

Modern applications of cellulose-based fibres and nanofibres

CHEM-E2140 Cellulose-based fibres

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Learning outcomes

After this lecture, the student will be able to:

- List the most prominent (potential) application areas of nanocellulose
- Describe the main challenges in utilizing nanocellulose
- Be aware of the contemporary case studies of nanocellulose and applications



Outline

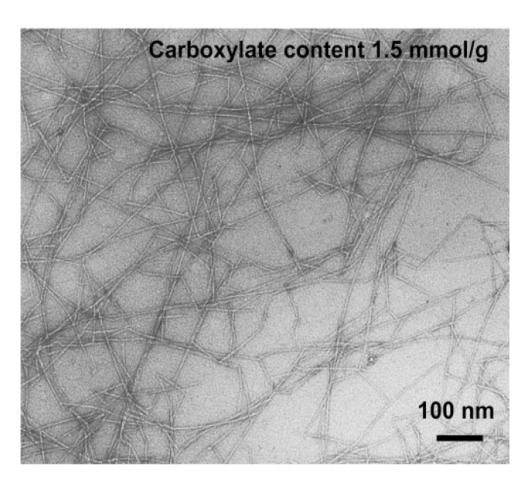
- (1) Nanocellulose: recap from the previous lecture
- (2) Nanopaper and other materials from nanofibrillar cellulose
 - Gas separation
 - Transistor supports
- (3) Cellulose nanocomposites
 - Cellulose nanofibre (CNF) composites
 - Cellulose nanocrystal (CNC) composites
- (4) CNF hydrogels
 - Biomedical applications
 - Solid-state cell factories



Basic types of nanocellulose

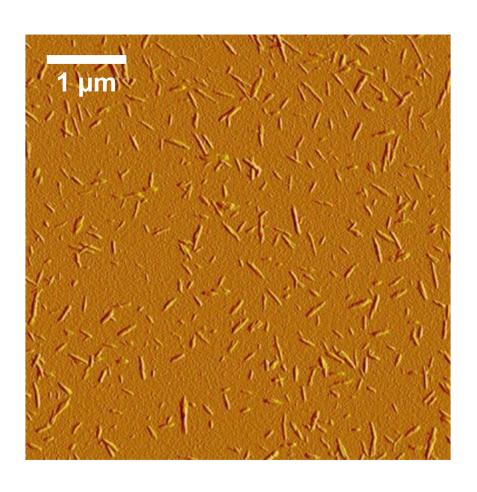
- (1) Cellulose nanofibres (CNFs)
 - Mechanically isolated microfibrils
 - Chemically isolated microfibrils (TEMPO-oxidation)
 - Bacterial cellulose
- (2) Cellulose nanocrystals (CNCs)
 - Rods of highly crystalline cellulose, isolated by acid hydrolysis

Nanofibrillar cellulose (NFC)



- Long threads of isolated cellulose microfibrils
- Very high aspect ratio
- Length: 0.5-5 µm
- Width: 3-50 nm
- Highly charged when prepared by chemical isolation with TEMPOoxidation
- Low charge density when prepared with mechanical isolation

Cellulose nanocrystals (CNCs)



- Rigid rods of crystalline cellulose
- Length: 50-1000 nm
- Width: 3-20 nm
- Usually charged with sulphate groups on the crystal surface
- Suspensions form spontaneously chiral nematic liquid crystal phases

Existing commercial applications of nanocellulose



Adult diapers (nanofibres, Japan)



Viscosity control in oil drilling (nanocrystals, Canada)



Ink dispersing in ball point pens (nanofibres, Japan)

Existing commercial applications from nanocellulose



Hydrogels for tissue growth applications (nanofibres, Finland)

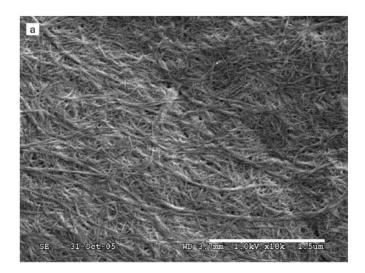
Nanopaper (networks of cellulose nanofibres)

Films from cellulose nanofibres

- Prepared by casting a film of cellulose nanofibres (CNF) by getting rid of the water in CNF suspension / gel in one way or another
- The resulting film (a.k.a. *nanopaper*) is often unusually tough and strong
- The CNF film is often aimed at being optically transparent
- High density of nanopaper results in good gas barrier properties
- Maintaining strength, barrier properties etc. under humid conditions is a current research challenge with nanopaper

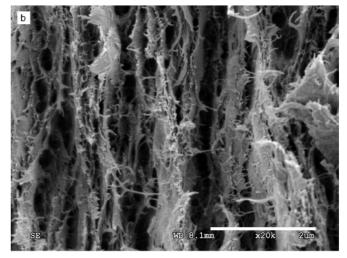


Nanopaper



Nanofibres obtained mechanically from bleached sulphite pulp after enzymatic pretreatment

The first nanopapers were prepared simply by vacuum filtration.

