

CHEM-E4109

MODERN METHODS IN **BIOCATALYSIS**

chapter #4: aminations & deaminations

11.3.2022

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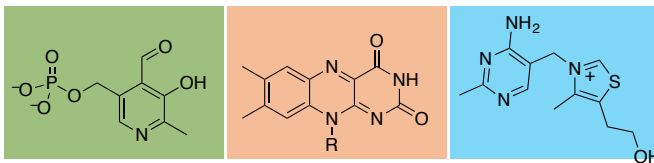
Jan Deska
Bioorganic
Chemistry

TODAY'S MENU

Nitrogen-containing organic molecules of particular importance in e.g. pharma or agrochemical products

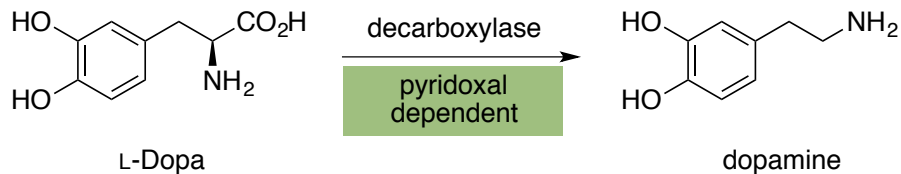
- oxidative enzymes can be exploited to prepare chiral amines
- focus on transferases
- similar to oxidoreductases: control of equilibria crucial for productive aminations

Amino acid metabolism

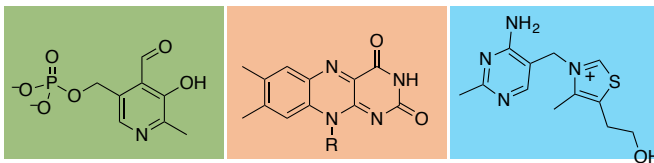


e.g. metabolic fate of Dopa

- L-Dopa used in Parkinson treatment
- source of dopamine (and other neurotransmitters)

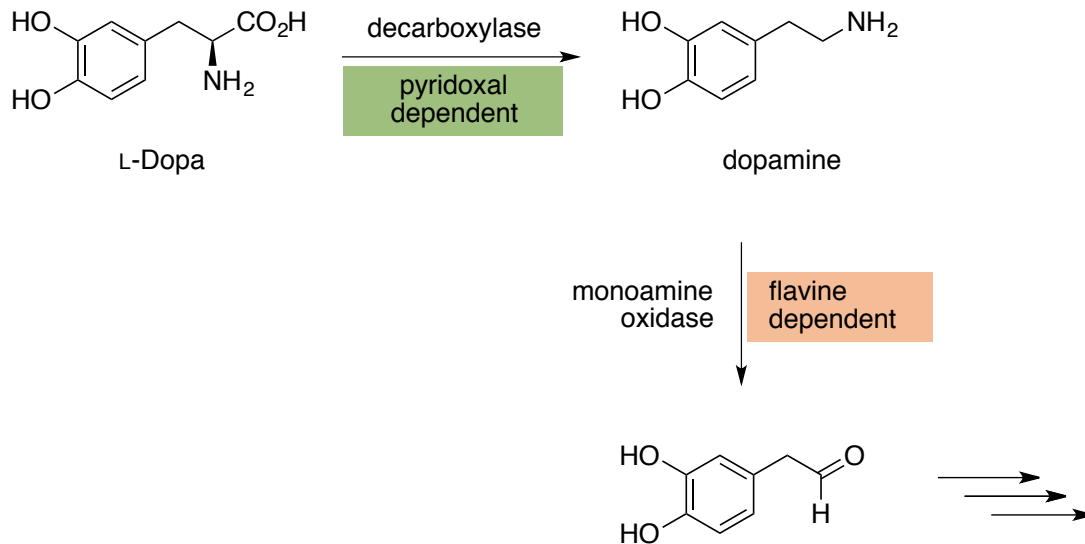


Amino acid metabolism

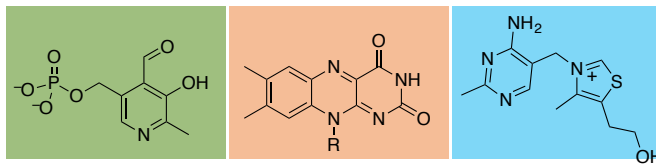


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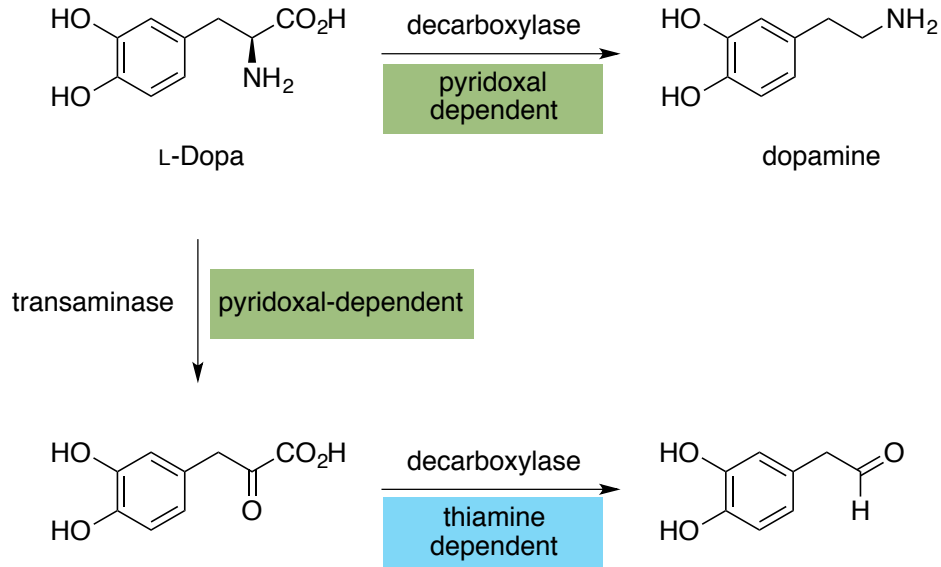


Amino acid metabolism



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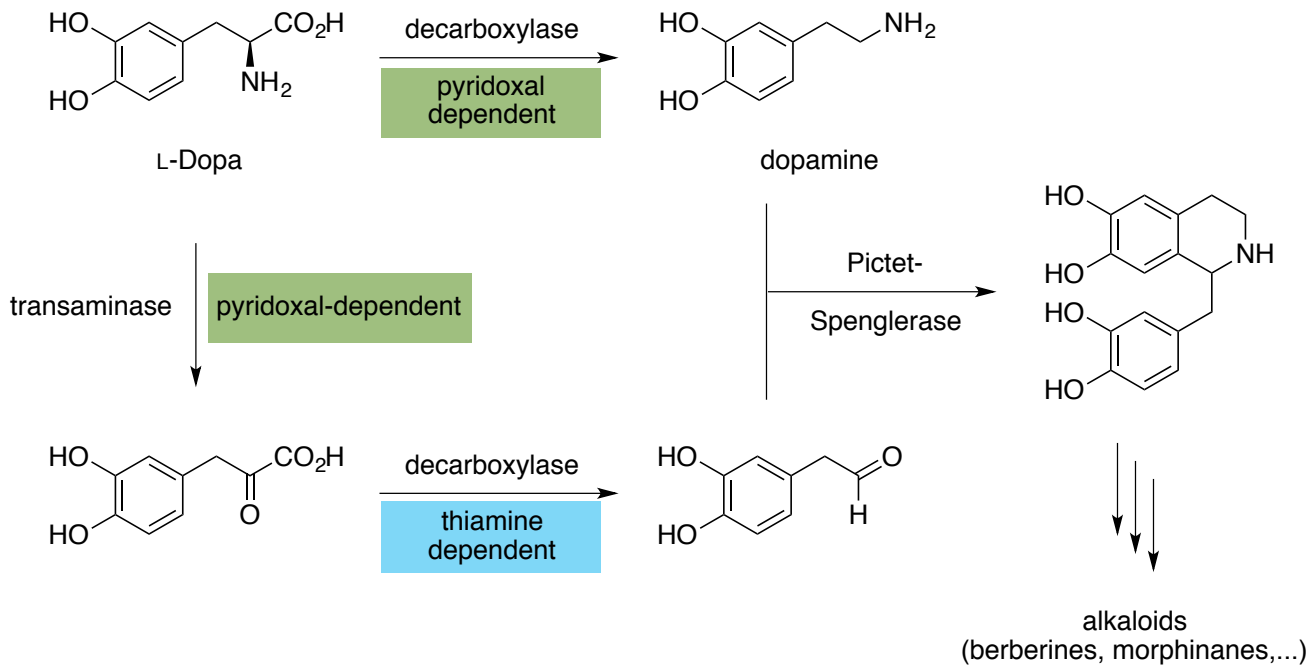
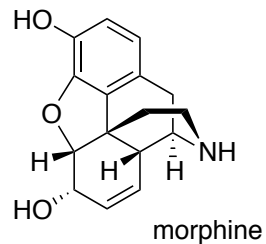
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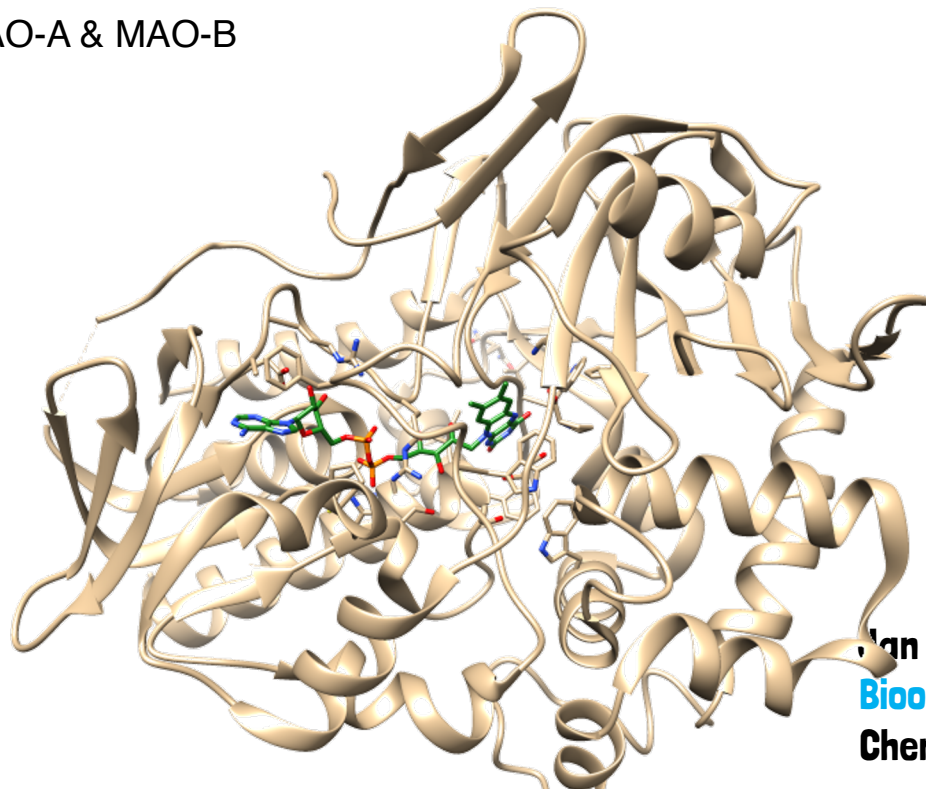
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Monoamine oxidases

