

Modern applications of nanocellulose



Aalto University
School of Chemical
Engineering

CHEM-E2140

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Learning outcomes

After this lecture, the student will be able to:

- **List the most prominent (potential) application areas of nanocellulose**
- **Describe the main challenges in utilizing nanocellulose**
- **Be aware of the contemporary case studies of nanocellulose and applications**

Outline

(1) Nanocellulose: recap from the previous lecture

(2) Nanopaper and other materials from nanofibrillar cellulose

- Gas separation
- Transistor supports

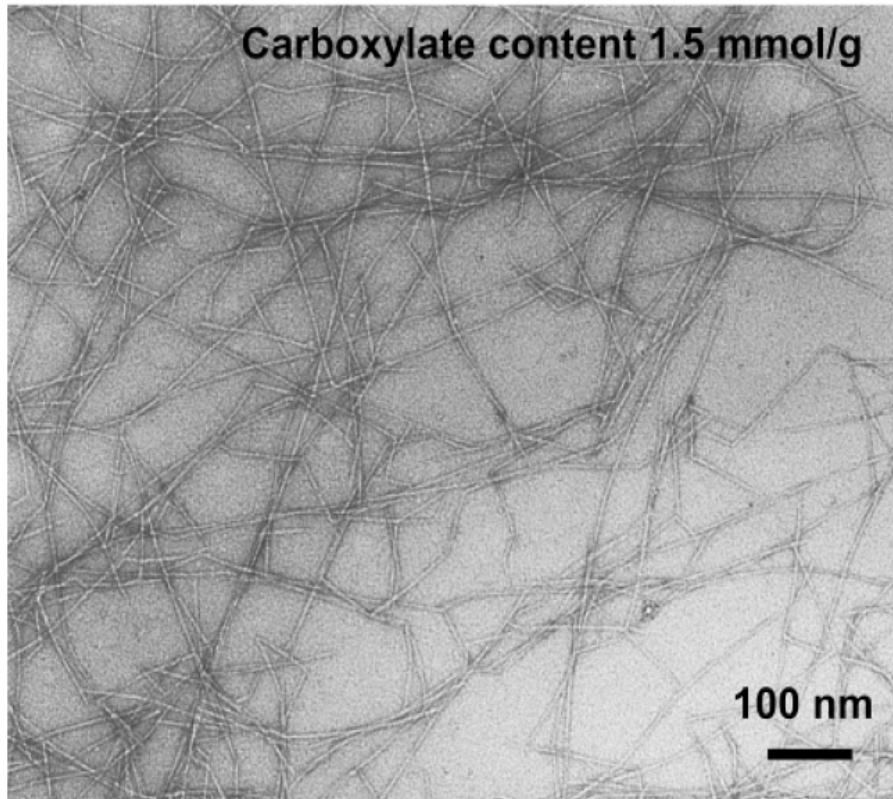
(3) Cellulose nanocomposites

- Cellulose nanofibre (CNF) composites
- Cellulose nanocrystal (CNC) composites

(4) CNF hydrogels

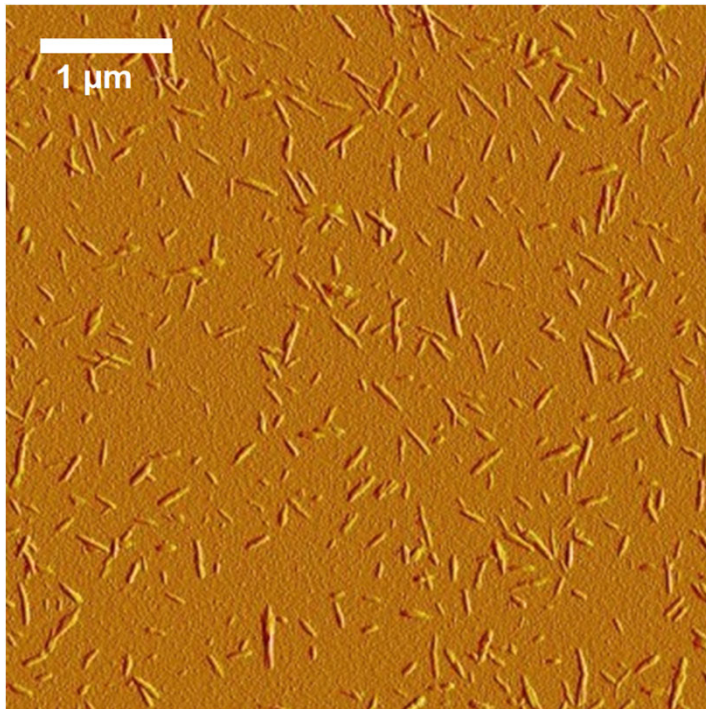
- Biomedical applications
- Solid-state cell factories

Cellulose nanofibres (CNF) - recap



- Long threads of isolated cellulose microfibrils
- Very high aspect ratio
- Length: 0.5-5 μm
- Width: 3-50 nm
- Highly charged when prepared by chemical isolation with TEMPO-oxidation
- Low charge density when prepared with mechanical isolation

Cellulose nanocrystals (CNCs) - recap



- Rigid rods of crystalline cellulose
- Length: 50-1000 nm
- Width: 3-20 nm
- Usually charged with sulphate groups on the crystal surface
- Suspensions form spontaneously chiral nematic liquid crystal phases

Existing commercial applications of nanocellulose



Adult diapers
(nanofibres, Japan)



Viscosity control in
oil drilling
(nanocrystals,
Canada)



Ink dispersing in
ball point pens
(nanofibres, Japan)

Existing commercial applications from nanocellulose



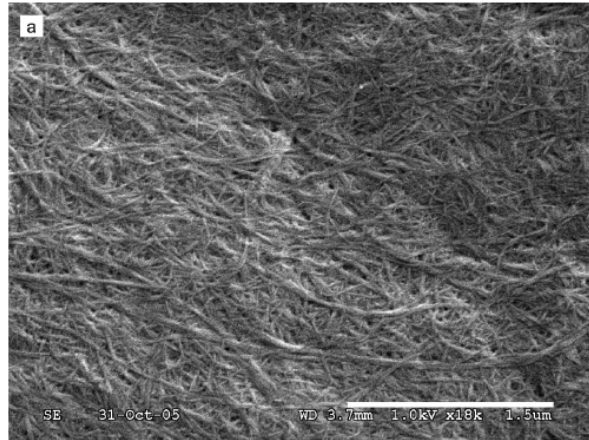
Hydrogels for tissue
growth and wound
healing applications

Nanopaper: network of cellulose nanofibres

Nanopaper

- Prepared by casting a film of cellulose nanofibres (CNF) by getting rid of the water in CNF suspension / gel
- The resulting film (a.k.a. *nanopaper*) is often unusually tough and strong
- The CNF film is often aimed at being optically transparent
- High density of nanopaper results in good gas barrier properties
- Maintaining strength, barrier properties etc. under humid conditions is a current research challenge with nanopaper

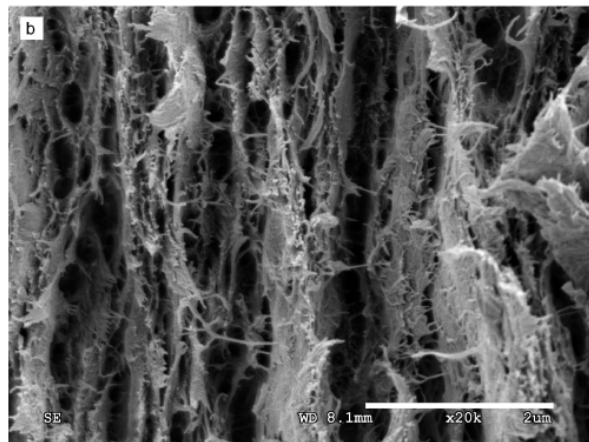
Nanopaper



Nanofibres obtained mechanically from bleached sulphite pulp after enzymatic pretreatment

The first nanopapers were prepared simply by vacuum filtration.