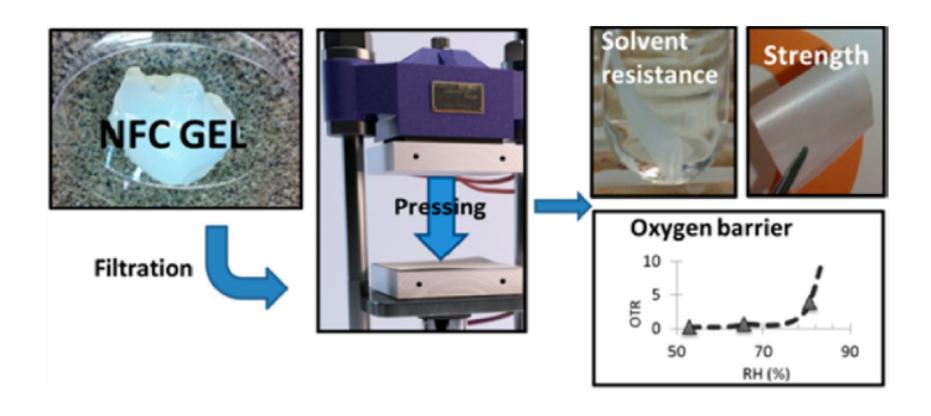


SEM image of the nanopaper surface

SEM image of the nanopaper fracture surface, exposing a layered structure.

Nanopaper preparation

- Generally prepared by a batch process with filtering
- Hot pressing is applied to squeeze residual water out of the CNF network

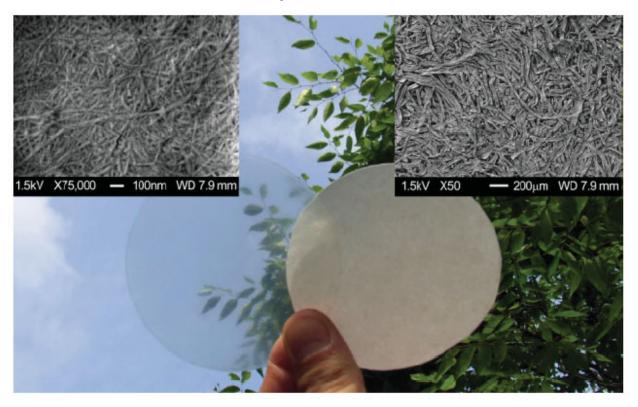




Nanopaper

Nanofibers obtained mechanically from wood fibres after chlorite delignification

Nanopaper

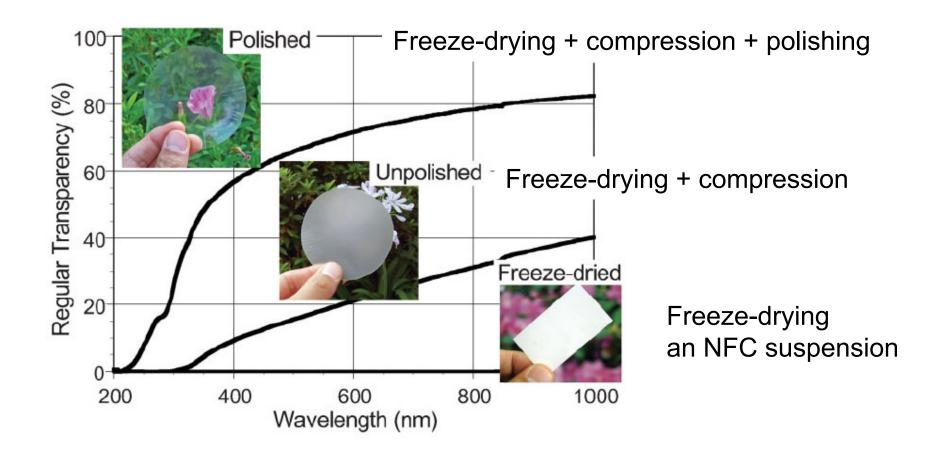


Paper

Figure 1. Optically transparent nanofiber paper (left) composed of 15 nm cellulose nanofibers (upper left, scale bar in inset: 100 nm) and conventional cellulose paper (right) composed of 30 μ m pulp fibers (upper right, scale bar in inset: 200 μ m).



Nanopaper

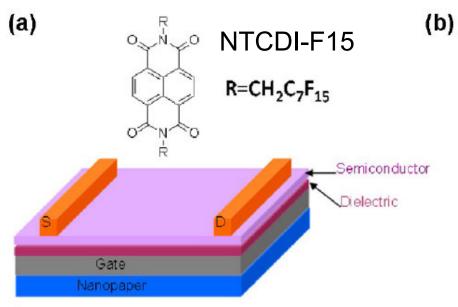


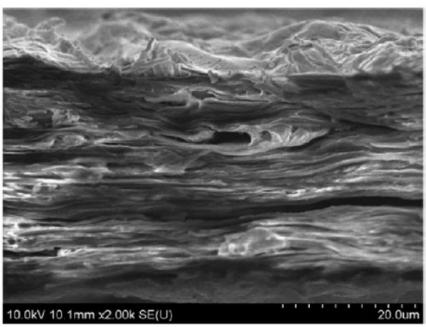
Surface roughness leads to light scattering → removed by polishing



Nanopaper as a transistor support

Cross sectional SEM from transistor





Semiconductor: NTCDI-F15 semiconductor film

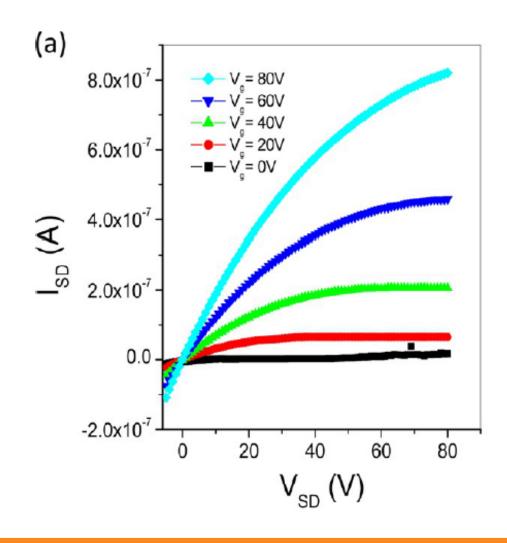
Dielectric: poly(methyl methacrylate) film

