  
225. [Fm-3m](#)    226. [Fm-3c](#)    227. [Fd-3m](#)    228. [Fd-3c](#)    229. [Im-3m](#)  
230. [Ia-3d](#)

From: <http://img.chem.ucl.ac.uk/sgp/large/sgp.htm>

All space group diagrams and tables online:  
<http://img.chem.ucl.ac.uk/sgp/large/sgp.htm>

$P4/mmm$

$D_{4h}^1$

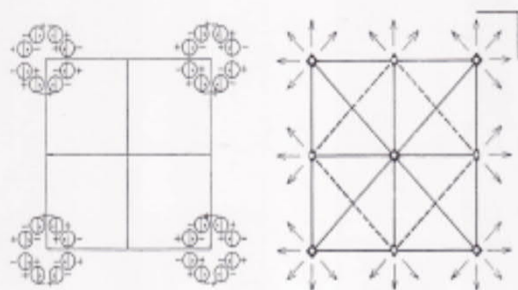
$4/mmm$

Tetragonal

No. 123

$P4/m2/m2/m$

Patterson symmetry  $P4/mmm$



Origin at centre ( $4/mmm$ )

Asymmetric unit  $0 \leq x \leq \frac{1}{2}; 0 \leq y \leq \frac{1}{2}; 0 \leq z \leq \frac{1}{2}; x \leq y$

Symmetry operations

- |                |                |                                    |                                    |
|----------------|----------------|------------------------------------|------------------------------------|
| (1) 1          | (2) 2 $0,0,z$  | (3) 4 <sup>+</sup> $0,0,z$         | (4) 4 <sup>-</sup> $0,0,z$         |
| (5) 2 $0,y,0$  | (6) 2 $x,0,0$  | (7) 2 $x,x,0$                      | (8) 2 $x,x,0$                      |
| (9) 1 $0,0,0$  | (10) m $x,y,0$ | (11) 4 <sup>+</sup> $0,0,z; 0,0,0$ | (12) 4 <sup>-</sup> $0,0,z; 0,0,0$ |
| (13) m $x,0,z$ | (14) m $0,y,z$ | (15) m $x,x,z$                     | (16) m $x,x,z$                     |

Maximal non-isomorphic subgroups

- I [2] $P422$  1; 2; 3; 4; 5; 6; 7; 8  
 [2] $P4/m11(P4/m)$  1; 2; 3; 4; 9; 10; 11; 12  
 [2] $P4mm$  1; 2; 3; 4; 13; 14; 15; 16  
 [2] $P42m$  1; 2; 5; 6; 11; 12; 15; 16  
 [2] $P4m2$  1; 2; 7; 8; 11; 12; 13; 14  
 [2] $P2/m2/m1(Pmmm)$  1; 2; 5; 6; 9; 10; 13; 14  
 [2] $P2/m12/m(Cmmm)$  1; 2; 7; 8; 9; 10; 15; 16
- IIa none
- IIb [2] $P4/mcc(c'=2c)$ ; [2] $P4_2/mcc(c'=2c)$ ; [2] $P4_2/mcm(c'=2c)$ ; [2] $C4/amd(a'=2a, b'=2b)(P4/bbw)$ ;  
 [2] $C4/mmd(a'=2a, b'=2b)(P4/bbm)$ ; [2] $C4/amw(a'=2a, b'=2b)(P4/mmw)$ ;  
 [2] $F4/mwm(a'=2a, b'=2b, c'=2c)(I4/mmm)$ ; [2] $F4/mmc(a'=2a, b'=2b, c'=2c)(I4/mcm)$

Maximal isomorphic subgroups of lowest index

- IIc [2] $P4/mmm(c'=2c)$ ; [2] $C4/mww(a'=2a, b'=2b)(P4/mmw)$

Minimal non-isomorphic supergroups

- I [3] $Pm3m$   
 II [2] $I4/mmm$

CONTINUED

No. 123

$P4/mmm$

Generators selected (1);  $r(1,0,0)$ ;  $r(0,1,0)$ ;  $r(0,0,1)$ ; (2); (3); (5); (9)

Positions

Multiplicity,  
Wyckoff letter,  
Site symmetry

Coordinates

Reflection conditions

16	a	1	(1) $x,y,z$	(2) $\bar{x},\bar{y},z$	(3) $\bar{y},x,z$	(4) $y,\bar{x},z$
			(5) $x,y,\bar{z}$	(6) $x,\bar{y},\bar{z}$	(7) $y,x,\bar{z}$	(8) $\bar{y},\bar{x},\bar{z}$
			(9) $\bar{x},\bar{y},z$	(10) $x,y,\bar{z}$	(11) $y,\bar{x},z$	(12) $\bar{y},x,\bar{z}$
			(13) $x,\bar{y},z$	(14) $\bar{x},y,z$	(15) $\bar{y},\bar{x},z$	(16) $y,x,z$

General:

no conditions

Special:

8	i	.m	$x,\bar{x},z$	$\bar{x},\bar{x},z$	$\bar{y},x,z$	$\bar{y},\bar{x},z$
			$\bar{x},\bar{x},\bar{z}$	$x,\bar{x},\bar{z}$	$\bar{y},x,\bar{z}$	$\bar{y},\bar{x},\bar{z}$

no extra conditions

8	s	.m	$x,0,z$	$\bar{x},0,z$	$0,x,z$	$0,\bar{x},z$
			$\bar{x},0,\bar{z}$	$x,0,\bar{z}$	$0,x,\bar{z}$	$0,\bar{x},\bar{z}$

no extra conditions

8	r	.m	$x,x,z$	$\bar{x},\bar{x},z$	$\bar{x},x,z$	$x,\bar{x},z$
			$\bar{x},x,\bar{z}$	$x,\bar{x},\bar{z}$	$x,x,\bar{z}$	$\bar{x},\bar{x},\bar{z}$

