

# SCHEDULE

	Date	Topic
1.	Tue 14.09.	Lec-1: Introduction
2.	Fri 17.09.	Lec-2: Crystal Chemistry & Tolerance parameter
3.	Fri 17.09.	EXERCISE 1
4.	Tue 21.09.	Lec-3: Crystal chemistry & BVS
5.	Fri 24.09.	Lec-4: Symmetry & Point Groups
6.	Fri 24.09.	EXERCISE 2
7.	Tue 28.10.	Lec-5: Crystallography & Space Groups
8.	Fri 01.10.	Lec-6: XRD & Reciprocal lattice
9.	Fri 01.10.	EXERCISE 3
10.	Tue 05.10.	Lec-7: ND & GI-XRD
11.	Fri 08.10.	Lec-8: Rietveld
12.	Fri 08.10	EXERCISE 4: Rietveld
13.	Tue 12.10.	Lec-9: Synchrotron rad. & XAS & RIXS
14.	Fri 15.10.	Lec-10: EXAFS & Mössbauer
15.	Fri 15.10.	EXERCISE 5
16.	Tue 19.10.	Seminars: XPS, FTIR, Raman, ED, HRTEM, SEM, AFM
17.	Fri 22.10.	Lec-11: XRR
18.	Fri 22.10.	EXERCISE 6: XRR

**EXAM: Friday, Oct. 29<sup>th</sup>, 2021**

# LECTURE 5: CRYSTALLOGRAPHY BASICS

- From "point-like" molecules to 3D crystals
- Translation in 3D crystals →  
**NEW SYMMETRY OPERATIONS (glide planes & screw axes)**
- From *Point groups* to *Space groups*
- Crystal lattice, lattice points & unit cell
- International Tables of Crystallography

# Symmetry elements (Schönflies / Hermann-Mauguin)

RECALL FROM PREVIOUS LECTURE:

Point/Molecular symmetry  One point remains unchanged

Identity  $E$

Symmetry/inversion center  $i / \bar{1}$

Rotation axis  $C_n / 1, 2, 3, \dots$

Reflection/mirror plane  $\sigma / m$

Improper rotation axis  $S_n / (\bar{1}, \bar{2}), \bar{3}, \bar{4}, \bar{6}$

NOTE:

- Improper rotation axis  $\bar{1}$  and inversion center  $\bar{1}$  are equivalent
- Mirror plane  $m$  and improper rotation axis  $\bar{2}$  are equivalent

# FROM MOLECULES TO **CRYSTALS**

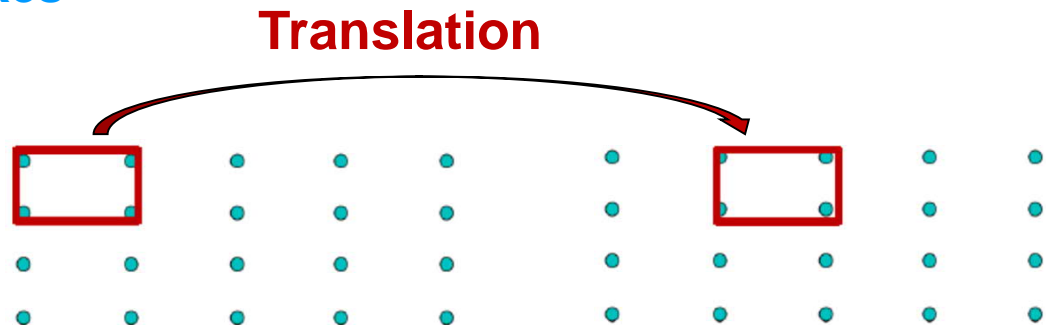
There are two things which make macroscopic (infinite) crystals different from discrete molecules in terms of symmetry: **Space-filling & Translation**

## **Space-filling**

- Macroscopic crystals need to continuously fill the space
- For molecules 5-fold rotation is possible, but not for crystals (except in quasicrystals)

## **Translation in crystals**

- Translation: move from one point to another (the entire object)
- This does not exist in molecules, but is the essence of macroscopic crystals exhibiting long-range order
- Combining translation with other symmetry operations/elements → new symmetry operations/elements (not included in point groups): **glide planes** & **screw axes**



# Additional translation symmetry elements in INFINITE LATTICES

## Translation

- Movement from one point to another point

## Screw axis

- Rotation plus translation

$n_m$  ( $2_1, 3_1, 3_2, 4_1, 4_2, 4_3, 6_1, 6_2, 6_3, 6_4, 6_5$ )

For example:  $2_1$ : rotation  $180^\circ$  and translation  $\frac{1}{2}(m/n)$

## Glide plane

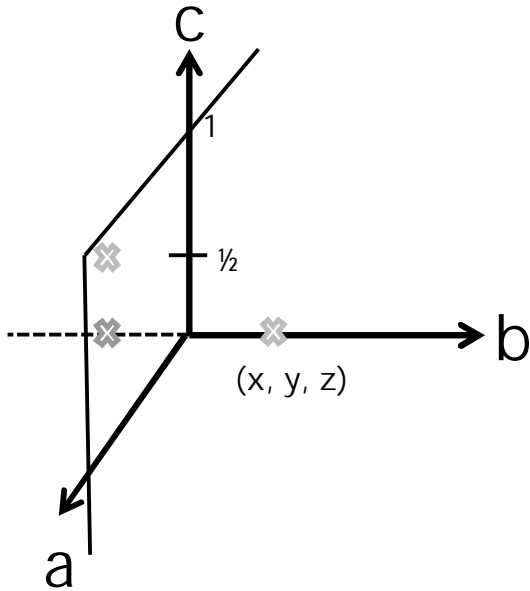
- Reflection against a mirror plane plus (half) translation parallel to the plane

Axis glide plane:  $a, b, c$  (translations by  $\frac{1}{2}a, \frac{1}{2}b, \frac{1}{2}c$  to each glide plane direction)

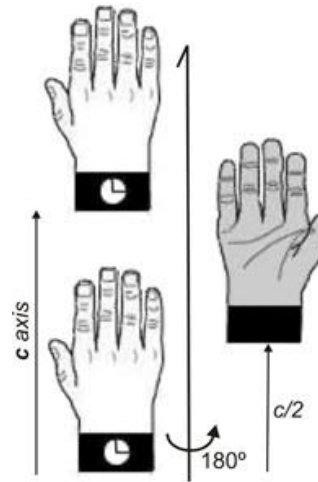
Diagonal glide plane:  $n$  [ $\frac{1}{2}(a+b), \frac{1}{2}(b+c), \frac{1}{2}(c+a)$ ]

$d$  [ $\frac{1}{4}(a+b), \frac{1}{4}(b+c), \frac{1}{4}(c+a)$ ] (so-called diamond glide plane)

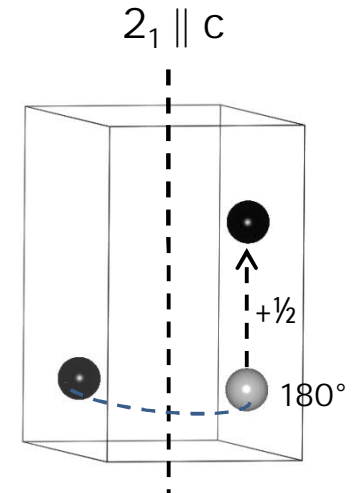
# GLIDE PLANE $c (\perp b)$



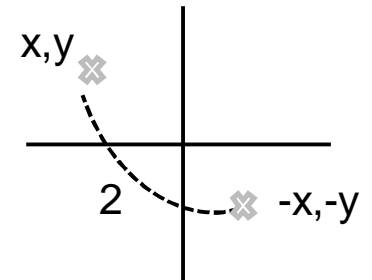
Rotation (c) followed by translation (t)



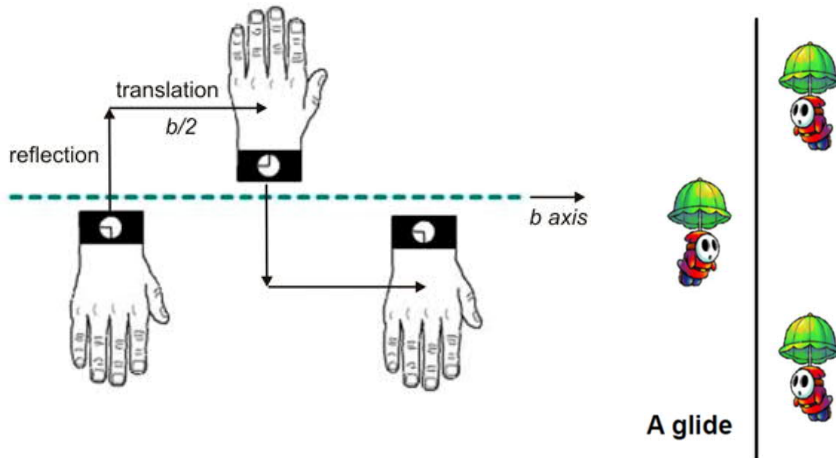
# SCREW AXIS



$$x, y, z \rightarrow -x, -y, z + \frac{1}{2}$$



Reflection (m) through ac-plane, followed by (half) translation (t) along c-axis



A glide

(from historical reasons)  
**DIFFERENT**

# SYMMETRY SYMBOLS

## ■ Schoenflies (S) symbols

- were developed first
- in molecular symmetry & spectroscopy

## ■ Hermann-Mauguin (H/M) symbols

- in crystallography
- long and short forms

## ■ Graphical symbols

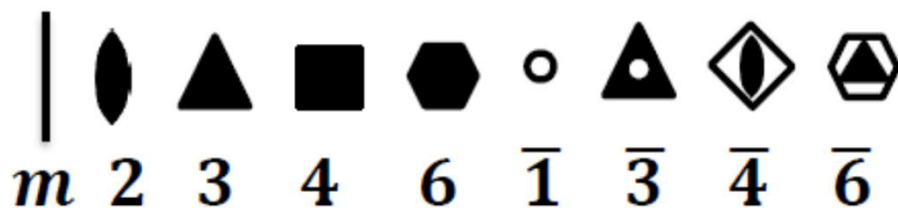


Table 1.1 Symmetry elements of crystal point groups.