Gate electrode: single-walled carbon nanotubes

Nanopaper: film from TEMPO-oxidized NFC

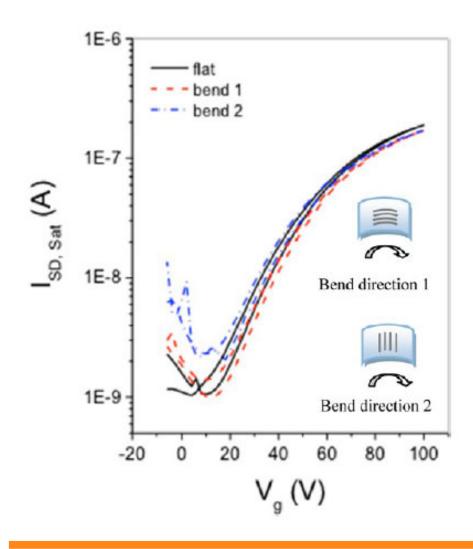


Nanopaper as a transistor support



 The constructed multilayer material works well as a field-effect transistor

Nanopaper as a transistor support



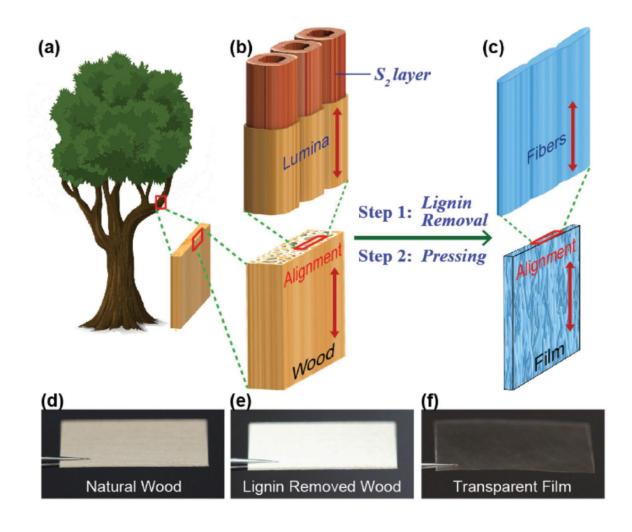
 Bending has little effect on the transistor performance of the material

Comparison of bending transistor supports

characteristics	nanopaper	traditional paper	plastic
surface roughness (nm)	5	5000-10000	5
porosity (%)	20-40	50	0
pore size (nm)	10-50	3000	0
optical transparency	90	20	90
at 550 nm (%)			
max loading stress (MPa)	200-400	6	50
coefficient of thermal expansion (CTE) (ppm K^{-1})	12—28.5	28—40	20—100
printability	good	excellent	poor
Young modulus (GPa)	7.4-14	0.5	2-2.7
bending radius (mm)	1	1	5
renewable	high	high	low



Nanopaper from aligned CNFs

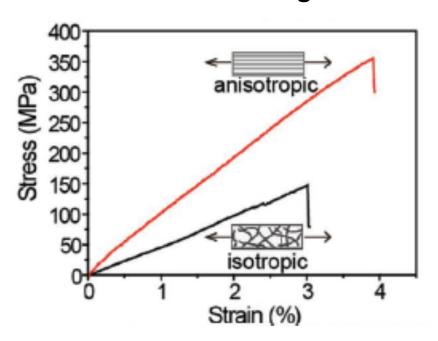


Wood structure of aligned fibres and aligned microfibrils (secondary wall) is preserved upon delignification, resulting in nanopaper of aligned CNFs



Aligned CNF vs. isotropic CNFs

Tensile strength



Isotropic nanopaper sample here may be deliberately downplayed (inferior strength)

- Aligned CNFs lead to a stronger "nanopaper" (350 MPa tensile strength)
- Modern nanopapers usually have tensile strengths at ~200-300 MPa



Cellulose nanocomposites



Nanocomposites – why?

- Nanoparticles are small
- → High surface area
- Stress transfer between reinforcing material and continuous matrix occurs via the interface
- → Much higher stress transfer from reinforcing nanoparticles than from bigger reinforcing particles

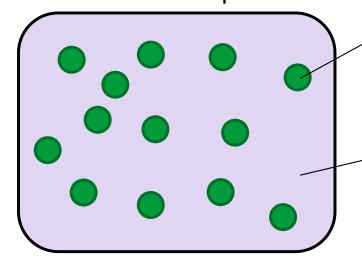


Cellulose nanocomposites

- Cellulose nanocomposites usually consist of cellulose nanofibres or cellulose nanocrystals embedded in a continuous polymer matrix
- The challenge is often to retain the nanometer dimensions of the cellulosic objects (i.e., prevent their aggregation)
- One of the big trends is to end up with materials that would be equal to plastics (or to surpass the properties of plastics)
- Another trend is to prepare stimuli-responsive composites for niche applications



Schematic cross section of a nanocomposite



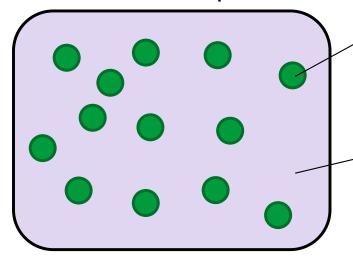
Reinforcing phase: Nanofibril or nanocrystal cross section (cellulose)

Continuous polymer matrix (preferrably from a biodegradable polymer)

OFTEN REFERRED TO SCENARIO: Poly(lactic acid) as a continuous matrix and nanofibrillar cellulose as the reinforcing phase. *Both* are biobased and biodegradable.