Monoamine oxidase N or MAO-N (*Aspergillus niger*)

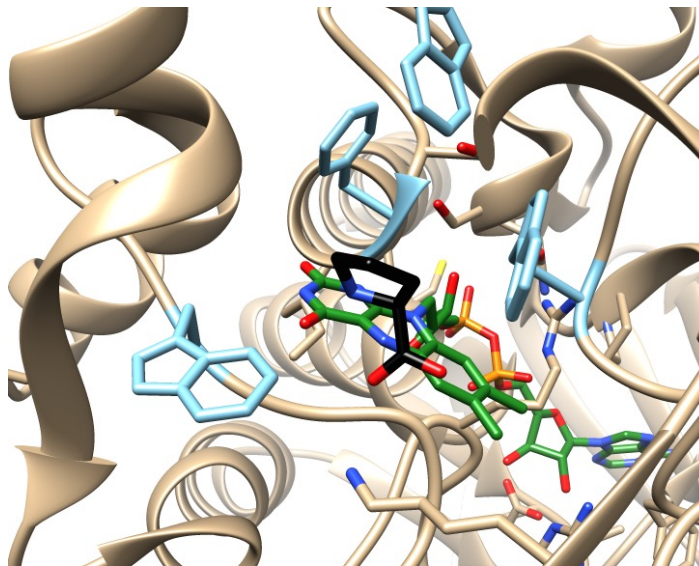
- homo-tetramer, subunits à 56 kDa, 495 amino acids
- MAO-N 24% homologous to human MAO-A & MAO-B
- flavoprotein
- catalyzes oxidation of amines to imines
- poor activity on non-natural substrates, but evolved proteins proved to accept a variety of synthetic amines



Monoamine oxidases

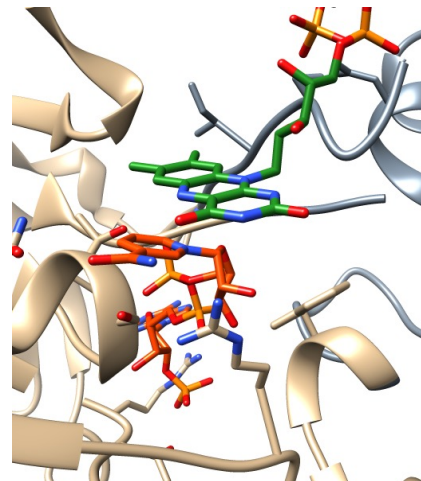
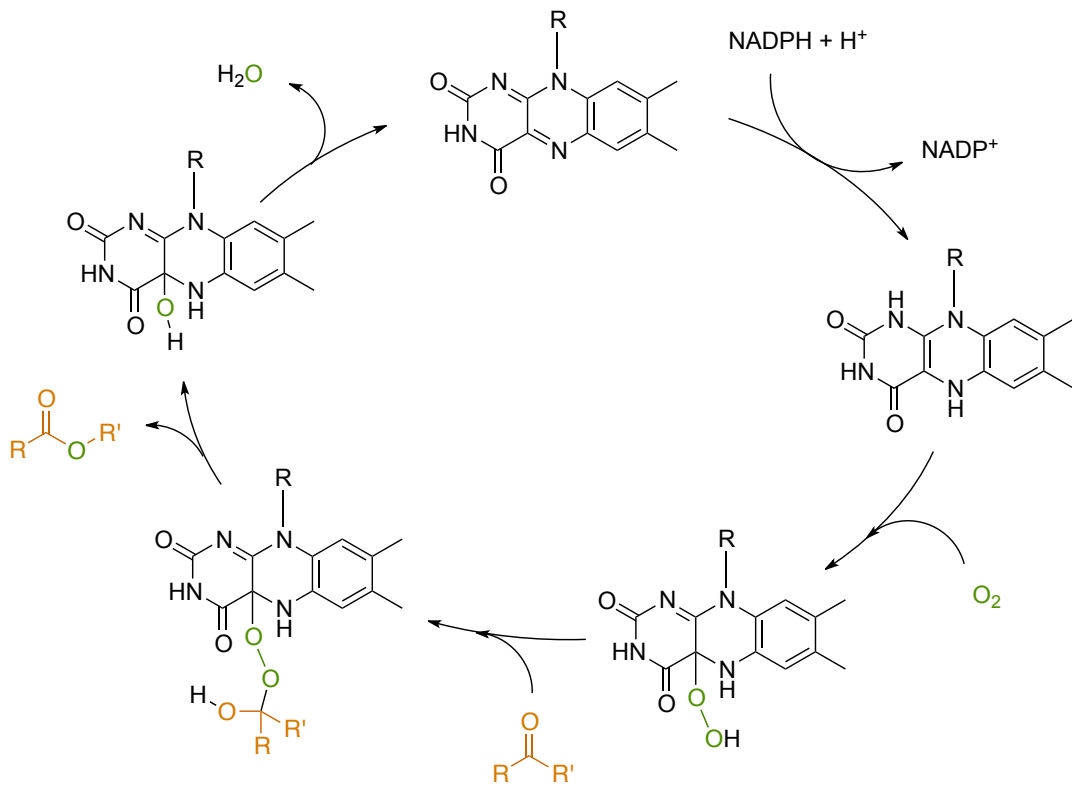
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- aromatic cage aligns substrate on top of the FAD



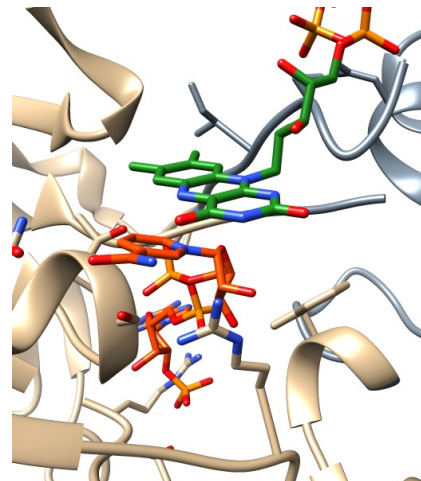
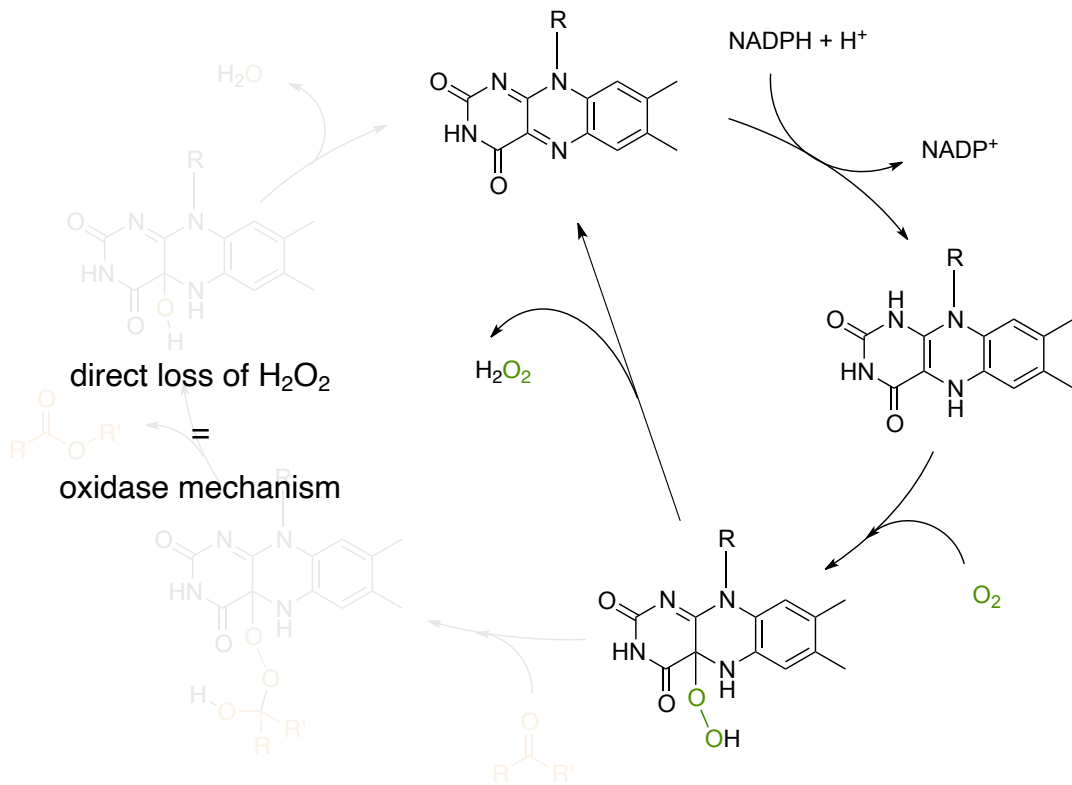
Recap: Baeyer–Villiger monooxygenases

