

CHEM-E4109

MODERN METHODS IN **BIOCATALYSIS**

chapter #3: oxygenative enzymes

8.3.2022

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Jan Deska
Bioorganic
Chemistry

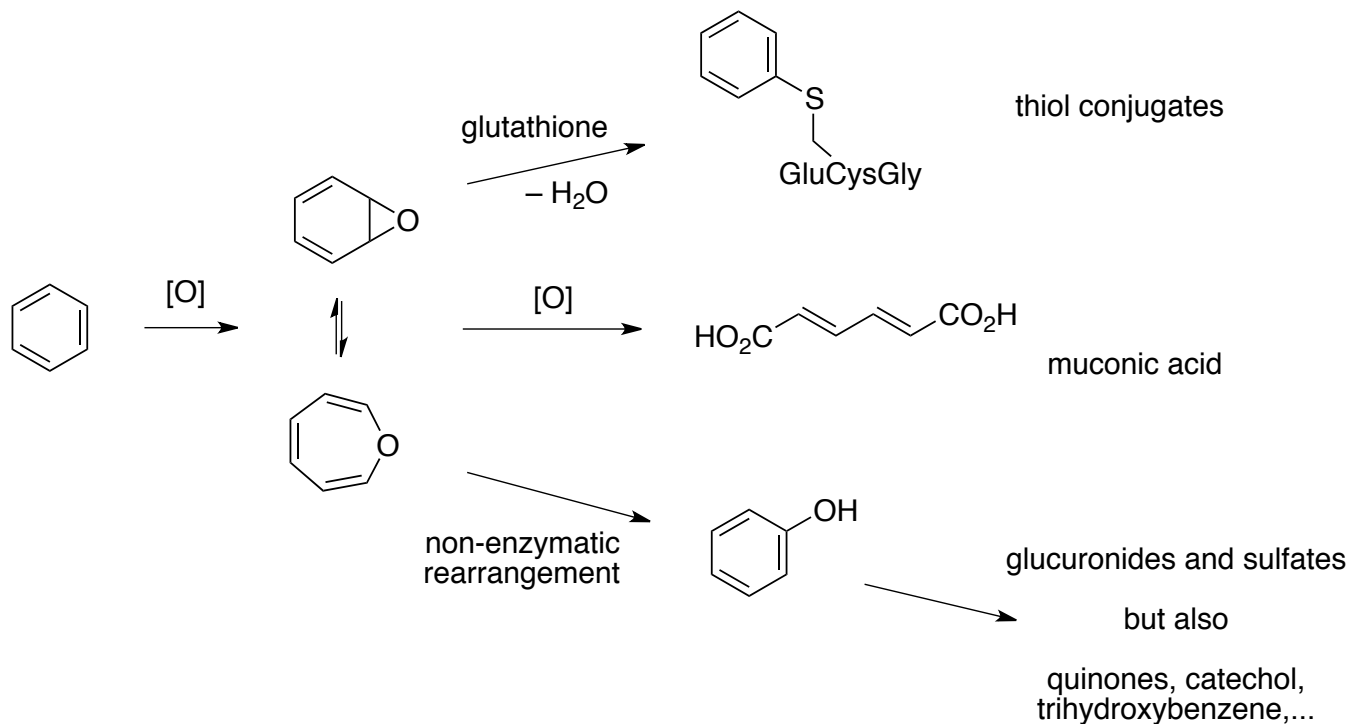
TODAY'S MENU

Content

- mono- & dioxygenases
- mechanistic comparison between iron-centred oxidizing enzymes and metal-free riboflavin-dependent biocatalysts
- blueprint for both transition metal-mediated and organocatalytic oxidations

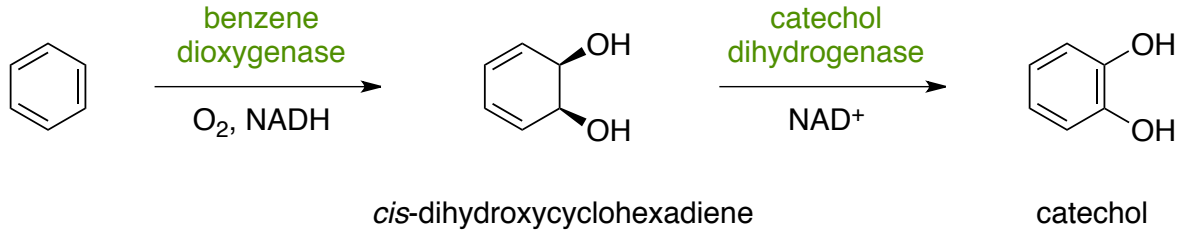
Arene dioxygenases

aromatic hydrocarbons like benzene are metabolized by mammals via initial oxygenation



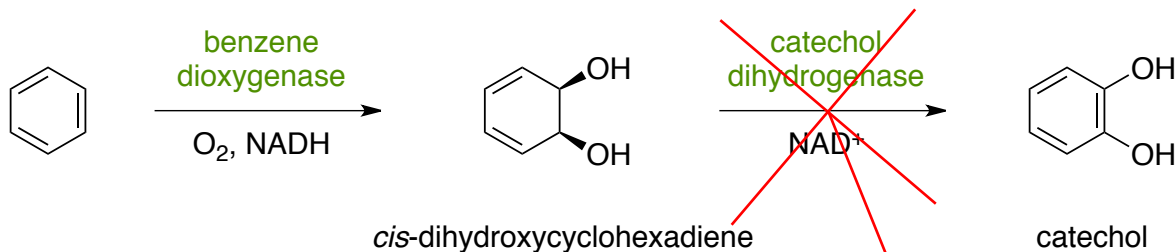
Arene dioxygenases

certain strains from *Pseudomonas putida* accumulate cis-dihydroxycyclohexadienes



Arene dioxygenases

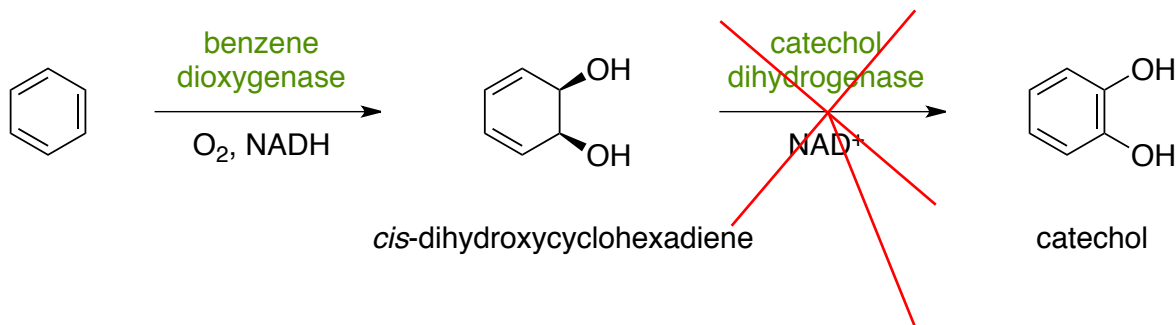
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P. putida 39/D and F1 mutants grow on toluene as carbon source but suffer from a somewhat sluggish dehydrogenase

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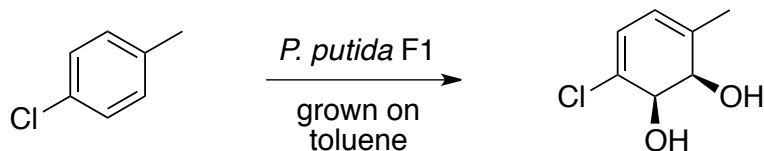


cis-dihydroxycyclohexadiene

catechol

P. putida 39/D and F1 mutants grow on toluene as carbon source but suffer from a somewhat sluggish dehydrogenase

- "biosynthesis" of complex organic building blocks

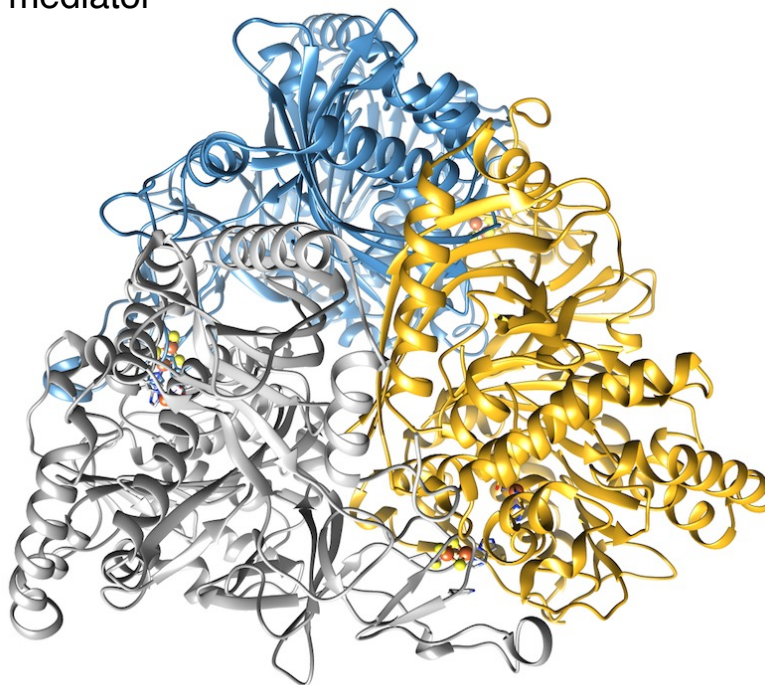


diastereoselectivity: pure *cis*
enantioselectivity: pure (+)

Arene dioxygenases

Naphthalene dioxygenase (*Pseudomonas* sp. NCIB 9816-4)

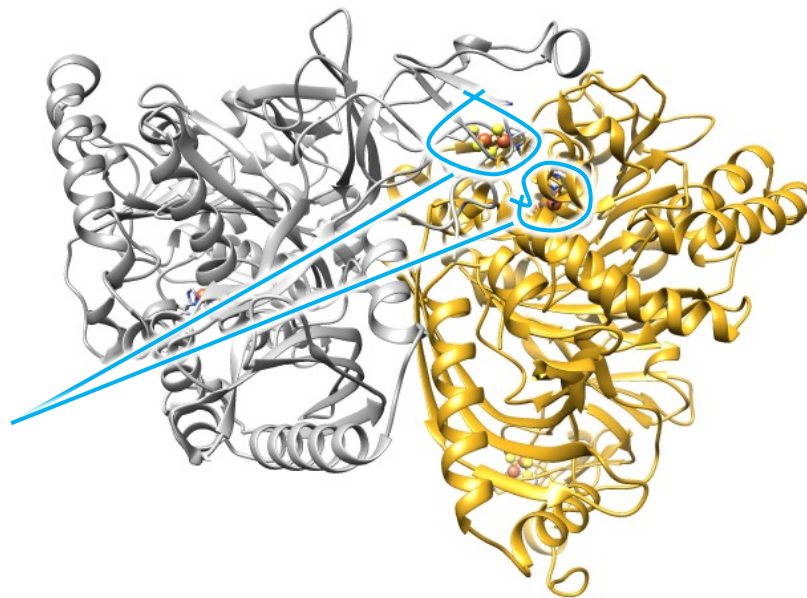
- hetero-hexamer, 210 kDa, $\alpha_3\beta_3$
- 3 identical iron-binding α -subunits, two different iron species
- Rieske iron-sulfur cluster as electron mediator
- non-heme mononuclear iron as oxygen binding center
- requires additional reductase
Fe-S-cluster + FAD
(not shown)
- NADH as terminal reductant



Arene dioxygenases

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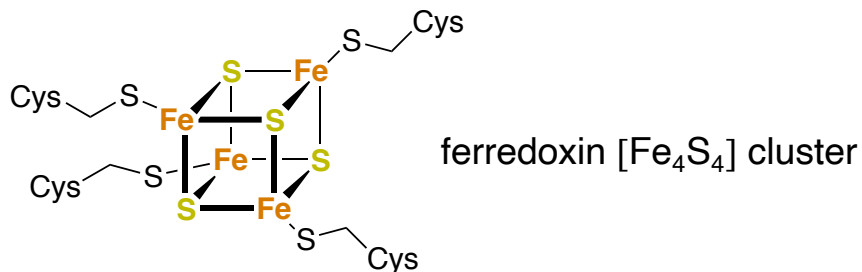
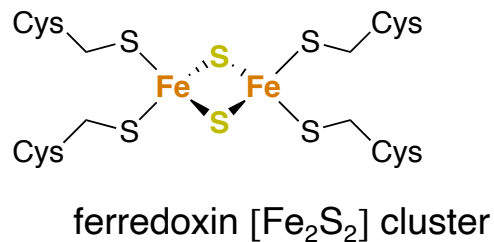
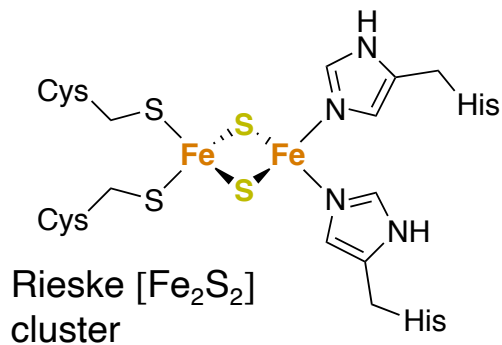
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Fe-S-cluster + FAD
(not shown)
- NADH as terminal reductant
- Rieske cluster from one α -subunit
teams up with mono-nuclear iron
from another α -subunits