System	Point group symmetry		Symmetry elements	Number of operations
	<i>S</i>	<i>H/M</i>		
Triclinic	$C_1$	1	$I = C_1$	1
	$C_1 = S_2$	$\bar{1}$	$I, i (= S_2)$	2
Monoclinic	$C_2$	2	$I, C_2$	2
	$C_2 = C_{1h} = C_{1v}$	$m$	$I, \sigma$	2
	$C_{2h}$	$2/m$	$I, C_2, \sigma_h, i$	4
Orthorhombic	$C_{2v}$	$mm2$	$I, C_2, 2\sigma$	4
	$D_2$	222	$I, 3C_2$	4
	$D_{2h}$	$mmm$	$I, 3C_2, 3\sigma, i$	8
Tetragonal	$C_4$	4	$I, C_4$	4
	$S_4$	$\bar{4}$	$I, S_4 (= C_2)$	4
	$D_4$	422	$I, C_4 (= C_2), 2C_2', 2C_2''$	8
	$C_{4v}$	$4mm$	$I, C_4, 2\sigma_v, 2\sigma_d$	8
	$C_{4h}$	$4/m$	$I, C_4 (= S_4), \sigma_h, i$	8
	$D_{2d}$	$\bar{4}2m$	$I, S_4 (= C_2), 2C_2', 2\sigma_d$	8
Trigonal	$D_{4h}$	$4/mmm$	$I, C_4 (= S_4), 2C_2', 2C_2'', 2\sigma_v, 2\sigma_d, i$	16
	$C_3$	3	$I, C_3$	3
	$C_{3i} = S_6$	$\bar{3}$	$I, S_6 (= C_3), i$	6
	$D_3$	32	$I, C_3, 3C_2$	6
	$C_{3v}$	$3m$	$I, C_3, 3\sigma_v$	6
	$D_{3d}$	$\bar{3}m$	$I, S_6 (= C_3), 3C_2, 3\sigma_d, i$	12
Hexagonal	$C_6$	6	$I, C_6$	6
	$C_{3h}$	$\bar{6}$	$I, S_3 (= C_3), \sigma_h$	6
	$D_6$	622	$I, C_6, 3C_2', 3C_2''$	12
	$D_{3h}$	$\bar{6}m2$	$I, C_3 (= S_3), 3C_2, 3\sigma_v, \sigma_h$	12
	$C_{6h}$	$6/m$	$I, C_6 (= S_6), \sigma_h, i$	12
	$C_{6v}$	$6mm$	$I, C_6, 3\sigma_v, 3\sigma_d$	12
Cubic	$D_{6h}$	$6/mmm$	$I, C_6 (= S_6), 3C_2', 3C_2'', 3\sigma_v, 3\sigma_d, \sigma_h, i$	24
	$T$	23	$I, 3C_2, 4C_3$	12
	$T_h$	$m\bar{3}$	$I, 3C_2, 4C_3 (= S_6), 3\sigma_h, i$	24
	$T_d$	$\bar{4}3m$	$I, 3C_2 (= S_4), 4C_3, 6\sigma_d$	24
	$O$	432	$I, 3C_2, 4C_3, 3C_4$	24
	$O_h$	$m\bar{3}m$	$I, 3C_2, 4C_3 (= S_6), 3C_4 (= S_4), 3\sigma_h, 6\sigma_d, i$	48

# WHAT WE LIKE TO KNOW ABOUT THE CRYSTAL STRUCTURE

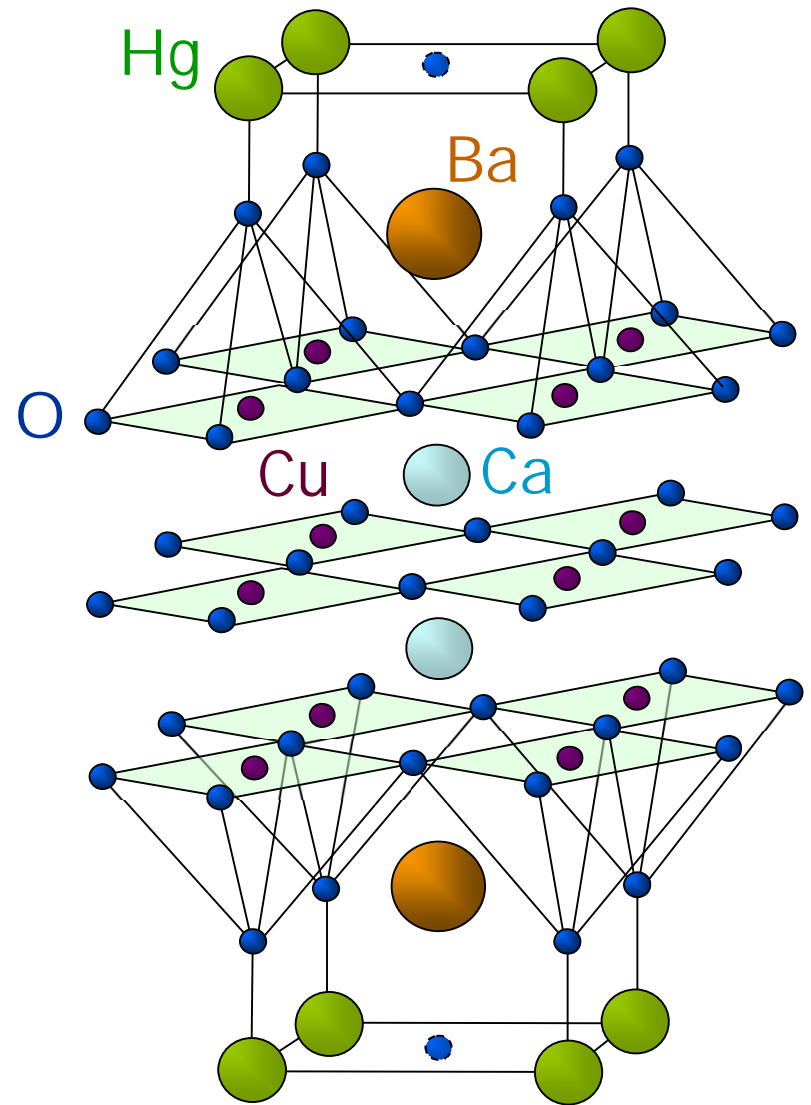
## CRYSTALLOGRAPHY

- symmetry
- unit cell
- lattice parameters
- number of formula units in unit cell
- space group
- etc.

## CRYSTAL CHEMISTRY

- coordination numbers
- coordination polyhedra
- bond lengths/angles
- occupation factors
- etc.

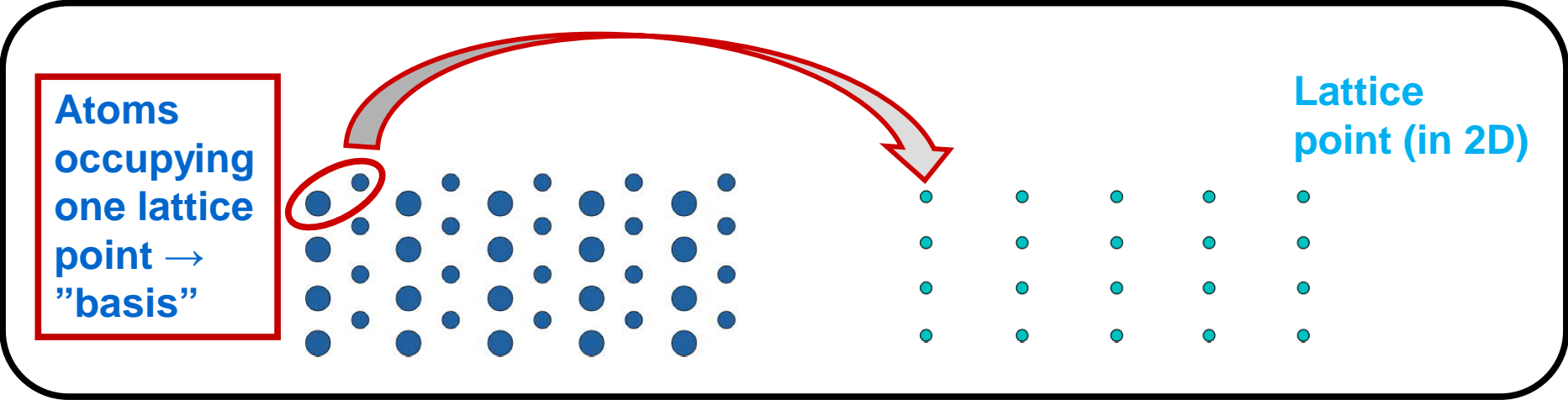
Discussed in Lecture 2!





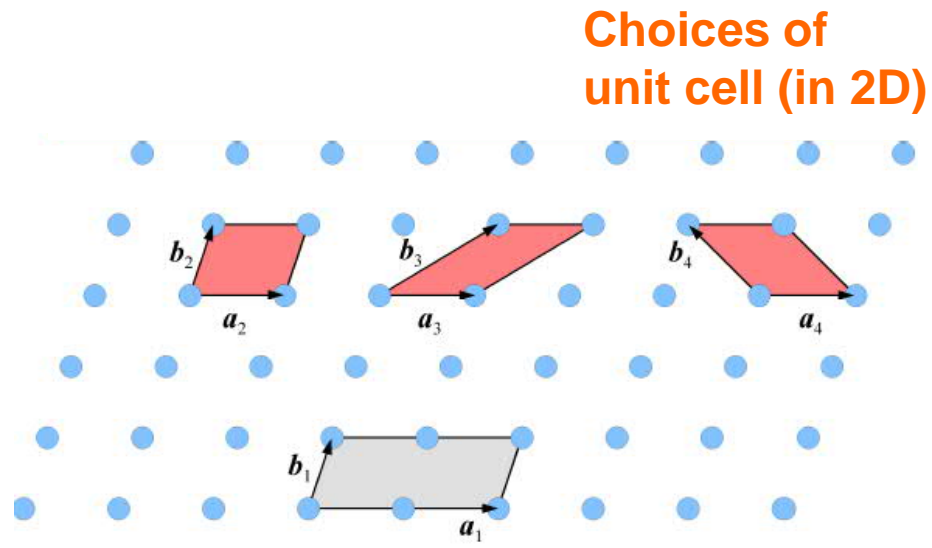
# CRYSTAL LATTICE

- Regular (infinite 3D) arrangement of lattice points
- **Lattice point**: consists of **one or more atoms (= basis)**
- Each lattice point has identical environment + symmetry properties (point group)

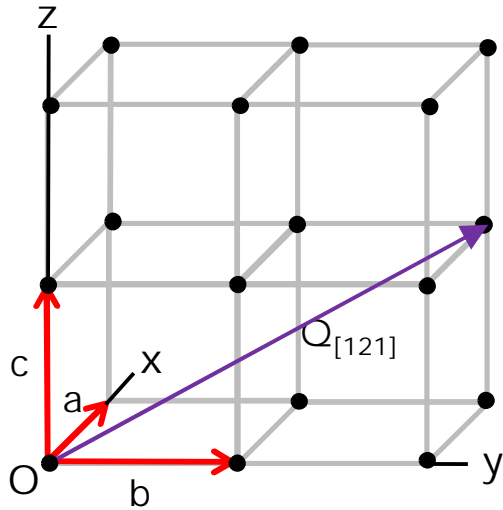


# UNIT CELL

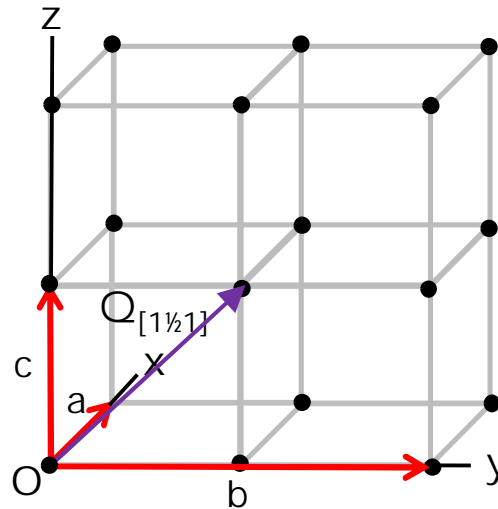
- Smallest possible microscopic 3D part of the crystal lattice that repeats itself periodically and completely fills the lattice volume, and is enough to describe the lattice perfectly
- Choice of the unit cell not always unambiguous: several possible choices



Each lattice point should be able to be described with the vector  $Q_{[uvw]} = ua + vb + wc$ , where  $u$ ,  $v$  and  $w$  all are INTEGER values



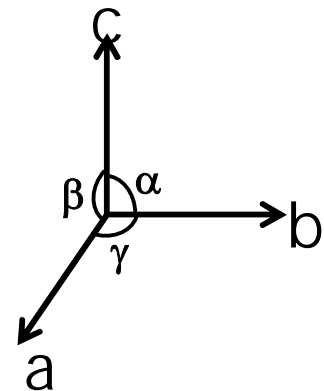
Unit cell: YES



Unit cell: NO

## What we need to tell about the UNIT CELL

- Shape & size of the unit cell *plus* the atomic positions in the unit cell
- Shape & size are given by:
  - Lattice parameters:  $a$ ,  $b$  ja  $c$
  - Angles between the axes:  $\alpha$ ,  $\beta$  ja  $\gamma$