Mechanism



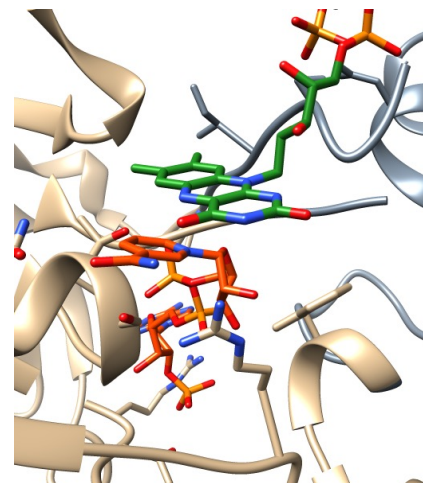
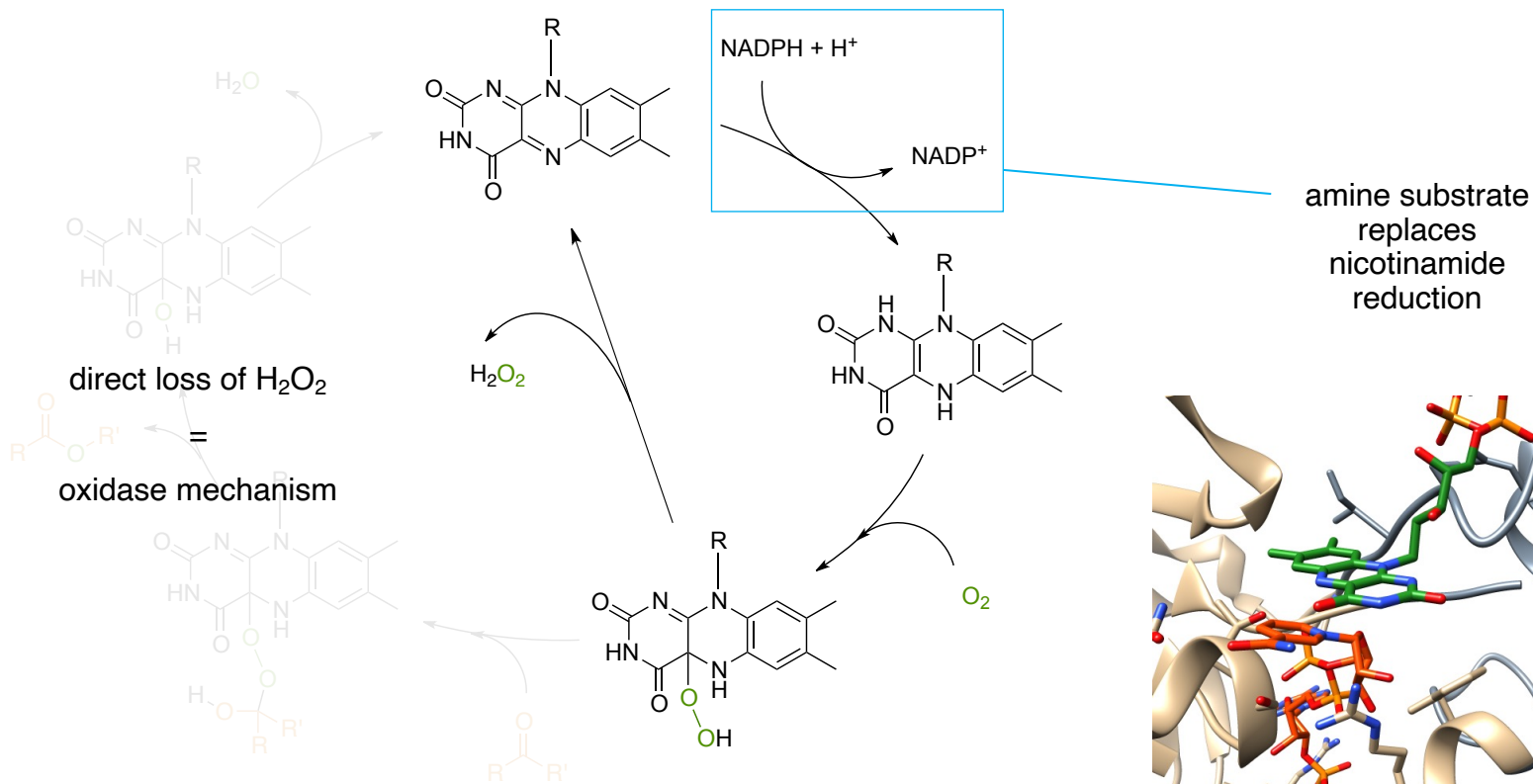
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Mechanism



Recap: Baeyer–Villiger monooxygenases

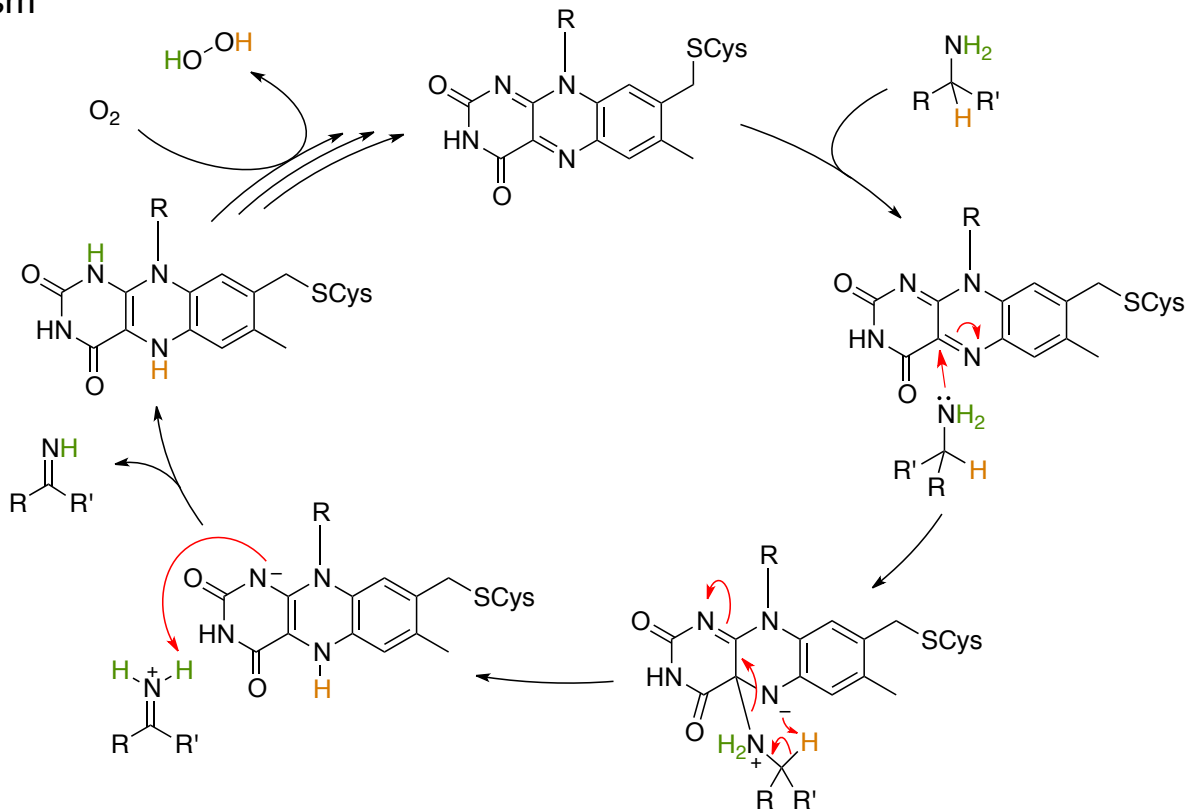
Mechanism



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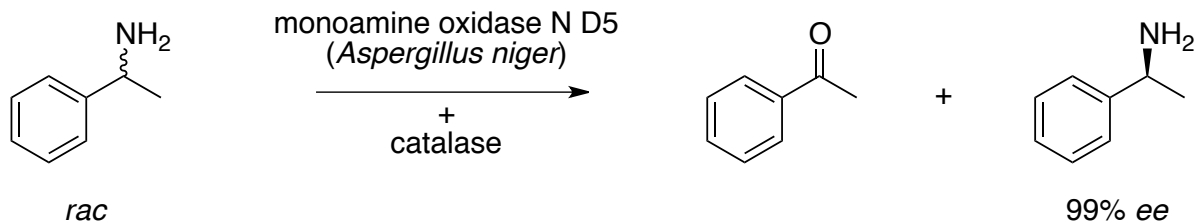
Monoamine oxidases