Arene dioxygenases

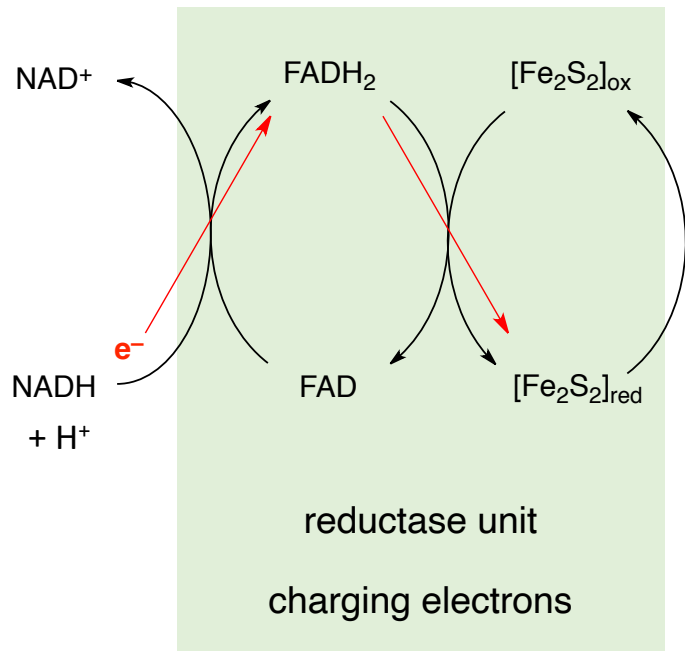
"biological capacitors"

- electron transfer often proceeds via cascade processes involving organic mediators (FAD/FMN) or metallic clusters (Rieske, ferredoxin); or both



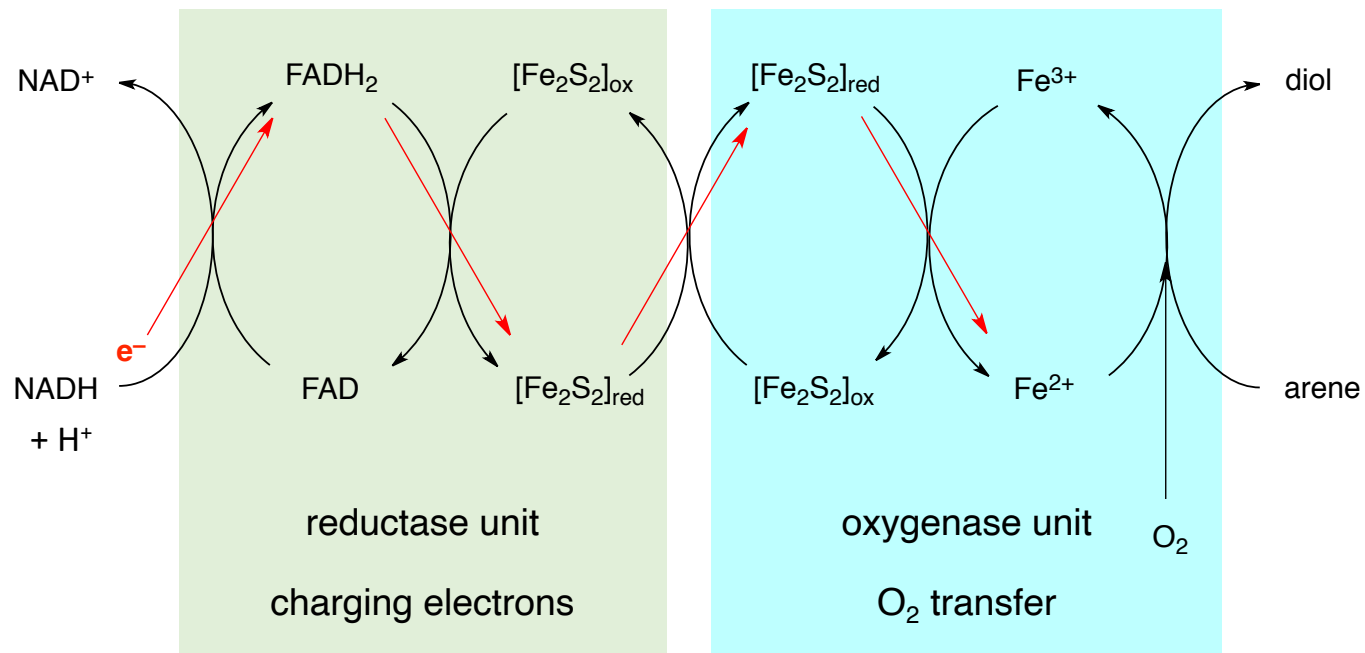
Arene dioxygenases

schematic electron flow in a naphthalene dioxygenation



Arene dioxygenases

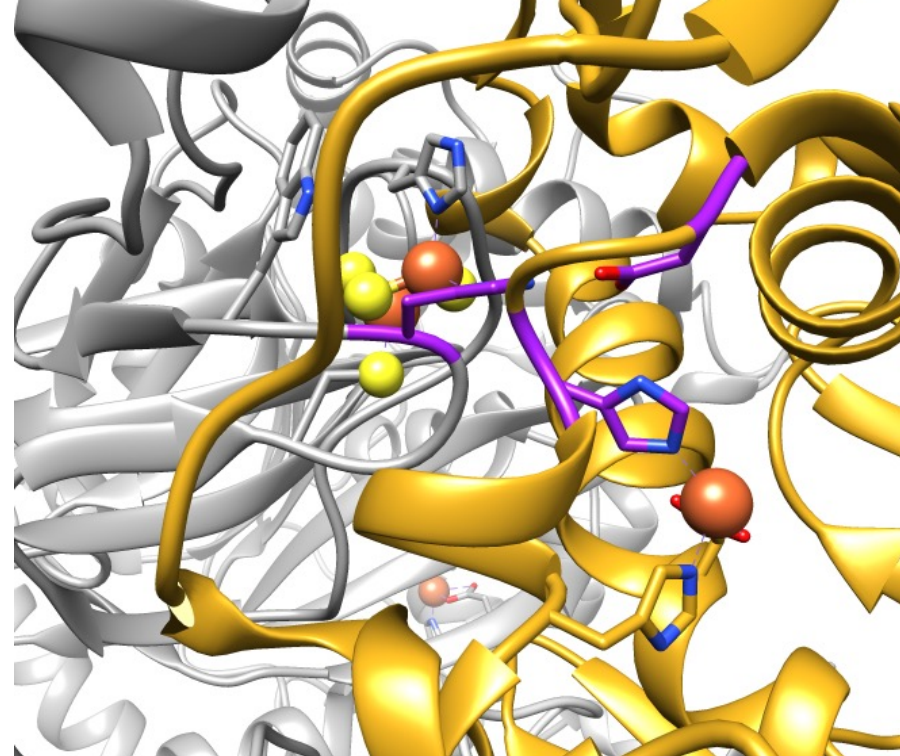
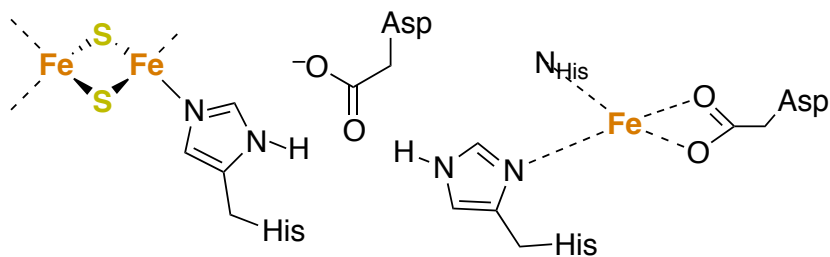
schematic electron flow in a naphthalene dioxygenation



Arene dioxygenases

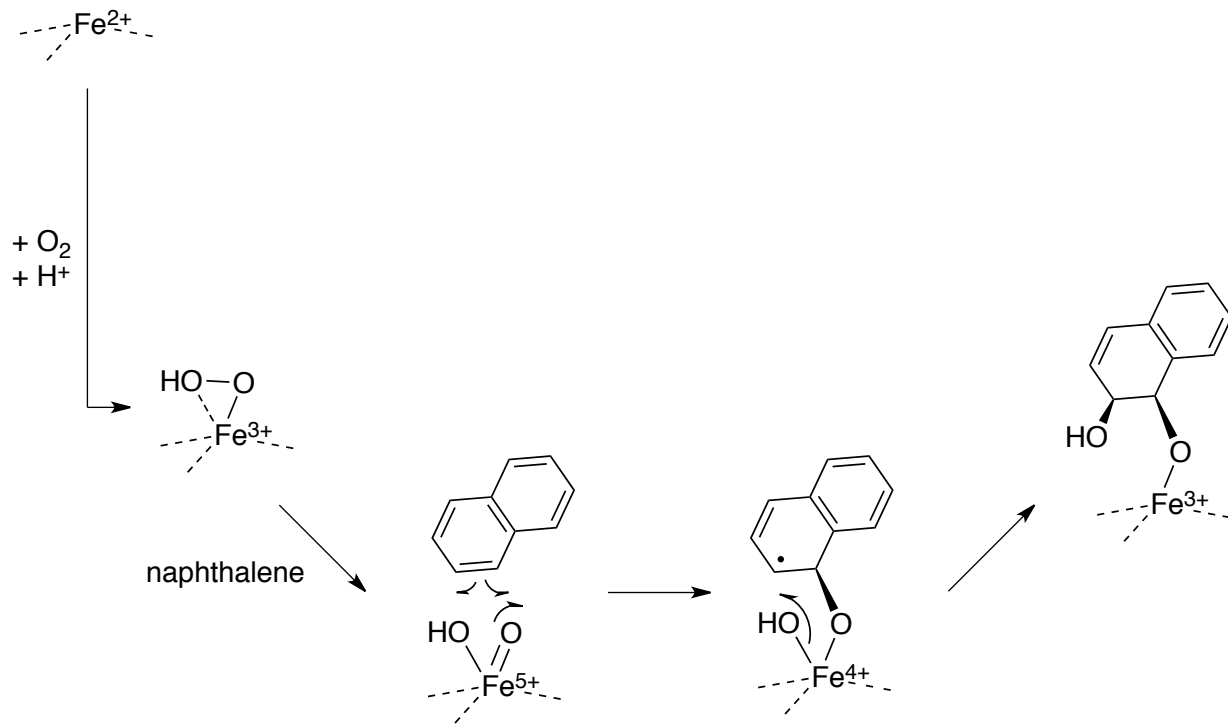
Mechanism

- charged $[\text{Fe}_2\text{S}_2]$ cluster transfers electron onto mono-nuclear iron
- communication between both centres (and both subunits!) via His-Asp-His alignment (purple)
- reduced iron centre recombines with O_2