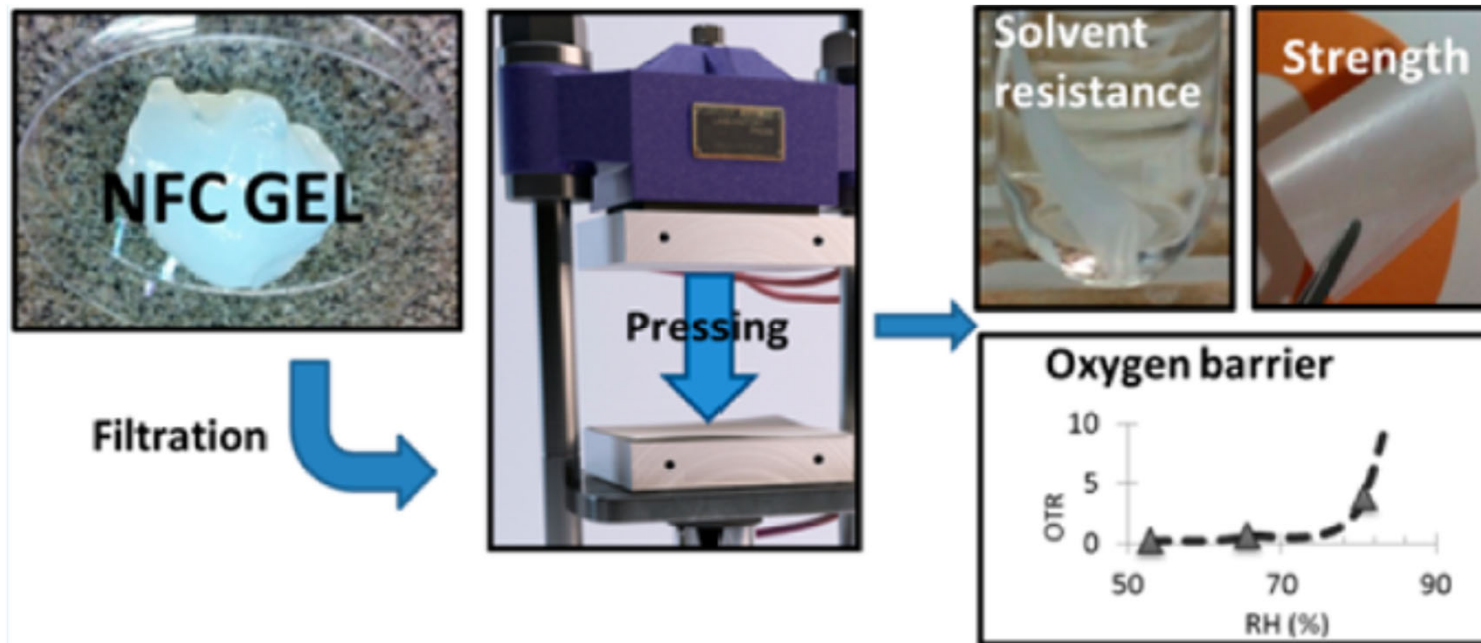
SEM image of the nanopaper surface



SEM image of the nanopaper fracture surface, exposing a layered structure.

Nanopaper: advanced preparation

- Generally prepared by a batch process with filtering
- Hot pressing is applied to squeeze residual water out of the CNF network

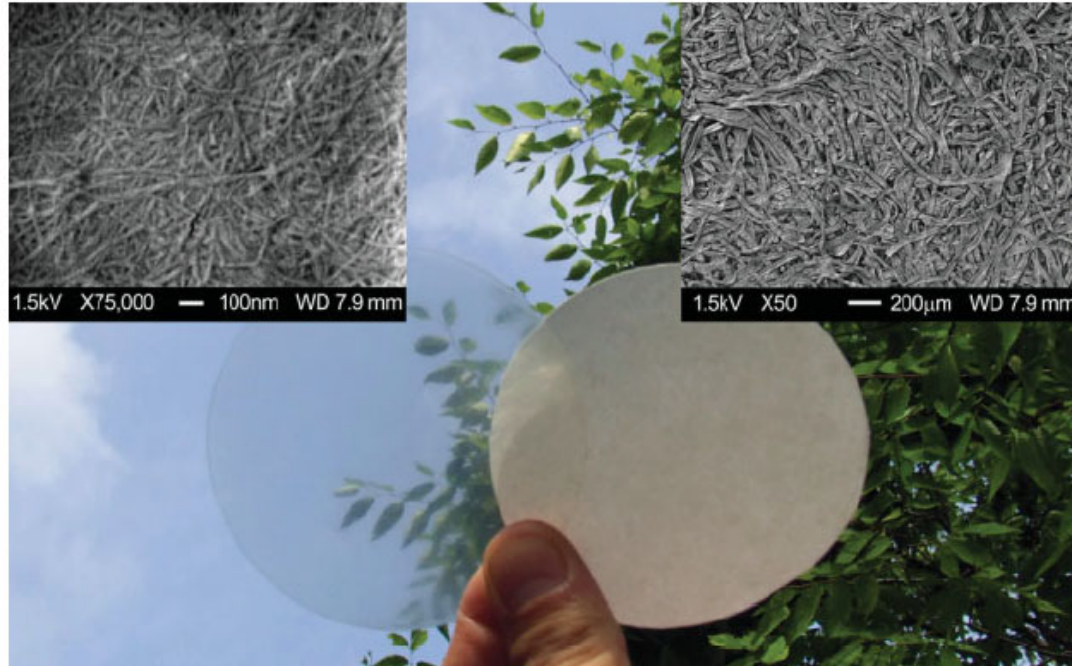


Nanopaper: transparency

Nanofibers obtained mechanically from wood fibres after chlorite delignification

When CNFs are narrow enough and non-aggregated, nanopapers can be transparent

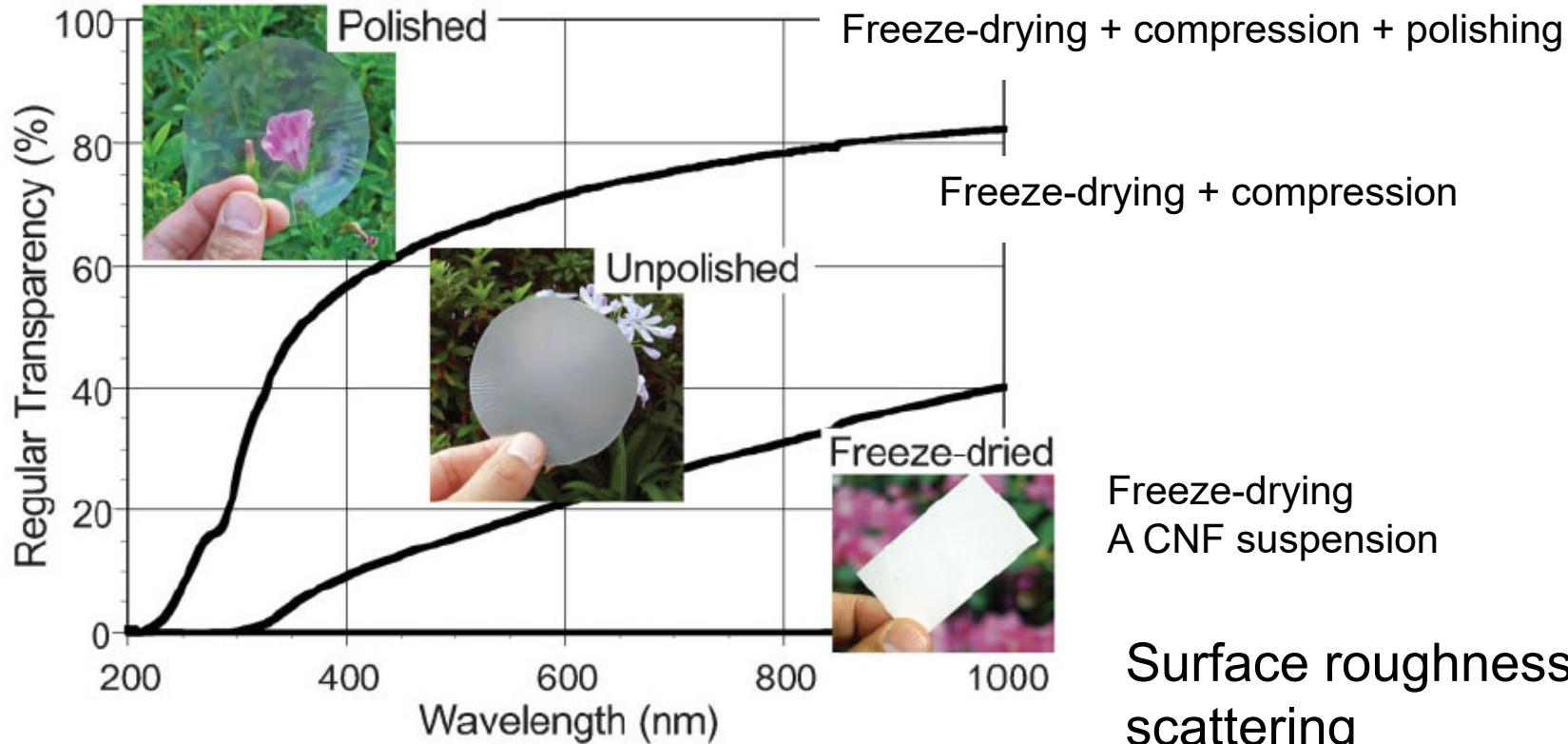
Nanopaper



Paper

Figure 1. Optically transparent nanofiber paper (left) composed of 15 nm cellulose nanofibers (upper left, scale bar in inset: 100 nm) and conventional cellulose paper (right) composed of 30 μm pulp fibers (upper right, scale bar in inset: 200 μm).