# CLASSIFICATIONS

- "Macroscopic shape of the crystal"  
→ "Point group for the lattice" → **7 CRYSTAL SYSTEMS**  
(can fill the space without holes; no information of the lattice points/atoms)

## CRYSTAL SYSTEM & LATTICE POINTS (historical importance)

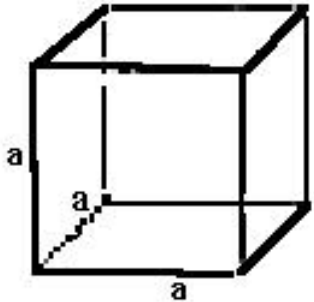
- Combination of crystal system and lattice type → **14 Bravais lattices**
- **Lattice type:** positions of lattice points ( $\neq$  atoms) within the cell considered: primitive (P), body-centered (I), face-centered (F), base-centered (A/B/C), rhombohedral (R)

## CRYSTAL SYSTEM & SYMMETRY (most important in crystallography)

- Possible combinations of point symmetry elements (which leave one point fixed)  
→ **32 (geometric) crystal classes**  
→ describe completely the symmetry of macroscopic crystals  
There are an infinite number of **three-dimensional point groups**, but the crystallographic restriction results in there being only 32 crystallographic point groups.
- Considering also the translational symmetry operations in 3D  
→ **230 space groups**

# CRYSTAL SYSTEMS (7)

No information of the positions of atoms

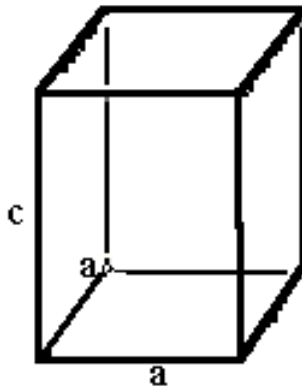


Cubic

$$a = b = c$$

$$\alpha = \beta = \gamma = 90^\circ$$

NaCl, MgAl<sub>2</sub>O<sub>4</sub>

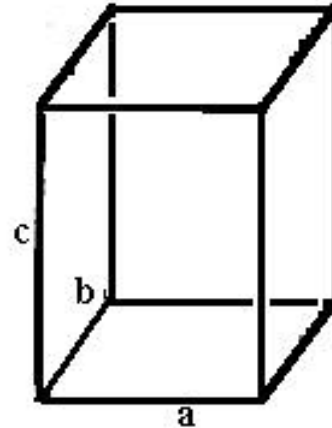


Tetragonal

$$a = b \neq c$$

$$\alpha = \beta = \gamma = 90^\circ$$

TiO<sub>2</sub>, K<sub>2</sub>NiF<sub>4</sub>

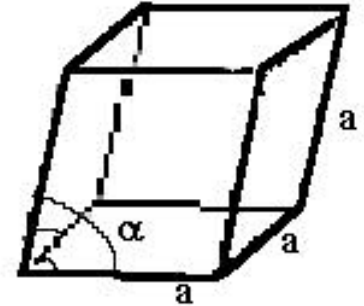


Orthorhombic

$$a \neq b \neq c$$

$$\alpha = \beta = \gamma = 90^\circ$$

YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7</sub>

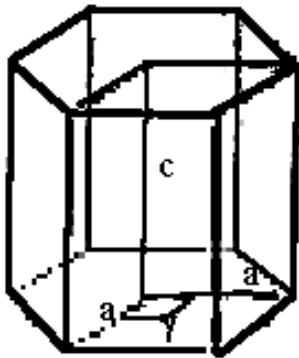


Rhombohedral

$$a = b = c$$

$$\alpha = \beta = \gamma \neq 90^\circ$$

BaTiO<sub>3</sub> (low-T)

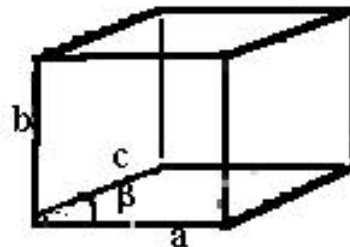


Hexagonal

$$a = b \neq c$$

$$\alpha = \beta = 90^\circ, \gamma = 120^\circ$$

LiNbO<sub>3</sub>

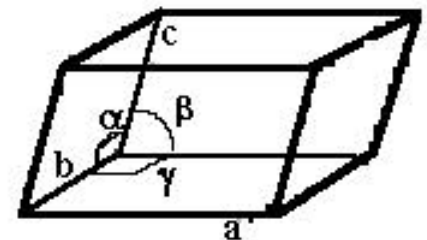


Monoclinic

$$a \neq b \neq c$$

$$\alpha = \gamma = 90^\circ, \beta \neq 90^\circ$$

KH<sub>2</sub>PO<sub>4</sub>



Triclinic

$$a \neq b \neq c$$

$$\alpha \neq \beta \neq \gamma \neq 90^\circ$$

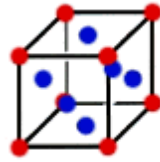
Element	What it does	Possible in crystal system
Identity (1)	-	All
Inversion ( $\bar{1}$ )	Inversion	All
Mirror plane (m)	Mirror	All but triclinic
2- fold rotation (2)	Rotate $180^\circ$	All but triclinic
3-fold rotation (3)	Rotate $120^\circ$	Trigonal, Hexagonal and Cubic
4-fold Rotation (4)	Rotate $90^\circ$	Tetragonal and Cubic
6-fold Rotation (6)	Rotate $60^\circ$	Hexagonal

# BRAVAIS LATTICES (14)

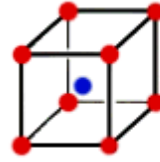
Positions of lattice sites (not atoms) included



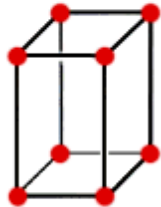
Simple cubic



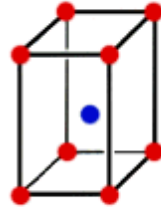
Face-centered cubic



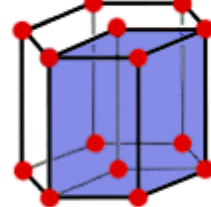
Body-centered cubic



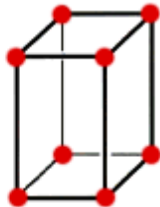
Simple tetragonal



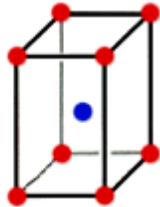
Body-centered tetragonal



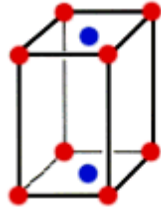
Hexagonal



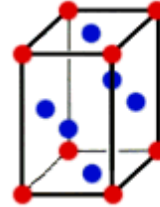
Simple orthorhombic



Body-centered orthorhombic



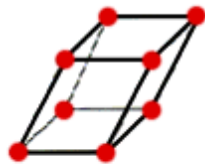
Base-centered orthorhombic



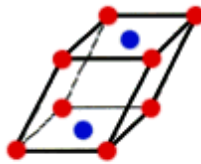
Face-centered orthorhombic



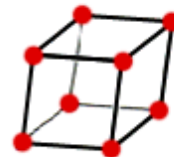
Rhomboidal



Simple Monoclinic



Base-centered monoclinic



Triclinic

Centering	Lattice points/cell	Abbreviation
Primitive	1	P
Base (A,B, or C) centered	2	A,B or C
Body centered	2	I
Hexagonal rhombohedral	3	hR
Face centered	4	F

# EXAMPLE

- What is the Bravais lattice type of NaCl: Cubic F (basis: Na-Cl)

Rock-salt (NaCl) structure

Paolo Fornasini  
Univ. Trento

NaCl	$a=5.64 \text{ \AA}$
KBr	$a=6.60 \text{ \AA}$
CaO	$a=4.81 \text{ \AA}$

conventional unit cell (8 atoms per cell)

Non-Bravais lattice

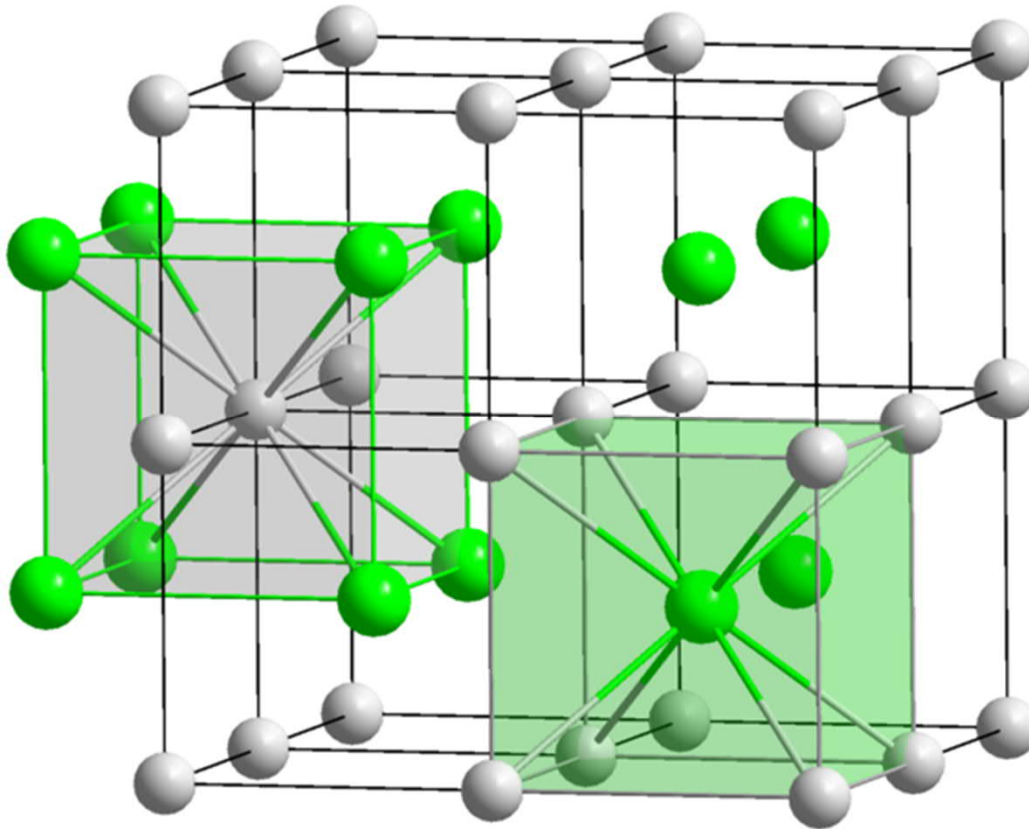
fcc Bravais lattice + 2-atom basis

$(0,0,0)$   $\left(\frac{1}{2}, \frac{1}{2}, \frac{1}{2}\right)$

Cordination number = 6

# Your **EXERCISE** question

- What is the Bravais lattice type of **CsCl**





# CRYSTAL CLASSES (32)

The 32 Point Groups			
1	4	$\bar{3}$	6mm
$\bar{1}$	$\bar{4}$	32	$\bar{6}m2$
2	4/m	3m	6/mmm
m	422	$\bar{3}m$	23
2/m	4mm	6	$m\bar{3}$
222	$\bar{4}2m$	$\bar{6}$	432
mm2	4/mmm	6/m	$\bar{4}3m$
mmm	3	622	$m\bar{3}m$