Schematic cross section of a nanocomposite



Reinforcing phase: Nanofibril or nanocrystal cross section (cellulose)

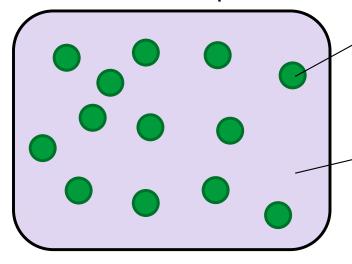
Continuous polymer matrix (preferrably from a biodegradable polymer)

WHY CELLULOSE?

Poly(lactic acid) is brittle and weak on its own. Cellulose with strong mechanical properties would boost its strength.



Schematic cross section of a nanocomposite



Reinforcing phase: Nanofibril or nanocrystal cross section (cellulose)

Continuous polymer matrix (preferrably from a biodegradable polymer)

WHY NANOCOMPOSITES?

Nanocellulose has a very large surface area, that is, a lot of contact points with the continuous poly(lactic acid) matrix, plus it is very strong.



$$\begin{bmatrix}
CH_3 & O \\
 & | \\
 & | \\
 & C - C - O
\end{bmatrix}$$

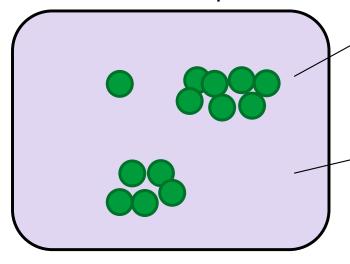
Poly(lactic acid)

Cellulose

Dissolves in hydrophobic solvents

Dissolves in very few solvents

Schematic cross section of a nanocomposite



Reinforcing phase: Nanofibril or nanocrystal cross section (cellulose)

Continuous polymer matrix (preferrably from a biodegradable polymer)

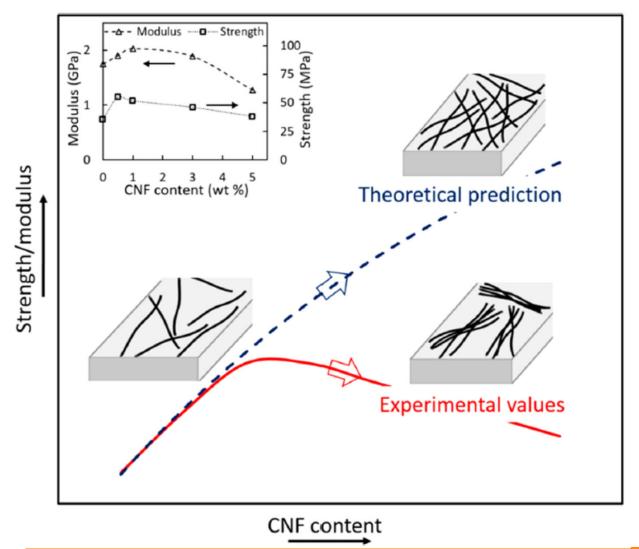
WHAT IS THE PROBLEM?

Cellulose aggregates easily with itself. It is incompatible with nearly anything else than the lignin/hemicellulose matrix in the plant cell wall.

 \rightarrow Loss in surface area \rightarrow No nanocomposite anymore



Aggregation of nanocellulose



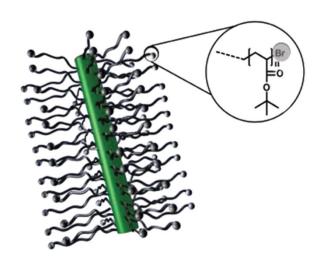
Because of aggregation, nanocellulose reinforcement often works well only with low nanocellulose contents.



Nanocellulose modification for nanocomposites

How does one solve the problem?

One approach: modify the surface of nanocellulose



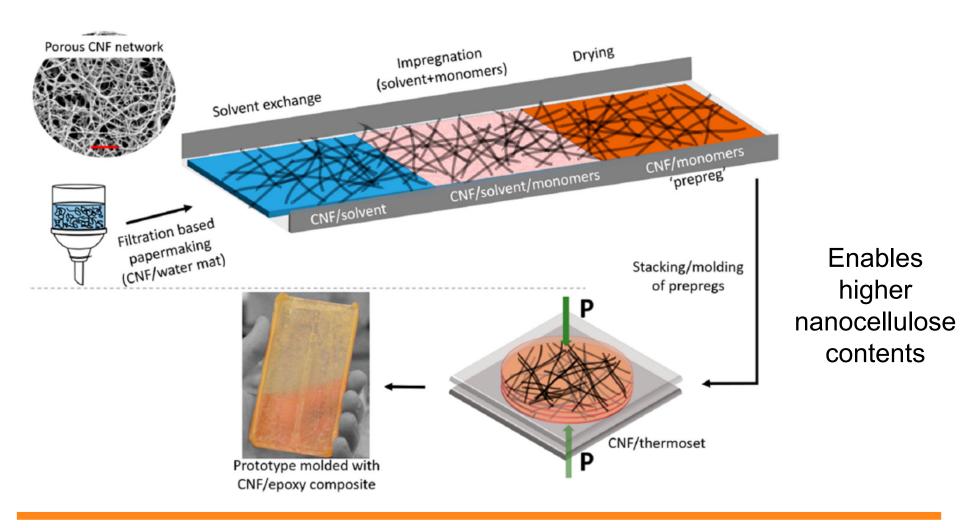
Surface modifiation:

- Surface of nanocellulose is more compatible with the surrounding polymer matrix in nanocomposites
- Crystalline core of cellulose stays intact
- → Strength properties of nanocellulose stay intact

At the moment, the methods to do this are expensive and not scalable.