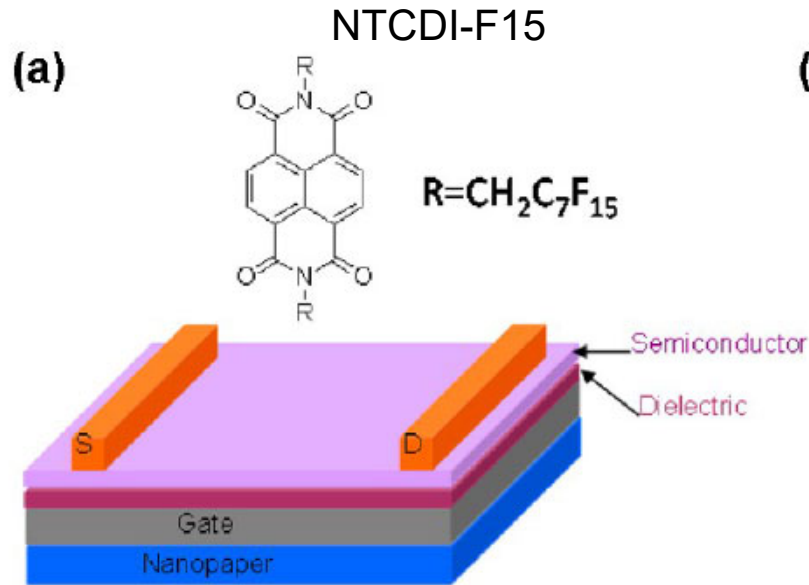
Nanopaper: transparency



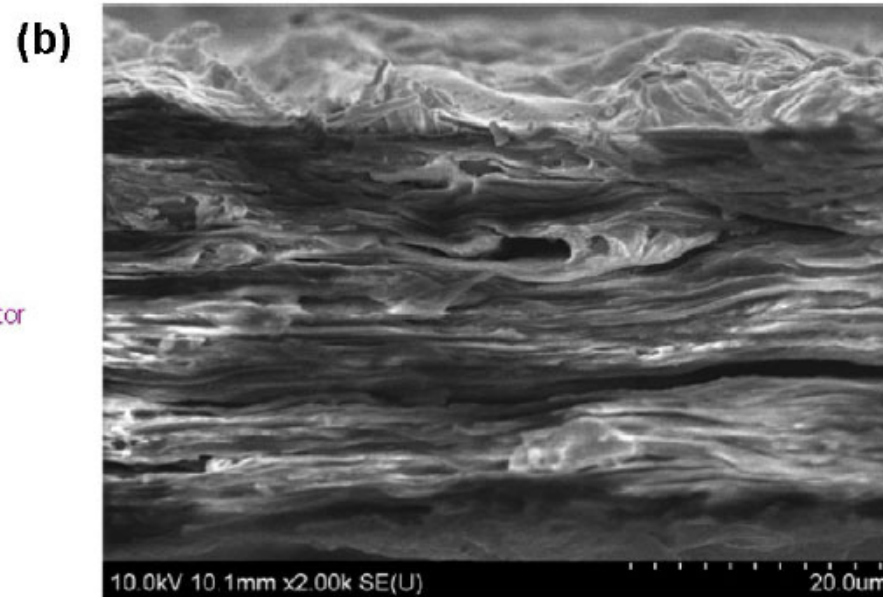
Freeze-drying
A CNF suspension

Surface roughness leads to light scattering
→ removed by polishing

Nanopapers: transistor supports



Cross sectional SEM from transistor



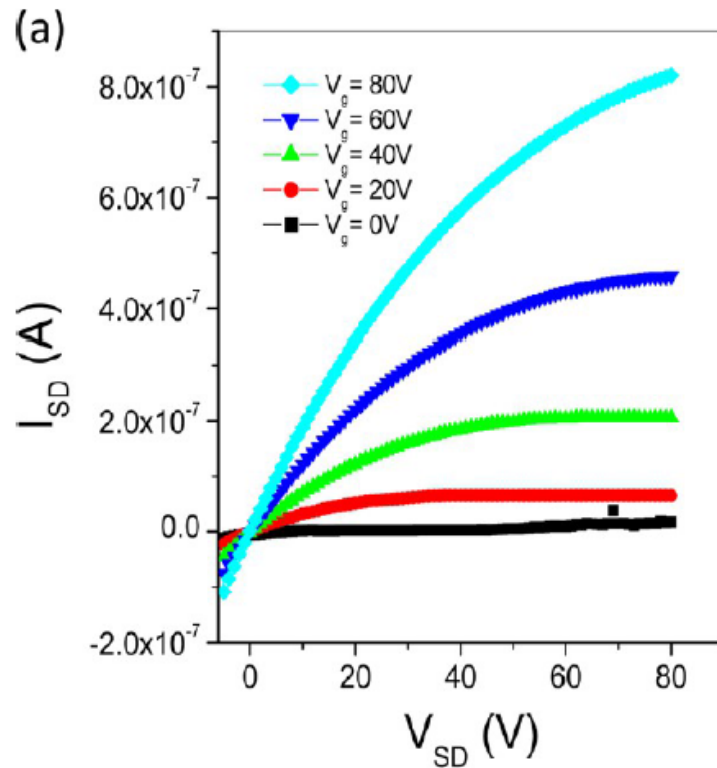
Semiconductor: NTCDI-F15 semiconductor film