Mechanism



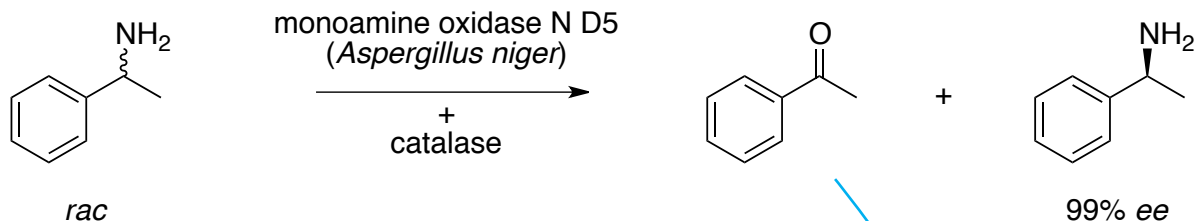
Asymmetric transformations

kinetic resolutions of primary and secondary, branched amines

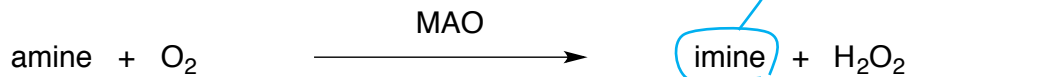


Asymmetric transformations

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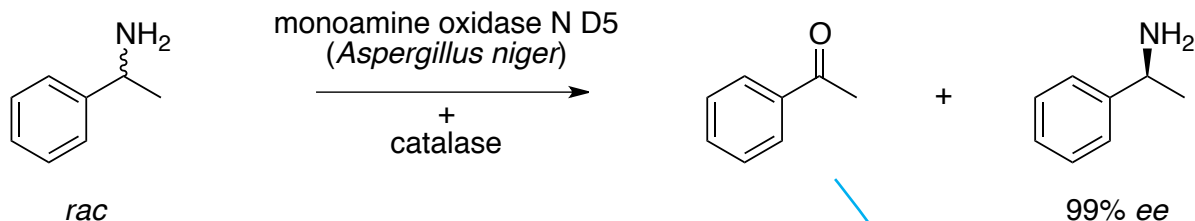


- hydrogen peroxide side product poisons the system

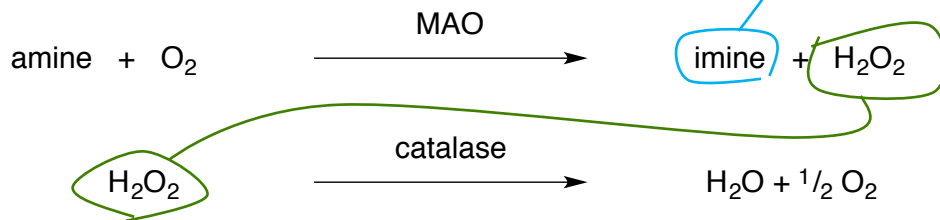


Asymmetric transformations

kinetic resolutions of primary and secondary, branched amines



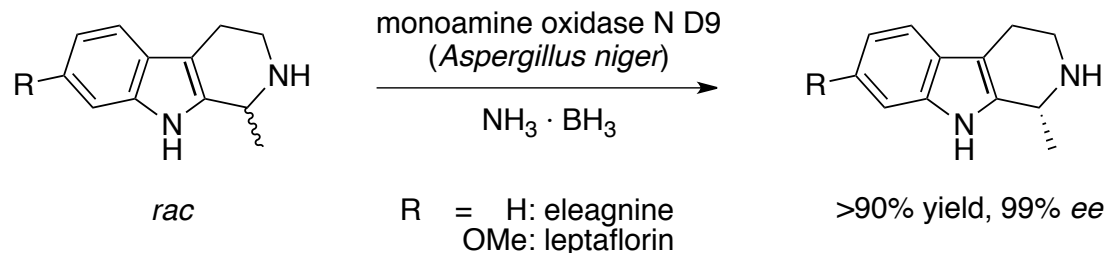
- hydrogen peroxide side product poisons the system
- catalase (heme enzyme) disproportionates H_2O_2



Asymmetric transformations

deracemization

- by non-enzymatic re-reduction of imine

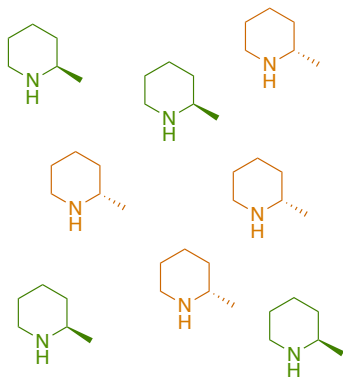


- in-situ formed achiral imine is reduced by BH_3 in a racemic fashion
 - H_2O_2 is destroyed by borane
 - over time, the non reactive enantiomer accumulates, without any side products
- = high yields & high ee

Asymmetric transformations

deracemization

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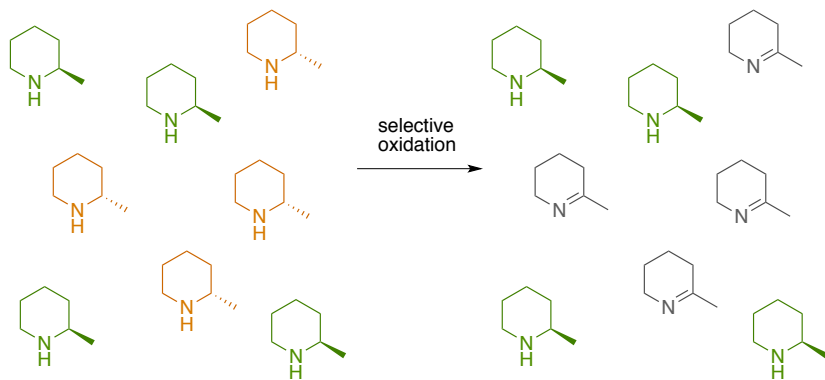


4 : 4

Asymmetric transformations

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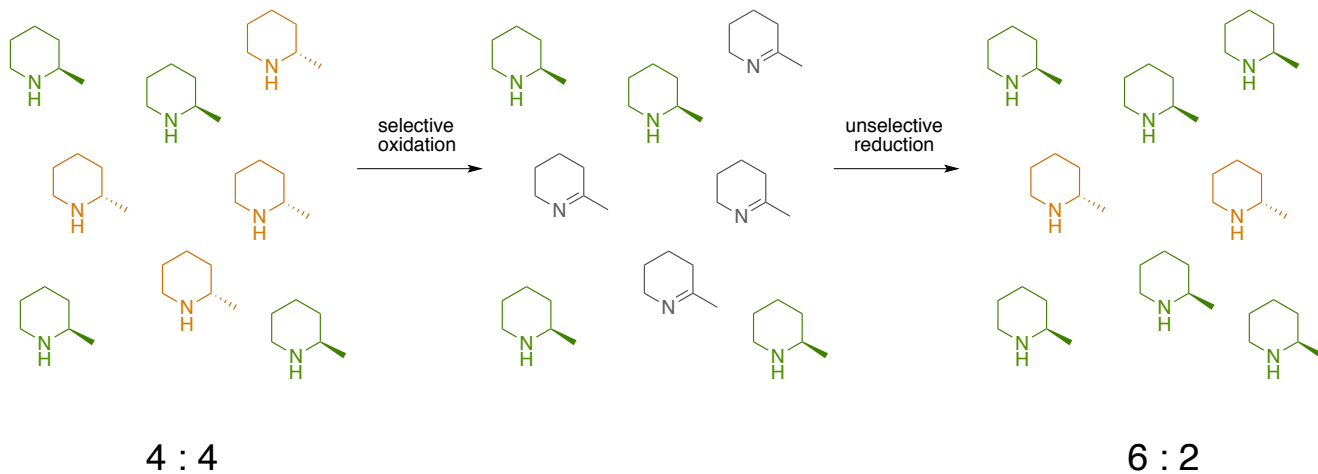


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Asymmetric transformations

deracemization

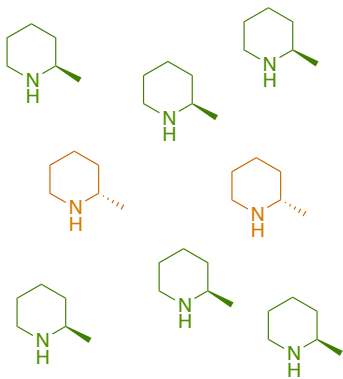
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Asymmetric transformations