no extra conditions

8	q	m..	$x,y,\bar{z}$	$\bar{x},\bar{y},\bar{z}$	$\bar{y},x,\bar{z}$	$y,\bar{x},\bar{z}$
			$\bar{x},y,\bar{z}$	$x,\bar{y},\bar{z}$	$y,x,\bar{z}$	$\bar{y},\bar{x},\bar{z}$

no extra conditions

8	p	m..	$x,y,0$	$\bar{x},\bar{y},0$	$\bar{y},x,0$	$y,\bar{x},0$
			$\bar{x},y,0$	$x,\bar{y},0$	$y,x,0$	$\bar{y},\bar{x},0$

no extra conditions

4	o	m2m	$x,\bar{x},z$	$\bar{x},\bar{x},z$	$\bar{y},x,z$	$\bar{y},\bar{x},z$
---	---	-----	---------------	---------------------	---------------	---------------------

no extra conditions

4	n	m2m	$x,\bar{x},0$	$\bar{x},\bar{x},0$	$\bar{y},x,0$	$\bar{y},\bar{x},0$
---	---	-----	---------------	---------------------	---------------	---------------------

no extra conditions

4	m	m2m	$x,0,z$	$\bar{x},0,z$	$0,x,z$	$0,\bar{x},z$
---	---	-----	---------	---------------	---------	---------------

no extra conditions

4	l	m2m	$x,0,0$	$\bar{x},0,0$	$0,x,0$	$0,\bar{x},0$
---	---	-----	---------	---------------	---------	---------------

no extra conditions

4	k	m.2m	$x,x,z$	$\bar{x},\bar{x},z$	$\bar{x},x,z$	$x,\bar{x},z$
---	---	------	---------	---------------------	---------------	---------------

no extra conditions

4	j	m.2m	$x,x,0$	$\bar{x},\bar{x},0$	$\bar{x},x,0$	$x,\bar{x},0$
---	---	------	---------	---------------------	---------------	---------------

no extra conditions

4	i	2mw	$0,\bar{x},z$	$\bar{0},x,z$	$0,\bar{x},\bar{z}$	$\bar{0},x,\bar{z}$
---	---	-----	---------------	---------------	---------------------	---------------------

$hk\bar{l} : h+k=2n$

2	h	4mw	$\bar{x},\bar{x},z$	$\bar{x},\bar{x},\bar{z}$
---	---	-----	---------------------	---------------------------

no extra conditions

2	g	4mw	$0,0,z$	$0,0,\bar{z}$
---	---	-----	---------	---------------

no extra conditions

2	f	mmm	$0,\bar{x},0$	$\bar{0},x,0$
---	---	-----	---------------	---------------

$hk\bar{l} : h+k=2e$

2	e	mmm	$0,\bar{x},\bar{z}$	$\bar{0},x,\bar{z}$
---	---	-----	---------------------	---------------------

$hk\bar{l} : h+k=2e$

1	d	4/mmm	$\bar{x},\bar{x},\bar{z}$
---	---	-------	---------------------------

no extra conditions

1	c	4/mww	$\bar{x},\bar{x},0$
---	---	-------	---------------------

no extra conditions

1	b	4/mmm	$0,0,\bar{z}$
---	---	-------	---------------

no extra conditions

1	a	4/mmm	$0,0,0$
---	---	-------	---------

no extra conditions

Symmetry of special projections

Along [001]  $p4mm$

$a'=a$   $b'=b$

Origin at  $0,0,z$

Along [100]  $p2mm$

$a'=b$   $b'=c$

Origin at  $x,0,0$

Along [110]  $p2mm$

$a'=b$   $b'=c$

Origin at  $x,x,0$

(Continued on preceding page)

## EXAMPLES of INFORMATION

- Space group number: 123
- Name: P4/mmm
- Complete name:  $P 4/m 2/m 2/m$  ; showing the symmetry elements (4-fold rotation axis, mirror planes)
- Crystal family: tetragonal
- Lattice type: P (primitive)
- Site symmetry of the highest-symmetry site:  $D_{4h}$
- Asymmetric unit: smallest closed part of space the entire space is filled by applying all symmetry operations



## On the second page:

- List of possible sites for the atoms
- These are indicated/named by: multiplicity, Wyckoff letter & site symmetry
- Not all sites are actually occupied by an atom
- On the top: general site (16u)
- At the bottom: the highest symmetry site (1a)
- Multiplicity: number of identical sites

Generators selected (1);  $t(1,0,0)$ ;  $t(0,1,0)$ ;  $t(0,0,1)$ ; (2); (3); (5); (9)