The 32 Point Groups (Schoenflies)			
1 ( $C_1$ )	4 ( $C_4$ )	$\bar{3}$ ( $C_{3i}$ )	6mm ( $C_{6\sigma v}$ )
$\bar{1}$ ( $C_i = S_2$ )	$\bar{4}$ ( $S_4$ )	32 ( $D_3$ )	$\bar{6}m2$ ( $D_{3\sigma h}$ )
2 ( $C_2$ )	4/m ( $C_{4\sigma h}$ )	3m ( $C_{3\sigma v}$ )	6/mmm ( $D_{6\sigma h}$ )
m ( $C_\sigma$ )	422 ( $D_4$ )	$\bar{3}m$ ( $D_{3d}$ )	23 (T)
2/m ( $C_{2\sigma h}$ )	4mm ( $C_{4\sigma v}$ )	6 ( $C_6$ )	$m\bar{3}$ ( $T_h$ )
222 ( $D_2$ )	$\bar{4}2m$ ( $D_{2d}$ )	$\bar{6}$ ( $C_{3\sigma h}$ )	432 (O)
2mm ( $C_{2\sigma v}$ )	4/mmm ( $D_{4h}$ )	6/m ( $C_{6\sigma h}$ )	$\bar{4}3m$ ( $T_d$ )
mmm ( $D_{2\sigma h}$ )	3 ( $C_3$ )	622 ( $D_6$ )	$m\bar{3}m$ ( $O_h$ )

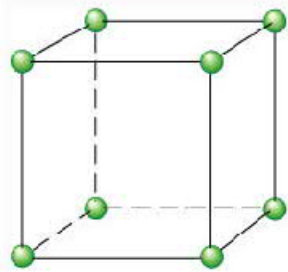
<b>Crystal System</b>	<b># of Point Groups</b>	<b># of Crystal Lattices</b>
<b>Cubic</b>	<b>5</b>	<b>3</b>
<b>Tetragonal</b>	<b>7</b>	<b>2</b>
<b>Orthorhombic</b>	<b>3</b>	<b>4</b>
<b>Monoclinic</b>	<b>3</b>	<b>2</b>
<b>Triclinic</b>	<b>2</b>	<b>1</b>
<b>Hexagonal</b>	<b>7</b>	<b>1</b>
<b>Trigonal</b>	<b>5</b>	<b>1</b>
<b>Totals</b>	<b>32</b>	<b>14</b>

System	Minimum Requirements
Cubic	Four 3-fold rotation axis
Tetragonal	One 4-fold rotation (or R1) axis
Orthorhombic	Three perpendicular 2-fold axis
Rhombohedral	One 3-fold rotation (or R1) axis
Hexagonal	One 6 fold rotation (or R1) axis
Monoclinic	One 2 fold rotation axis or mirror plane
Triclinic	none

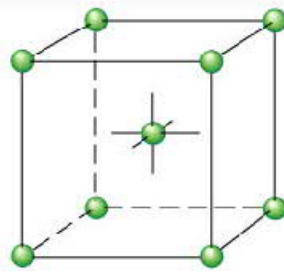
System	Point groups
Cubic	23, $m\bar{3}$ , 432, $\bar{4}3m$ , $m\bar{3}m$
Tetragonal	4, $\bar{4}$ , 4/m, 422, 4mm, $\bar{4}2m$ , 4/mmm
Orthorhombic	222, 2mm, mmm
Trigonal	3, $\bar{3}$ , 32, 3m, $\bar{3}m$
Hexagonal	6, $\bar{6}$ , 6/m, 622, 6mm, $\bar{6}m2$ , 6/mmm
Monoclinic	2, m, 2/m
Triclinic	1, $\bar{1}$

# NUMBER of FORMULA UNITS in UNIT CELL (Z)

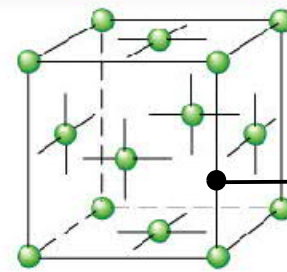
- typically 1 – 6, but can be tens or even hundreds
- atom inside unit cell: belongs only to one unit cell
- atom on unit cell face: belongs to two unit cells
- atom on unit cell edge: belongs to four unit cells
- atom on unit cell corner: belongs to eight unit cells



Simple cubic

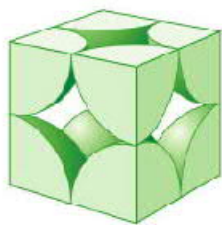


Body-centered cubic

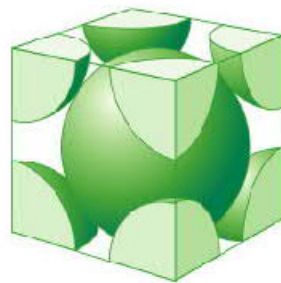


Face-centered cubic

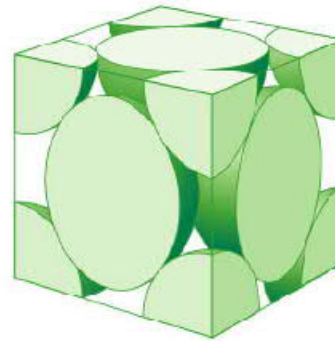
EDGE SITE



Simple cubic



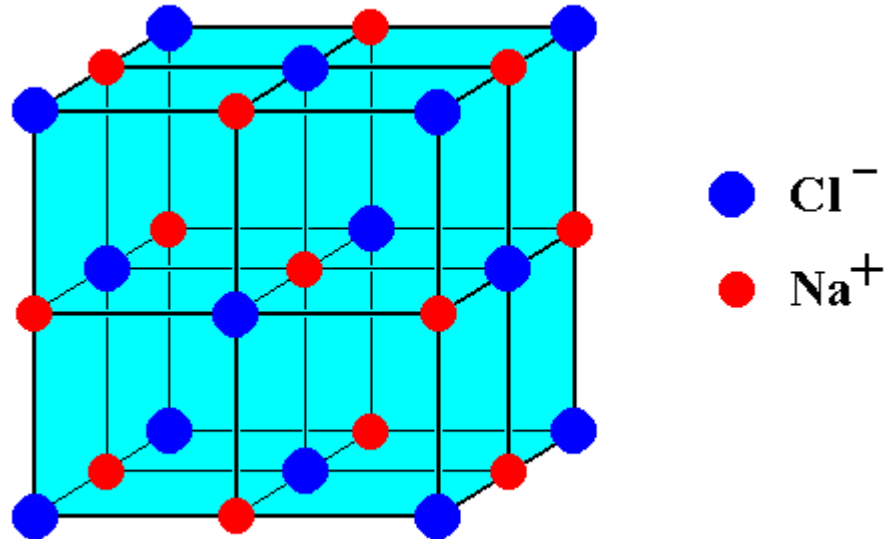
Body-centered cubic



Face-centered cubic

# EXAMPLE

- How many NaCl formula units in unit cell?
- Answer: 4

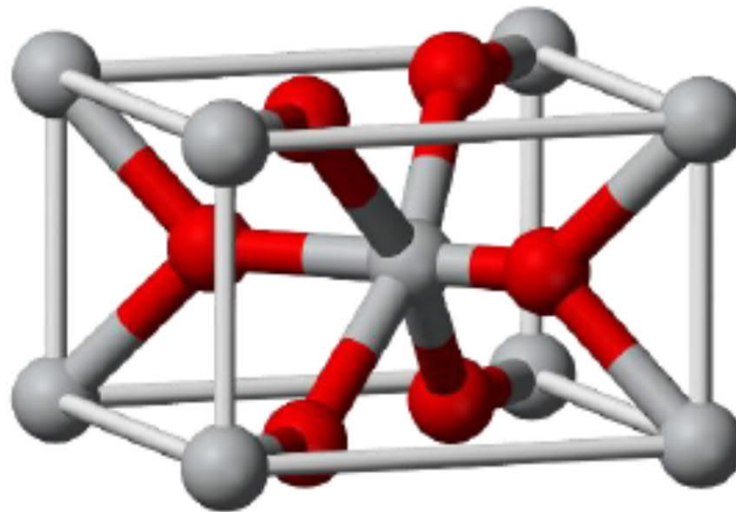


## Your **EXERCISE** question

White balls are Ti atoms, red balls are oxygen atoms.

Unit cell parameters:  $a = b = 4.5937 \text{ \AA}$ ,  $c = 2.9587 \text{ \AA}$ ; all angles  $90^\circ$ .

- (a)** What is the crystal system?
- (b)** What is the formula of the compound ?
- (c)** Please calculate the density.



# SPACE GROUPS

- From finite molecule (or macroscopic crystal) to infinite lattice → translation symmetry must be included
- Possible combinations of symmetry elements (including the translation symmetry elements): **230**  
→ **230 space groups**
- Space groups (and the characteristic information in 2 pages) are listed in **International Tables for Crystallography**  
→ **BIBLE of CRYSTALLOGRAPHY**
- Next-next slide: Space Group P4/mmm as an example

<https://it.iucr.org/>



## Triclinic

(For the enlarged unit cells, click [here](#))

1. [P1](#)      2. [P-1](#)

## Monoclinic

(For a fuller list with alternative unique axes, origins, or enlarged unit cells click [here](#))

3. [P121](#)    4. [P1211](#)    5. [C121](#)    6. [P1m1](#)    7. [P1c1](#)  
8. [C1m1](#)    9. [C1c1](#)    10. [P12/m1](#)    11. [P121/m1](#)    12. [C12/m1](#)  
13. [P12/c1](#)    14. [P121/c1](#)    15. [C12/c1](#)

## Orthorhombic

(For a fuller list with alternative axes and origins click [here](#))

16. [P222](#)    17. [P2221](#)    18. [P21212](#)    19. [P212121](#)    20. [C2221](#)  
21. [C222](#)    22. [F222](#)    23. [I222](#)    24. [I212121](#)    25. [Pmm2](#)  
26. [Pmc21](#)    27. [Pcc2](#)    28. [Pma2](#)    29. [Pca21](#)    30. [Pnc2](#)  
31. [Pmn21](#)    32. [Pba2](#)    33. [Pna21](#)    34. [Pnn2](#)    35. [Cmm2](#)  
36. [Cmc21](#)    37. [Ccc2](#)    38. [Amn2](#)    39. [Abm2](#)    40. [Ama2](#)  
41. [Aba2](#)    42. [Fmm2](#)    43. [Fdd2](#)    44. [Imn2](#)    45. [Iba2](#)  
46. [Ima2](#)    47. [Pmmm](#)    48. [Pnnn](#)    49. [Pccm](#)    50. [Pban](#)  
51. [Pmma](#)    52. [Pnna](#)    53. [Pmna](#)    54. [Pcca](#)    55. [Pbam](#)  
56. [Pccn](#)    57. [Pbcn](#)    58. [Pnnm](#)    59. [Pmnn](#)    60. [Pbcn](#)  
61. [Pbca](#)    62. [Pnma](#)    63. [Cmcm](#)    64. [Cmca](#)    65. [Cmmm](#)  
66. [Cccm](#)    67. [Cnma](#)    68. [Ccca](#)    69. [Fmmm](#)    70. [Fddd](#)  
71. [Immm](#)    72. [Ibam](#)    73. [Ibca](#)    74. [Imma](#)

## Tetragonal

(For the enlarged C- and F-centred unit cells, click [here](#))