Another solution: preparing nanocellulose network before composite preparation

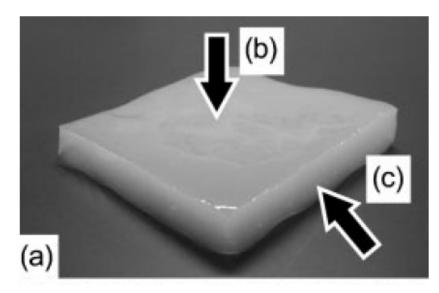




Composites with nanofibrillar cellulose



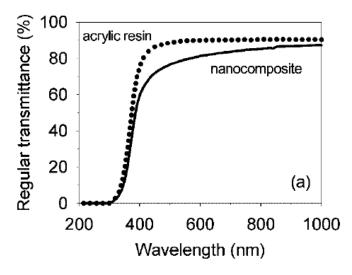
Acrylic resin / bacterial cellulose

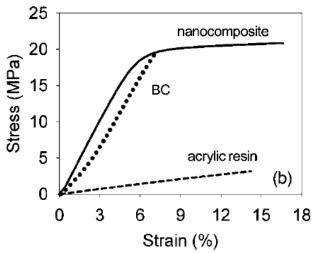


Bacterial cellulose pellicle

Water in a bacterial cellulose pellicle is replaced gradually with ethanol and impregnated with acrylic resin which is subsequently cured.

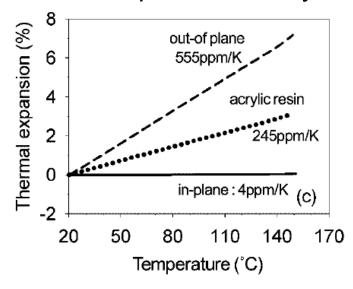
Acrylic resin / bacterial cellulose





Light transmittance of the resin is retained while the mechanical properties are enhanced.

Thermal expansion is very low.



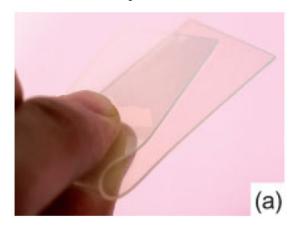


Acrylic resin / bacterial cellulose



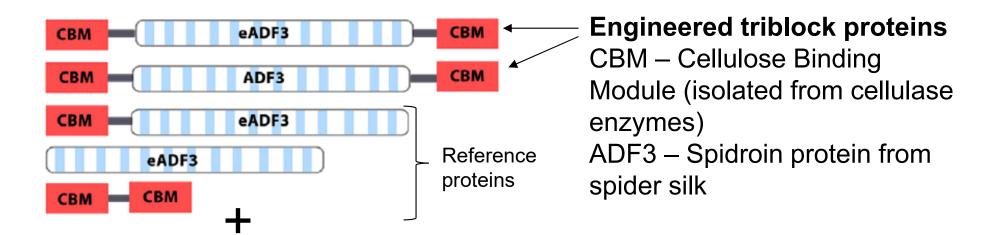
Figure 1. Luminescence of an organic light-emitting diode deposited onto a transparent BC nanocomposite. The luminescence area is $40 \times 25 \text{ mm}^2$. This work was carried out in collaboration with Mitsubishi Chemical Corporation and Pioneer Corporation.

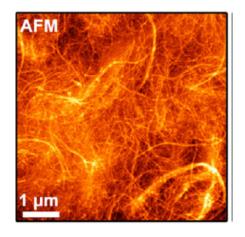
Potential usage for foldable flat panel displays in the electronics device industry.



Most plastics have too large a thermal expansion for electronics devices.

CNF with silk proteins



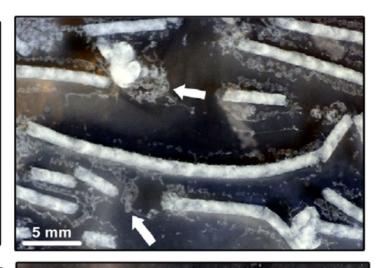


Wood-based CNF



CNF with silk proteins: fibre extrusion

eADF3 & CNF



Proteins without CBM block

- → Poor adhesion between CNF and protein
- → Fragmented fibres

CBM-eADF3-CBM & CNF

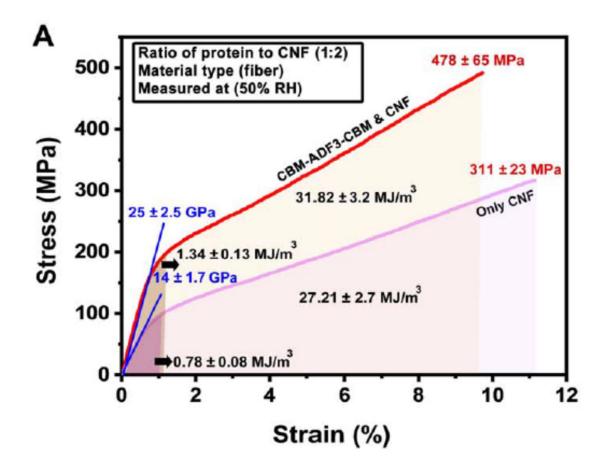


Ratio CBM-eADF3-CBM:CNF 1:2

Proteins with CBM block

- → Good adhesion between CNF and protein
- → Fibres with good integrity

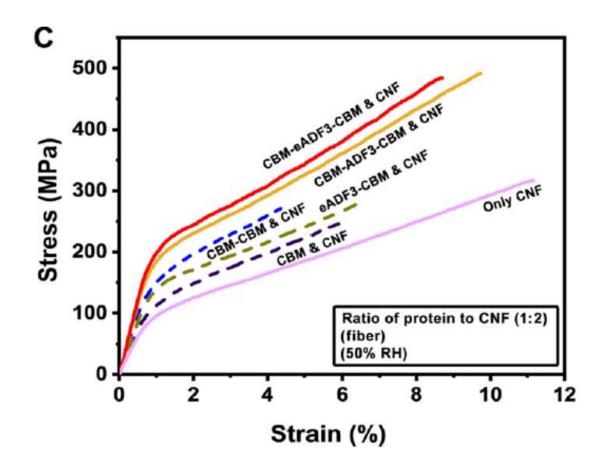
CNF with silk proteins



Composite fibres have a far higher tensile strength than pure CNF fibres



CNF with silk proteins



Triblock proteins with cellulose binding domains (CBM) are necessary for high strength