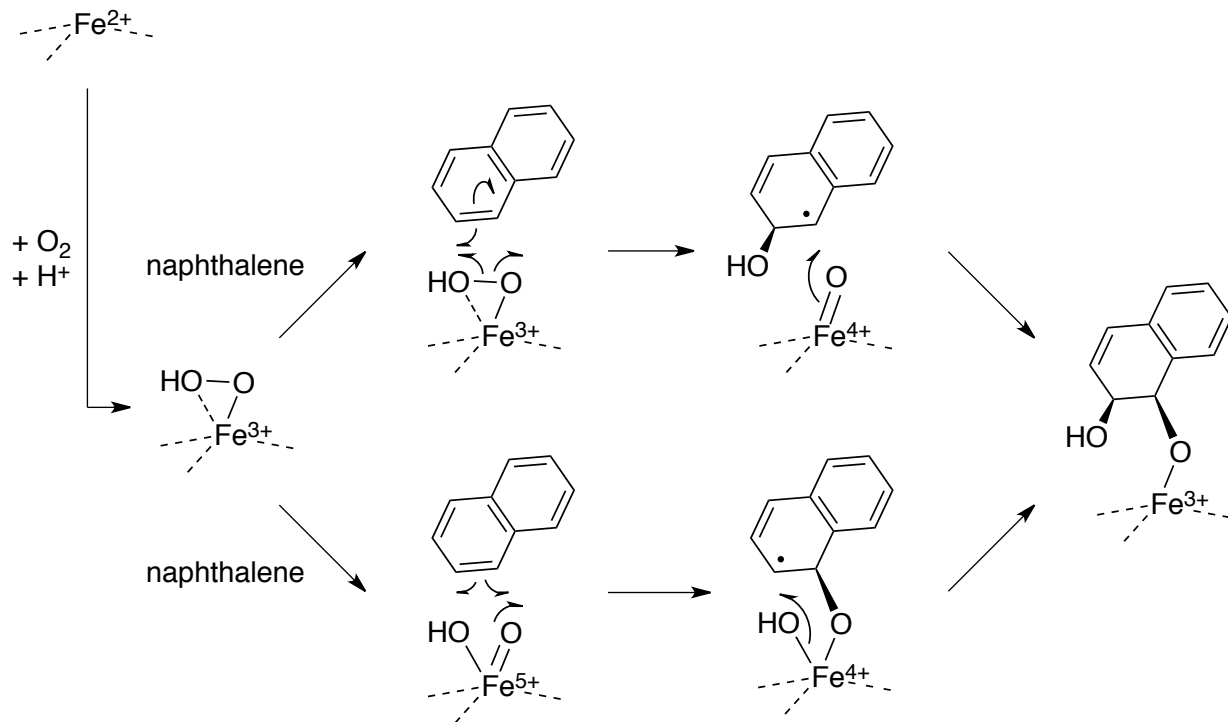
Arene dioxygenases

Mechanism: two proposed pathways for oxygen transfer



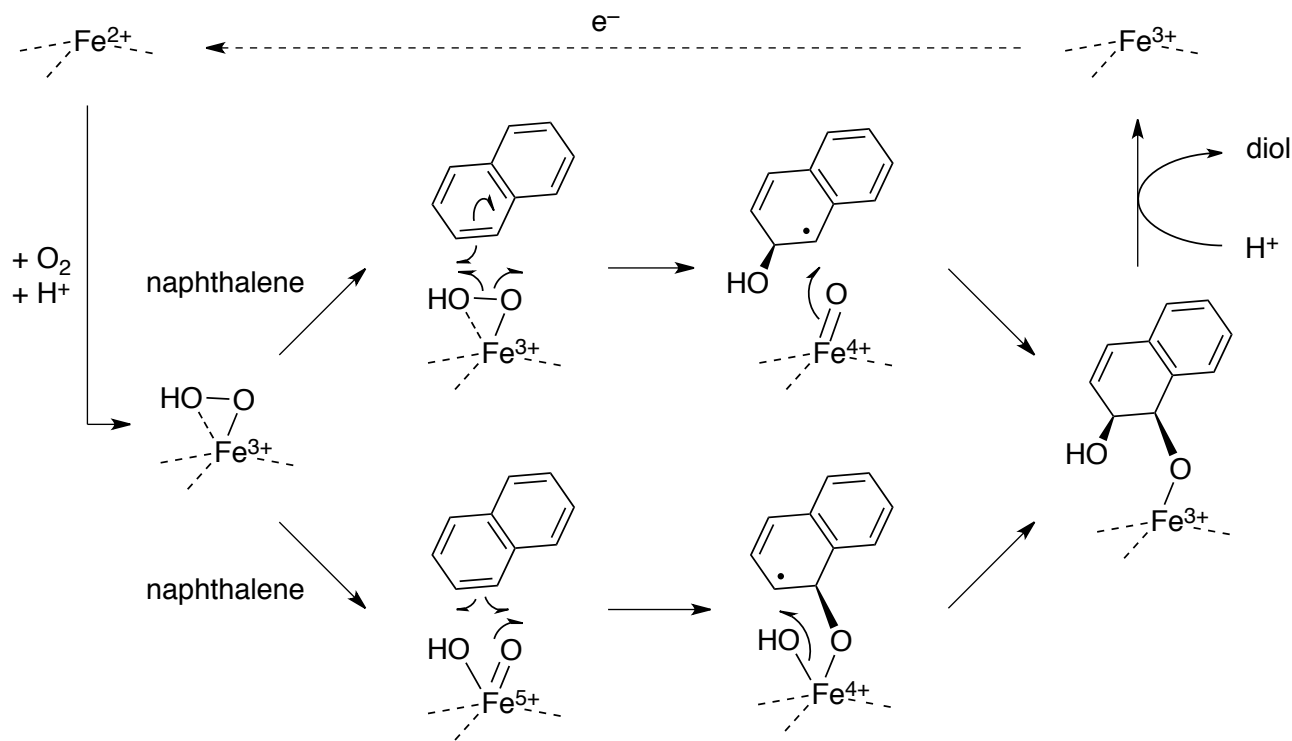
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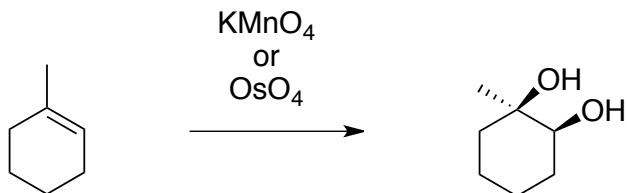
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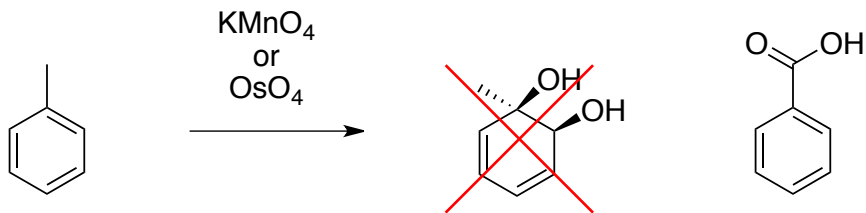


Synthetic use of arene dioxygenases

- nowadays various arene DOs with broad substrate scope known and working (benzene DO, toluene DO, naphthalene DO)
- particular knock-out mutants allow for selective production of cis-diols without catechol side products (biotech. removal of catechol dehydrogenase)
- absolutely no equivalent chemical method known!



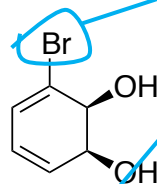
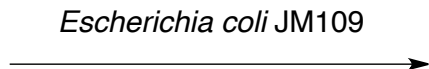
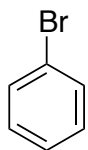
but



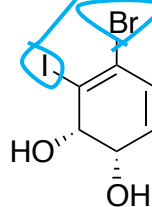
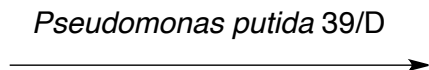
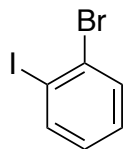
Synthetic use of arene dioxygenases

fermentative production of basic synthetic building blocks

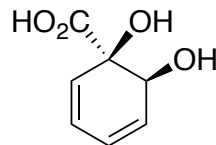
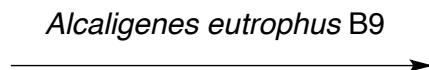
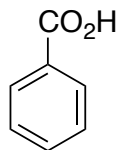
excellent for
follow-up
chemistry



99% ee
99% cis



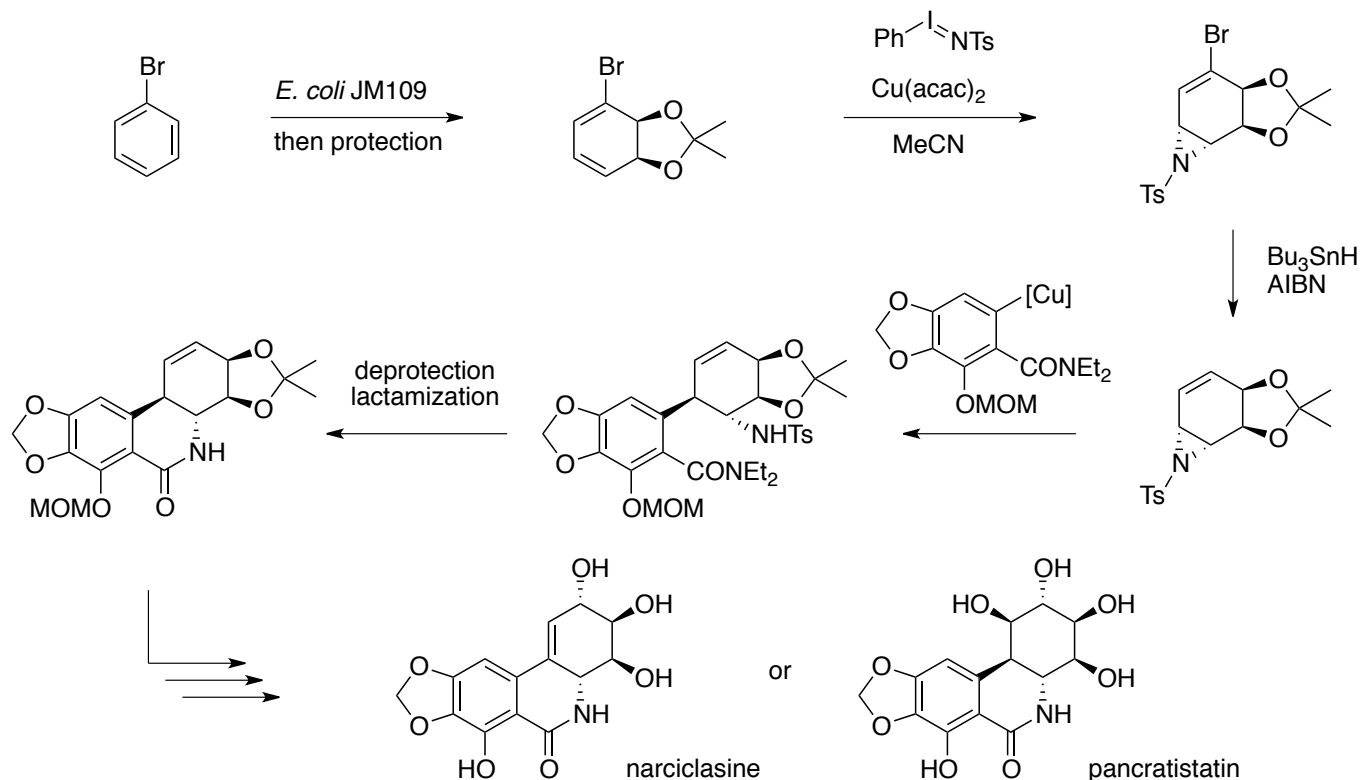
99% ee
99% cis



occasionally also
1,2-selectivity

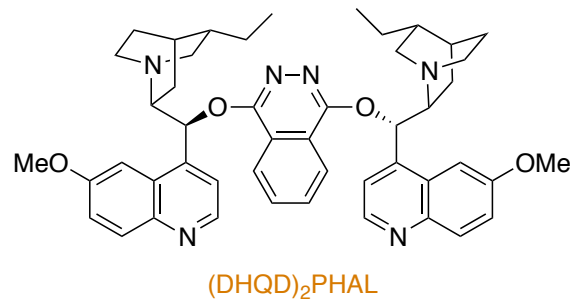
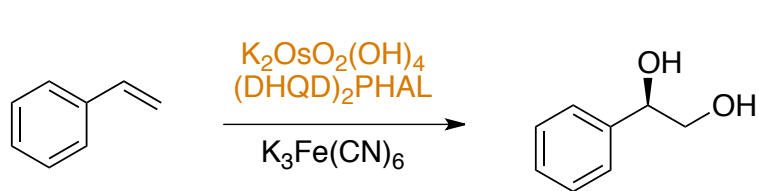
Synthetic use of arene dioxygenases

Synthesis of antitumor agents (pancratistatin family), Hudlicky 1996

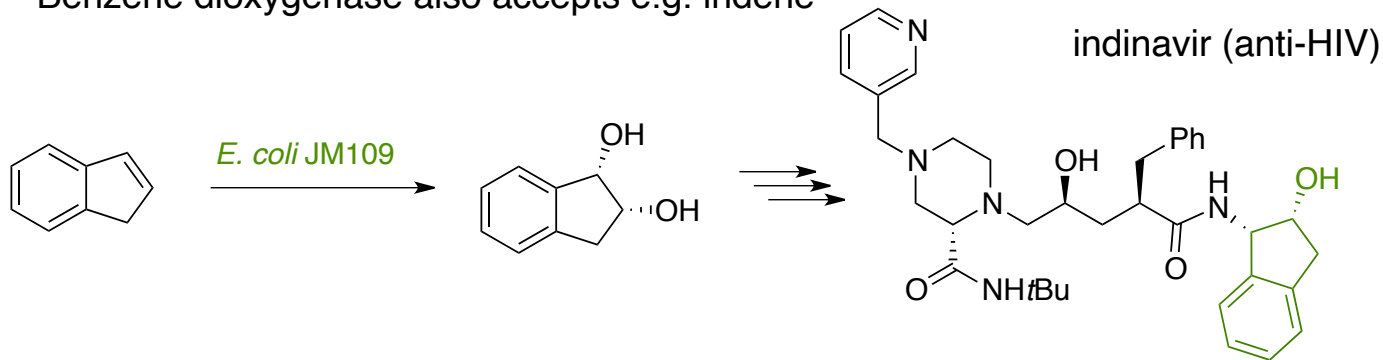


Activity on non-arene substrates

Chemical asymmetric *cis*-dihydroxylation (Sharpless dihydroxylation)



Benzene dioxygenase also accepts e.g. indene



Monooxygenases

flavin dependent monooxygenases

- purely "organocatalytic" activation via flavin-hydroperoxide
- catalyzes oxygen-transfer processes such as sulfoxidation and N-oxygenation but also Baeyer-Villiger reactions and epoxidations of activated olefins
- NAD(P)H-dependent, without co-enzyme (reductase) requirement

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heme-dependent monooxygenases

- iron porphyrine as terminal reactive species
- catalyzes all kinds of oxygen-transfer processes but also radical initiations, dealkylations, etc.
- NAD(P)H-dependent
- electron transfer requires often complex cascades involving flavin and ferredoxin-containing domains and additional reductases

Baeyer-Villiger monooxygenases

Baeyer-Villiger reaction

- formation of esters or lactones from acyclic or cyclic ketones, respectively
- most common oxidant: peracids
- reaction proceeds via acylperoxy adduct