75. [P4](#)      76. [P41](#)      77. [P42](#)      78. [P43](#)      79. [I4](#)  
80. [I41](#)    81. [P-4](#)    82. [I-4](#)    83. [P4/m](#)    84. [P42/m](#)  
85. [P4/n](#)    86. [P42/n](#)    87. [I4/m](#)    88. [I4/a](#)    89. [P422](#)    168. [P6](#)    169. [P61](#)    170. [P62](#)    171. [P62](#)    172. [P64](#)  
90. [P422](#)    91. [P4122](#)    92. [P41212](#)    93. [P4222](#)    94. [P42212](#)    173. [P63](#)    174. [P-6](#)    175. [P6/m](#)    176. [P62/m](#)    177. [P622](#)  
95. [P4122](#)    96. [P41212](#)    97. [I422](#)    98. [I4122](#)    99. [P4mm](#)    178. [P6122](#)    179. [P6222](#)    180. [P6222](#)    181. [P6422](#)    182. [P6222](#)  
100. [P4bm](#)    101. [P42cm](#)    102. [P42nm](#)    103. [P4cc](#)    104. [P4nc](#)    183. [P6mm](#)    184. [P6cc](#)    185. [P63cm](#)    186. [P63mc](#)    187. [P-6m2](#)  
105. [P42mc](#)    106. [P42bc](#)    107. [I4mm](#)    108. [I4cm](#)    109. [I4md](#)    188. [P-6c2](#)    189. [P-62m](#)    190. [P-62c](#)    191. [P6/mmm](#)    192. [P6/mcc](#)  
110. [I4cd](#)    111. [P-42m](#)    112. [P-42c](#)    113. [P-421m](#)    114. [P-421c](#)    193. [P63/mcm](#)    194. [P63/mmc](#)  
115. [P-4m2](#)    116. [P-4c2](#)    117. [P-4b2](#)    118. [P-4n2](#)    119. [I-4m2](#)  
120. [I-4c2](#)    121. [I-42m](#)    122. [I-42d](#)    123. [P4/mmm](#)    124. [P4/mcc](#)  
125. [P4/nbm](#)    126. [P4/nnc](#)    127. [P4/nbm](#)    128. [P4/mnc](#)    129. [P4/nmm](#)  
130. [P4/ncc](#)    131. [P42/mmc](#)    132. [P42/mcm](#)    133. [P42/nbc](#)    134. [P42/nnm](#)  
135. [P42/mbc](#)    136. [P42/mnm](#)    137. [P42/nmc](#)    138. [P42/nccm](#)    139. [I4/mmm](#)  
140. [I4/mcm](#)    141. [I4/amd](#)    142. [I4/acd](#)

## Trigonal

(For the R-centred cells with hexagonal axes and the larger H-centred trigonal cells, click [here](#))

143. [P3](#)      144. [P31](#)      145. [P32](#)      146. [R3](#)      147. [P-3](#)  
148. [R-3](#)    149. [P312](#)    150. [P321](#)    151. [P3112](#)    152. [P3211](#)  
153. [P3212](#)    154. [P3221](#)    155. [R32](#)    156. [P3m1](#)    157. [P31m](#)  
158. [P3c1](#)    159. [P31c](#)    160. [R3m](#)    161. [R3c](#)    162. [P-31m](#)  
163. [P-31c](#)    164. [P-3m1](#)    165. [P-3c1](#)    166. [R-3m](#)    167. [R-3c](#)

## Hexagonal

195. [P23](#)    196. [F23](#)    197. [I23](#)    198. [P213](#)    199. [I213](#)  
200. [Pm-3](#)    201. [Pn-3](#)    202. [Fm-3](#)    203. [Fd-3](#)    204. [Im-3](#)  
205. [Pa-3](#)    206. [Ia-3](#)    207. [P432](#)    208. [P432](#)    209. [F432](#)  
210. [F432](#)    211. [I432](#)    212. [P432](#)    213. [P432](#)    214. [I432](#)  
215. [P-43m](#)    216. [F-43m](#)    217. [I-43m](#)    218. [P-43n](#)    219. [F-43c](#)  
220. [I-43d](#)    221. [Pm-3m](#)    222. [Pn-3n](#)    223. [Pm-3n](#)    224. [Pn-3m](#)  
225. [Fm-3m](#)    226. [Fm-3c](#)    227. [Fd-3m](#)    228. [Fd-3c](#)    229. [Im-3m](#)  
230. [Ia-3d](#)

## Cubic

From: <http://img.chem.ucl.ac.uk/sgp/large/sgp.htm>

All space group diagrams and tables online:  
<http://img.chem.ucl.ac.uk/sgp/large/sgp.htm>



$P4/mmm$

$D_{4h}^1$

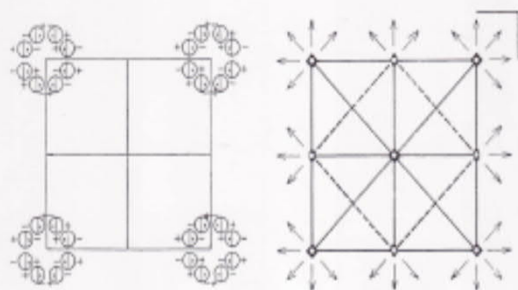
$4/mmm$

Tetragonal

No. 123

$P4/m2/m2/m$

Patterson symmetry  $P4/mmm$



Origin at centre ( $4/mmm$ )

Asymmetric unit  $0 \leq x \leq \frac{1}{2}; 0 \leq y \leq \frac{1}{2}; 0 \leq z \leq \frac{1}{2}; x \leq y$

Symmetry operations

- |                  |                  |                                 |                                 |
|------------------|------------------|---------------------------------|---------------------------------|
| (1) 1            | (2) 2 $0,0,z$    | (3) $4^+$ $0,0,z$               | (4) $4^-$ $0,0,z$               |
| (5) 2 $0,y,0$    | (6) 2 $x,0,0$    | (7) 2 $x,x,0$                   | (8) 2 $x,x,0$                   |
| (9) 1 $0,0,0$    | (10) $m$ $x,y,0$ | (11) $\bar{4}^+$ $0,0,z; 0,0,0$ | (12) $\bar{4}^-$ $0,0,z; 0,0,0$ |
| (13) $m$ $x,0,z$ | (14) $m$ $0,y,z$ | (15) $m$ $x,x,z$                | (16) $m$ $x,x,z$                |

Maximal non-isomorphic subgroups

- I** [2] $P422$  1; 2; 3; 4; 5; 6; 7; 8  
 [2] $P4/m11(P4/m)$  1; 2; 3; 4; 9; 10; 11; 12  
 [2] $P4mm$  1; 2; 3; 4; 13; 14; 15; 16  
 [2] $P\bar{4}2m$  1; 2; 5; 6; 11; 12; 15; 16  
 [2] $P\bar{4}m2$  1; 2; 7; 8; 11; 12; 13; 14  
 [2] $P2/m2/m1(Pmmm)$  1; 2; 5; 6; 9; 10; 13; 14  
 [2] $P2/m12/m(Cmmm)$  1; 2; 7; 8; 9; 10; 15; 16
- IIa** none
- IIb** [2] $P4/mcc(c'=2c)$ ; [2] $P4_2/mcc(c'=2c)$ ; [2] $P4_2/mcm(c'=2c)$ ; [2] $C4/mc2(a'=2a, b'=2b)(P4/bbw)$ ;  
 [2] $C4/mcd(a'=2a, b'=2b)(P4/bbm)$ ; [2] $C4/mww(a'=2a, b'=2b)(P4/nmm)$ ;  
 [2] $F4/mww(a'=2a, b'=2b, c'=2c)(I4/mmm)$ ; [2] $F4/mmc(a'=2a, b'=2b, c'=2c)(I4/mcm)$

Maximal isomorphic subgroups of lowest index

- IIc** [2] $P4/mmm(c'=2c)$ ; [2] $C4/mww(a'=2a, b'=2b)(P4/mmm)$

Minimal non-isomorphic supergroups

- I** [3] $Pm\bar{3}m$   
**II** [2] $I4/mmm$

CONTINUED

No. 123

$P4/mmm$

Generators selected (1);  $r(1,0,0)$ ;  $r(0,1,0)$ ;  $r(0,0,1)$ ; (2); (3); (5); (9)

Positions

Multiplicity,  
Wyckoff letter,  
Site symmetry

Coordinates

Reflection conditions

16	$a$	1	(1) $x,y,z$	(2) $\bar{x},\bar{y},z$	(3) $\bar{y},x,z$	(4) $y,\bar{x},z$
			(5) $x,y,\bar{z}$	(6) $x,\bar{y},\bar{z}$	(7) $y,x,\bar{z}$	(8) $\bar{y},\bar{x},\bar{z}$
			(9) $\bar{x},\bar{y},z$	(10) $x,y,\bar{z}$	(11) $y,\bar{x},z$	(12) $\bar{y},x,\bar{z}$
			(13) $x,\bar{y},z$	(14) $\bar{x},y,z$	(15) $\bar{y},\bar{x},z$	(16) $y,x,z$

General:

no conditions

Special:

8	$i$	$.m$	$x,\bar{z}$	$\bar{x},\bar{z}$	$\bar{y},z$	$\bar{y},z$
			$\bar{x},z$	$x,\bar{z}$	$\bar{y},z$	$\bar{y},z$

no extra conditions

8	$s$	$.m$	$x,0,z$	$\bar{x},0,z$	$0,x,z$	$0,\bar{x},z$
			$\bar{x},z$	$x,0,z$	$0,x,z$	$0,\bar{x},z$

no extra conditions

8	$r$	$.m$	$x,x,z$	$\bar{x},\bar{x},z$	$\bar{x},x,z$	$x,\bar{x},z$
			$\bar{x},z$	$x,\bar{x},z$	$\bar{x},x,z$	$x,\bar{x},z$

no extra conditions

8	$q$	$m..$	$x,y,\bar{z}$	$\bar{x},\bar{y},\bar{z}$	$\bar{y},x,\bar{z}$	$y,\bar{x},\bar{z}$
			$\bar{y},\bar{z}$	$x,\bar{y},\bar{z}$	$y,x,\bar{z}$	$\bar{y},\bar{x},\bar{z}$

no extra conditions

8	$p$	$m..$	$x,y,0$	$\bar{x},\bar{y},0$	$\bar{y},x,0$	$y,\bar{x},0$
			$\bar{y},0$	$x,\bar{y},0$	$y,x,0$	$\bar{y},\bar{x},0$

no extra conditions

4	$o$	$m2m$	$x,\bar{z}$	$\bar{x},\bar{z}$	$\bar{y},z$	$\bar{y},z$
			$\bar{x},z$	$x,\bar{z}$	$\bar{y},z$	$\bar{y},z$

no extra conditions

4	$n$	$m2m$	$x,\bar{z}$	$\bar{x},\bar{z}$	$\bar{y},z$	$\bar{y},z$
			$\bar{x},z$	$x,\bar{z}$	$\bar{y},z$	$\bar{y},z$

no extra conditions

4	$m$	$m2m$	$x,0,z$	$\bar{x},0,z$	$0,x,z$	$0,\bar{x},z$
			$\bar{x},z$	$x,0,z$	$0,x,z$	$0,\bar{x},z$

no extra conditions

4	$l$	$m2m$	$x,0,z$	$\bar{x},0,z$	$0,x,z$	$0,\bar{x},z$
			$\bar{x},z$	$x,0,z$	$0,x,z$	$0,\bar{x},z$

no extra conditions

4	$k$	$m.2m$	$x,x,z$	$\bar{x},\bar{x},z$	$\bar{x},x,z$	$x,\bar{x},z$
			$\bar{x},z$	$x,\bar{x},z$	$\bar{x},x,z$	$x,\bar{x},z$

no extra conditions

4	$j$	$m.2m$	$x,x,0$	$\bar{x},\bar{x},0$	$\bar{x},x,0$	$x,\bar{x},0$
			$\bar{x},0$	$x,\bar{x},0$	$\bar{x},x,0$	$x,\bar{x},0$

no extra conditions

4	$i$	$2mw$	$0,\bar{z}$	$\bar{0},z$	$0,\bar{z}$	$\bar{0},z$
			$\bar{z}$	$0,z$	$0,\bar{z}$	$\bar{0},z$