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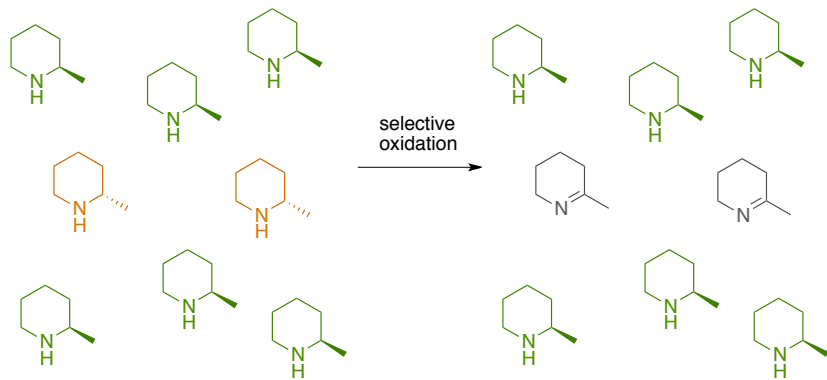


6 : 2

Asymmetric transformations

deracemization

- by non-enzymatic re-reduction of imine

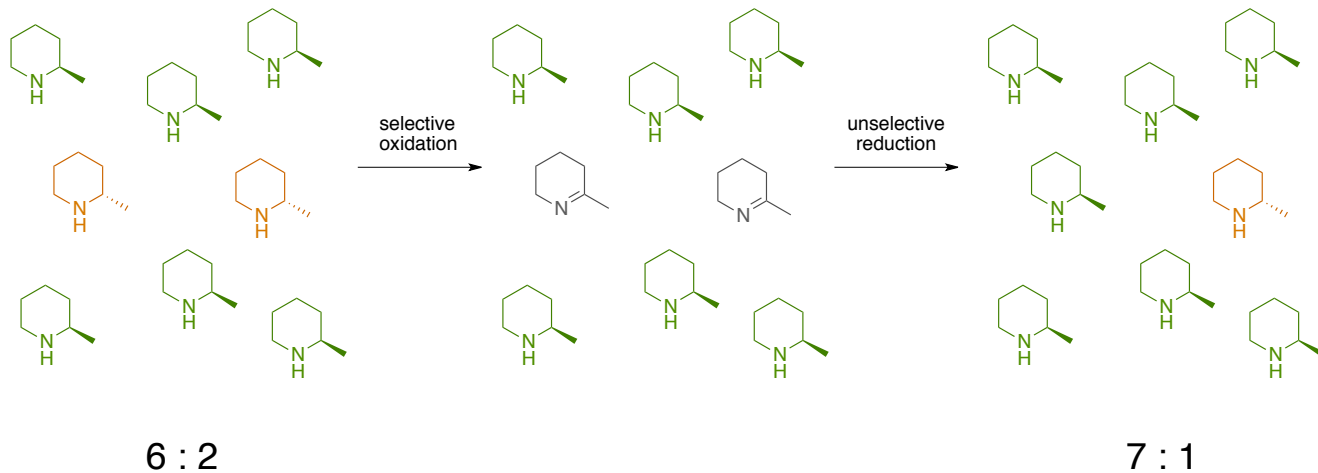


6 : 2

Asymmetric transformations

deracemization

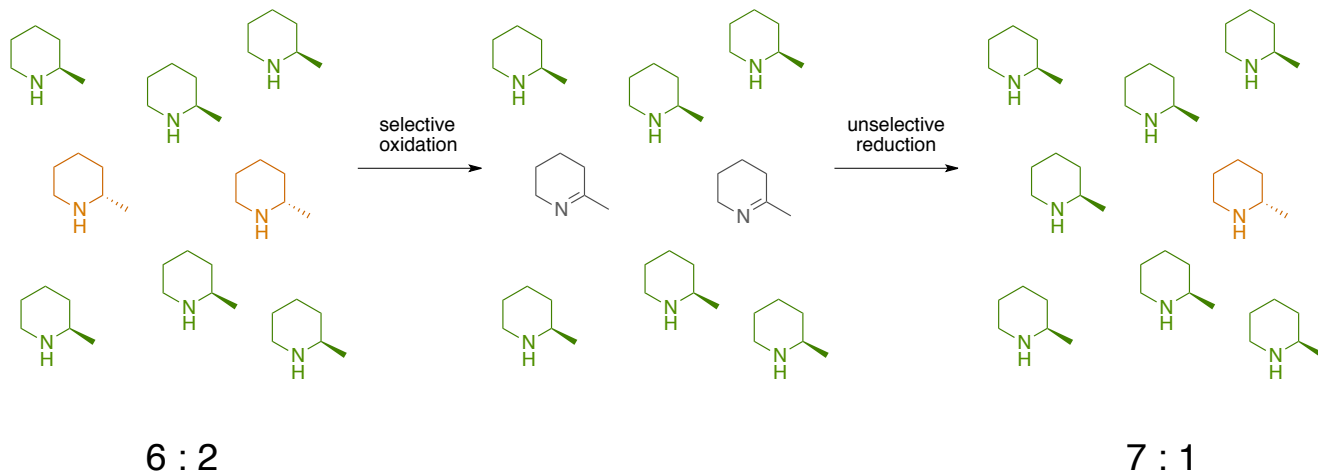
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Asymmetric transformations

deracemization

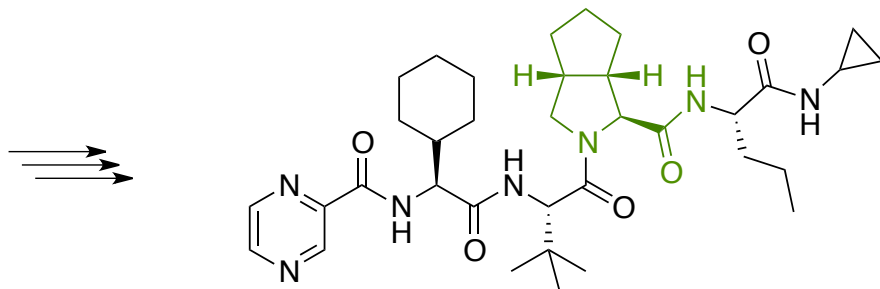
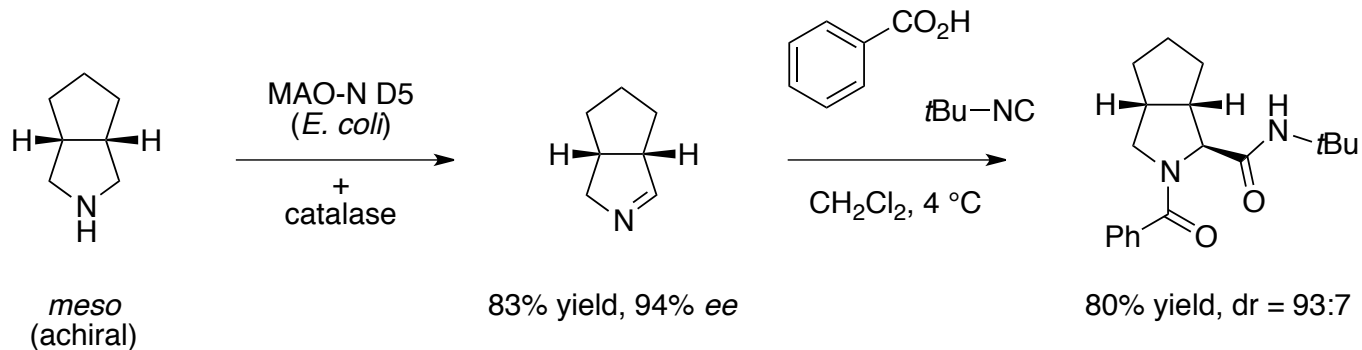
- by non-enzymatic re-reduction of imine



continuously repeated until only one enantiomer remains

Asymmetric transformations

desymmetrization



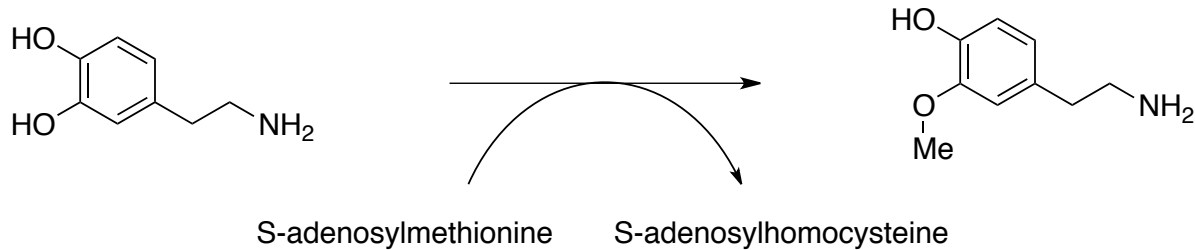
telaprevir ("Incivek" or "Incivo")
anti hepatitis C