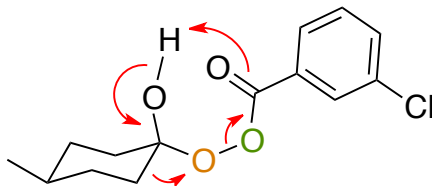
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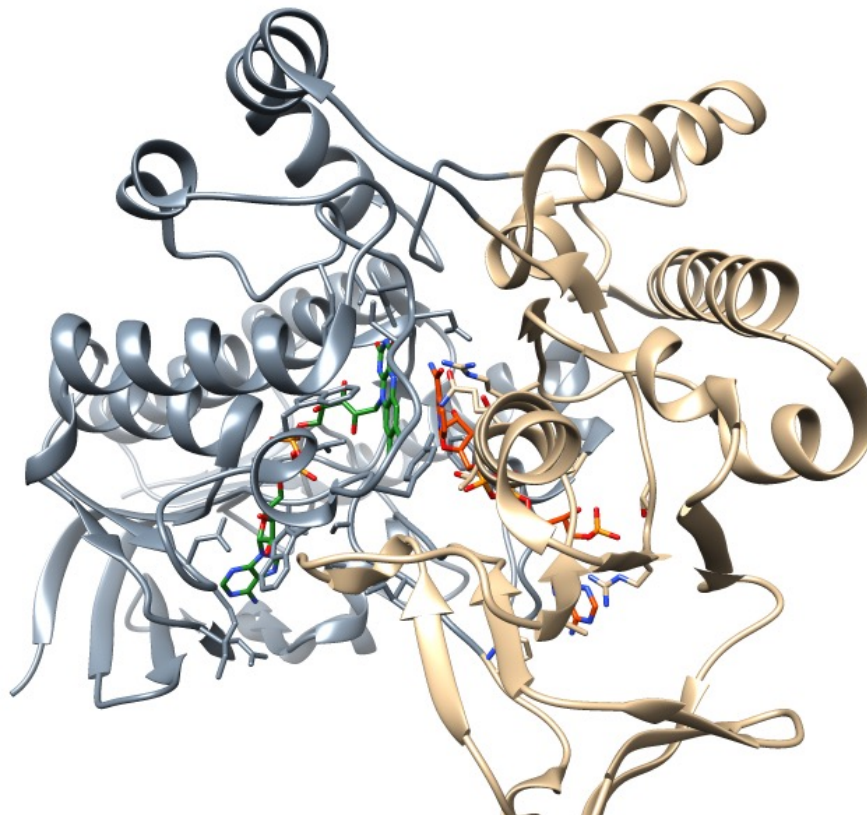
- Criegee rearrangement as key step



Baeyer–Villiger monooxygenases

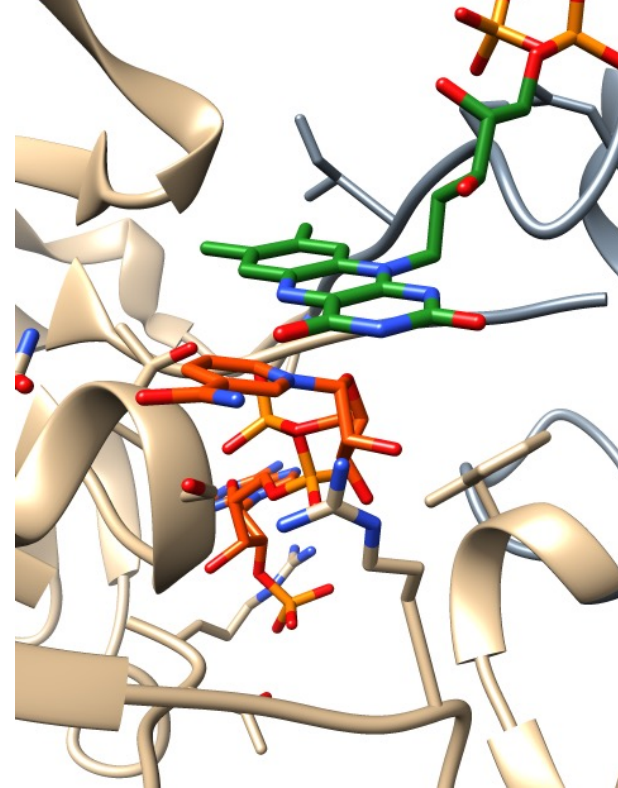
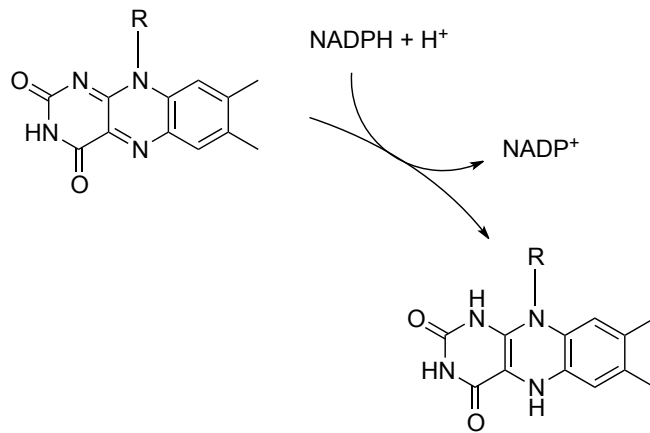
Cyclohexanone monooxygenase (*Acinetobacter* sp. NCIB 9871)

- monomer, 59 kDa
- large FAD-binding domain
- smaller NADPH-binding domain
- flavin and nicotinamide stacked close to each other
- but quite dynamic with large domain movements



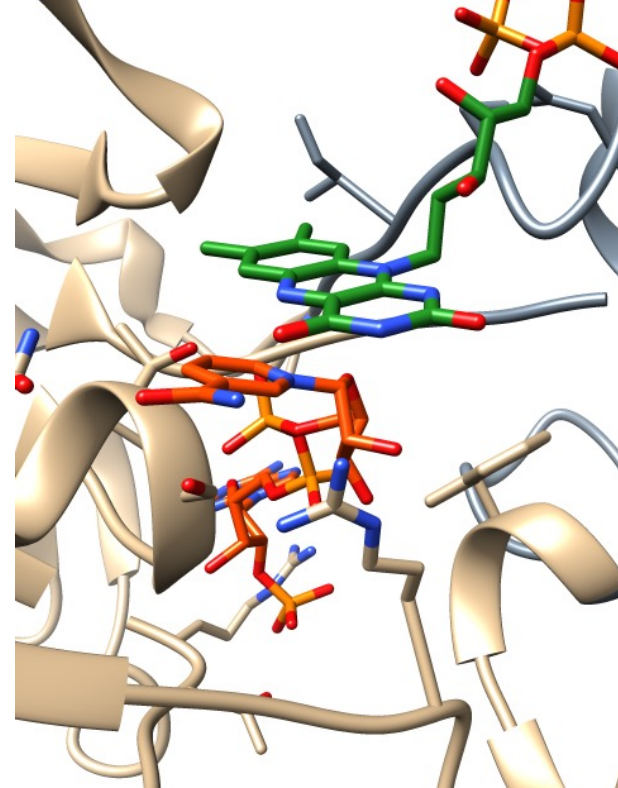
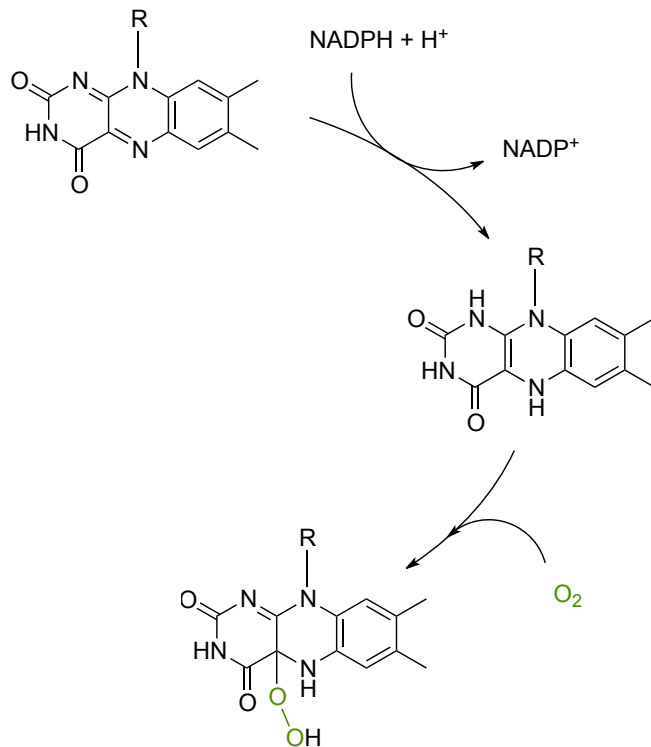
Baeyer-Villiger monooxygenases

Mechanism



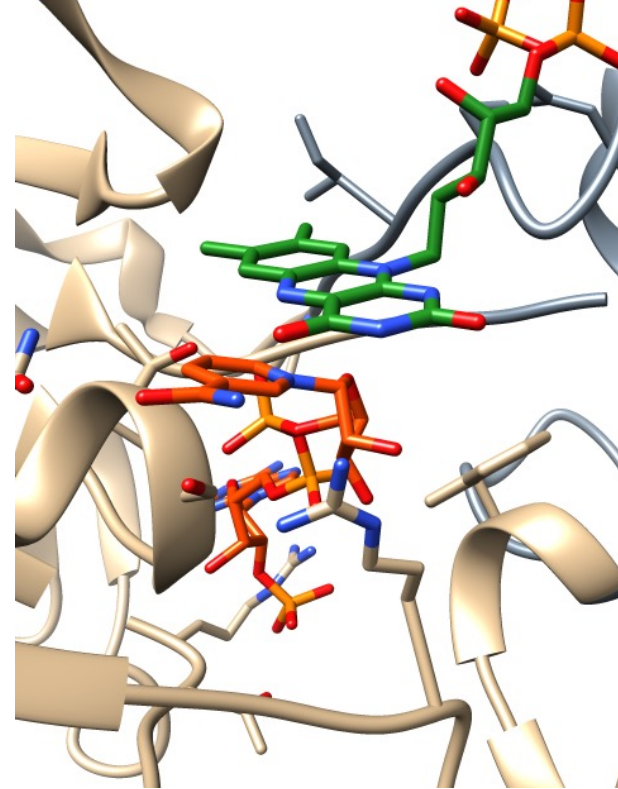
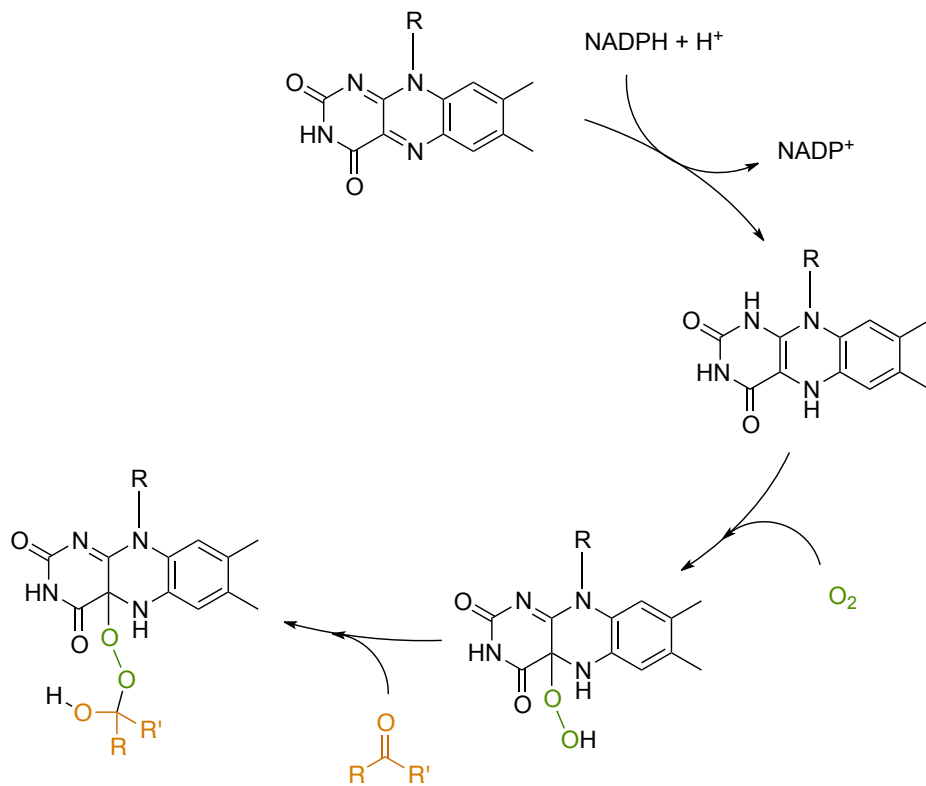
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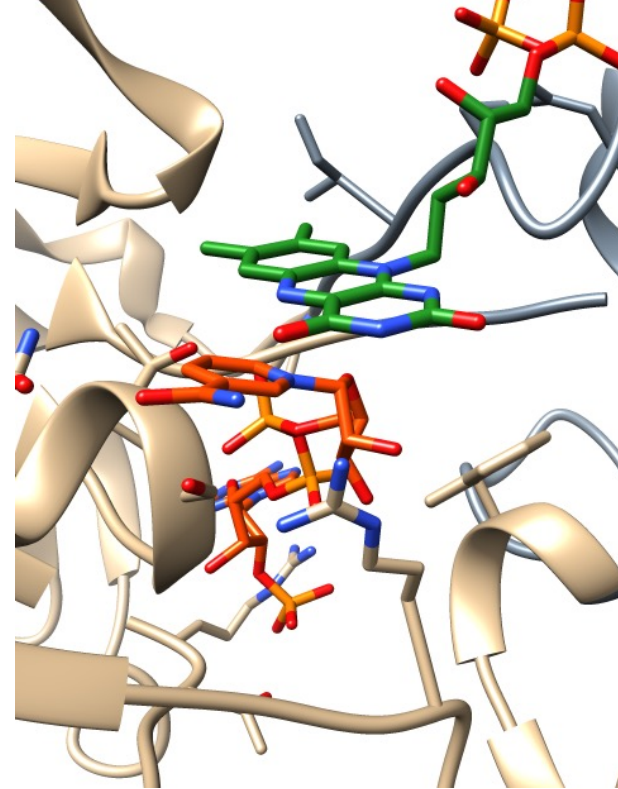
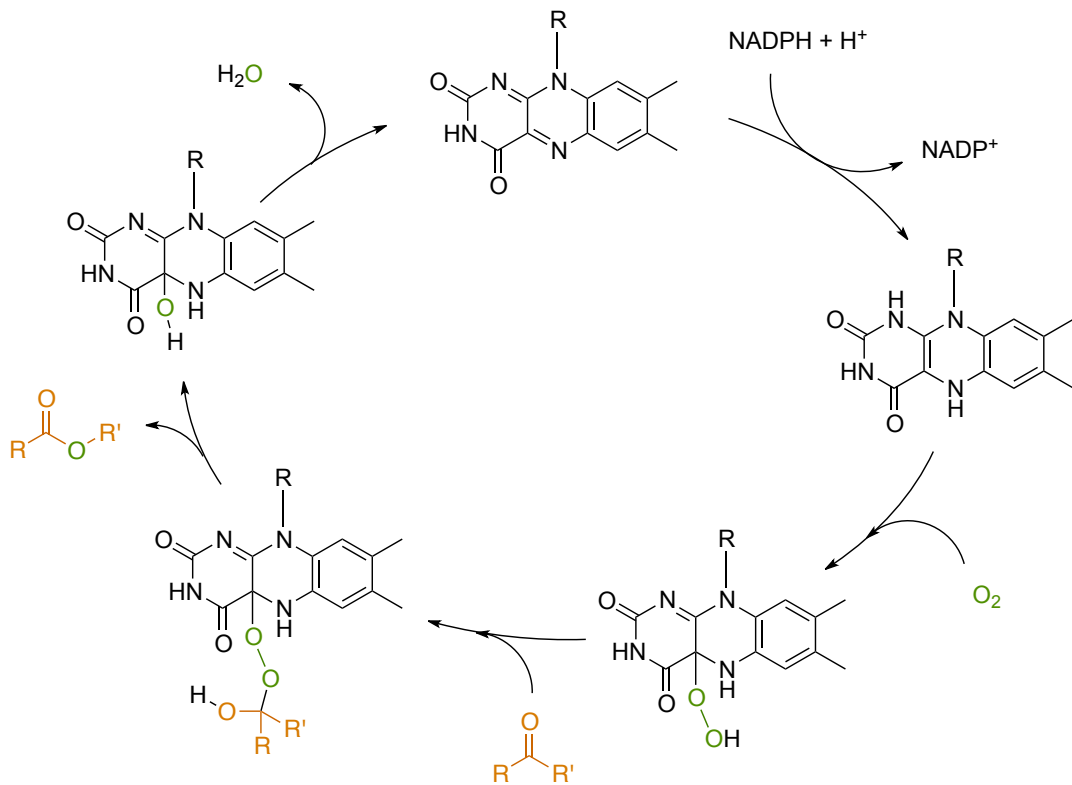
Baeyer–Villiger monooxygenases