Positions		Coordinates				Reflection conditions
Multiplicity	Wyckoff letter					General:
						no conditions
						Special:
						no extra conditions
16	u 1	(1) $x, y, z$	(2) $\bar{x}, \bar{y}, z$	(3) $\bar{y}, x, z$	(4) $y, \bar{x}, z$	
		(5) $\bar{x}, y, \bar{z}$	(6) $x, \bar{y}, \bar{z}$	(7) $y, x, \bar{z}$	(8) $\bar{y}, \bar{x}, \bar{z}$	
		(9) $\bar{x}, \bar{y}, \bar{z}$	(10) $x, y, z$	(11) $y, \bar{x}, \bar{z}$	(12) $\bar{y}, x, \bar{z}$	
		(13) $x, \bar{y}, z$	(14) $\bar{x}, y, z$	(15) $\bar{y}, \bar{x}, z$	(16) $y, x, z$	
8	t .m .	$x, \frac{1}{2}, z$	$\bar{x}, \frac{1}{2}, z$	$\frac{1}{2}, x, z$	$\frac{1}{2}, \bar{x}, z$	
		$\bar{x}, \frac{1}{2}, \bar{z}$	$x, \frac{1}{2}, \bar{z}$	$\frac{1}{2}, x, \bar{z}$	$\frac{1}{2}, \bar{x}, \bar{z}$	
8	s .m .	$x, 0, z$	$\bar{x}, 0, z$	$0, x, z$	$0, \bar{x}, z$	
		$\bar{x}, 0, \bar{z}$	$x, 0, \bar{z}$	$0, x, \bar{z}$	$0, \bar{x}, \bar{z}$	
8	r .m	$x, x, z$	$\bar{x}, \bar{x}, z$	$\bar{x}, x, z$	$x, \bar{x}, z$	
		$\bar{x}, x, \bar{z}$	$x, \bar{x}, \bar{z}$	$x, x, \bar{z}$	$\bar{x}, \bar{x}, \bar{z}$	
8	q m . .	$x, y, \frac{1}{2}$	$\bar{x}, \bar{y}, \frac{1}{2}$	$\bar{y}, x, \frac{1}{2}$	$y, \bar{x}, \frac{1}{2}$	
		$\bar{x}, y, \frac{1}{2}$	$x, \bar{y}, \frac{1}{2}$	$y, x, \frac{1}{2}$	$\bar{y}, \bar{x}, \frac{1}{2}$	
8	p m . .	$x, y, 0$	$\bar{x}, \bar{y}, 0$	$\bar{y}, x, 0$	$y, \bar{x}, 0$	
		$\bar{x}, y, 0$	$x, \bar{y}, 0$	$y, x, 0$	$\bar{y}, \bar{x}, 0$	
4	o m 2m .	$x, \frac{1}{2}, \frac{1}{2}$	$\bar{x}, \frac{1}{2}, \frac{1}{2}$	$\frac{1}{2}, x, \frac{1}{2}$	$\frac{1}{2}, \bar{x}, \frac{1}{2}$	
4	n m 2m .	$x, \frac{1}{2}, 0$	$\bar{x}, \frac{1}{2}, 0$	$\frac{1}{2}, x, 0$	$\frac{1}{2}, \bar{x}, 0$	
4	m m 2m .	$x, 0, \frac{1}{2}$	$\bar{x}, 0, \frac{1}{2}$	$0, x, \frac{1}{2}$	$0, \bar{x}, \frac{1}{2}$	
4	l m 2m .	$x, 0, 0$	$\bar{x}, 0, 0$	$0, x, 0$	$0, \bar{x}, 0$	
4	k m . 2m	$x, x, \frac{1}{2}$	$\bar{x}, \bar{x}, \frac{1}{2}$	$\bar{x}, x, \frac{1}{2}$	$x, \bar{x}, \frac{1}{2}$	
4	j m . 2m	$x, x, 0$	$\bar{x}, \bar{x}, 0$	$\bar{x}, x, 0$	$x, \bar{x}, 0$	
4	i 2m m .	$0, \frac{1}{2}, z$	$\frac{1}{2}, 0, z$	$0, \frac{1}{2}, \bar{z}$	$\frac{1}{2}, 0, \bar{z}$	$hkl : h+k=2n$
2	h 4m m	$\frac{1}{2}, \frac{1}{2}, z$	$\frac{1}{2}, \frac{1}{2}, \bar{z}$			no extra conditions
2	g 4m m	$0, 0, z$	$0, 0, \bar{z}$			no extra conditions
2	f m m m .	$0, \frac{1}{2}, 0$	$\frac{1}{2}, 0, 0$			$hkl : h+k=2n$
2	e m m m .	$0, \frac{1}{2}, \frac{1}{2}$	$\frac{1}{2}, 0, \frac{1}{2}$			$hkl : h+k=2n$
1	d 4/m m m	$\frac{1}{2}, \frac{1}{2}, \frac{1}{2}$				no extra conditions
1	c 4/m m m	$\frac{1}{2}, \frac{1}{2}, 0$				no extra conditions
1	b 4/m m m	$0, 0, \frac{1}{2}$				no extra conditions
1	a 4/m m m	$0, 0, 0$				no extra conditions

### Symmetry of special projections

Along [001]  $p 4m m$   
 $a' = a$   $b' = b$   
 Origin at  $0, 0, z$

Along [100]  $p 2m m$   
 $a' = b$   $b' = c$   
 Origin at  $x, 0, 0$

Along [110]  $p 2m m$   
 $a' = \frac{1}{2}(-a+b)$   $b' = c$   
 Origin at  $x, x, 0$

(Continued on preceding page)

## EXAMPLE: Potassium tetrachloroplatinate(II): $K_2PtCl_4$

Space group:  $P4/mmm$  (No. 123)

Lattice parameters:  $a = b = 7.023\text{\AA}$ ,  $c = 4.1486\text{\AA}$

Atomic positions:	Pt	1a:	0,0,0
	K	2e:	0,1/2,1/2
	Cl	4j:	x,x,0 ; x = 0.23247

- Draw the unit cell with the atoms.
- Draw the projection of the unit cell in  $c$ -axis direction.
- Theoretical density is  $3.37\text{ g/cm}^3$ . Calculate  $Z$ ?  
( $N_A = 6.022 \times 10^{23}$ ; atomic weights: K 39.098; Pt 195.22; Cl 35.453)
- Calculate the distances: Pt-Pt, Pt-K, Pt-Cl.
- What is the coordination number of platinum ?
- What is the site symmetry of platinum ?