Dielectric: poly(methyl methacrylate) film

Gate electrode: single-walled carbon nanotubes

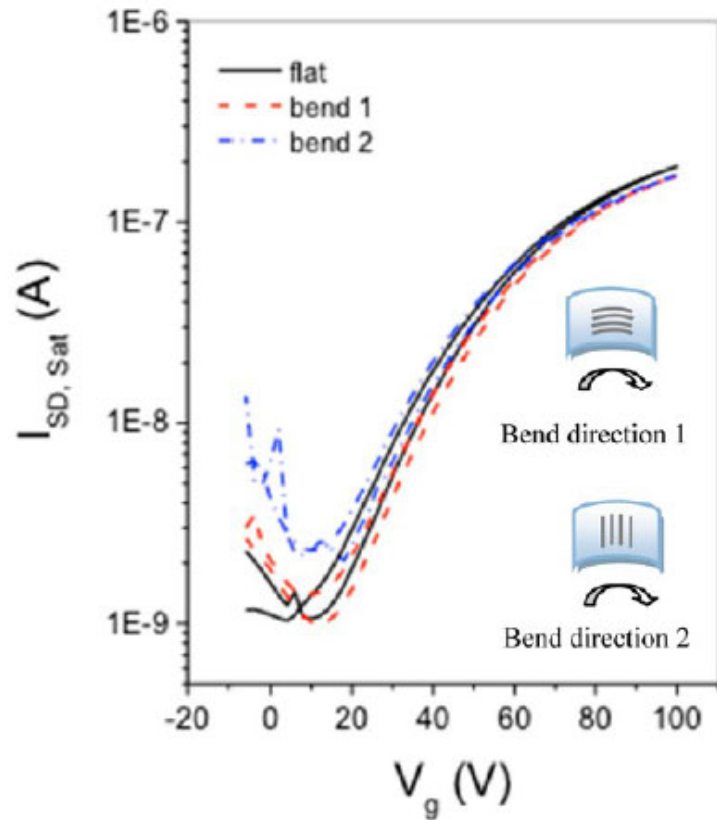
Nanopaper: film from TEMPO-oxidized NFC

Nanopapers: transistor support



- The constructed multilayer material works well as a field-effect transistor

Nanopapers: transistor support



- Bending has little effect on the transistor performance of the material

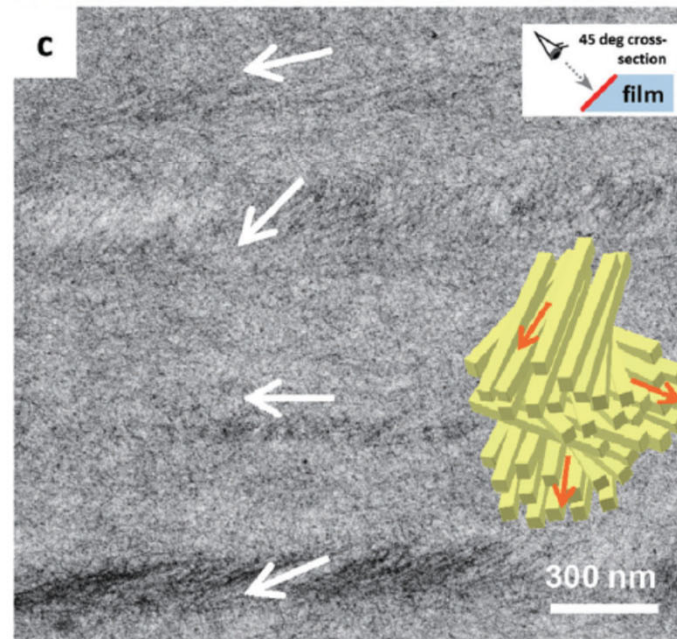
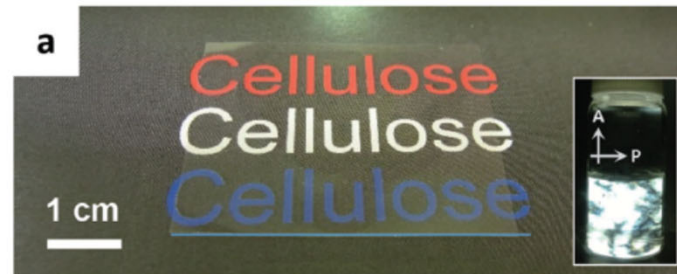
Comparison with bending transistor supports

characteristics	nanopaper	traditional paper	plastic
surface roughness (nm)	5	5000—10000	5
porosity (%)	20—40	50	0
pore size (nm)	10—50	3000	0
optical transparency at 550 nm (%)	90	20	90
max loading stress (MPa)	200—400	6	50
coefficient of thermal expansion (CTE) (ppm K ⁻¹)	12—28.5	28—40	20—100
printability	good	excellent	poor
Young modulus (GPa)	7.4—14	0.5	2—2.7
bending radius (mm)	1	1	5
renewable	high	high	low

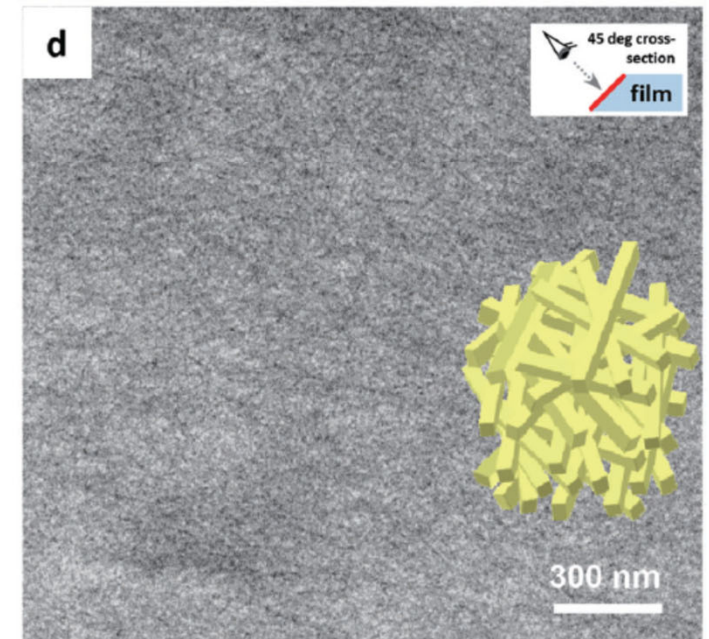
Nanopaper from aligned CNF

Nematic liquid crystal order in concentrated CNF results in nanopapers with oriented CNFs

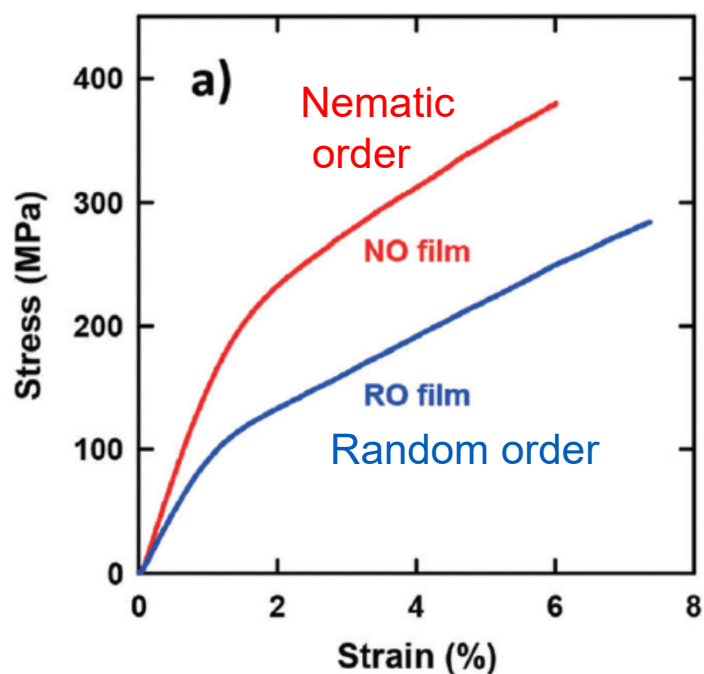
Evaporation-condensed nematic-ordered CNF



Vacuum-filtrated randomly-oriented CNF



Nanopaper from aligned CNF



Tensile properties

	NO film	RO film
Young's modulus E (GPa)	15.4 ± 1.8	12.0 ± 0.5
Yield stress σ_y (MPa)	228 ± 0.2	117 ± 1.2
Tensile strength σ_b (MPa)	349 ± 46	259 ± 43
Elongation at break ε_b (%)	5.7 ± 0.4	6.6 ± 0.5
Light transmittance at 600 nm (%)	84.9 ± 0.7	74.3 ± 0.2
Oxygen gas permeability ($\text{mL } \mu\text{m m}^{-2} \text{ day}^{-1} \text{ kPa}^{-1}$)	0.15	0.59

Heat transfer properties

Diffusivity α ($10^{-7} \text{ m}^2 \text{ s}^{-1}$)	1.03 ± 0.03	0.89 ± 0.07
Conductivity k ($\text{W m}^{-1} \text{ K}^{-1}$)	0.14 ± 0.01	0.11 ± 0.01
Electrical resistivity ($10^8 \Omega$ per square)	1.39 ± 0.13	0.93 ± 0.10

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Cellulose nanocomposites

Nanocomposites – why?