$hk\bar{l} : h+k=2n$

2	$h$	$4mw$	$\bar{z}$	$\bar{z}$	$\bar{z}$	$\bar{z}$
			$z$	$z$	$z$	$z$

no extra conditions

2	$g$	$4mw$	$0,z$	$0,z$	$0,z$	$0,z$
			$z$	$z$	$z$	$z$

no extra conditions

2	$f$	$mmm$	$0,\bar{z}$	$\bar{0},z$	$0,\bar{z}$	$\bar{0},z$
			$z$	$z$	$z$	$z$

$hk\bar{l} : h+k=2e$

2	$e$	$mmm$	$0,\bar{z}$	$\bar{0},z$	$0,\bar{z}$	$\bar{0},z$
			$z$	$z$	$z$	$z$

$hk\bar{l} : h+k=2n$

Symmetry of special projections

Along [001]  $p4mm$   
 $a'=a$   $b'=b$   
 Origin at  $0,0,z$

Along [100]  $p2mm$   
 $a'=b$   $b'=c$   
 Origin at  $x,0,0$

Along [110]  $p2mm$   
 $a'=b(-a+b)$   $b'=c$   
 Origin at  $x,x,0$

(Continued on preceding page)

## EXAMPLES of INFORMATION

- Space group number: 123
- Name: P4/mmm
- Complete name:  $P 4/m 2/m 2/m$  ; showing the symmetry elements (4-fold rotation axis, mirror planes)
- Crystal system: tetragonal
- Lattice type: P (primitive)
- Site symmetry of the highest-symmetry site:  $D_{4h}$
- Asymmetric unit: smallest closed part of space the entire space is filled by applying all symmetry operations

## On the second page:

- List of **possible** sites for the atoms
- These are indicated/named by: multiplicity, Wyckoff letter & site symmetry
- Not all sites are actually occupied by an atom
- On the top: general site (16u)
- At the bottom: the highest symmetry site (1a)
- Multiplicity: number of identical sites

Generators selected (1);  $t(1,0,0)$ ;  $t(0,1,0)$ ;  $t(0,0,1)$ ; (2); (3); (5); (9)

Positions		Coordinates				Reflection conditions	
Multiplicity	Wyckoff letter					General:	
Site symmetry						no conditions	
16	u	1	(1) $x, y, z$ (5) $\bar{x}, y, \bar{z}$ (9) $\bar{x}, \bar{y}, \bar{z}$ (13) $x, \bar{y}, z$	(2) $\bar{x}, \bar{y}, z$ (6) $x, \bar{y}, \bar{z}$ (10) $x, y, \bar{z}$ (14) $\bar{x}, y, z$	(3) $\bar{y}, x, z$ (7) $y, x, \bar{z}$ (11) $y, \bar{x}, \bar{z}$ (15) $\bar{y}, \bar{x}, z$	(4) $y, \bar{x}, z$ (8) $\bar{y}, \bar{x}, \bar{z}$ (12) $\bar{y}, x, \bar{z}$ (16) $y, x, z$	Special: no extra conditions
8	t	.m.	$x, \frac{1}{2}, z$ $\bar{x}, \frac{1}{2}, \bar{z}$	$\bar{x}, \frac{1}{2}, z$ $x, \frac{1}{2}, \bar{z}$	$\frac{1}{2}, x, z$ $\frac{1}{2}, x, \bar{z}$	$\frac{1}{2}, \bar{x}, z$ $\frac{1}{2}, \bar{x}, \bar{z}$	no extra conditions
8	s	.m.	$x, 0, z$ $\bar{x}, 0, \bar{z}$	$\bar{x}, 0, z$ $x, 0, \bar{z}$	$0, x, z$ $0, x, \bar{z}$	$0, \bar{x}, z$ $0, \bar{x}, \bar{z}$	no extra conditions
8	r	.m	$x, x, z$ $\bar{x}, x, \bar{z}$	$\bar{x}, \bar{x}, z$ $x, \bar{x}, \bar{z}$	$\bar{x}, x, z$ $x, x, \bar{z}$	$x, \bar{x}, z$ $\bar{x}, \bar{x}, \bar{z}$	no extra conditions
8	q	m..	$x, y, \frac{1}{2}$ $\bar{x}, y, \frac{1}{2}$	$\bar{x}, \bar{y}, \frac{1}{2}$ $x, \bar{y}, \frac{1}{2}$	$\bar{y}, x, \frac{1}{2}$ $y, x, \frac{1}{2}$	$y, \bar{x}, \frac{1}{2}$ $\bar{y}, \bar{x}, \frac{1}{2}$	no extra conditions
8	p	m..	$x, y, 0$ $\bar{x}, y, 0$	$\bar{x}, \bar{y}, 0$ $x, \bar{y}, 0$	$\bar{y}, x, 0$ $y, x, 0$	$y, \bar{x}, 0$ $\bar{y}, \bar{x}, 0$	no extra conditions
4	o	m 2m.	$x, \frac{1}{2}, \frac{1}{2}$	$\bar{x}, \frac{1}{2}, \frac{1}{2}$	$\frac{1}{2}, x, \frac{1}{2}$	$\frac{1}{2}, \bar{x}, \frac{1}{2}$	no extra conditions
4	n	m 2m.	$x, \frac{1}{2}, 0$	$\bar{x}, \frac{1}{2}, 0$	$\frac{1}{2}, x, 0$	$\frac{1}{2}, \bar{x}, 0$	no extra conditions
4	m	m 2m.	$x, 0, \frac{1}{2}$	$\bar{x}, 0, \frac{1}{2}$	$0, x, \frac{1}{2}$	$0, \bar{x}, \frac{1}{2}$	no extra conditions
4	l	m 2m.	$x, 0, 0$	$\bar{x}, 0, 0$	$0, x, 0$	$0, \bar{x}, 0$	no extra conditions
4	k	m . 2m	$x, x, \frac{1}{2}$	$\bar{x}, \bar{x}, \frac{1}{2}$	$\bar{x}, x, \frac{1}{2}$	$x, \bar{x}, \frac{1}{2}$	no extra conditions
4	j	m . 2m	$x, x, 0$	$\bar{x}, \bar{x}, 0$	$\bar{x}, x, 0$	$x, \bar{x}, 0$	no extra conditions
4	i	2m m.	$0, \frac{1}{2}, z$	$\frac{1}{2}, 0, z$	$0, \frac{1}{2}, \bar{z}$	$\frac{1}{2}, 0, \bar{z}$	$hkl : h+k=2n$
2	h	4m m	$\frac{1}{2}, \frac{1}{2}, z$	$\frac{1}{2}, \frac{1}{2}, \bar{z}$			no extra conditions
2	g	4m m	$0, 0, z$	$0, 0, \bar{z}$			no extra conditions
2	f	m m m.	$0, \frac{1}{2}, 0$	$\frac{1}{2}, 0, 0$			$hkl : h+k=2n$
2	e	m m m.	$0, \frac{1}{2}, \frac{1}{2}$	$\frac{1}{2}, 0, \frac{1}{2}$			$hkl : h+k=2n$
1	d	4/m m m	$\frac{1}{2}, \frac{1}{2}, \frac{1}{2}$				no extra conditions
1	c	4/m m m	$\frac{1}{2}, \frac{1}{2}, 0$				no extra conditions
1	b	4/m m m	$0, 0, \frac{1}{2}$				no extra conditions
1	a	4/m m m	$0, 0, 0$				no extra conditions

### Symmetry of special projections

Along [001]  $p 4m m$   
 $a' = a$   $b' = b$   
 Origin at  $0, 0, z$

Along [100]  $p 2m m$   
 $a' = b$   $b' = c$   
 Origin at  $x, 0, 0$

Along [110]  $p 2m m$   
 $a' = \frac{1}{2}(-a+b)$   $b' = c$   
 Origin at  $x, x, 0$

(Continued on preceding page)

## EXAMPLE: Potassium tetrachloroplatinate(II): $K_2PtCl_4$

Space group:  $P4/mmm$  (No. 123)

Lattice parameters:  $a = b = 7.023\text{\AA}$ ,  $c = 4.1486\text{\AA}$

Atomic positions:	Pt	1a:	0,0,0
	K	2e:	$0, \frac{1}{2}, \frac{1}{2}$
	Cl	4j:	$x, x, 0$ ; $x = 0.23247$

- Draw the unit cell with the atoms.
- Draw the projection of the unit cell in  $c$ -axis direction.
- Theoretical density is  $3.37\text{ g/cm}^3$ . Calculate  $Z$ ?  
( $N_A = 6.022 \times 10^{23}$ ; atomic weights: K 39.098; Pt 195.22; Cl 35.453)
- Calculate the distances: Pt-Pt, Pt-K, Pt-Cl.
- What is the coordination number of platinum ?
- What is the site symmetry of platinum ?

$P4/mmm$

$D_{4h}^1$

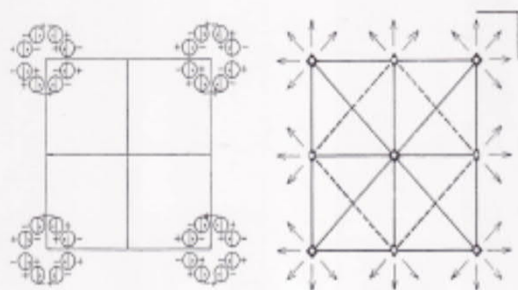
$4/mmm$

Tetragonal

No. 123

$P4/m\ 2/m\ 2/m$

Patterson symmetry  $P4/mmm$



Origin at centre ( $4/mmm$ )

Asymmetric unit  $0 \leq x \leq \frac{1}{2}; 0 \leq y \leq \frac{1}{2}; 0 \leq z \leq \frac{1}{2}; x \leq y$

Symmetry operations

- |                 |                 |                                 |                                 |
|-----------------|-----------------|---------------------------------|---------------------------------|
| (1) 1           | (2) 2 $0,0,z$   | (3) $4^+$ $0,0,z$               | (4) $4^-$ $0,0,z$               |
| (5) 2 $0,y,0$   | (6) 2 $x,0,0$   | (7) 2 $x,x,0$                   | (8) 2 $x,x,0$                   |
| (9) 1 $0,0,0$   | (10) $m\ x,y,0$ | (11) $\bar{4}^+$ $0,0,z; 0,0,0$ | (12) $\bar{4}^-$ $0,0,z; 0,0,0$ |
| (13) $m\ x,0,z$ | (14) $m\ 0,y,z$ | (15) $m\ x,x,z$                 | (16) $m\ x,x,z$                 |