EC 2: TRANSFERASES

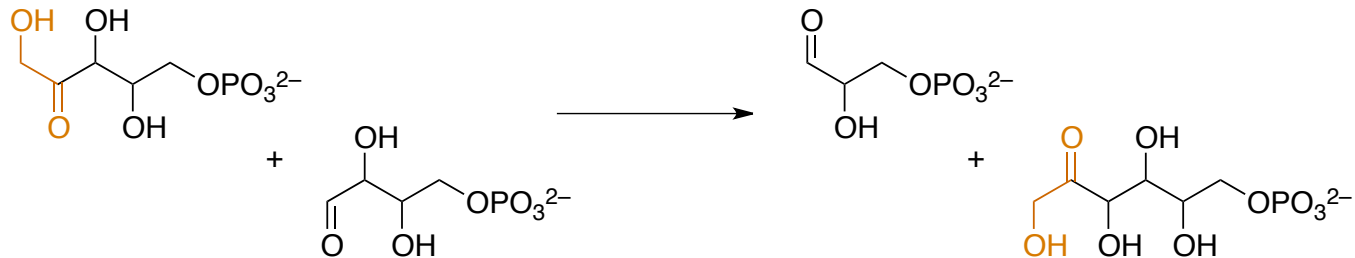
Transferases - features

enzymes of class EC 2 catalyze the transfer of functional groups from one donor substrate to an acceptor substrate

methyltransferases



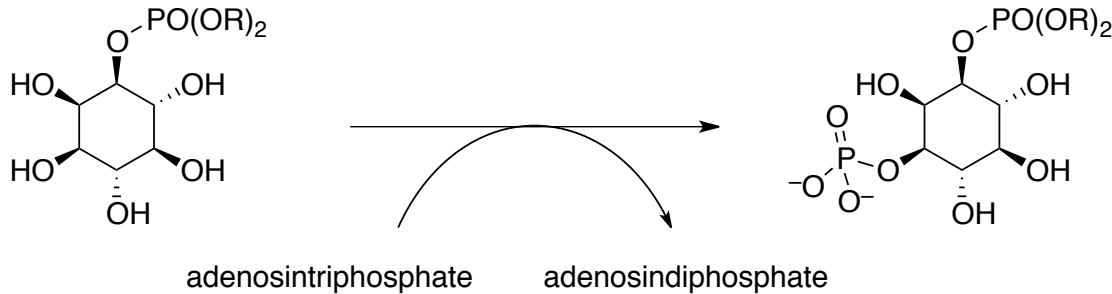
transketolases



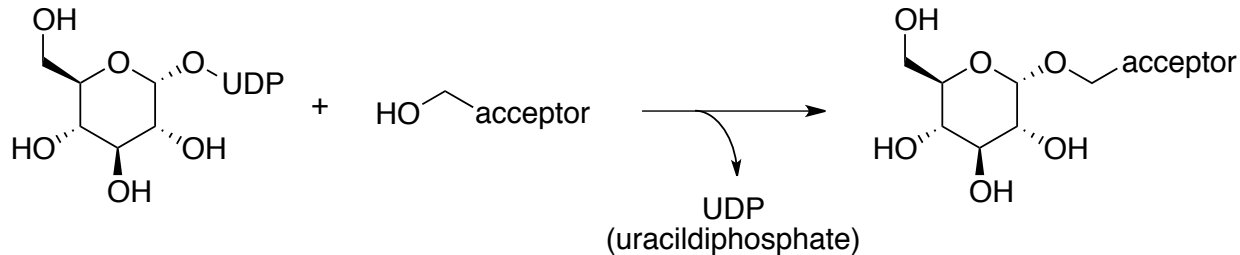
Transferases - features

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kinases

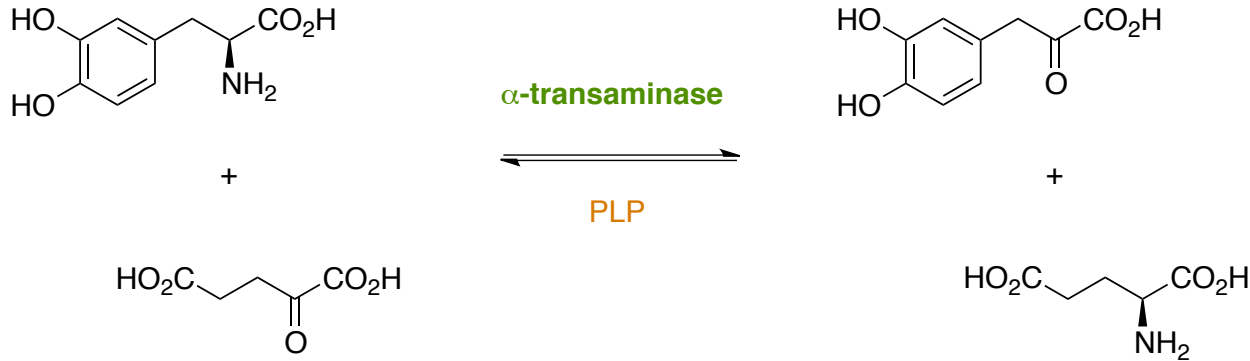


glycosyltransferases



Transaminases

exchange of amino groups between amino acids and α -ketocarboxylic acids

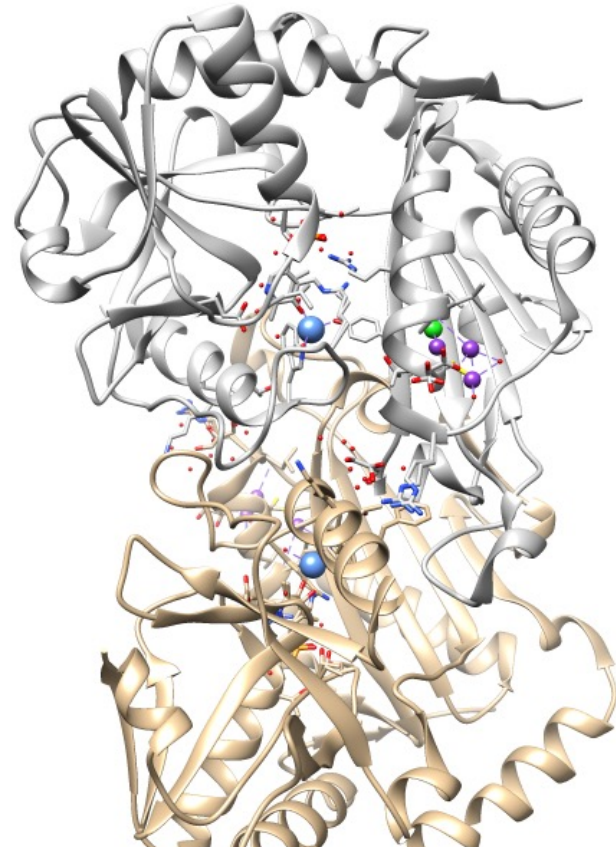


- α -transaminases important for biosynthesis of non-proteinogenic amino acids
- also involved in the breakdown of amino acids
- = synthetically more interesting: ω -transaminases (do not require carboxylate substrate)

ω -Transaminases

transaminase from *Aspergillus fumigatus*

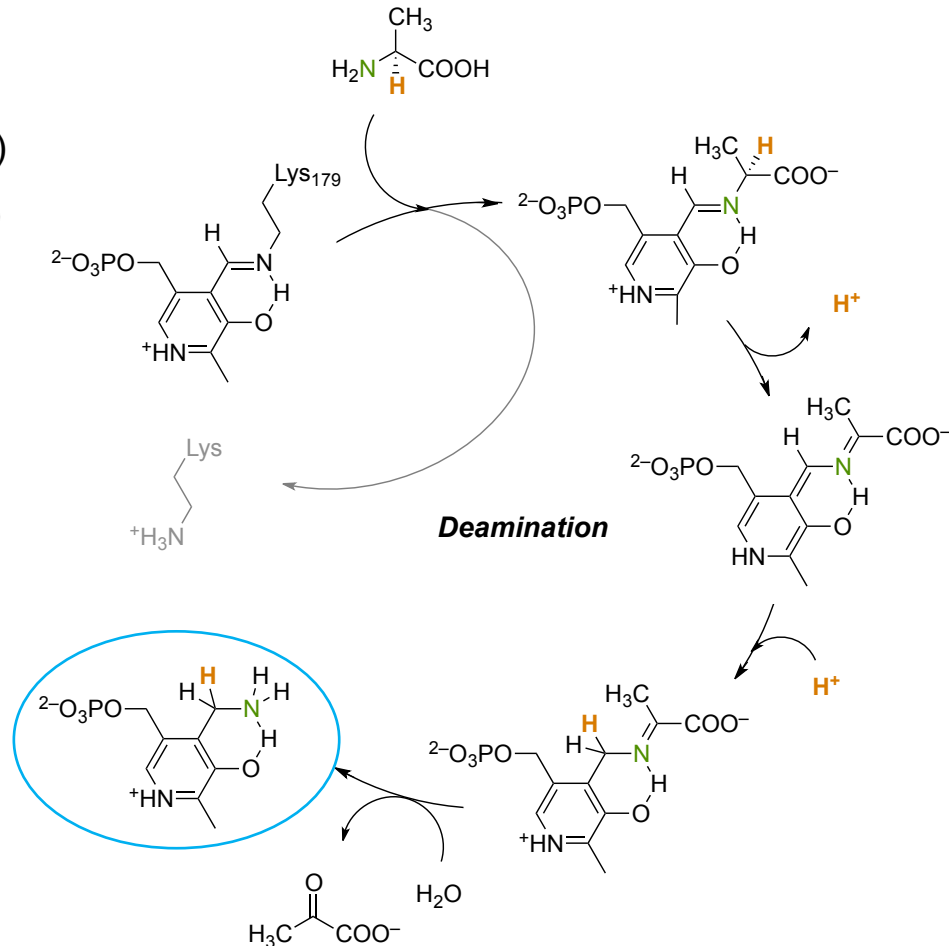
- homo-dimer, subunits à 326 amino acids
- pyridoxal-dependent
- contains structural potassium (purple) and catalytic sodium (blue) ions
- catalyzes aminations and deaminations