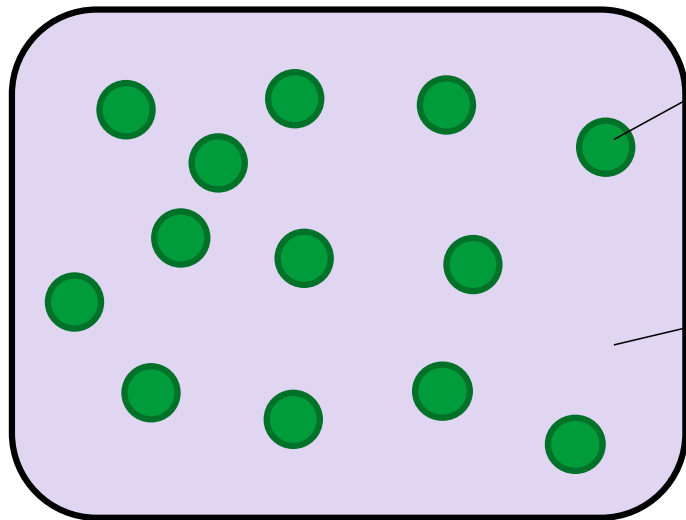
- **Nanoparticles are small**
 - **High surface area**
- **Stress transfer between reinforcing material and continuous matrix occurs via the interface**
 - **Much higher stress transfer from reinforcing nanoparticles than from bigger reinforcing particles**

Cellulose nanocomposites

- **Cellulose nanocomposites usually consist of cellulose nanofibres or cellulose nanocrystals embedded in a continuous polymer matrix**
- **The challenge is often to retain the nanometer dimensions of the cellulosic objects (i.e., prevent their aggregation)**
- **One of the big trends is to end up with materials that would be equal to plastics (or to surpass the properties of plastics)**
- **Another trend is to prepare stimuli-responsive composites for niche applications**

Cellulose nanocomposites

Schematic cross section
of a nanocomposite



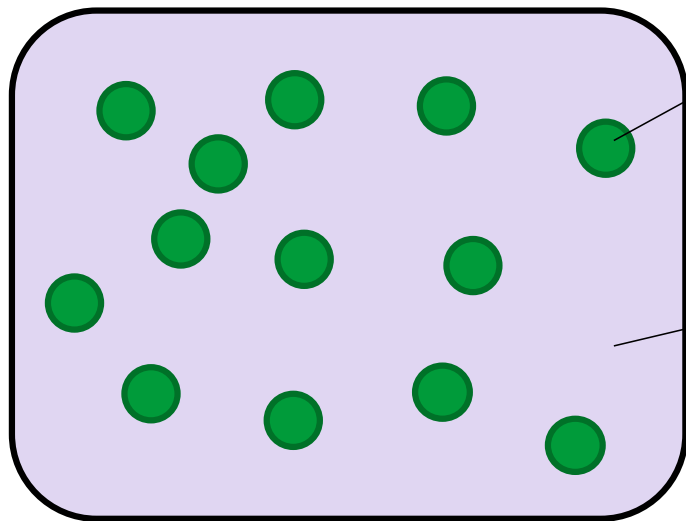
Reinforcing phase: Nanofibril or nanocrystal cross section (cellulose)

Continuous polymer matrix
(preferably from a biodegradable polymer)

OFTEN REFERRED TO SCENARIO: Poly(lactic acid) as a continuous matrix and nanofibrillar cellulose as the reinforcing phase. *Both* are bio-based and biodegradable.

Cellulose nanocomposites

Schematic cross section
of a nanocomposite



Reinforcing phase: Nanofibril or nanocrystal cross section (cellulose)

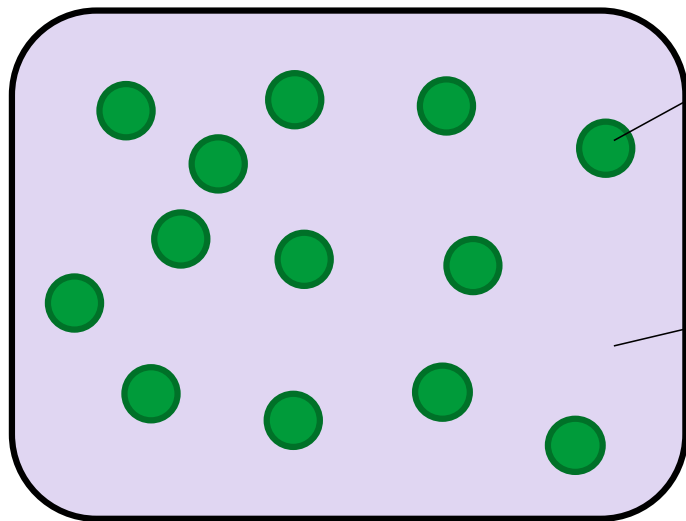
Continuous polymer matrix
(preferably from a biodegradable polymer)

WHY CELLULOSE?

Poly(lactic acid) is brittle and weak on its own. Cellulose with strong mechanical properties would boost its strength

Cellulose nanocomposites

Schematic cross section
of a nanocomposite



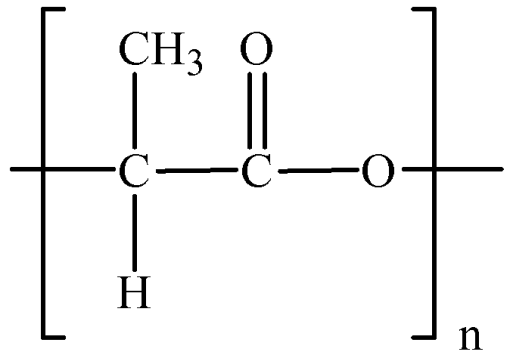
Reinforcing phase: Nanofibril or nanocrystal cross section (cellulose)

Continuous polymer matrix
(preferably from a biodegradable polymer)

WHY NANOCOMPOSITES?

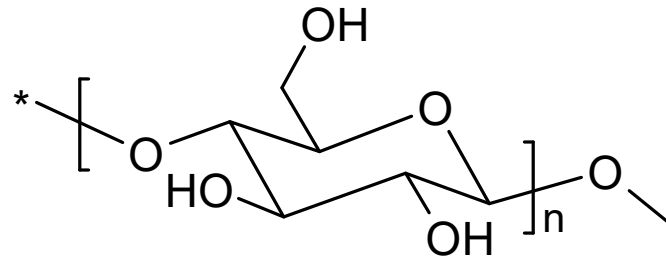
Nanocellulose has a very large surface area, that is, a lot of contact points with the continuous poly(lactic acid) matrix, plus it is very strong.

Cellulose nanocomposites



Poly(lactic acid)

Dissolves in hydrophobic solvents

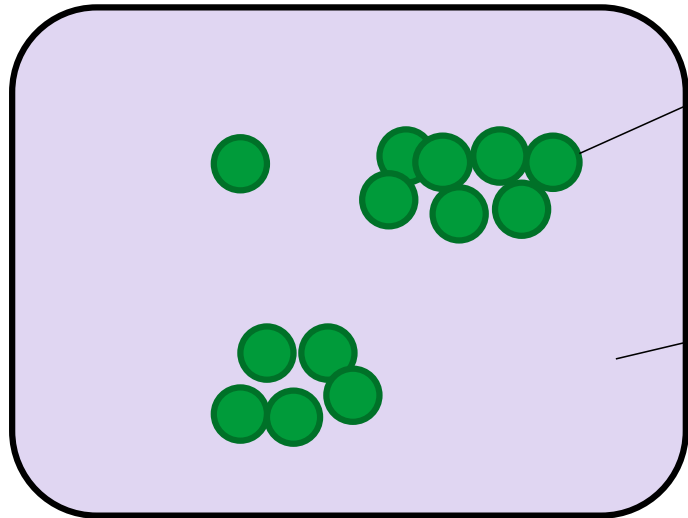


Cellulose

Dissolves in very few solvents

Cellulose nanocomposites

Schematic cross section
of a nanocomposite



Reinforcing phase: Nanofibril or nanocrystal cross section (cellulose)