Mechanism



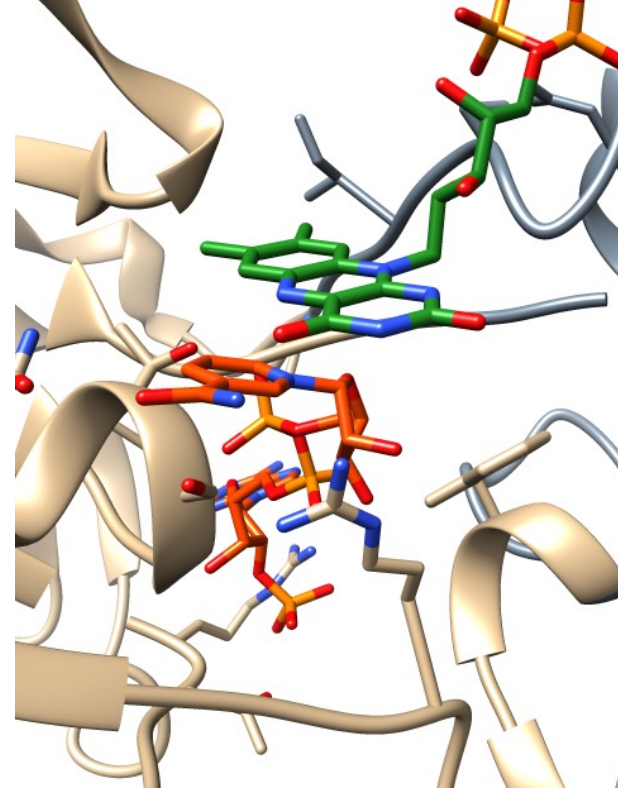
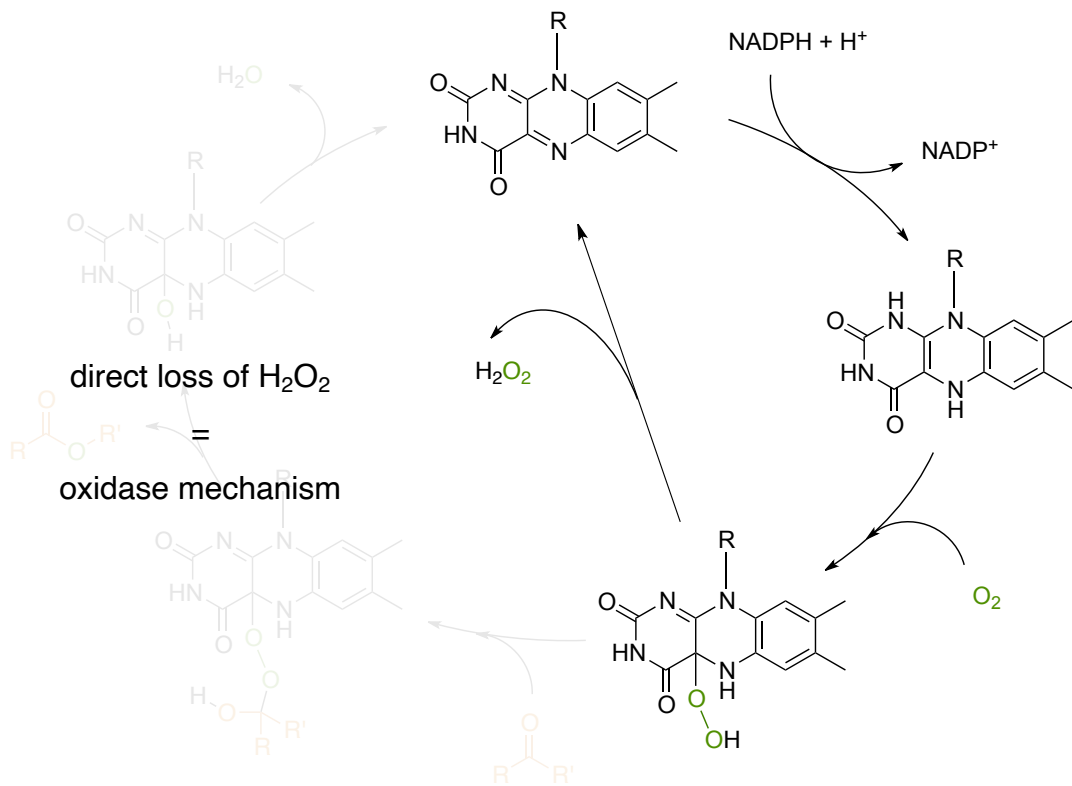
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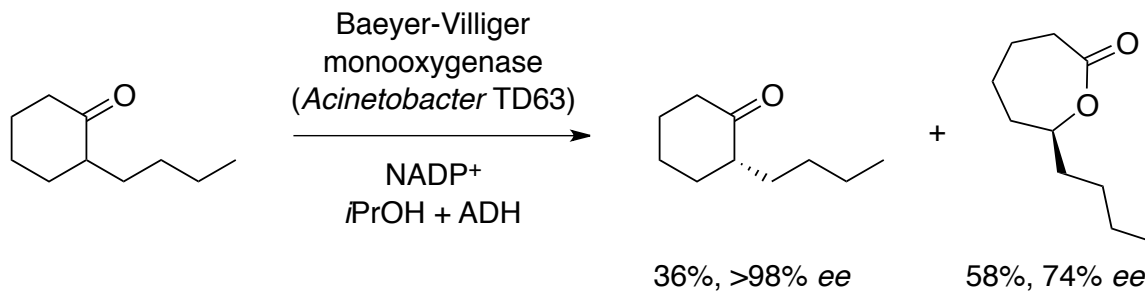
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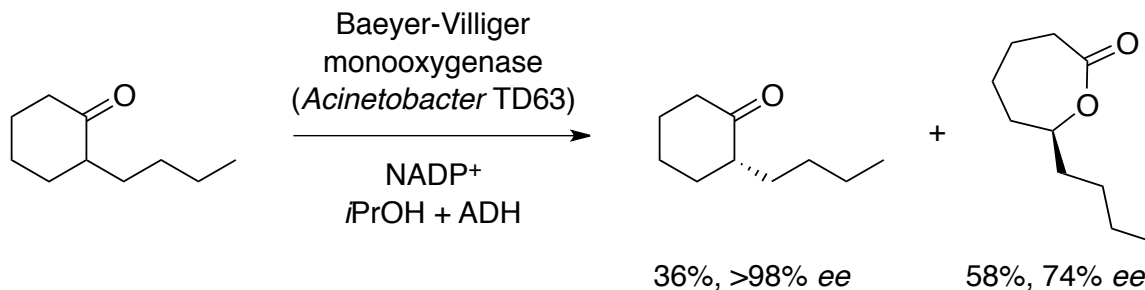
Asymmetric Baeyer–Villiger transformations

kinetic resolution of chiral ketones

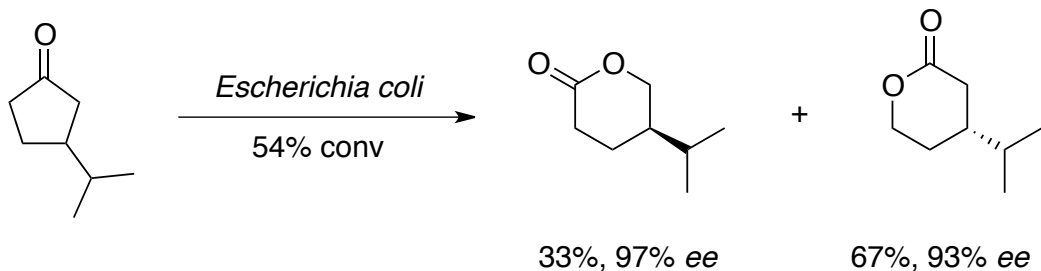


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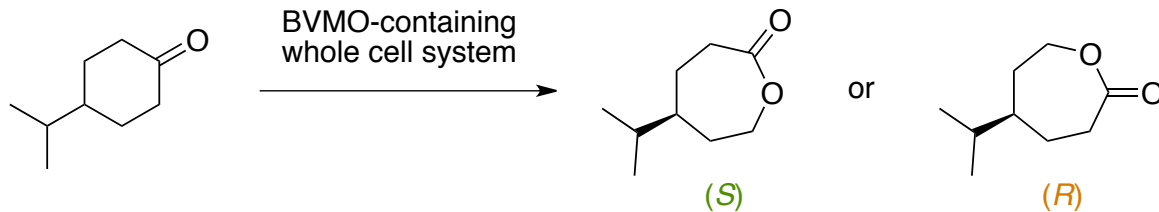


occasionally both enantiomers react... to give two different products



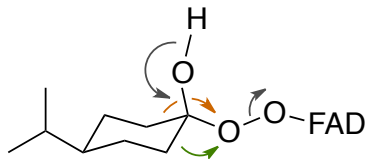
Asymmetric Baeyer–Villiger transformations

desymmetrization of achiral ketones



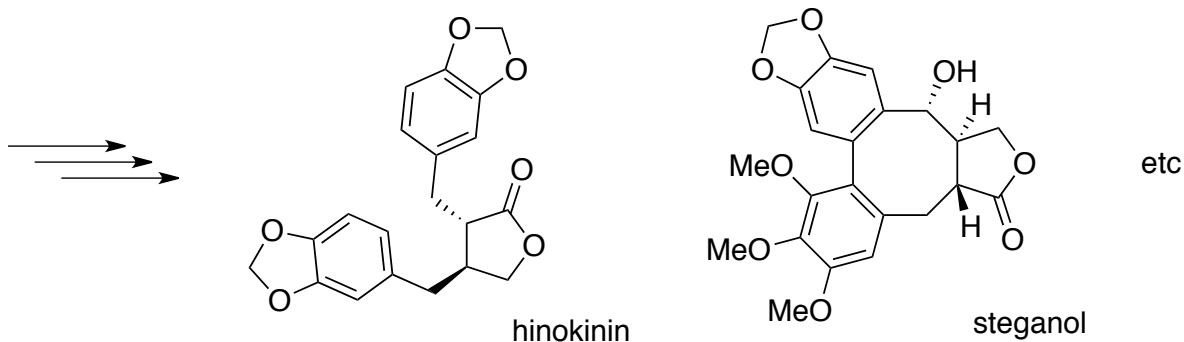
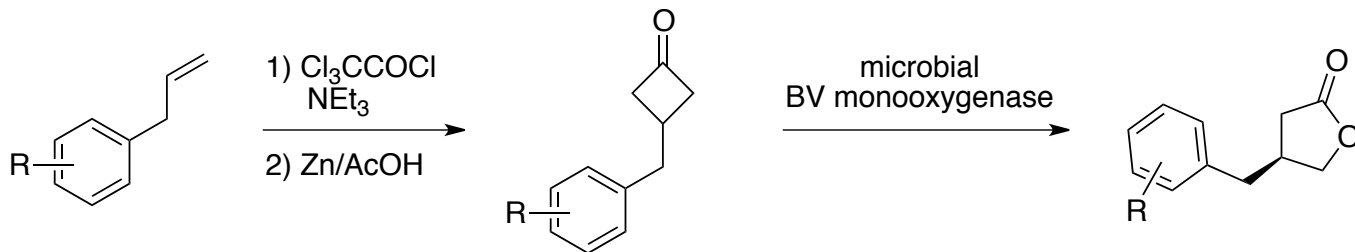
Acinetobacter sp. NCIB9871: 94% ee (*S*)