$P4/mmm$

$D_{4h}^1$

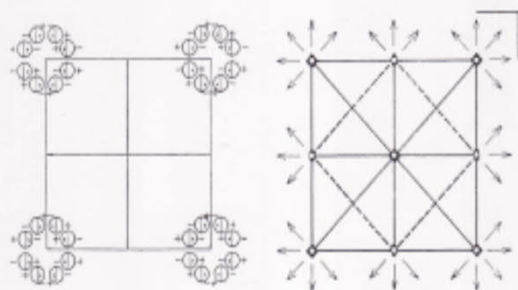
$4/mmm$

Tetragonal

No. 123

$P4/m\ 2/m\ 2/m$

Patterson symmetry  $P4/mmm$



Origin at centre ( $4/mmm$ )

Asymmetric unit  $0 \leq x \leq \frac{1}{2}; 0 \leq y \leq \frac{1}{2}; 0 \leq z \leq \frac{1}{2}; x \leq y$

Symmetry operations

- |                |                |                                    |                                    |
|----------------|----------------|------------------------------------|------------------------------------|
| (1) 1          | (2) 2 $0,0,z$  | (3) 4 <sup>+</sup> $0,0,z$         | (4) 4 <sup>-</sup> $0,0,z$         |
| (5) 2 $0,y,0$  | (6) 2 $x,0,0$  | (7) 2 $x,x,0$                      | (8) 2 $x,x,0$                      |
| (9) 1 $0,0,0$  | (10) m $x,y,0$ | (11) 4 <sup>+</sup> $0,0,z; 0,0,0$ | (12) 4 <sup>-</sup> $0,0,z; 0,0,0$ |
| (13) m $x,0,z$ | (14) m $0,y,z$ | (15) m $x,x,z$                     | (16) m $x,x,z$                     |

Maximal non-isomorphic subgroups

- I [2] $P4\ 2\ 2$  1; 2; 3; 4; 5; 6; 7; 8  
 [2] $P4/m\ 1\ 1$  ( $P4/m$ ) 1; 2; 3; 4; 9; 10; 11; 12  
 [2] $P4mm$  1; 2; 3; 4; 13; 14; 15; 16  
 [2] $P4\ 2m$  1; 2; 5; 6; 11; 12; 15; 16  
 [2] $P4m\ 2$  1; 2; 7; 8; 11; 12; 13; 14  
 [2] $P2/m\ 2/m\ 1$  ( $Pmmm$ ) 1; 2; 5; 6; 9; 10; 13; 14  
 [2] $P2/m\ 1\ 2/m$  ( $Cmmm$ ) 1; 2; 7; 8; 9; 10; 15; 16
- IIa none
- IIb [2] $P4/m\ cc$  ( $c' = 2c$ ); [2] $P4_2/m\ mc$  ( $c' = 2c$ ); [2] $P4_2/m\ cm$  ( $c' = 2c$ ); [2] $C4/m\ ad$  ( $a' = 2a, b' = 2b$ ) ( $P4/nb\ w$ ); [2] $C4/m\ md$  ( $a' = 2a, b' = 2b$ ) ( $P4/m\ bm$ ); [2] $C4/am\ w$  ( $a' = 2a, b' = 2b$ ) ( $P4/nmm$ ); [2] $F4/m\ ww$  ( $a' = 2a, b' = 2b, c' = 2c$ ) ( $I4/m\ mm$ ); [2] $F4/m\ mc$  ( $a' = 2a, b' = 2b, c' = 2c$ ) ( $I4/m\ cm$ )

Maximal isomorphic subgroups of lowest index

- IIc [2] $P4/m\ mm$  ( $c' = 2c$ ); [2] $C4/m\ ww$  ( $a' = 2a, b' = 2b$ ) ( $P4/m\ mm$ )

Minimal non-isomorphic supergroups

- I [3] $Pm\ 3m$   
 II [2] $I4/m\ mm$

CONTINUED

No. 123

$P4/mmm$

Generators selected (1);  $r(1,0,0)$ ;  $r(0,1,0)$ ;  $r(0,0,1)$ ; (2); (3); (5); (9)

Positions

Multiplicity,  
Wyckoff letter,  
Site symmetry

Coordinates

Reflection conditions

- |    |   |   |                         |                         |                          |                               |
|----|---|---|-------------------------|-------------------------|--------------------------|-------------------------------|
| 16 | a | 1 | (1) $x,y,z$             | (2) $\bar{x},\bar{y},z$ | (3) $\bar{y},x,z$        | (4) $y,\bar{x},z$             |
|    |   |   | (5) $x,y,\bar{z}$       | (6) $x,\bar{y},\bar{z}$ | (7) $y,x,\bar{z}$        | (8) $\bar{y},\bar{x},\bar{z}$ |
|    |   |   | (9) $\bar{x},\bar{y},z$ | (10) $x,y,\bar{z}$      | (11) $y,\bar{x},z$       | (12) $\bar{y},x,\bar{z}$      |
|    |   |   | (13) $x,\bar{y},z$      | (14) $\bar{x},y,z$      | (15) $\bar{y},\bar{x},z$ | (16) $y,x,z$                  |

General:

no conditions

Special:

- |   |   |    |                           |                     |                           |                           |
|---|---|----|---------------------------|---------------------|---------------------------|---------------------------|
| 8 | i | .m | $x,\bar{x},z$             | $\bar{x},\bar{x},z$ | $\bar{y},\bar{y},z$       | $\bar{y},\bar{y},z$       |
|   |   |    | $\bar{x},\bar{x},\bar{z}$ | $x,\bar{x},\bar{z}$ | $\bar{y},\bar{y},\bar{z}$ | $\bar{y},\bar{y},\bar{z}$ |

no extra conditions

- |   |   |    |                     |               |               |                     |
|---|---|----|---------------------|---------------|---------------|---------------------|
| 8 | s | .m | $x,0,z$             | $\bar{x},0,z$ | $0,x,z$       | $0,\bar{x},z$       |
|   |   |    | $\bar{x},0,\bar{z}$ | $x,0,\bar{z}$ | $0,x,\bar{z}$ | $0,\bar{x},\bar{z}$ |

no extra conditions

- |   |   |    |                     |                     |               |                           |
|---|---|----|---------------------|---------------------|---------------|---------------------------|
| 8 | r | .m | $x,x,z$             | $\bar{x},\bar{x},z$ | $\bar{x},x,z$ | $x,\bar{x},z$             |
|   |   |    | $\bar{x},x,\bar{z}$ | $x,\bar{x},\bar{z}$ | $x,x,\bar{z}$ | $\bar{x},\bar{x},\bar{z}$ |