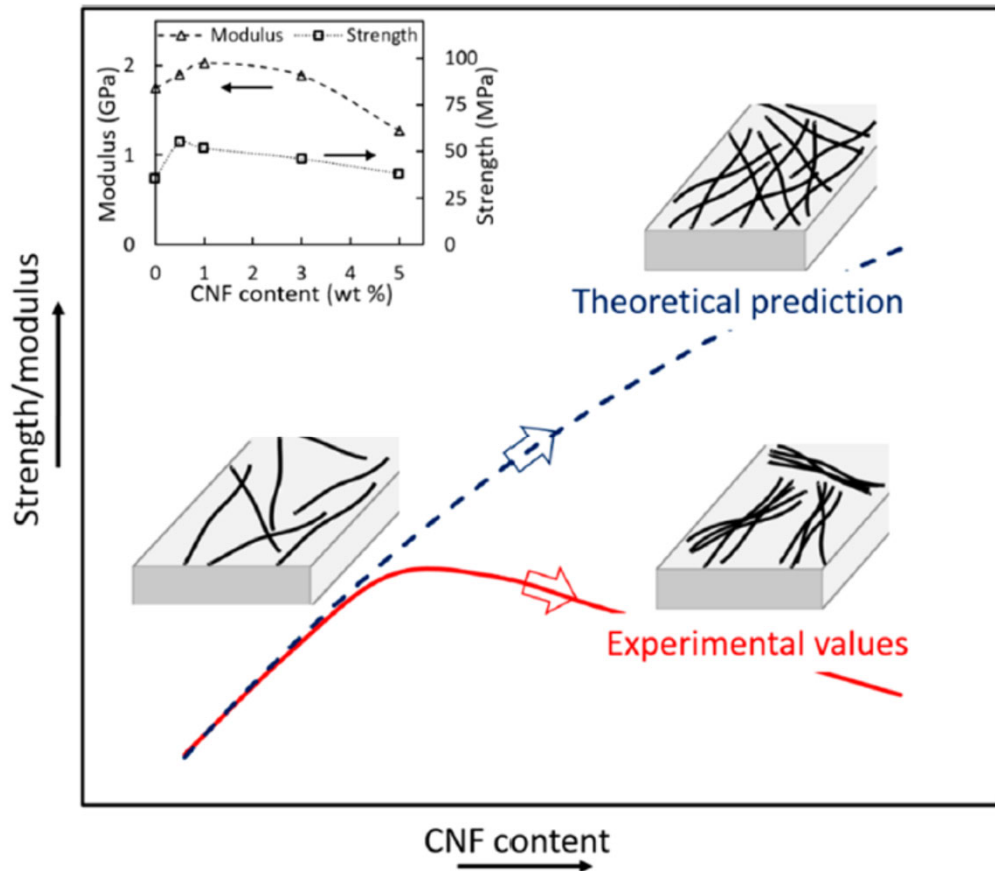
Continuous polymer matrix
(preferably from a biodegradable polymer)

WHAT IS THE PROBLEM?

Cellulose aggregates easily with itself. It is incompatible with nearly anything else than the lignin/hemicellulose matrix in the plant cell wall.

→ Loss in surface area → No nanocomposite anymore

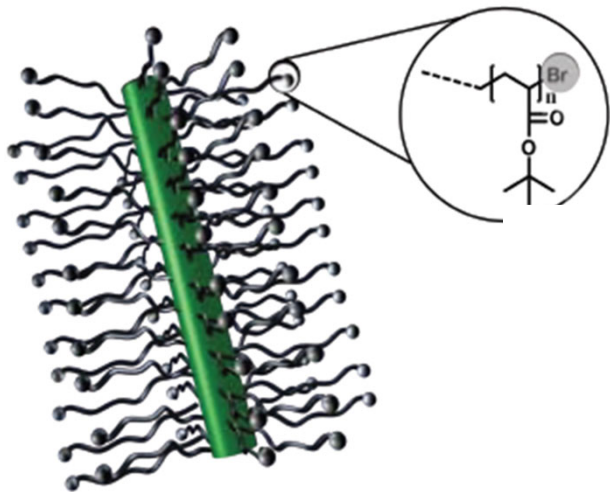
Aggregation of nanocellulose



Because of aggregation, nanocellulose reinforcement often works well only with low nanocellulose contents.

Overcoming aggregation: modification

How does one solve the problem of aggregation in cellulose nanocomposites?
One approach: modify the surface of nanocellulose

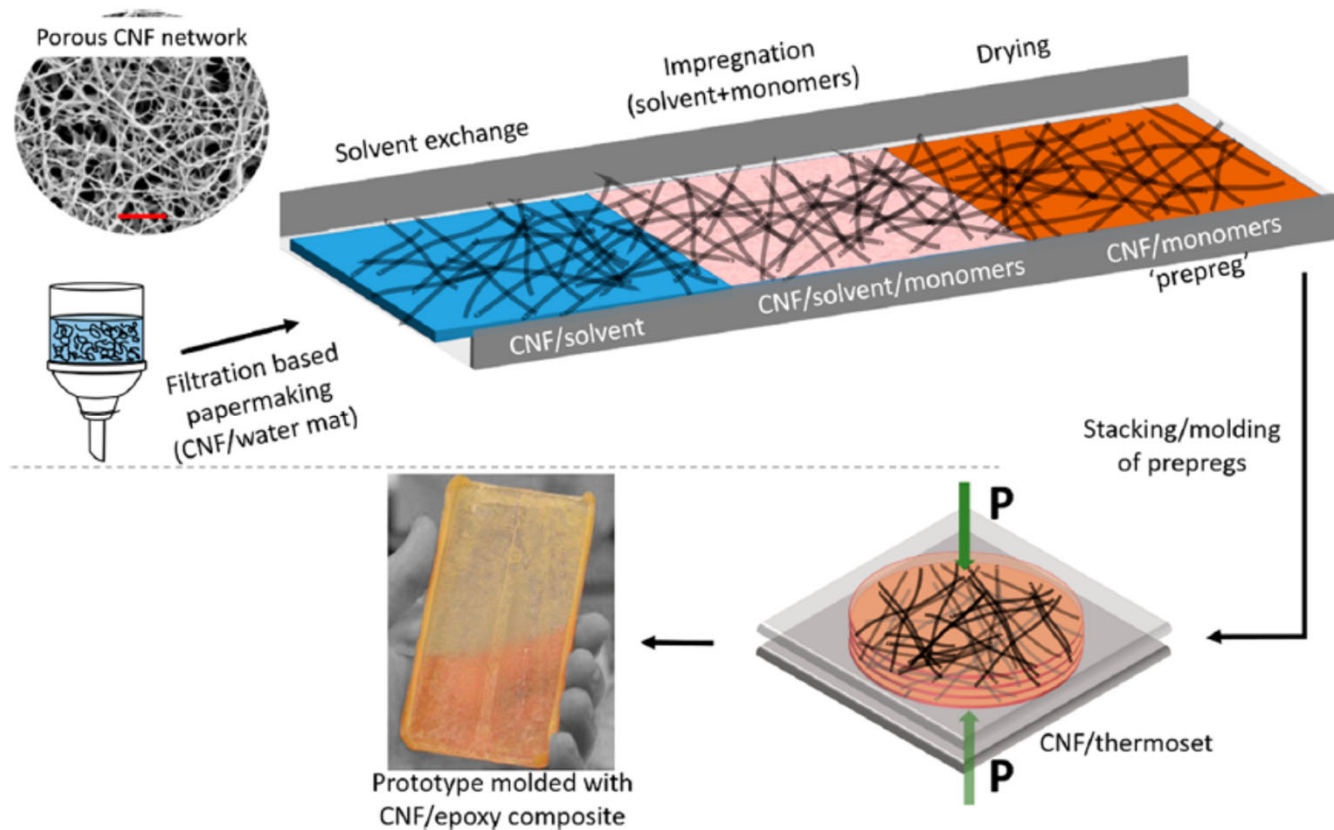


Surface modification:

- Surface of nanocellulose is more compatible with the surrounding polymer matrix in nanocomposites
- Crystalline core of cellulose stays intact
→ Strength properties of nanocellulose stay intact

At the moment, the methods to do this are expensive and not scalable.