Aspergillus flavus: 84% ee (*R*)



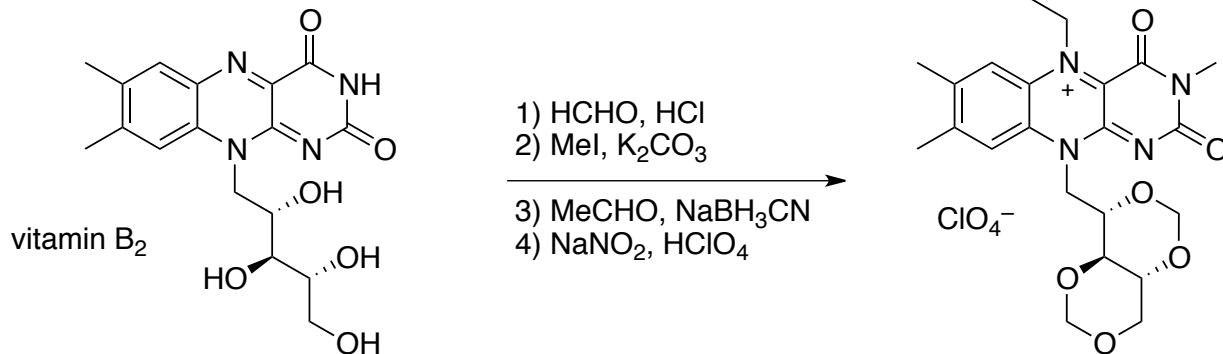
Asymmetric Baeyer–Villiger transformations

total synthesis of bioactive butyrolactones



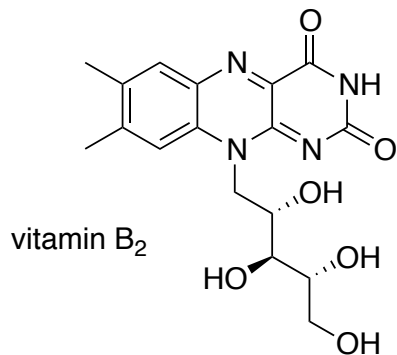
Biomimetic catalysis

modified flavins as organocatalysts

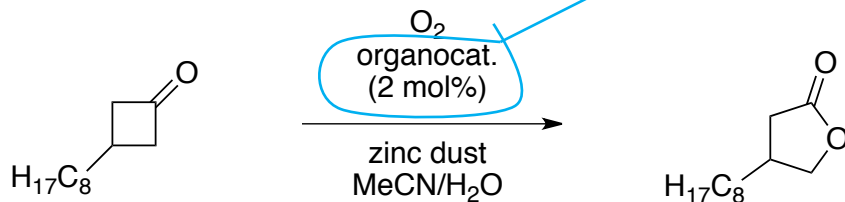
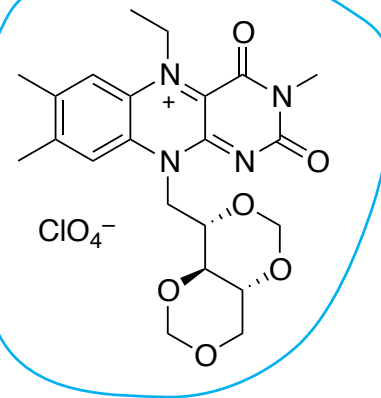


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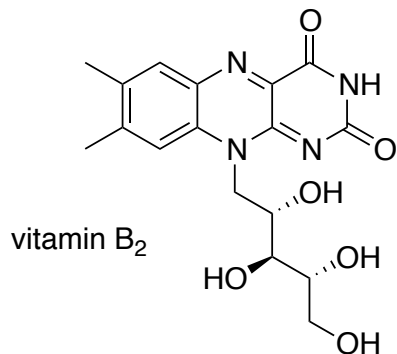


- 1) HCHO, HCl
- 2) MeI, K₂CO₃
- 3) MeCHO, NaBH₃CN
- 4) NaNO₂, HClO₄

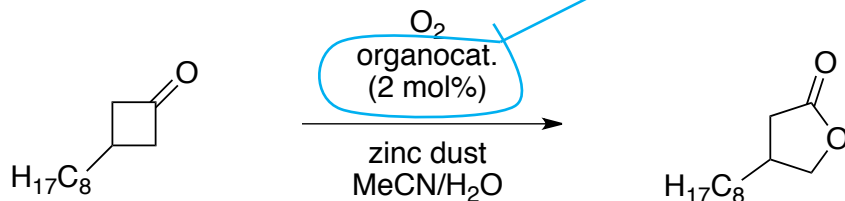
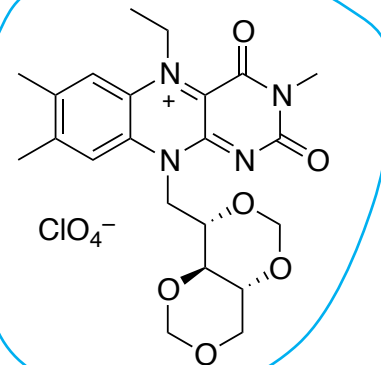


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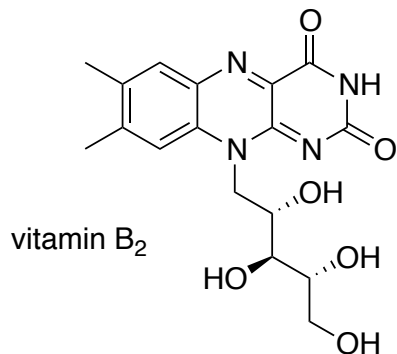
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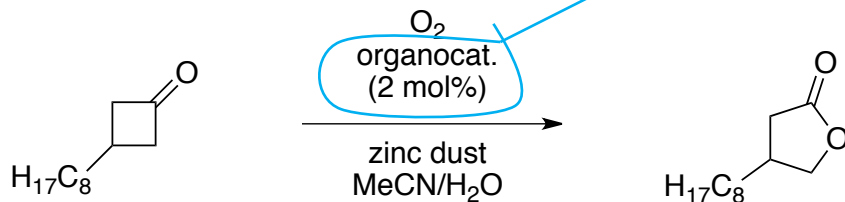
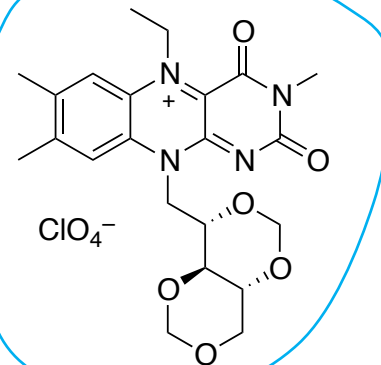
- no chiral induction (yet)
- but: more selective than other BV systems (e.g. olefins not epoxidized)

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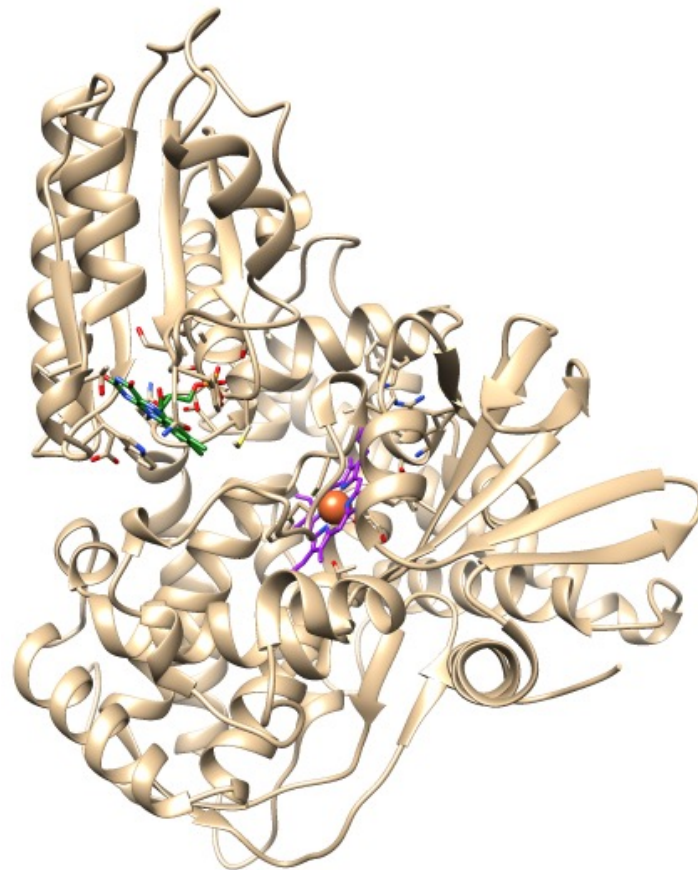


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Cytochrome P450 monooxygenases

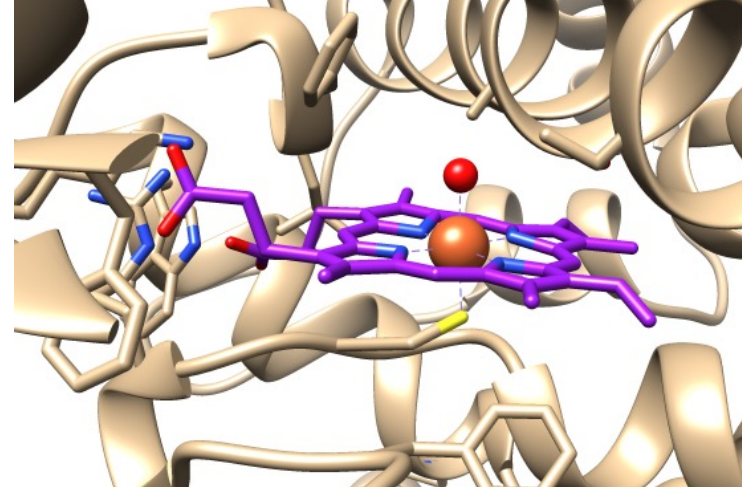
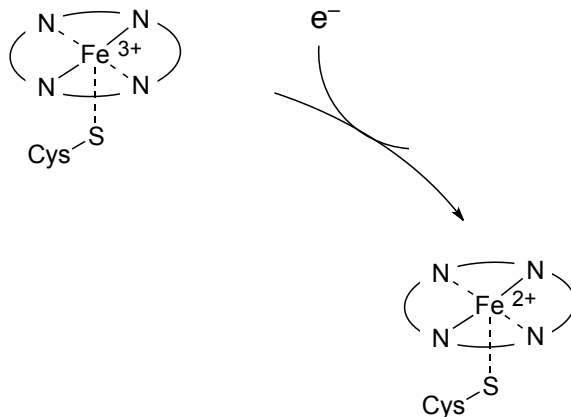
Cytochrome P450 BM3 (*Bacillus megaterium*)

- monomer, 119 kDa
- flavocytochrome
- FMN-binding domain fused to
- porphyrine-binding domain
- great exception among other CYPs as it requires no lengthy electron cascades commonly found for cytochrome systems (e.g. ferredoxin, ferredoxin reductase, cytochrome reductase,...)
- catalyzes epoxidations, sulfoxidations and CH-oxygenations, among others



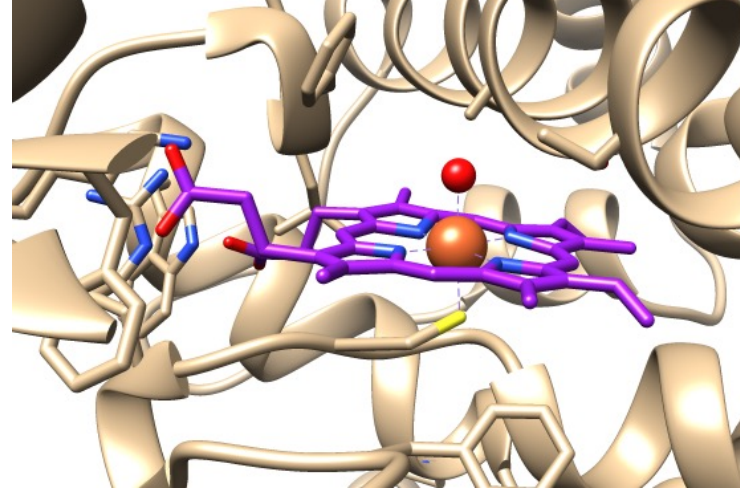
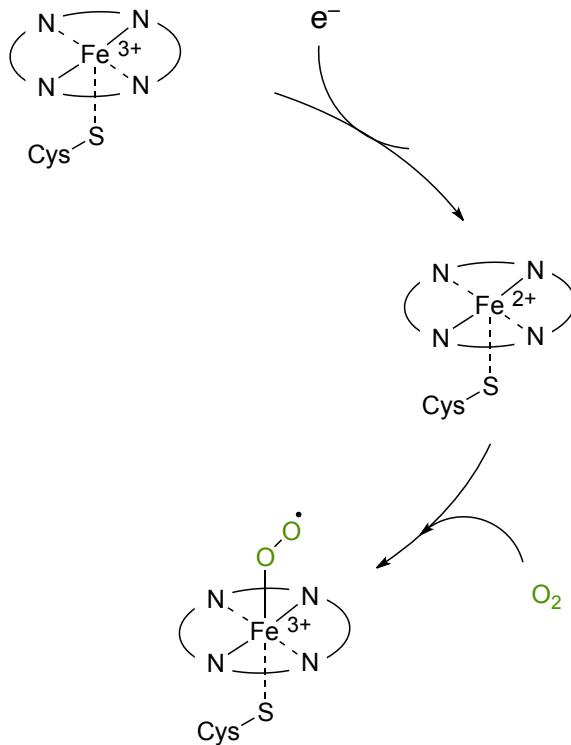
Cytochrome P450 monooxygenases