ω -Transaminases

mechanism

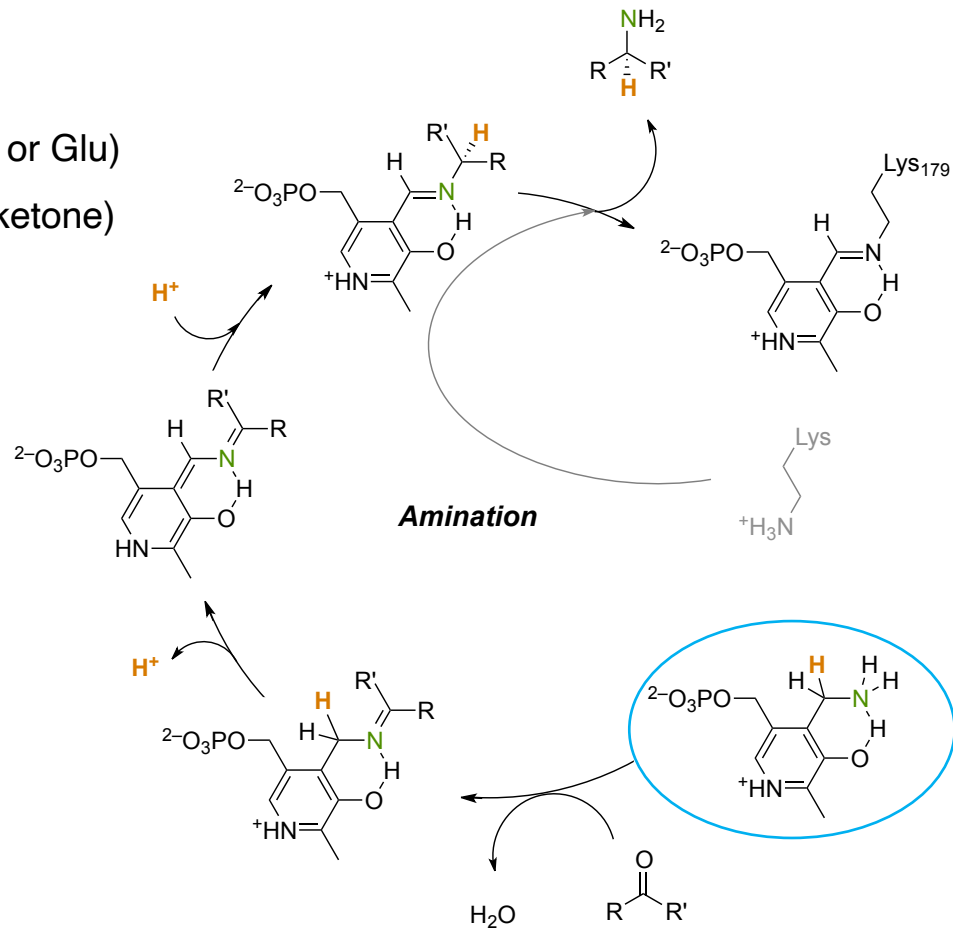
- deamination of donor (Ala or Glu)
+ amination of substrate (ketone)



ω -Transaminases

mechanism

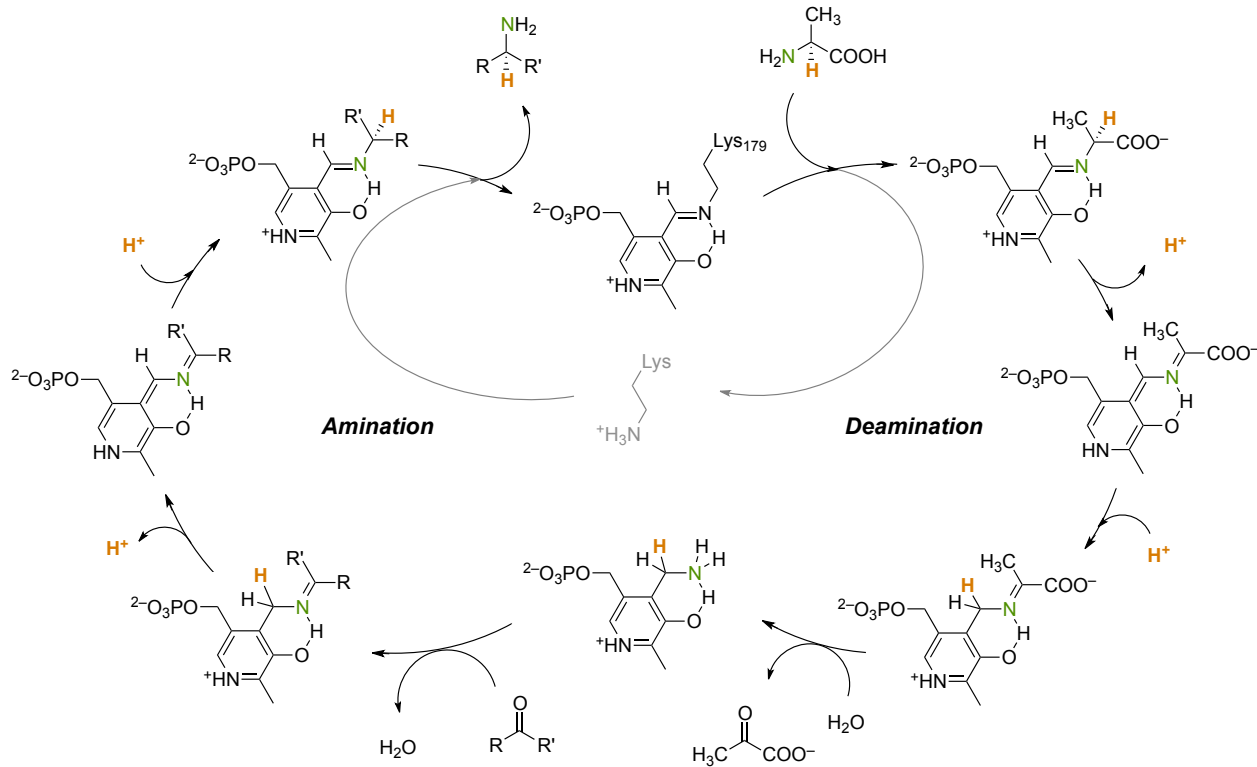
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ω -Transaminases

mechanism

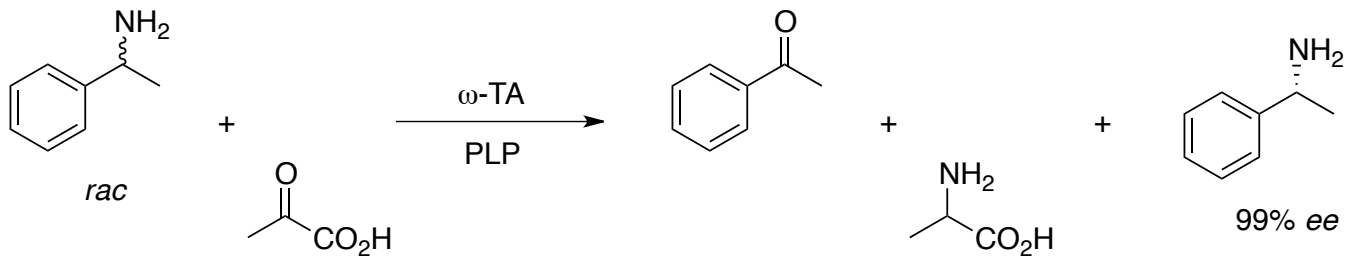
- deamination of donor (Ala or Glu) + amination of substrate (ketone)



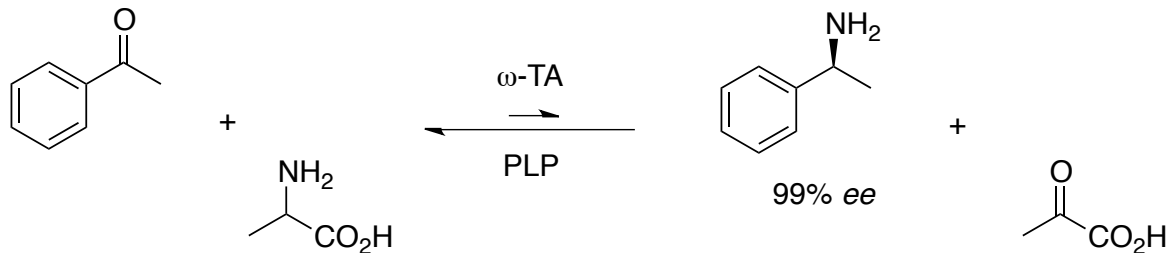
ω -Transaminases

concepts for the application of ω -TAs

- enantiodiscriminative deamination = kinetic resolution



- asymmetric amination = enantioselective synthesis

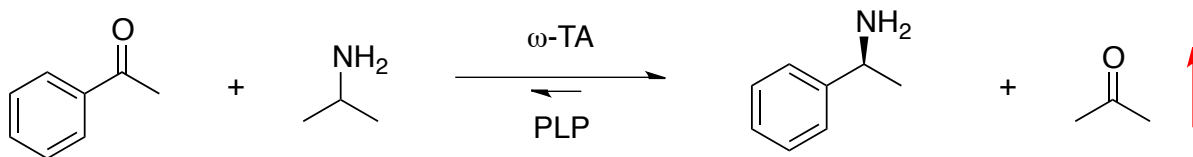


unfavourable equilibrium!