Maximal non-isomorphic subgroups

- I** [2] $P422$  1; 2; 3; 4; 5; 6; 7; 8  
 [2] $P4/m\ 11(P4/m)$  1; 2; 3; 4; 9; 10; 11; 12  
 [2] $P4mm$  1; 2; 3; 4; 13; 14; 15; 16  
 [2] $P\bar{4}2m$  1; 2; 5; 6; 11; 12; 15; 16  
 [2] $P\bar{4}m2$  1; 2; 7; 8; 11; 12; 13; 14  
 [2] $P2/m\ 2/m\ 1(Pmmm)$  1; 2; 5; 6; 9; 10; 13; 14  
 [2] $P2/m\ 12/m(Cmmm)$  1; 2; 7; 8; 9; 10; 15; 16
- IIa** none
- IIb** [2] $P4/mcc(c'=2c)$ ; [2] $P4_2/mcc(c'=2c)$ ; [2] $P4_2/mcm(c'=2c)$ ; [2] $C4/mc\ d(a'=2a, b'=2b)(P4/bbw)$ ;  
 [2] $C4/mcd(a'=2a, b'=2b)(P4/bbm)$ ; [2] $C4/mww(a'=2a, b'=2b)(P4/nmm)$ ;  
 [2] $F4/mww(a'=2a, b'=2b, c'=2c)(I4/mmm)$ ; [2] $F4/mmc(a'=2a, b'=2b, c'=2c)(I4/mcm)$

Maximal isomorphic subgroups of lowest index

- IIc** [2] $P4/mmm(c'=2c)$ ; [2] $C4/mww(a'=2a, b'=2b)(P4/mmm)$

Minimal non-isomorphic supergroups

- I** [3] $Pm\ \bar{3}m$   
**II** [2] $I4/mmm$

CONTINUED

No. 123

$P4/mmm$

Generators selected (1);  $r(1,0,0)$ ;  $r(0,1,0)$ ;  $r(0,0,1)$ ; (2); (3); (5); (9)

Positions

Multiplicity,  
Wyckoff letter,  
Site symmetry

Coordinates

Reflection conditions

16	$a\ 1$	(1) $x,y,z$	(2) $\bar{x},\bar{y},z$	(3) $\bar{y},x,z$	(4) $y,\bar{x},z$
		(5) $x,y,\bar{z}$	(6) $x,\bar{y},\bar{z}$	(7) $y,x,\bar{z}$	(8) $\bar{y},\bar{x},\bar{z}$
		(9) $\bar{x},\bar{y},z$	(10) $x,y,\bar{z}$	(11) $y,\bar{x},z$	(12) $\bar{y},x,\bar{z}$
		(13) $x,\bar{y},z$	(14) $\bar{x},y,z$	(15) $\bar{y},\bar{x},z$	(16) $y,x,z$

General:

no conditions

Special:

8	$i\ .m$	$x,\bar{z}$	$\bar{x},\bar{z}$	$\bar{y},z$	$\bar{y},z$
		$\bar{x},z$	$x,\bar{z}$	$\bar{y},z$	$\bar{y},z$

no extra conditions

8	$s\ .m$	$x,0,z$	$\bar{x},0,z$	$0,x,z$	$0,\bar{x},z$
		$\bar{x},z$	$x,0,z$	$0,x,z$	$0,\bar{x},z$

no extra conditions

8	$r\ .m$	$x,x,z$	$\bar{x},\bar{x},z$	$\bar{x},z$	$x,\bar{x},z$
		$\bar{x},z$	$x,\bar{x},z$	$\bar{x},z$	$x,\bar{x},z$

no extra conditions

8	$q\ m..$	$x,y,\bar{z}$	$\bar{x},\bar{y},\bar{z}$	$\bar{y},x,\bar{z}$	$y,\bar{x},\bar{z}$
		$\bar{y},\bar{z}$	$x,\bar{y},\bar{z}$	$y,x,\bar{z}$	$\bar{y},\bar{z}$

no extra conditions

8	$p\ m..$	$x,y,0$	$\bar{x},\bar{y},0$	$\bar{y},x,0$	$y,\bar{x},0$
		$\bar{y},0$	$x,\bar{y},0$	$y,x,0$	$\bar{y},0$

no extra conditions

4	$o\ m2m$	$x,\bar{z}$	$\bar{x},\bar{z}$	$\bar{y},z$	$\bar{y},z$
---	----------	-------------	-------------------	-------------	-------------

no extra conditions

4	$n\ m2m$	$x,\bar{z}$	$\bar{x},\bar{z}$	$\bar{y},z$	$\bar{y},z$
---	----------	-------------	-------------------	-------------	-------------

no extra conditions

4	$m\ m2m$	$x,0,z$	$\bar{x},0,z$	$0,x,z$	$0,\bar{x},z$
---	----------	---------	---------------	---------	---------------

no extra conditions

4	$l\ m2m$	$x,0,0$	$\bar{x},0,0$	$0,x,0$	$0,\bar{x},0$
---	----------	---------	---------------	---------	---------------

no extra conditions

4	$k\ m.2m$	$x,x,\bar{z}$	$\bar{x},\bar{x},\bar{z}$	$\bar{x},z$	$x,\bar{x},z$
---	-----------	---------------	---------------------------	-------------	---------------

no extra conditions

4	$j\ m.2m$	$x,x,0$	$\bar{x},\bar{x},0$	$\bar{x},z$	$x,\bar{x},z$
---	-----------	---------	---------------------	-------------	---------------

no extra conditions

4	$i\ 2mw$	$0,\bar{z}$	$\bar{0},z$	$0,\bar{z}$	$\bar{0},z$
---	----------	-------------	-------------	-------------	-------------

$hk\ell: h+k=2n$

2	$h\ 4mw$	$\bar{z}$	$\bar{z}$	$\bar{z}$	$\bar{z}$
---	----------	-----------	-----------	-----------	-----------

no extra conditions

2	$g\ 4mw$	$0,z$	$0,z$	$z$	$z$
---	----------	-------	-------	-----	-----

no extra conditions

2	$f\ mmm$	$0,\bar{z}$	$\bar{0},z$	$z$	$z$
---	----------	-------------	-------------	-----	-----

$hk\ell: h+k=2e$

2	$e\ mmm$	$0,\bar{z}$	$\bar{0},z$	$z$	$z$
---	----------	-------------	-------------	-----	-----

$hk\ell: h+k=2e$

1	$d\ 4/mmm$	$\bar{z}$	$\bar{z}$	$z$	$z$
---	------------	-----------	-----------	-----	-----

no extra conditions

1	$c\ 4/mww$	$\bar{z}$	$\bar{z}$	$z$	$z$
---	------------	-----------	-----------	-----	-----

no extra conditions

1	$b\ 4/mmm$	$0,\bar{z}$	$\bar{0},z$	$z$	$z$
---	------------	-------------	-------------	-----	-----

no extra conditions

1	$a\ 4/mmm$	$0,0,\bar{z}$	$\bar{0},z$	$z$	$z$
---	------------	---------------	-------------	-----	-----

no extra conditions

Symmetry of special projections

Along [001]  $p4mm$

$a'=a\ b'=b$

Origin at  $0,0,z$

Along [100]  $p2mm$

$a'=b\ b'=c$

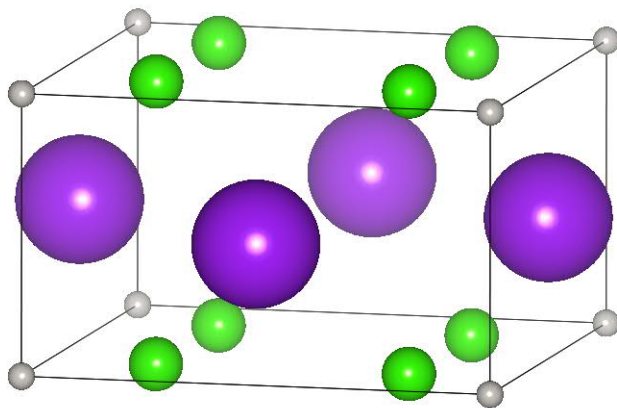
Origin at  $x,0,0$

Along [110]  $p2mm$

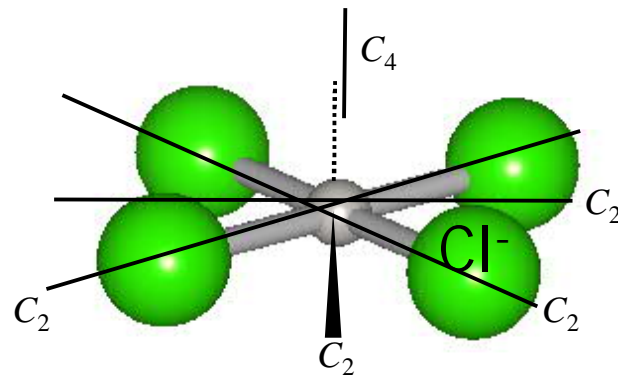
$a'=b(-a+b)\ b'=c$

Origin at  $x,x,0$

(Continued on preceding page)



**Pt: 1 atom in unit cell**  
**K: 2 atoms in unit cell**  
**Cl: 4 atoms in unit cell**



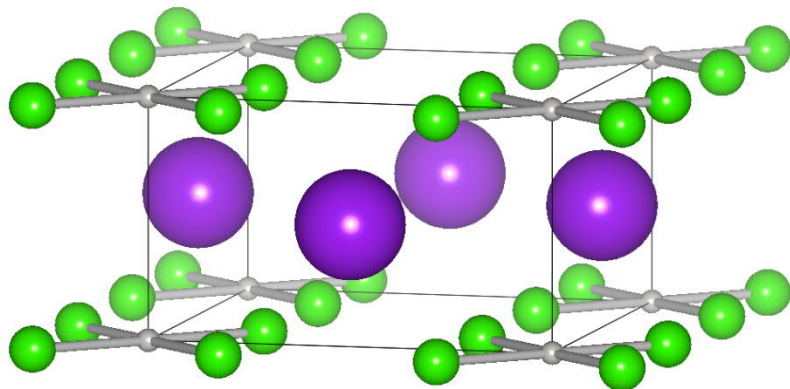
Site symmetry of Pt:  $D_{4h}$

**Bond lengths:**

**Pt-Pt:  $(1-0)^2 \cdot 4.15\text{\AA}$**

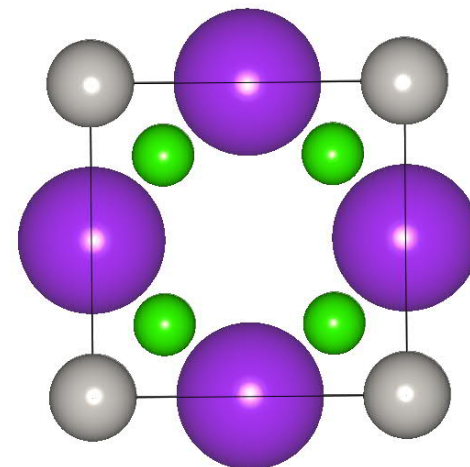
**Pt-K:  $\sqrt{[(0.5-0)^2 \cdot 7.023\text{\AA} + (0.5-0)^2 \cdot 4.149\text{\AA}]} = 4.08\text{\AA}$**

**Pt-Cl:  $\sqrt{[(0.232-0)^2 \cdot 7.023\text{\AA} + (0.232-0)^2 \cdot 7.023\text{\AA}]} = 2.30\text{\AA}$**



*ab*-projectio

(seen from *c*-direction)



## **K<sub>2</sub>PtCl<sub>4</sub>**

- $\rho = 3.37 \times 10^6 \text{ g/m}^3$
- $V = 7.023 \text{ \AA} \times 7.023 \text{ \AA} \times 4.1486 \text{ \AA} = 204.62 \times 10^{-30} \text{ m}^3$
- $M = (2 \times 39.098 + 195.22 + 4 \times 35.453) \text{ g/mol} = 415.228 \text{ g/mol}$
- $Z = (V \times \rho \times N_A) / M = 1$
- Distances:
  - Pt-Pt: 4.15 Å
  - Pt-K: 4.08 Å
  - Pt-Cl: 2.31 Å (→ chemical bond)
- CN(Pt) = 4
- Pl site symmetry:  $D_{4h}$

# EXAMPLE: Chromium oxychloride CrOCl

Space group *Pmmn* (No. 59)

Lattice parameters:  $a = 3.88 \text{ \AA}$ ,  $b = 3.20 \text{ \AA}$ ,  $c = 7.72 \text{ \AA}$  ( $Z = 2$ )

Atomic positions:	Cr	$2a$	$z = 0.109$
	Cl	$2b$	$z = 0.327$
	O	$2b$	$z = 0.960$

(a) Draw the unit cell.