no extra conditions

- |   |   |     |                     |                           |                     |                     |
|---|---|-----|---------------------|---------------------------|---------------------|---------------------|
| 8 | q | m.. | $x,y,\bar{z}$       | $\bar{x},\bar{y},\bar{z}$ | $\bar{y},x,\bar{z}$ | $y,\bar{x},\bar{z}$ |
|   |   |     | $\bar{x},\bar{y},z$ | $x,\bar{y},z$             | $y,x,z$             | $\bar{y},\bar{x},z$ |

no extra conditions

- |   |   |     |               |                     |               |                     |
|---|---|-----|---------------|---------------------|---------------|---------------------|
| 8 | p | m.. | $x,y,0$       | $\bar{x},\bar{y},0$ | $\bar{y},x,0$ | $y,\bar{x},0$       |
|   |   |     | $\bar{x},y,0$ | $x,\bar{y},0$       | $y,x,0$       | $\bar{y},\bar{x},0$ |

no extra conditions

- |   |   |     |               |                     |               |                     |
|---|---|-----|---------------|---------------------|---------------|---------------------|
| 4 | o | m2m | $x,\bar{x},z$ | $\bar{x},\bar{x},z$ | $\bar{y},x,z$ | $\bar{y},\bar{x},z$ |
|---|---|-----|---------------|---------------------|---------------|---------------------|

no extra conditions

- |   |   |     |               |                     |               |                     |
|---|---|-----|---------------|---------------------|---------------|---------------------|
| 4 | n | m2m | $x,\bar{x},0$ | $\bar{x},\bar{x},0$ | $\bar{y},x,0$ | $\bar{y},\bar{x},0$ |
|---|---|-----|---------------|---------------------|---------------|---------------------|

no extra conditions

- |   |   |     |               |                     |               |                     |
|---|---|-----|---------------|---------------------|---------------|---------------------|
| 4 | m | m2m | $x,0,\bar{z}$ | $\bar{x},0,\bar{z}$ | $0,x,\bar{z}$ | $0,\bar{x},\bar{z}$ |
|---|---|-----|---------------|---------------------|---------------|---------------------|

no extra conditions

- |   |   |     |         |               |         |               |
|---|---|-----|---------|---------------|---------|---------------|
| 4 | l | m2m | $x,0,0$ | $\bar{x},0,0$ | $0,x,0$ | $0,\bar{x},0$ |
|---|---|-----|---------|---------------|---------|---------------|

no extra conditions

- |   |   |      |               |                           |                     |                     |
|---|---|------|---------------|---------------------------|---------------------|---------------------|
| 4 | k | m.2m | $x,x,\bar{z}$ | $\bar{x},\bar{x},\bar{z}$ | $\bar{x},x,\bar{z}$ | $x,\bar{x},\bar{z}$ |
|---|---|------|---------------|---------------------------|---------------------|---------------------|

no extra conditions

- |   |   |      |         |                     |               |               |
|---|---|------|---------|---------------------|---------------|---------------|
| 4 | j | m.2m | $x,x,0$ | $\bar{x},\bar{x},0$ | $\bar{x},x,0$ | $x,\bar{x},0$ |
|---|---|------|---------|---------------------|---------------|---------------|

no extra conditions

- |   |   |     |               |             |                     |                   |
|---|---|-----|---------------|-------------|---------------------|-------------------|
| 4 | i | 2mw | $0,\bar{z},z$ | $\bar{z},z$ | $0,\bar{z},\bar{z}$ | $\bar{z},\bar{z}$ |
|---|---|-----|---------------|-------------|---------------------|-------------------|

$hk\bar{l} : h+k=2n$

- |   |   |     |                     |                           |
|---|---|-----|---------------------|---------------------------|
| 2 | h | 4mw | $\bar{z},\bar{z},z$ | $\bar{z},\bar{z},\bar{z}$ |
|---|---|-----|---------------------|---------------------------|

no extra conditions

- |   |   |     |         |               |
|---|---|-----|---------|---------------|
| 2 | g | 4mw | $0,0,z$ | $0,0,\bar{z}$ |
|---|---|-----|---------|---------------|

no extra conditions

- |   |   |     |               |               |
|---|---|-----|---------------|---------------|
| 2 | f | mmm | $0,\bar{z},0$ | $\bar{z},0,0$ |
|---|---|-----|---------------|---------------|

$hk\bar{l} : h+k=2e$

- |   |   |     |                     |                     |
|---|---|-----|---------------------|---------------------|
| 2 | e | mmm | $0,\bar{z},\bar{z}$ | $\bar{z},0,\bar{z}$ |
|---|---|-----|---------------------|---------------------|

$hk\bar{l} : h+k=2e$

- |   |   |       |                           |
|---|---|-------|---------------------------|
| 1 | d | 4/mmm | $\bar{z},\bar{z},\bar{z}$ |
|---|---|-------|---------------------------|

no extra conditions

- |   |   |       |                     |
|---|---|-------|---------------------|
| 1 | c | 4/mmm | $\bar{z},\bar{z},0$ |
|---|---|-------|---------------------|

no extra conditions

- |   |   |       |               |
|---|---|-------|---------------|
| 1 | b | 4/mmm | $0,0,\bar{z}$ |
|---|---|-------|---------------|

no extra conditions

- |   |   |       |         |
|---|---|-------|---------|
| 1 | a | 4/mmm | $0,0,0$ |
|---|---|-------|---------|

no extra conditions

Symmetry of special projections

Along [001]  $p4mm$   
 $a' = a$   $b' = b$

Along [100]  $p2mm$   
 $a' = b$   $b' = c$

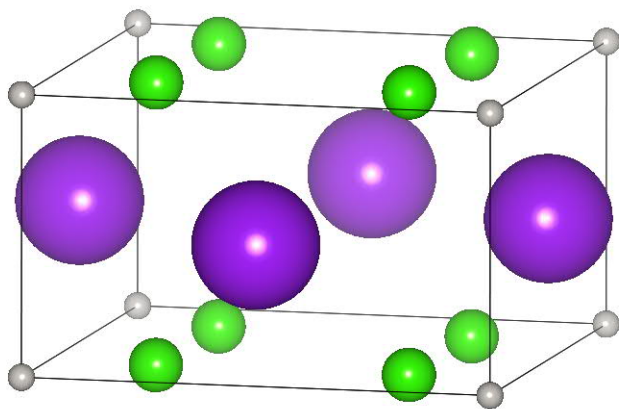
Along [110]  $p2mm$   
 $a' = \frac{1}{2}(-a+b)$   $b' = c$