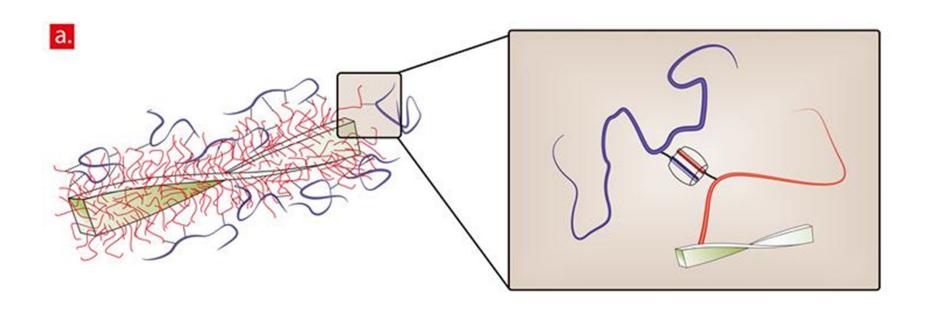


Composites with cellulose nanocrystals

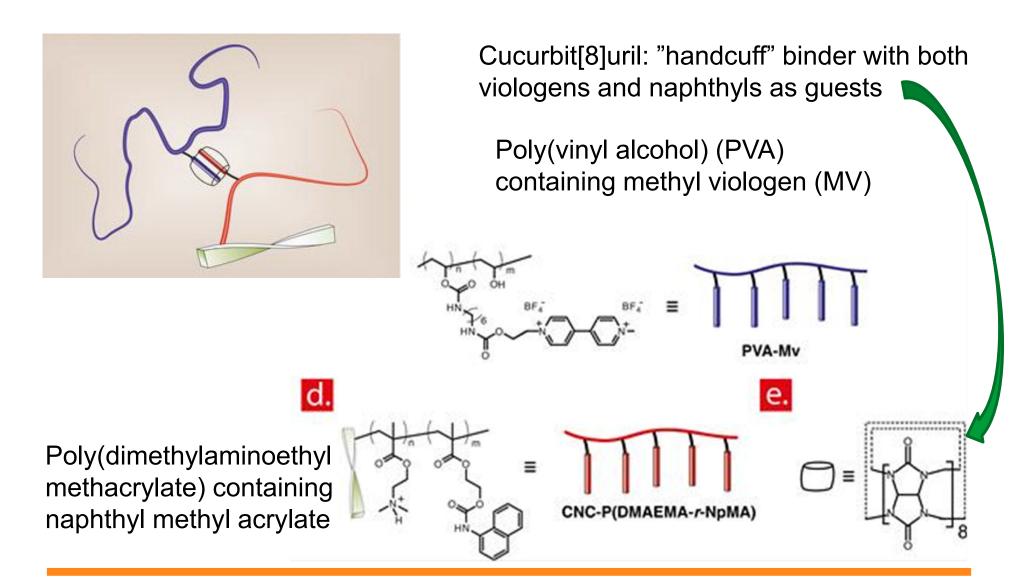


Basic idea:

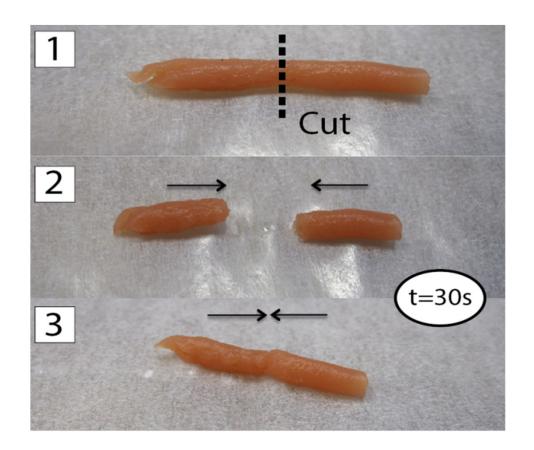
- (i) Graft polymers on CNCs
- (ii) Blend grafted CNCs within a continuous polymer network
- (iii) Use strong cross linking agents that enable strength and self-healing





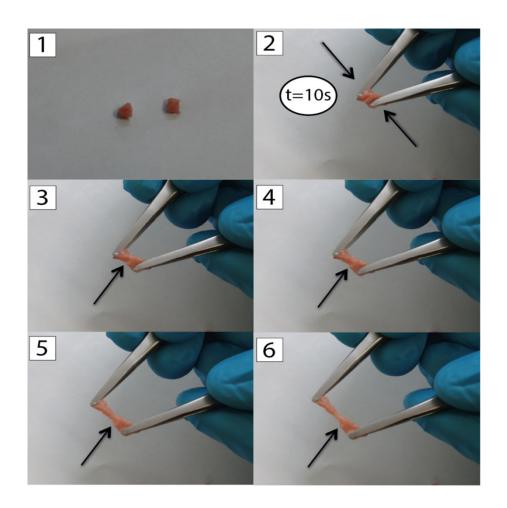






Self-healing immediately after cutting:

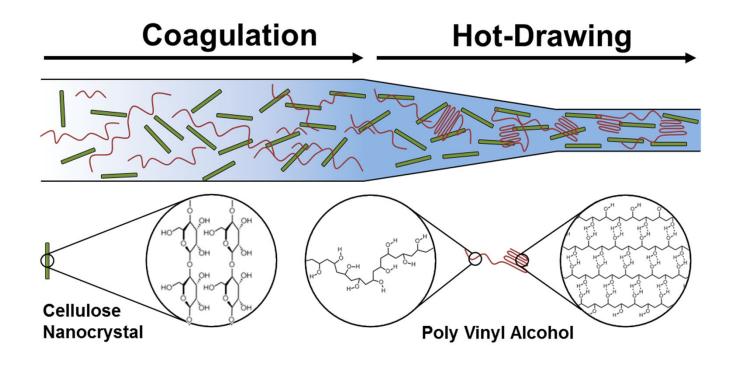
Supramolecular binding with cucurbit[8]uril is fast and strong.



Self-healing after 4 months storage:

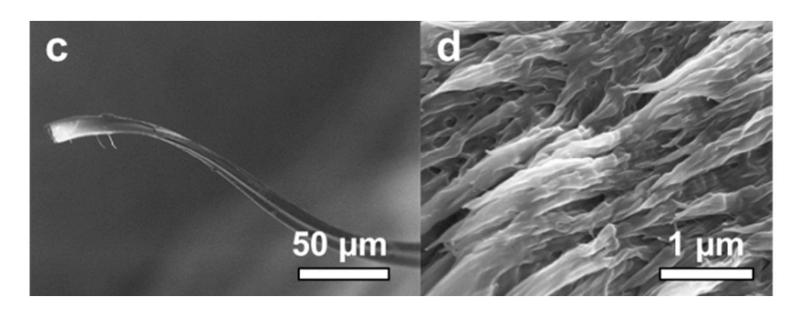
The mechanism with cucurbit[8]uril still works and it is still very fast.

- PVA with loads of hydroxyl groups is a compatible matrix with CNCs
- ightarrow No compatibilizing modifications on CNCs required





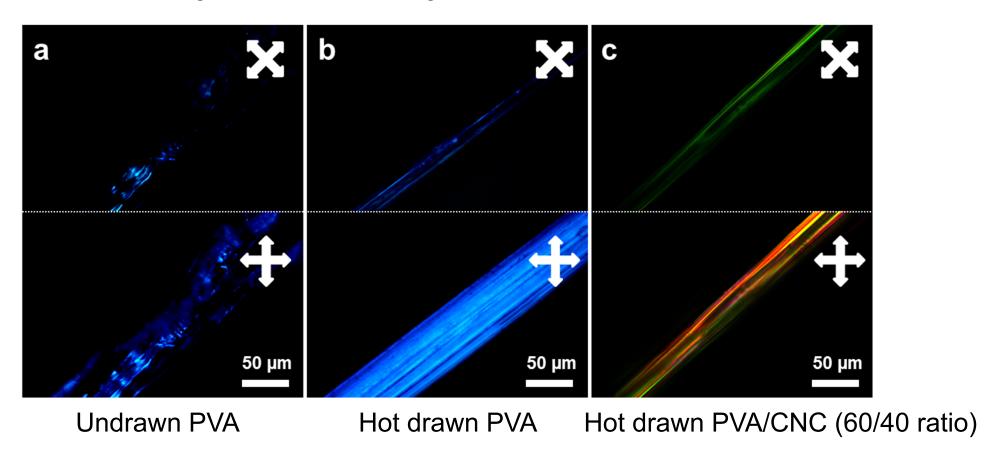
Result: CNC/PVA composite fibers with tens of µm diameter



- CNC concentration upon fibre spinning is fairly high
- → Above the concentration for liquid crystal formation
- → Liquid crystal phase induces alignment of CNCs in resulting fibres

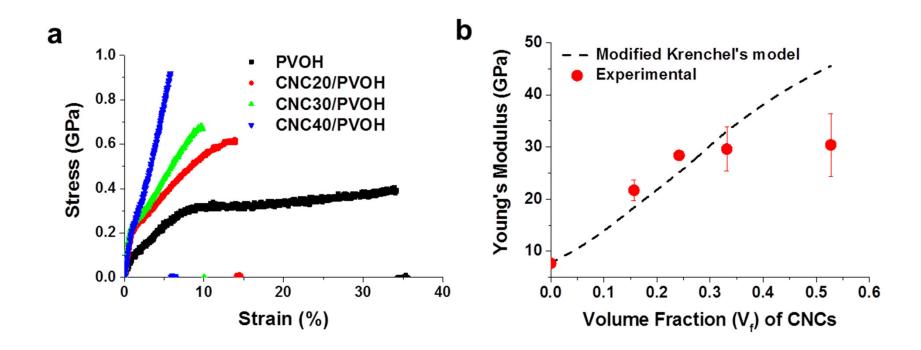


Hot drawing further induces alignment and order in PVA/CNC fibres





 Very high mechanical properties for the resulting PVA/CNC fibres: tensile strength of close to 1 MPa with 40% CNC loading





Applications of nanocellulose hydrogels

Nanocellulose hydrogels

- Particularly CNFs form gels at low concentrations in water
- Gel formation can be an advantage, utilized in specific applications
- Most biomedical applications utilise the gel formation properties of CNF