(b) Give for chromium:

- bond lengths
- coordination numbers
- site symmetry

(c) Calculate BVS for chromium.

[ $R^0$  values: Cr<sup>III</sup>-O<sup>II</sup>: 1.724, Cr<sup>III</sup>-Cl<sup>I</sup>: 2.08]



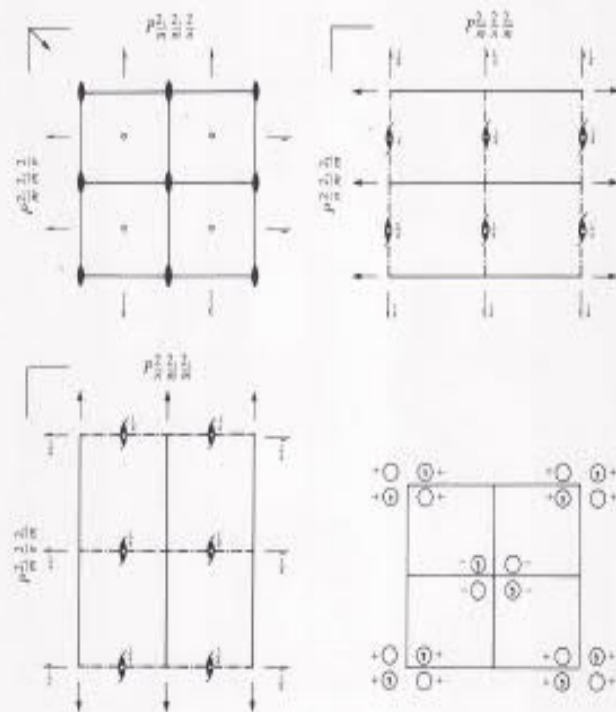
$Pmmn$  $D_{2h}^{13}$  $mmm$ 

Orthorhombic

No. 59

 $P2_1/m 2_1/m 2/n$ Patterson symmetry  $Pmmm$ 

ORIGIN CHOICE 1

Origin at  $mb/2$ , at  $\frac{1}{2}, \frac{1}{2}, 0$  from  $\bar{1}$ Asymmetric unit  $0 \leq x \leq \frac{1}{2}; 0 \leq y \leq \frac{1}{2}; 0 \leq z \leq \frac{1}{2}$ 

Symmetry operations

- |                                   |                                                |                                              |                                                |
|-----------------------------------|------------------------------------------------|----------------------------------------------|------------------------------------------------|
| (1) $\bar{1}$                     | (2) $2(0,0,z)$                                 | (3) $2(0,\frac{1}{2},0)$ $\frac{1}{2}, y, 0$ | (4) $2(\frac{1}{2}, 0, 0)$ $x, \frac{1}{2}, 0$ |
| (5) $\bar{1}$ $1, \frac{1}{2}, 0$ | (6) $m(\frac{1}{2}, \frac{1}{2}, 0)$ $x, y, 0$ | (7) $m(x, 0, z)$                             | (8) $m(0, y, z)$                               |

CONTINUED

No. 59

 $Pmmn$ Generators selected (1);  $t(1,0,0)$ ;  $t(0,1,0)$ ;  $t(0,0,1)$ ; (2); (3); (5)

Positions

Multiplicity,  
Wyckoff letter,  
site symmetry

Coordinates

Reflection conditions

- |                 |                                             |                                       |                                       |                                             |
|-----------------|---------------------------------------------|---------------------------------------|---------------------------------------|---------------------------------------------|
| 8 $g$ $\bar{1}$ | (1) $x, y, z$                               | (2) $\bar{x}, \bar{y}, z$             | (3) $x+\frac{1}{2}, y+\frac{1}{2}, z$ | (4) $x+\frac{1}{2}, \bar{y}+\frac{1}{2}, z$ |
|                 | (5) $x+\frac{1}{2}, y+\frac{1}{2}, \bar{z}$ | (6) $x+\frac{1}{2}, y+\frac{1}{2}, z$ | (7) $x, y, z$                         | (8) $\bar{x}, y, z$                         |

General:

- $hk0: h+k=2n$   
 $h00: h=2n$   
 $0k0: k=2n$

Special: as above, plus

- 4  $f$   $m$ :  $x, 0, z$   $\bar{x}, 0, z$   $x+\frac{1}{2}, \frac{1}{2}, z$   $x+\frac{1}{2}, \bar{\frac{1}{2}}, z$

- 4  $e$   $m$ ..:  $0, y, z$   $0, \bar{y}, z$   $\frac{1}{2}, y+\frac{1}{2}, z$   $\frac{1}{2}, \bar{y}+\frac{1}{2}, z$

- 4  $d$   $\bar{1}$ :  $\frac{1}{2}, \frac{1}{2}, \frac{1}{2}$   $\frac{1}{2}, \frac{1}{2}, \bar{\frac{1}{2}}$   $\frac{1}{2}, \frac{1}{2}, \frac{1}{2}$   $\frac{1}{2}, \frac{1}{2}, \bar{\frac{1}{2}}$

- 4  $c$   $\bar{1}$ :  $\frac{1}{2}, \frac{1}{2}, 0$   $\frac{1}{2}, \frac{1}{2}, 0$   $\frac{1}{2}, \frac{1}{2}, 0$   $\frac{1}{2}, \frac{1}{2}, 0$

- 2  $b$   $mm2$   $0, \frac{1}{2}, z$   $\frac{1}{2}, 0, z$

- 2  $a$   $mm2$   $0, 0, z$   $\frac{1}{2}, \frac{1}{2}, z$

no extra conditions

no extra conditions

 $hkl: h, k=2n$  $hkl: h, k=2n$ 

no extra conditions

no extra conditions

Symmetry of special projections

Along [001]  $c2mm$  $a'=a$   $b'=b$ Origin at  $0, 0, z$ Along [100]  $p2mg$  $a'=b$   $b'=c$ Origin at  $x, \frac{1}{2}, 0$ Along [010]  $p2gm$  $a'=c$   $b'=a$ Origin at  $\frac{1}{2}, y, 0$ 

Maximal non-isomorphic subgroups

- I [2] $P2_12_12_1$  1; 2; 3; 4  
 [2] $P112/n(P2/c)$  1; 2; 5; 6  
 [2] $P12/m1(P2/m)$  1; 3; 5; 7  
 [2] $P2/m11(P2/m)$  1; 4; 5; 8  
 [2] $Pmm2$  1; 2; 7; 8  
 [2] $Pm2_1(n(Pmn2_1))$  1; 3; 6; 8  
 [2] $P2_1mn(Pmn2_1)$  1; 4; 6; 7

IIa none

- IIb [2] $Pcmn(c'=2c)(Pnma)$ ; [2] $Pmcn(c'=2c)(Pnma)$ ; [2] $Pccn(c'=2c)$

Maximal isomorphic subgroups of lowest index

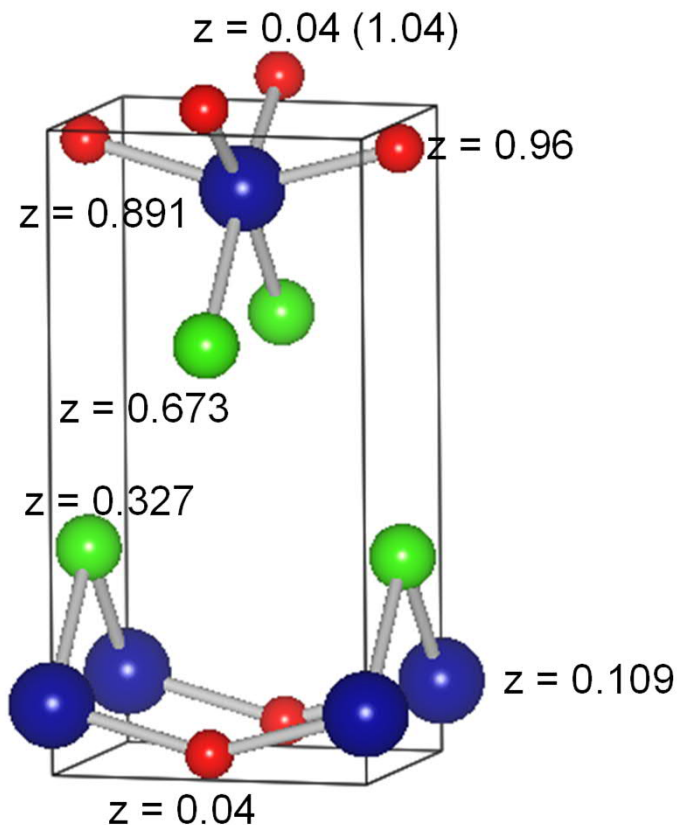
- IIc [3] $Pmmn(a'=3a \text{ or } b'=3b)$ ; [2] $Pmmn(c'=2c)$

Minimal non-isomorphic supergroups

I [2] $P4/nmm$ ; [2] $P4/mnc$ 

- II [2] $Ammm(Cmcm)$ ; [2] $Bmmb(Cmcm)$ ; [2] $Cmmm$ ; [2] $Immm$ ; [2] $Pmmb(2a'=a)(Pnma)$ ; [2] $Pnna(2b'=b)$

# CrOCl



## Chromium bonding

$$2 \times \text{Cr-Cl: } \sqrt{\{(0.891-0.673) \times 7.72 \text{ \AA}\}^2 + \{0.5 \times 3.20 \text{ \AA}\}^2} = 2.3222 \text{ \AA}$$

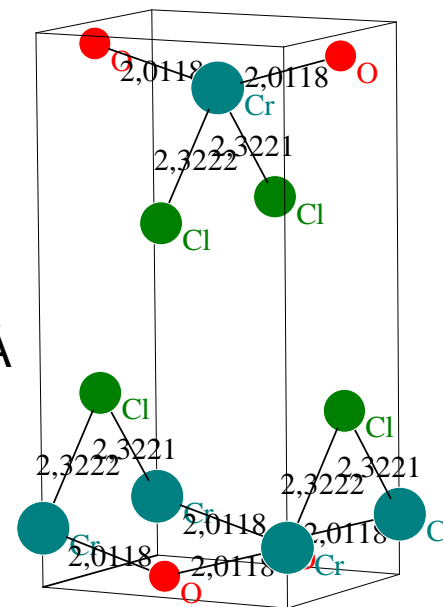
$$2 \times \text{Cr-O: } \sqrt{\{(0.960-0.891) \times 7.72 \text{ \AA}\}^2 + \{0.5 \times 3.88 \text{ \AA}\}^2} = 2.0118 \text{ \AA}$$

$$2 \times \text{Cr-O: } \sqrt{\{[(1-0.891)+0.04] \times 7.72 \text{ \AA}\}^2 + \{0.5 \times 3.20 \text{ \AA}\}^2} = 1.9706 \text{ \AA}$$

$$\text{CN}(\text{Cr}) = 6$$

Cr site symmetry:  $C_{2v}$

$$\text{BVS}(\text{Cr}): +2.985$$



# CrOCl: simulated XRD pattern based on the structure data