Origin at  $0,0,z$

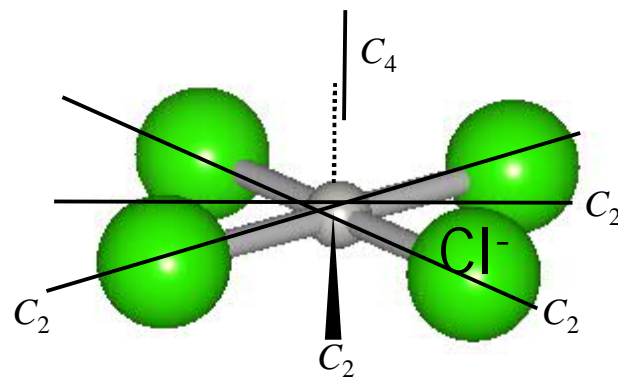
Origin at  $x,0,0$

Origin at  $x,x,0$

(Continued on preceding page)



**Pt: 1 atom in unit cell**  
**K: 2 atoms in unit cell**  
**Cl: 4 atoms in unit cell**



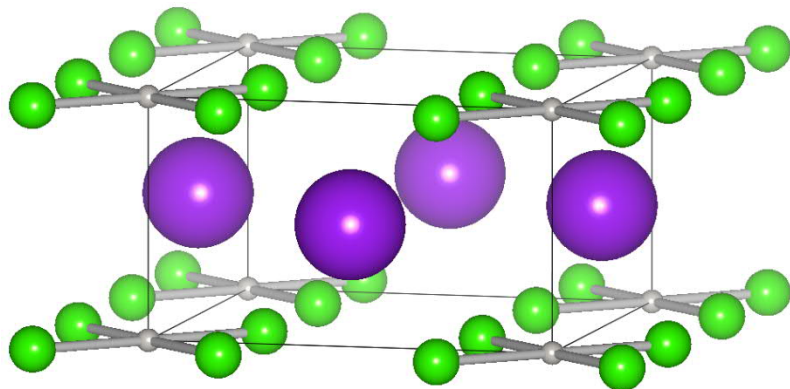
Site symmetry of Pt:  $D_{4h}$

**Bond lengths:**

**Pt-Pt:  $(1-0)^2 \cdot 4.15\text{\AA}$**

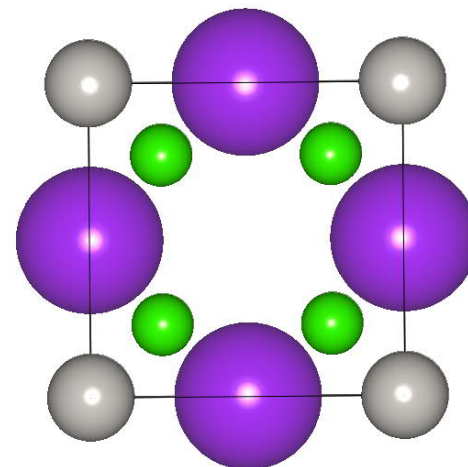
**Pt-K:  $\sqrt{[(0.5-0)^2 \cdot 7.023\text{\AA} + (0.5-0)^2 \cdot 4.149\text{\AA}]} = 4.08\text{\AA}$**

**Pt-Cl:  $\sqrt{[(0.232-0)^2 \cdot 7.023\text{\AA} + (0.232-0)^2 \cdot 7.023\text{\AA}]} = 2.30\text{\AA}$**



*ab*-projectio

(seen from *c*-direction)



## **K<sub>2</sub>PtCl<sub>4</sub>**

- $\rho = 3.37 \times 10^6 \text{ g/m}^3$
- $V = 7.023 \text{ \AA} \times 7.023 \text{ \AA} \times 4.1486 \text{ \AA} = 204.62 \times 10^{-30} \text{ m}^3$
- $M = (2 \times 39.098 + 195.22 + 4 \times 35.453) \text{ g/mol} = 415.228 \text{ g/mol}$
- $Z = (V \times \rho \times N_A) / M = 1$
- Distances:
  - Pt-Pt: 4.15 Å
  - Pt-K: 4.08 Å
  - Pt-Cl: 2.31 Å (→ chemical bond)
- CN(Pt) = 4
- Pl site symmetry:  $D_{4h}$

# EXAMPLE: Chromium oxychloride CrOCl

Space group *Pmmn* (No. 59)

Lattice parameters:  $a = 3.88 \text{ \AA}$ ,  $b = 3.20 \text{ \AA}$ ,  $c = 7.72 \text{ \AA}$  ( $Z = 2$ )

Atomic positions:	Cr	$2a$	$z = 0.109$
	Cl	$2b$	$z = 0.327$
	O	$2b$	$z = 0.960$

(a) Draw the unit cell.

(b) Give for chromium:

- bond lengths
- coordination numbers
- site symmetry

(c) Calculate BVS for chromium.