Mechanism



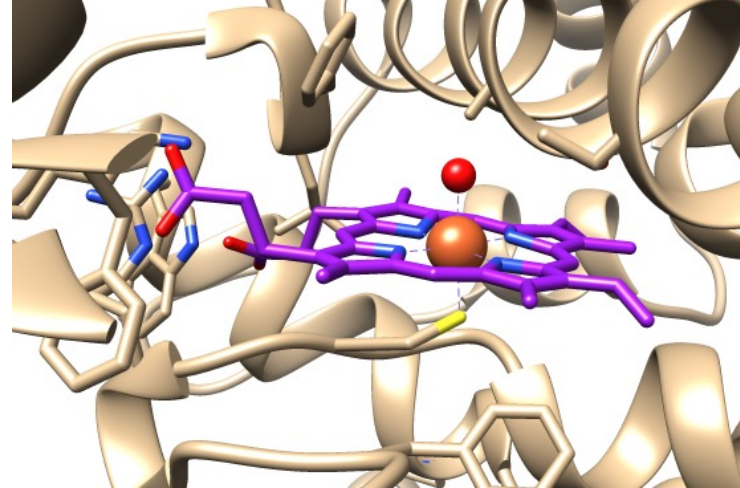
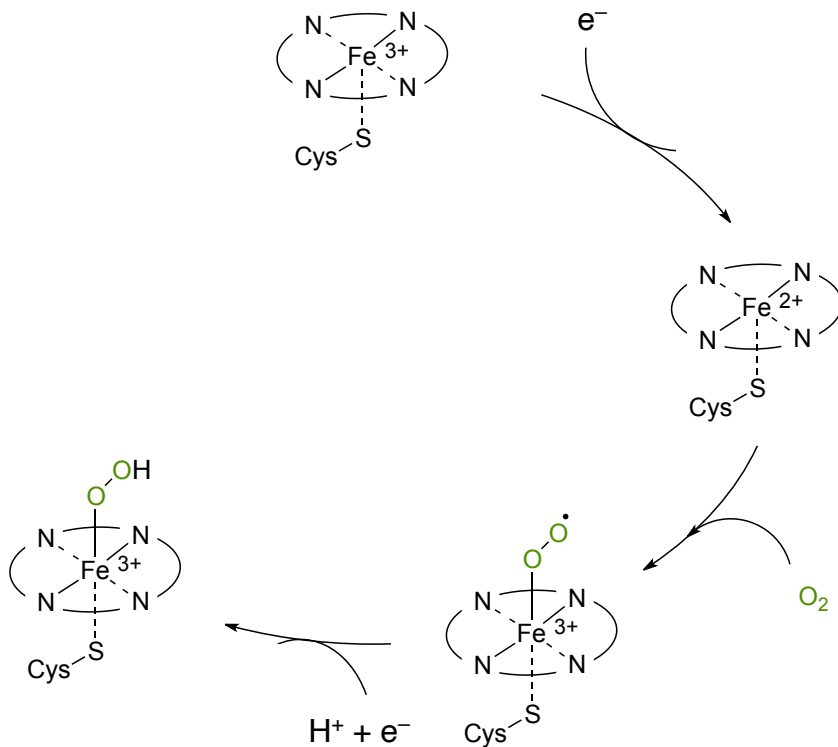
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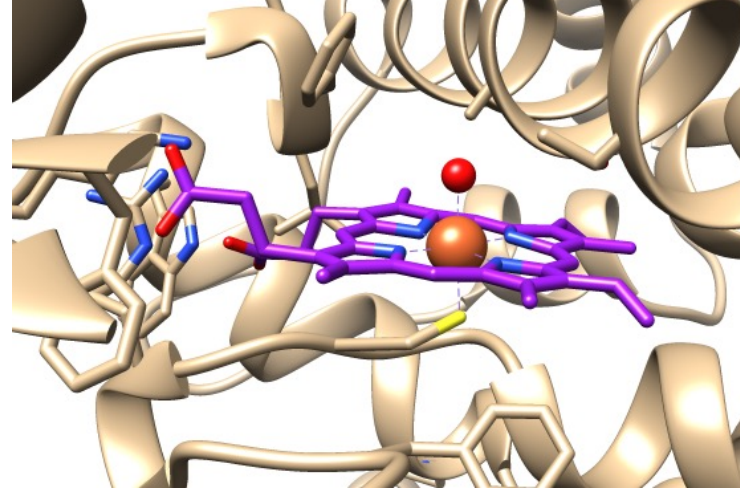
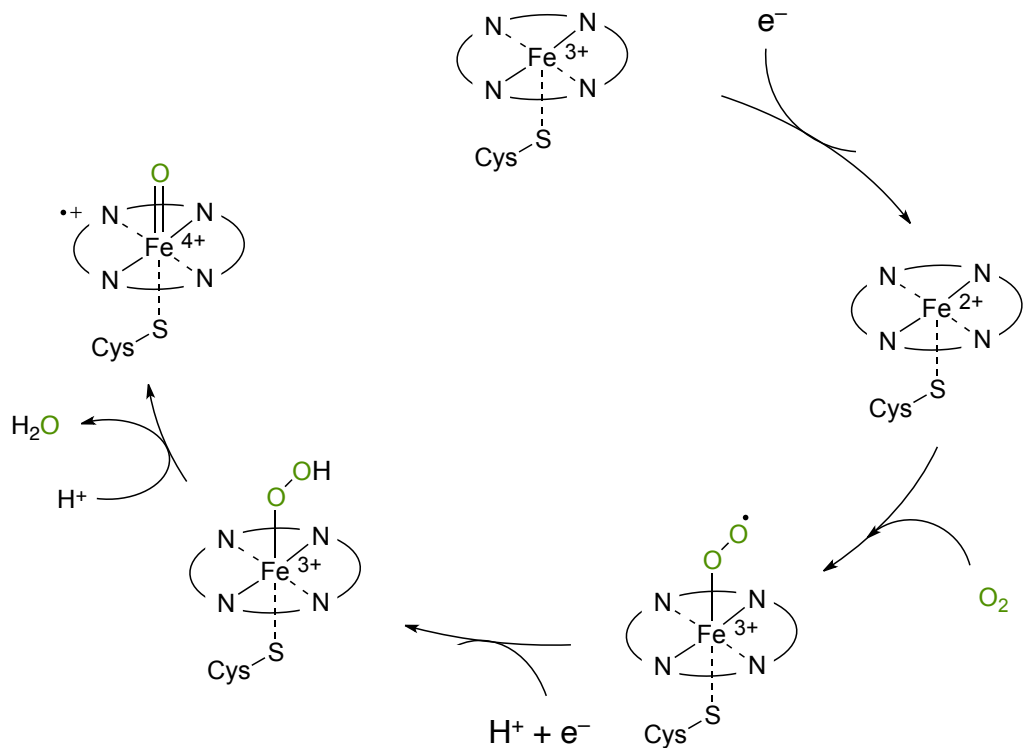
Cytochrome P450 monooxygenases

Mechanism



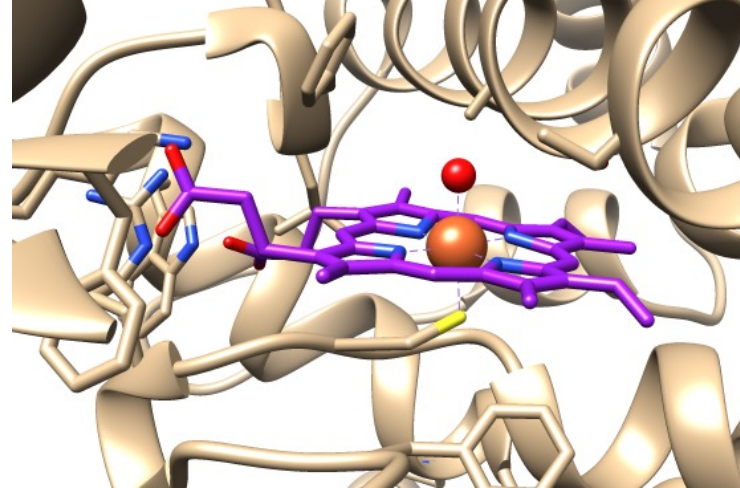
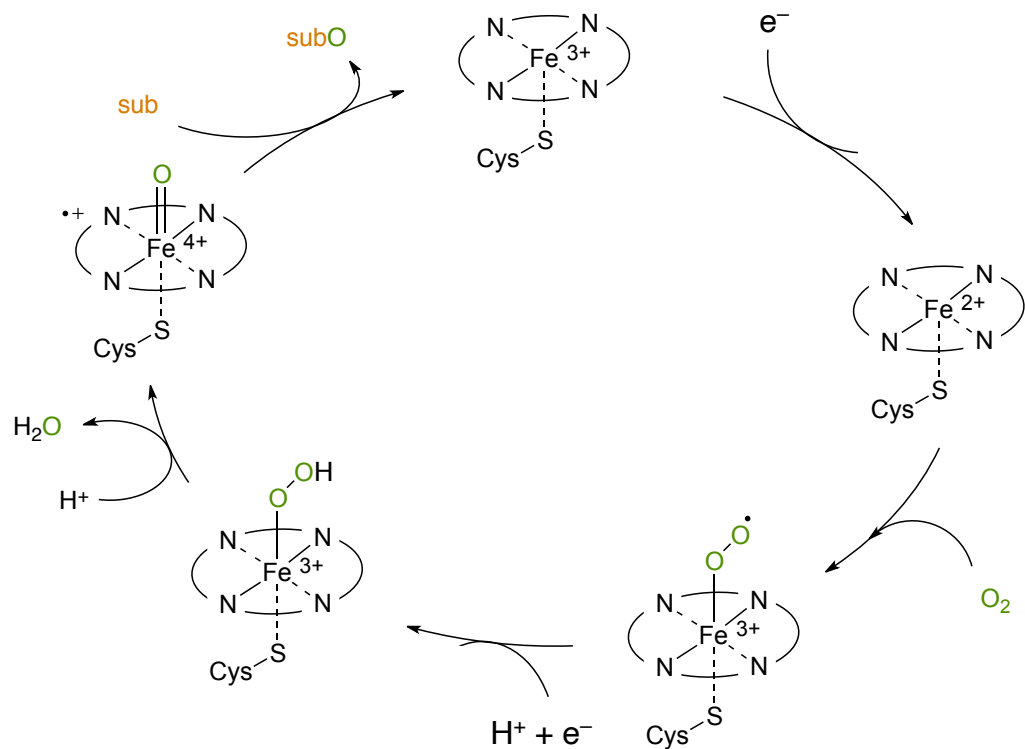
Cytochrome P450 monooxygenases

Mechanism



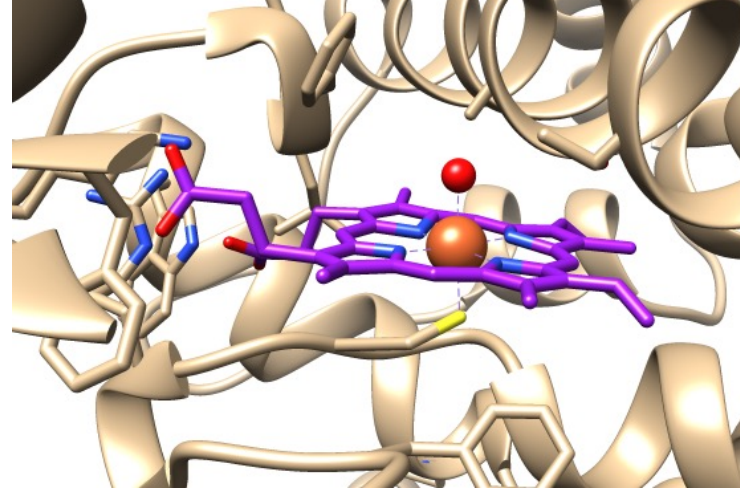
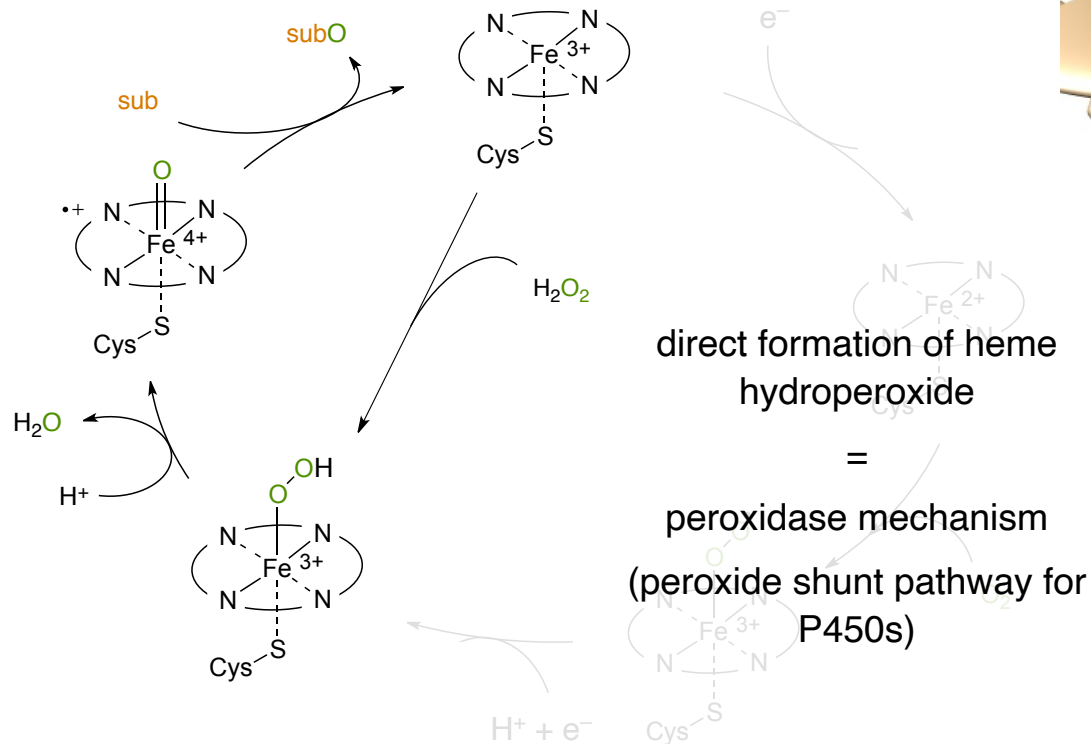
Cytochrome P450 monooxygenases