ω -Transaminases

equilibrium shift towards the desired amine products

volatile co-products

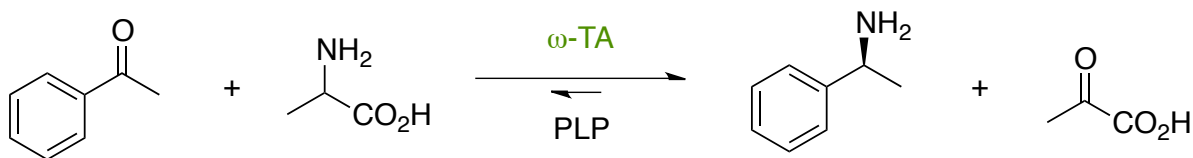


- isopropylamine is deaminated to acetone
- acetone is less soluble in water, thus can be removed by slight vacuum, elevated temperature or N_2 -purge
- requires relatively high concentrations of isopropylamine

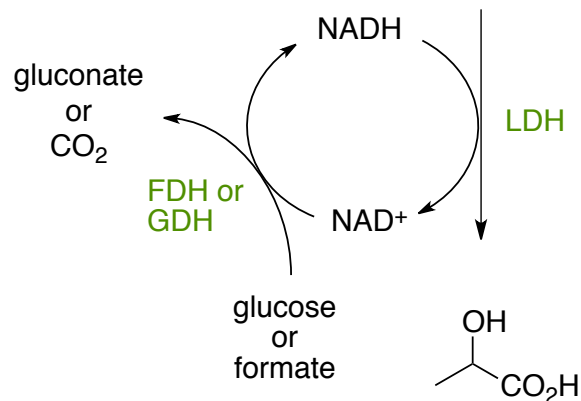
ω -Transaminases

equilibrium shift towards the desired amine products

enzyme-coupled pyruvate removal



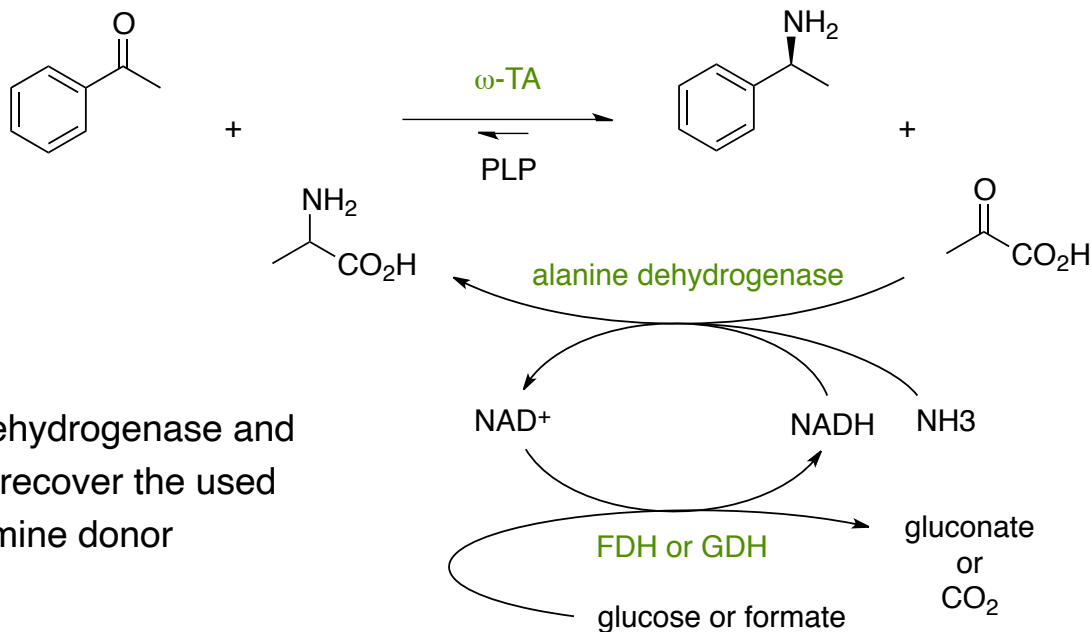
- lactate dehydrogenase (LDH) removes pyruvate
- requires cofactor recycling systems such as GDH and FDH to provide sufficient NADH



ω -Transaminases

equilibrium shift towards the desired amine products

enzyme-coupled alanine recycling

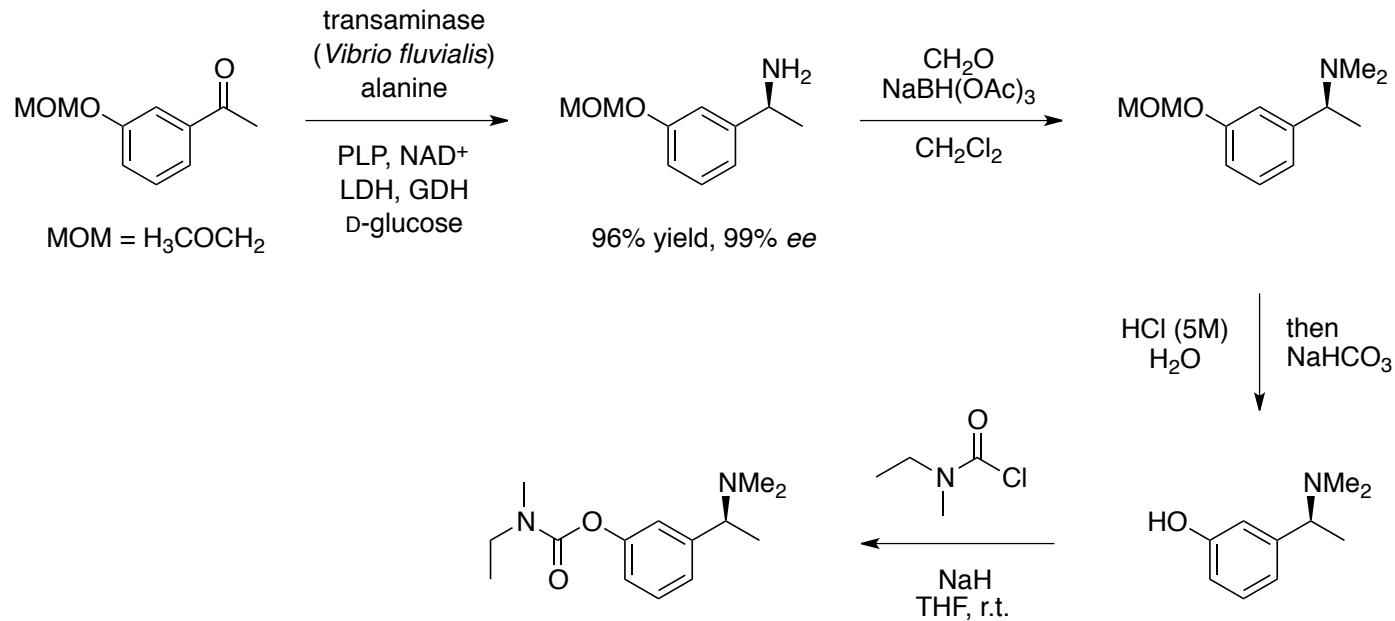


- alanine dehydrogenase and ammonia recover the used alanine amine donor

- relatively high price of AlaDH

ω -Transaminases

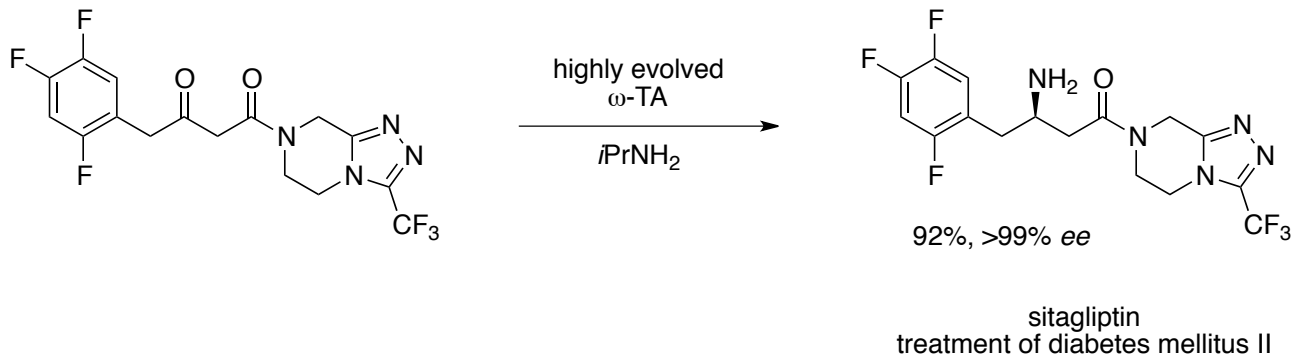
synthetic applications



rivastigmine ("Exelon")
treatment of dementia, Alzheimer's and Parkinson's

ω -Transaminases

synthetic applications



- late-stage biotransformation
- up to 200 g/L
- original industrial process: ruthenium-catalyzed enamine reduction

SUMMARY

amine oxidases

- niche biocatalysts with some quite unique applications
- flavin-dependent mechanism with similarities to Baeyer-Villiger monooxygenases

transaminases

- blockbuster enzyme class
- very developed, first industrial applications
- PLP-dependent mechanism that relies on a series of redox processes
- yet, not classified as oxidoreductase