[ $R^0$  values: Cr<sup>III</sup>-O<sup>II</sup>: 1.724, Cr<sup>III</sup>-Cl<sup>I</sup>: 2.08]

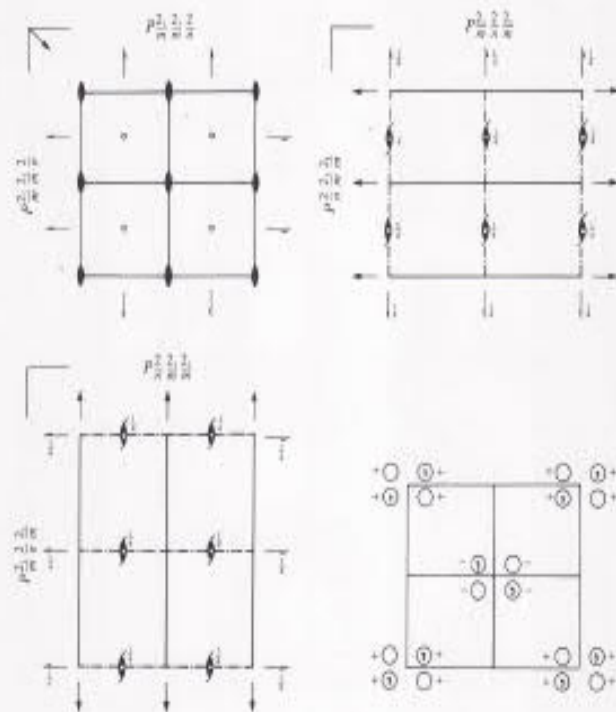
$Pmmn$  $D_{2h}^{13}$  $mmm$ 

Orthorhombic

No. 59

 $P2_1/m 2_1/m 2/n$ Patterson symmetry  $Pmmm$ 

ORIGIN CHOICE 1

Origin at  $mc/2$ , at  $\frac{1}{2}, \frac{1}{2}, 0$  from  $\bar{1}$ Asymmetric unit  $0 \leq x \leq \frac{1}{2}$ ;  $0 \leq y \leq \frac{1}{2}$ ;  $0 \leq z \leq \frac{1}{2}$ 

Symmetry operations

- |                       |  |  |  |
|-----------------------|--|--|--|
| (1) $\bar{1}$         | (2) $2(0,0,z)$                             | (3) $2(0,\frac{1}{2},0)$ $\frac{1}{2},y,0$ | (4) $2(\frac{1}{2},0,0)$ $x,\frac{1}{2},0$ |
| (5) $\bar{1}$ $1,1,0$ | (6) $m(\frac{1}{2},\frac{1}{2},0)$ $x,y,0$ | (7) $m(x,0,z)$                             | (8) $m(0,y,z)$                             |

CONTINUED

No. 59

 $Pmmn$ Generators selected (1);  $t(1,0,0)$ ;  $t(0,1,0)$ ;  $t(0,0,1)$ ; (2); (3); (5)

Positions

Multiplicity,  
Wyckoff letter,  
site symmetry

Coordinates

Reflection conditions

- |                 |   |                                     |                                     |   |
|-----------------|---|-------------------------------------|-------------------------------------|---|
| 8 $g$ $\bar{1}$ | (1) $x,y,z$                               | (2) $\bar{x},\bar{y},z$             | (3) $x+\frac{1}{2},y+\frac{1}{2},z$ | (4) $x+\frac{1}{2},\bar{y}+\frac{1}{2},z$ |
|                 | (5) $x+\frac{1}{2},y+\frac{1}{2},\bar{z}$ | (6) $x+\frac{1}{2},y+\frac{1}{2},z$ | (7) $x,\bar{y},z$                   | (8) $x,y,\bar{z}$                         |

General:

- $hk0: h+k=2n$   
 $h00: h=2n$   
 $0k0: k=2n$

Special: as above, plus

- |           |         |               |                         |                   |
|-----------|---------|---------------|-------------------------|-------------------|
| 4 $f$ $m$ | $x,0,z$ | $\bar{x},0,z$ | $x+\frac{1}{2},\bar{z}$ | $x+\frac{1}{2},z$ |
|-----------|---------|---------------|-------------------------|-------------------|

no extra conditions

- |           |         |               |                               |                                     |
|-----------|---------|---------------|-------------------------------|-------------------------------------|
| 4 $e$ $m$ | $0,y,z$ | $0,\bar{y},z$ | $\frac{1}{2},y+\frac{1}{2},z$ | $\frac{1}{2},\bar{y}+\frac{1}{2},z$ |
|-----------|---------|---------------|-------------------------------|-------------------------------------|

no extra conditions

- |                 |                             |                                   |                             |                                   |
|-----------------|-----------------------------|-----------------------------------|-----------------------------|-----------------------------------|
| 4 $d$ $\bar{1}$ | $\frac{1}{2},\frac{1}{2},z$ | $\frac{1}{2},\frac{1}{2},\bar{z}$ | $\frac{1}{2},\frac{1}{2},z$ | $\frac{1}{2},\frac{1}{2},\bar{z}$ |
|-----------------|-----------------------------|-----------------------------------|-----------------------------|-----------------------------------|

 $hkl: h,k=2n$ 

- |                 |                             |                             |                             |                             |
|-----------------|-----------------------------|-----------------------------|-----------------------------|-----------------------------|
| 4 $c$ $\bar{1}$ | $\frac{1}{2},\frac{1}{2},0$ | $\frac{1}{2},\frac{1}{2},0$ | $\frac{1}{2},\frac{1}{2},0$ | $\frac{1}{2},\frac{1}{2},0$ |
|-----------------|-----------------------------|-----------------------------|-----------------------------|-----------------------------|