CNF hydrogels as scaffolds for tissue engineering

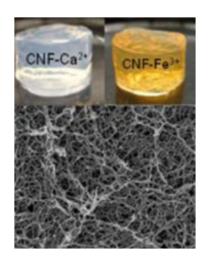
- Tissue engineering scaffolds provide 3D foundation to direct cellular attachment, proliferation, and differentiation ultimately tissue formation
- Requirements: biocompatibility, non-toxicity, biodegradability, sufficient (tunable) porosity and mechanical properties

CNF hydrogels as tissue scaffolds:

- Cellulose lacks the adhesive sites necessary for cell signalling and migration
- Tuning the porosity and mechanical properties is not always straightforward



CNF hydrogels as tissue growth scaffold



Cross-linking TEMPO-oxidized CNFs by multivalent metal ions

→ Stronger gels

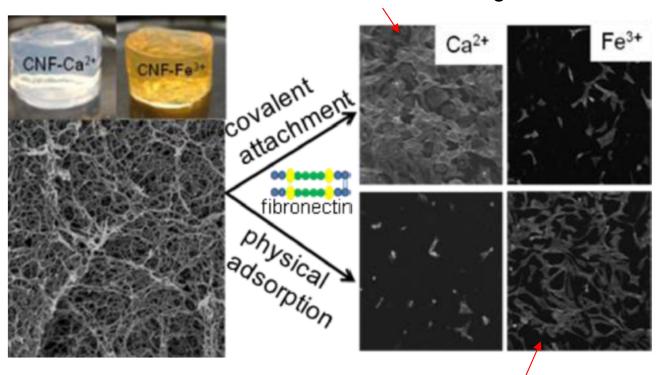
Physical adsorption or covalent attachment of fibronectin protein

→ Supports biorecognition

CNF hydrogels as tissue growth scaffold: fibroblast cell growth

Good cell proliferation:

Ca cross linking + covalent fibronectin attachment



Good cell proliferation:

Fe cross linking + physical fibronectin attachment

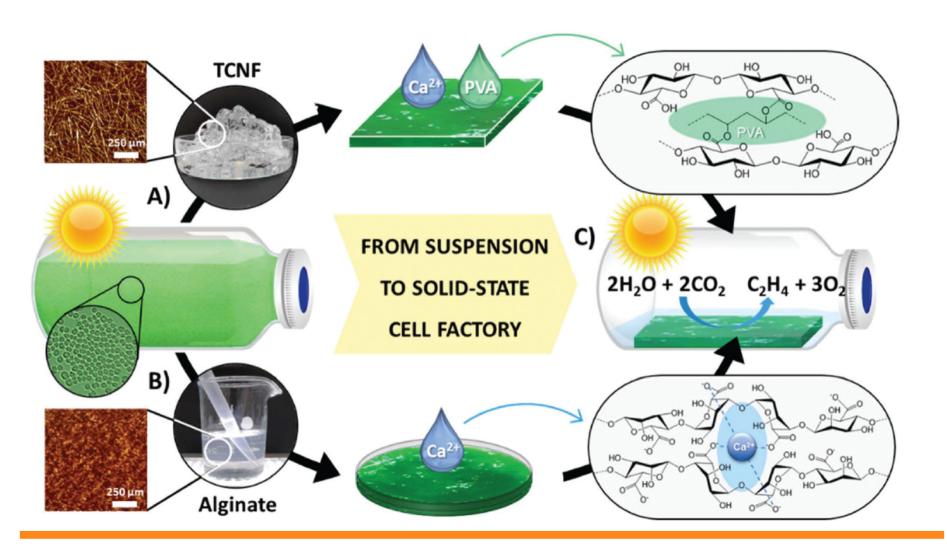


Solid-state cell factories

- Photosynthetic cell factories are platforms where microbial cells of cyanobacteria or green algae are used as biocatalysts to produce biofuels or platform chemicals
- They generally consist of pools of water where the cells are located (suspension culture)
- Application for nanocellulose hydrogels: "solid state" cell factories where the photosynthetic cells are trapped in a nanocellulose gel matrix
 - Enables better control over the chemicals production and collection
 - Harmful proliferation of cells (instead of chemical production) can be better controlled
 - Enables better control over light-to-production efficiency

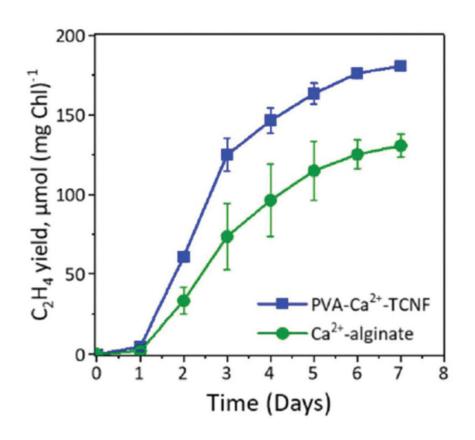


Solid state cell factories





Solid state cell factories



Production of ethylene by photosynthetic cells:

 Higher conversion by cross linked CNF matrix than by alginate matrix

(Note: alginate is currently the only other option for "solid state" cell factories)



Conclusions

- Materials incorporating nanocellulosics is a highly visible research area
- Some applications require just film formation (e.g., gas barriers, transistor supports), some require complex chemistry and recognition mechanisms (e.g., self-healing composites, tissue growth scaffolds)

Examples of versatility in research:

- Transparent films
- Plastic surrogates
- Supports for electronics
- Tissue growth scaffolds
- Solid state cell factories



Conclusions

Nanocellulose:

- High potential: high strength, bio-based origin, biodegradability, nontoxicity
- Difficulties: compatibility with composite matrix, realisation of potential, susceptibility to water
- Native properties of cellulose in the plant cell wall are underutilised