Network formation beforehand



Another solution: prepare a nanocellulose network before introducing the continuous polymer matrix

Enables higher nanocellulose contents

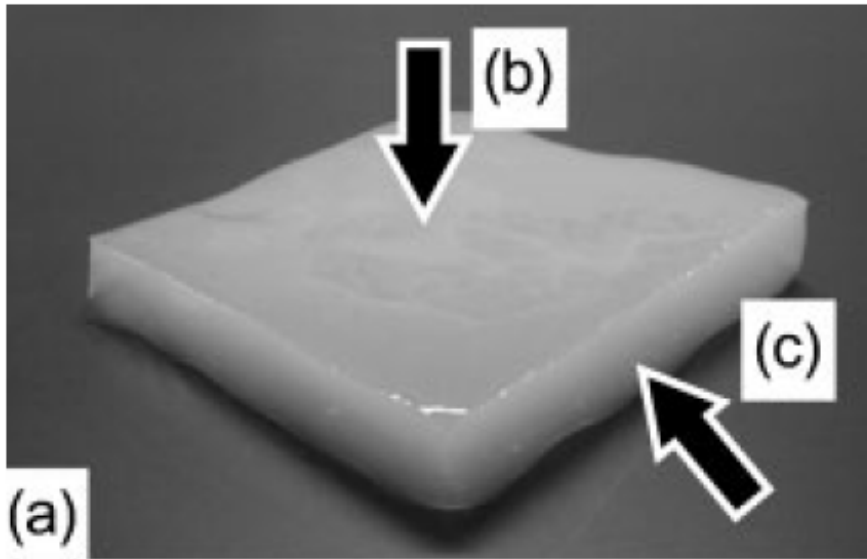
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Composites with CNFs

Acrylic resin / bacterial cellulose



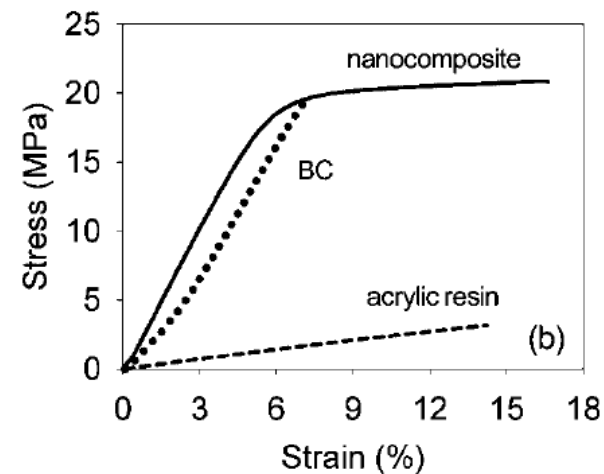
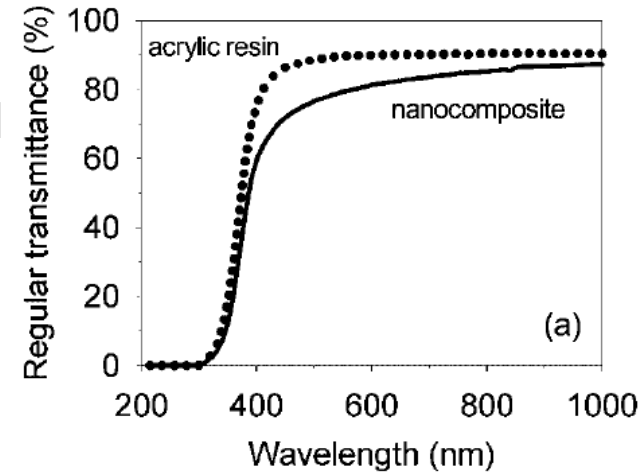
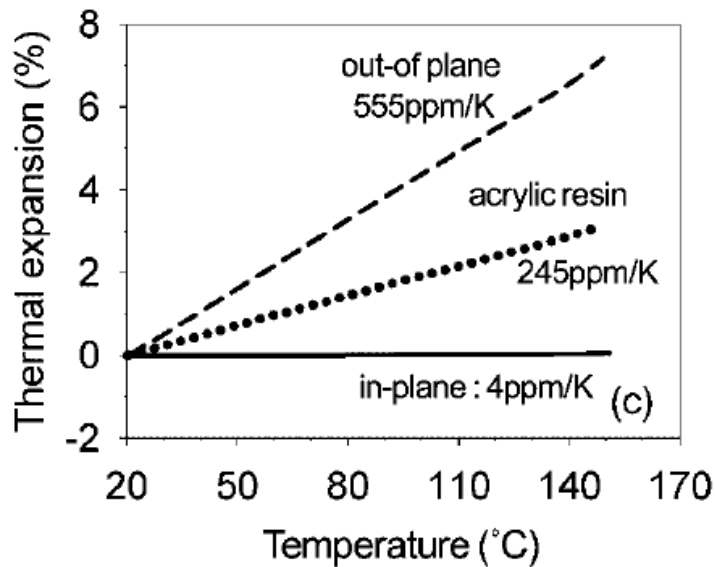
Bacterial cellulose pellicle

Water in a bacterial cellulose pellicle is replaced gradually with ethanol and impregnated with acrylic resin which is subsequently cured.

Acrylic resin / bacterial cellulose

Light transmittance of the resin is retained while the mechanical properties are enhanced.

Thermal expansion is very low.

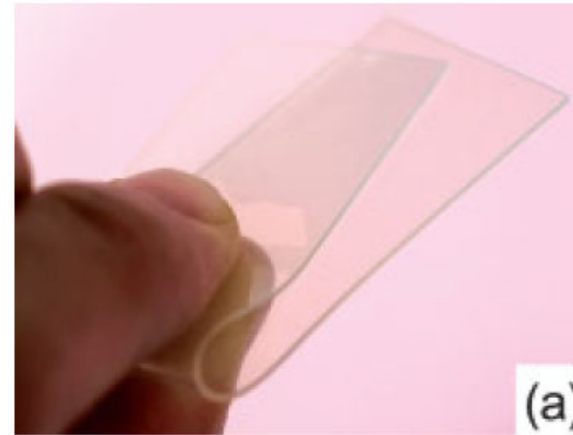


Acrylic resin / bacterial cellulose



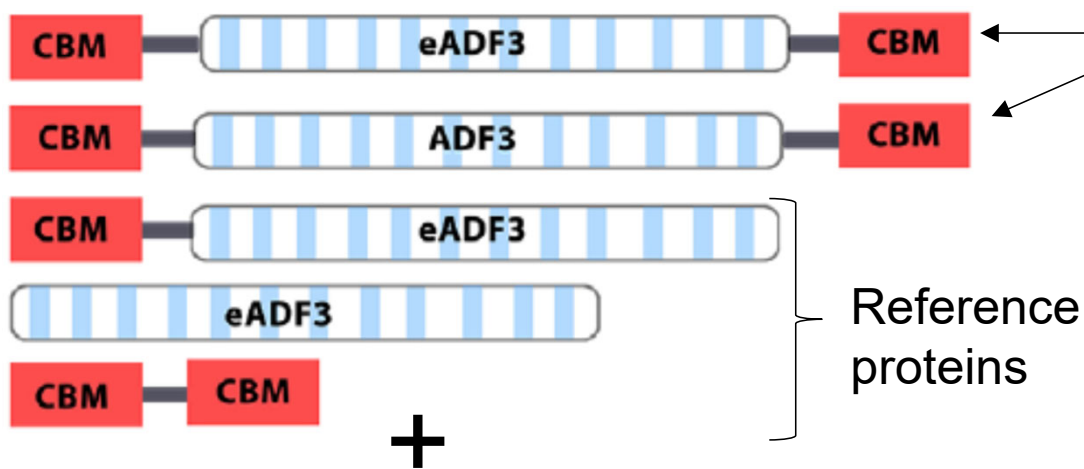
Figure 1. Luminescence of an organic light-emitting diode deposited onto a transparent BC nanocomposite. The luminescence area is $40 \times 25 \text{ mm}^2$. This work was carried out in collaboration with Mitsubishi Chemical Corporation and Pioneer Corporation.

Potential usage for foldable flat panel displays in the electronics device industry.



Most plastics have too large a thermal expansion for electronics devices.