 $hkl: h,k=2n$ 

- |             |                   |                 |
|-------------|-------------------|-----------------|
| 2 $b$ $mm2$ | $0,\frac{1}{2},z$ | $\frac{1}{2},z$ |
|-------------|-------------------|-----------------|

no extra conditions

- |             |         |                 |
|-------------|---------|-----------------|
| 2 $a$ $mm2$ | $0,0,z$ | $\frac{1}{2},z$ |
|-------------|---------|-----------------|

no extra conditions

Symmetry of special projections

Along [001]  $c2mm$  $a'=a$   $b'=b$ Origin at  $0,0,z$ Along [100]  $p2mg$  $a'=b$   $b'=c$ Origin at  $x,\frac{1}{2}$ Along [010]  $p2gm$  $a'=c$   $b'=a$ Origin at  $\frac{1}{2},y,0$ 

Maximal non-isomorphic subgroups

- |   |                        |         |
|---|------------------------|---------|
| I | [2] $P2_12_12_1$       | 1;2;3;4 |
|   | [2] $P112/n(P2/c)$     | 1;2;5;6 |
|   | [2] $P12_1/m1(P2_1/m)$ | 1;3;5;7 |
|   | [2] $P2_1/m11(P2_1/m)$ | 1;4;5;8 |
|   | [2] $Pmm2$             | 1;2;7;8 |
|   | [2] $Pm2_1n(Pmn2_1)$   | 1;3;6;8 |
|   | [2] $P2_1mn(Pmn2_1)$   | 1;4;6;7 |

IIa none

- |     |   |
|-----|---|
| IIb | [2] $Pcmn(c'=2c)(Pnma)$ ; [2] $Pmcn(c'=2c)(Pnma)$ ; [2] $Pccn(c'=2c)$ |
|-----|---|

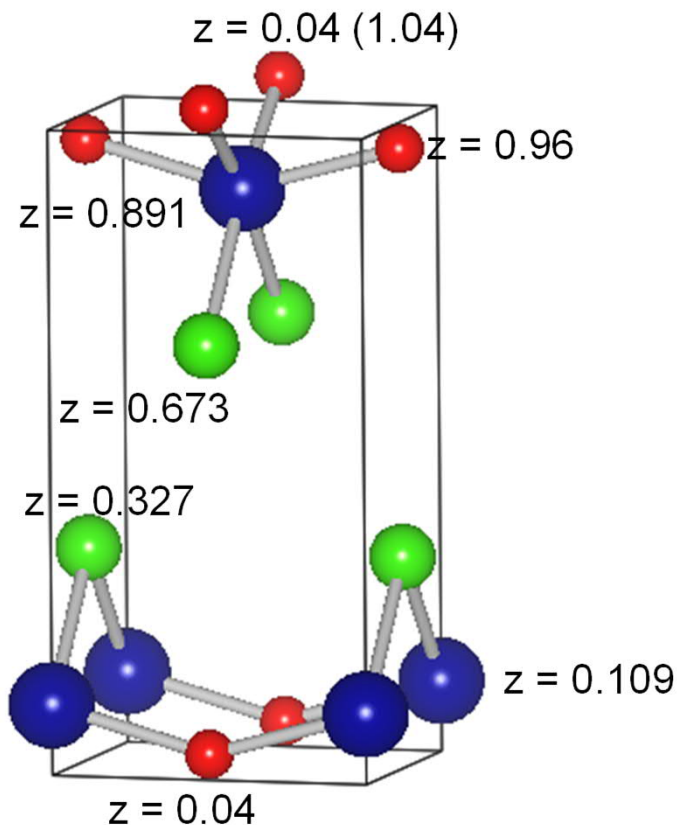
Maximal isomorphic subgroups of lowest index

- |     |   |
|-----|---|
| IIc | [3] $Pmmn(a'=3a \text{ or } b'=3b)$ ; [2] $Pmmn(c'=2c)$ |
|-----|---|

Minimal non-isomorphic supergroups

- |    |   |
|----|---|
| I  | [2] $P4/nmm$ ; [2] $P4/mnc$   |
| II | [2] $Ammm(Cmcm)$ ; [2] $Bmmb(Cmcm)$ ; [2] $Cmmm$ ; [2] $Immm$ ; [2] $Pmmb(2a'=a)(Pnma)$ ; [2] $Pnna(2b'=b)$ |

# CrOCl



## Chromium bonding

$$2 \times \text{Cr-Cl: } \sqrt{\{(0.891-0.673) \times 7.72 \text{ \AA}\}^2 + \{0.5 \times 3.20 \text{ \AA}\}^2} = 2.3222 \text{ \AA}$$

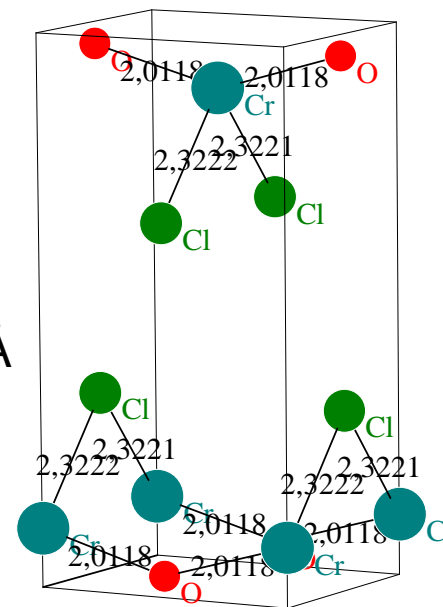
$$2 \times \text{Cr-O: } \sqrt{\{(0.960-0.891) \times 7.72 \text{ \AA}\}^2 + \{0.5 \times 3.88 \text{ \AA}\}^2} = 2.0118 \text{ \AA}$$

$$2 \times \text{Cr-O: } \sqrt{\{[(1-0.891)+0.04] \times 7.72 \text{ \AA}\}^2 + \{0.5 \times 3.20 \text{ \AA}\}^2} = 1.9706 \text{ \AA}$$

$$\text{CN}(\text{Cr}) = 6$$

Cr site symmetry:  $C_{2v}$

$$\text{BVS}(\text{Cr}): +2.985$$





# CrOCl: simulated XRD pattern based on the structure data