Mechanism



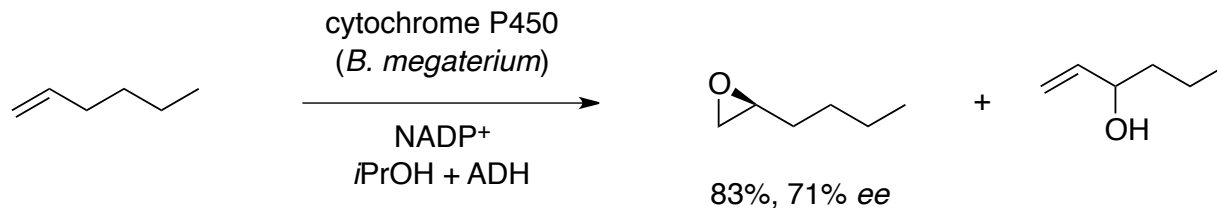
Cytochrome P450 monooxygenases

Mechanism

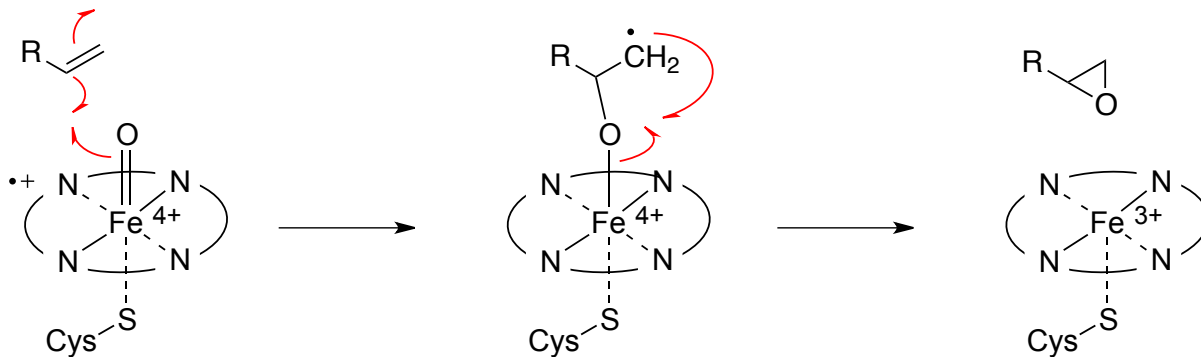


Cytochrome P450 monooxygenases

Epoxidations

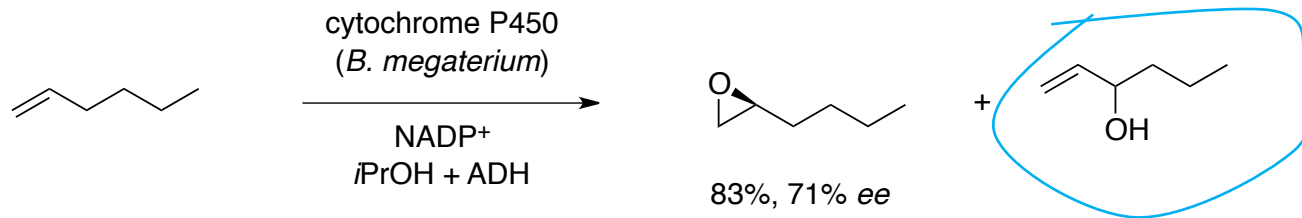


proposed oxygen transfer



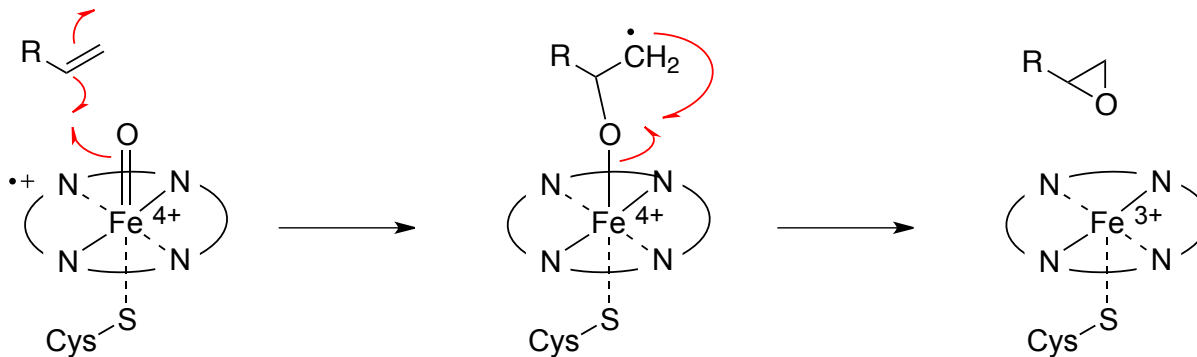
Cytochrome P450 monooxygenases

Epoxidations



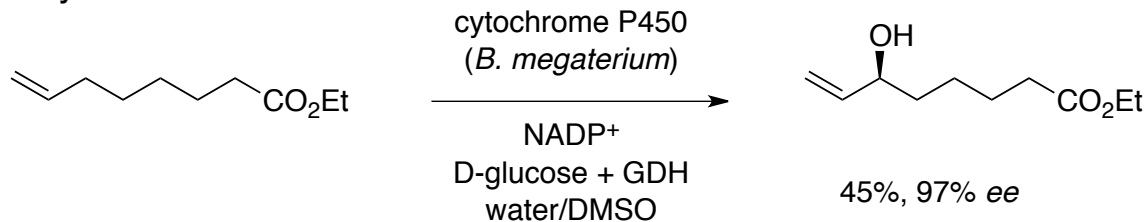
hydroxylation side product

proposed oxygen transfer

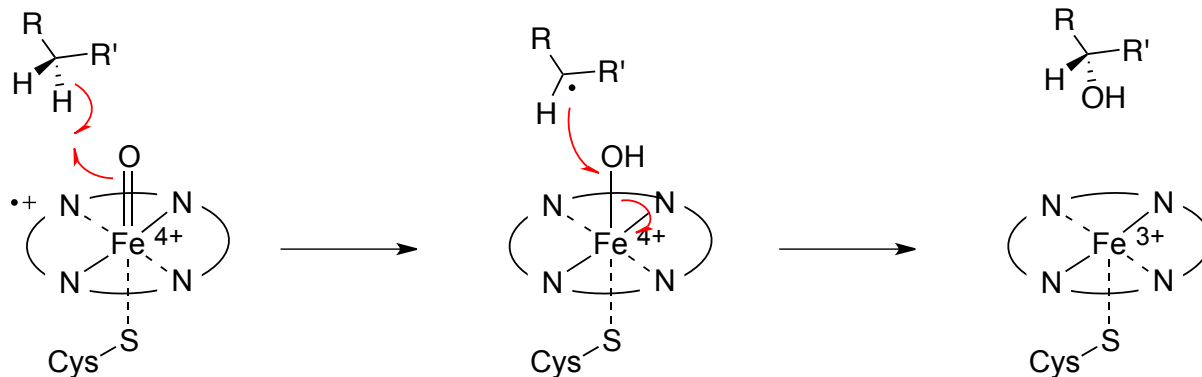


Cytochrome P450 monooxygenases

Hydroxylations



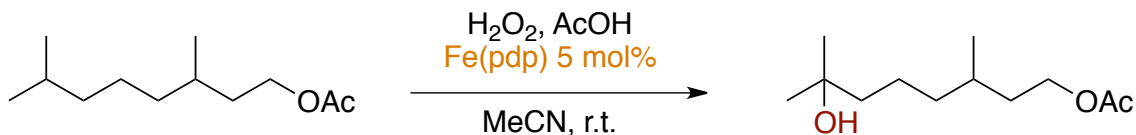
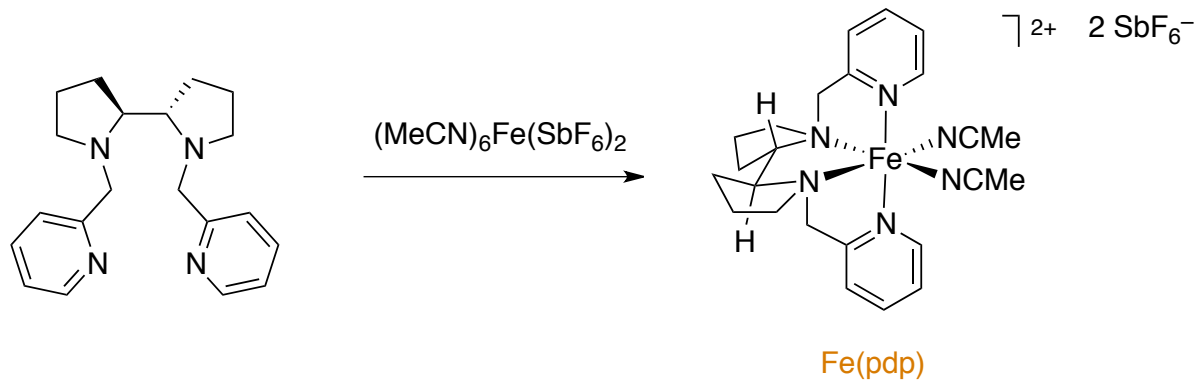
- not limited to allylic systems, but alkane CH-hydroxylation yet untamed



Biomimetic Iron-based hydroxylations

multiple non-heme iron complexes have been tested for epoxidation and hydroxylation

- rather successful: White-Chen catalysts (Fe(pdp))



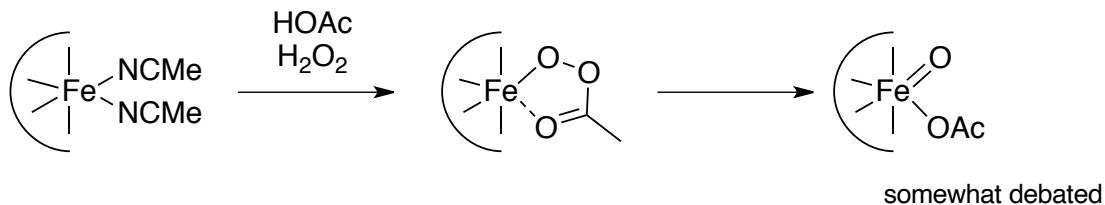
Gelalcha, *Adv. Synth. Catal.* **2014**, 356, 261-299.

Chen, White, *Science* **2007**, 318, 783-787.

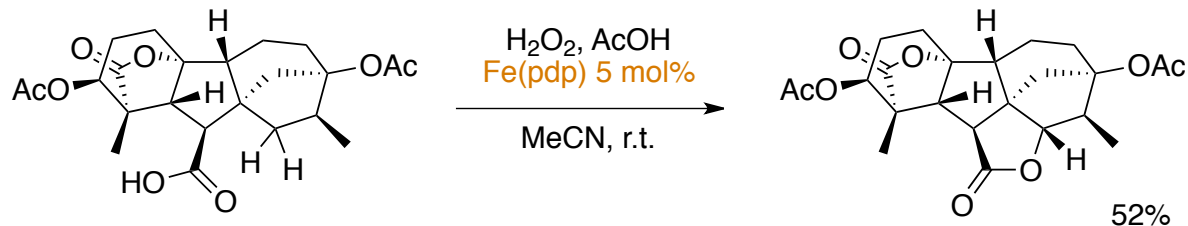
Biomimetic Iron-based hydroxylations

multiple non-heme iron complexes have been tested for epoxidation and hydroxylation

- rather successful: White-Chen catalysts (Fe(pdp))



- ✓ internal carboxylate as directing group: very high regioselectivities



SUMMARY

- dioxygenases very limited in application, but arene dihydroxylation provides unique reactivity
- molecular capacitors often involved in reductive preactivation
- both "organocatalytic" and "metal-catalyzed" versions of monooxygenases as relatively established biocatalysts also in synthetic applications