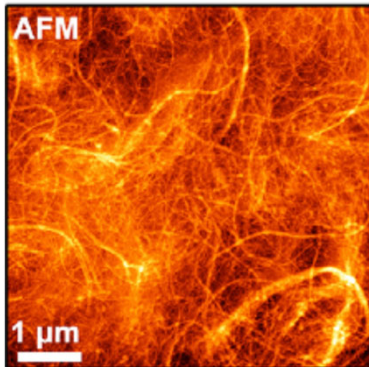
CNF with silk proteins



Engineered triblock proteins

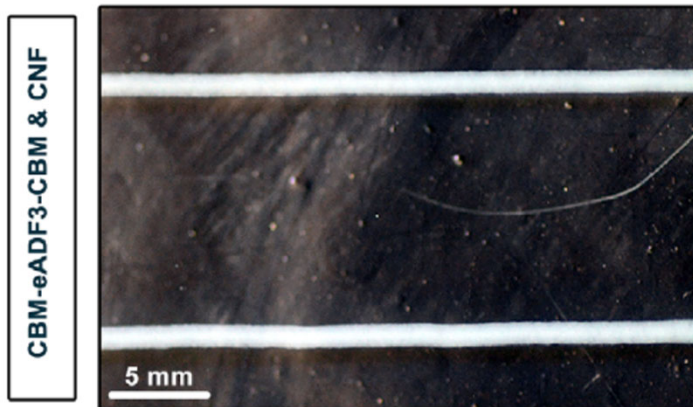
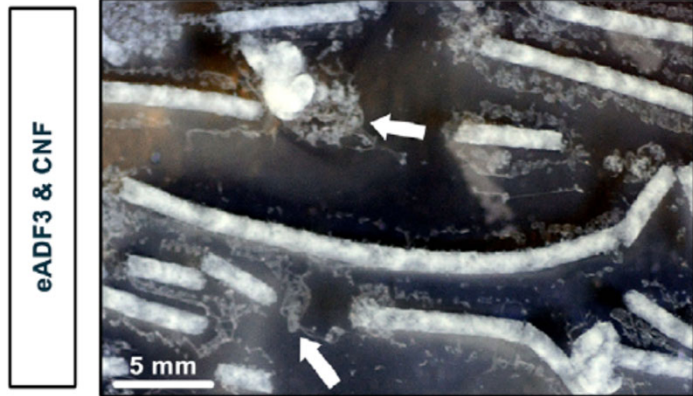
CBM – Cellulose Binding Module
(isolated from cellulase enzymes)

ADF3 – Spidroin protein from spider
silk



Wood-based CNF

CNFs with silk proteins



Ratio CBM-eADF3-CBM:CNF 1:2

EXTRUDED FIBRES

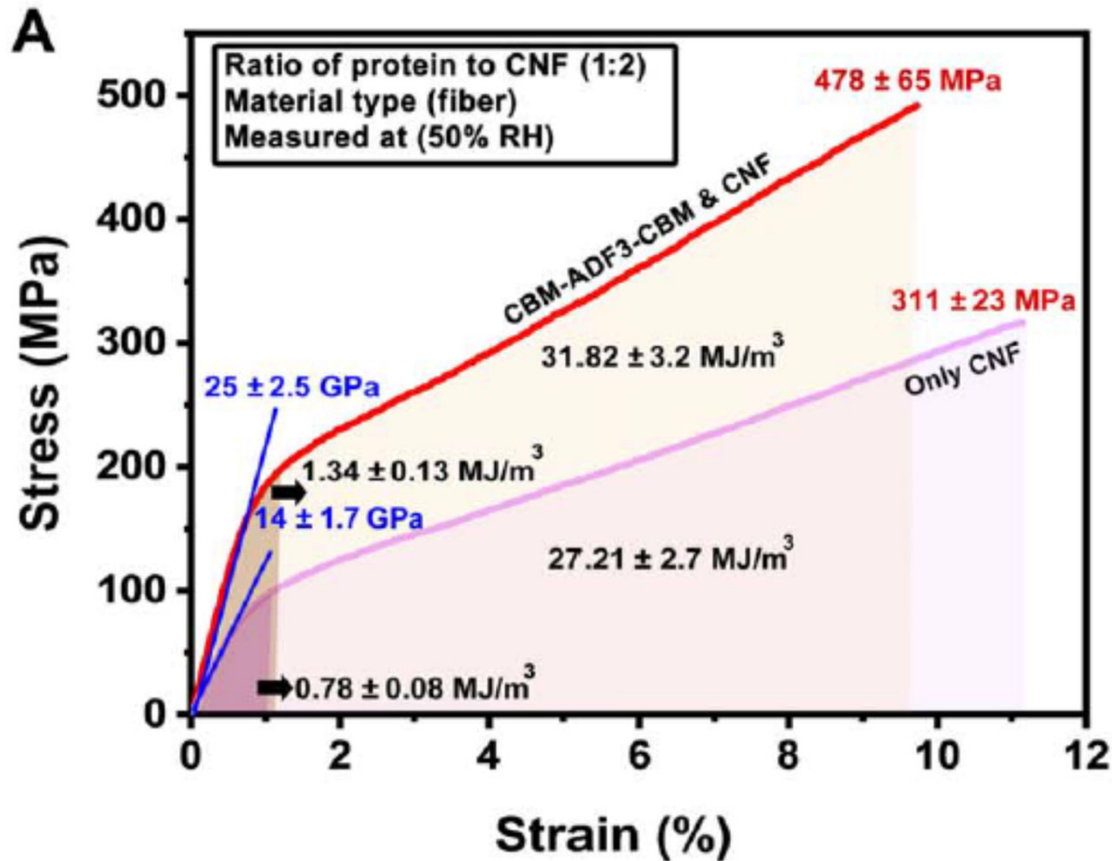
Proteins without CBM block

- Poor adhesion between CNF and protein
- Fragmented fibres

Proteins with CBM block

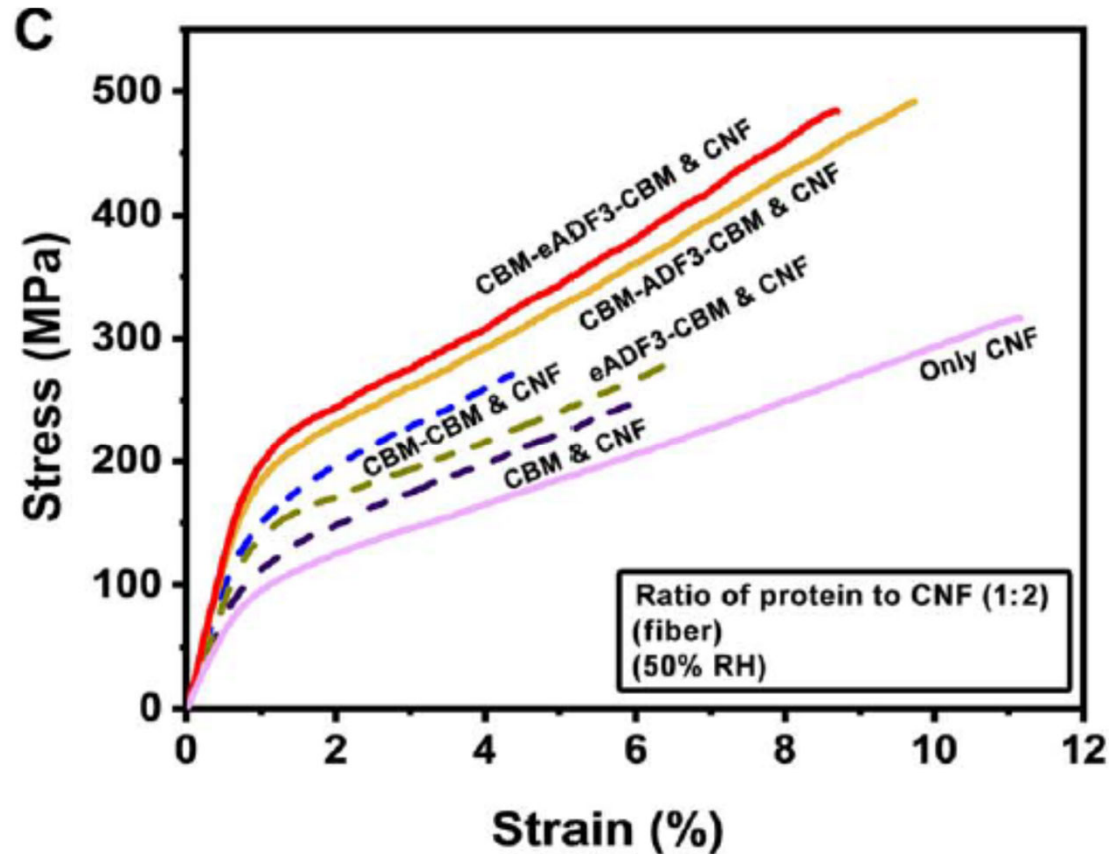
- Good adhesion between CNF and protein
- Fibres with good integrity

CNFs with silk proteins



Composite fibres have a far higher tensile strength than pure CNF fibres

CNFs with silk proteins



Triblock proteins with cellulose binding domains (CBM) are necessary for high strength

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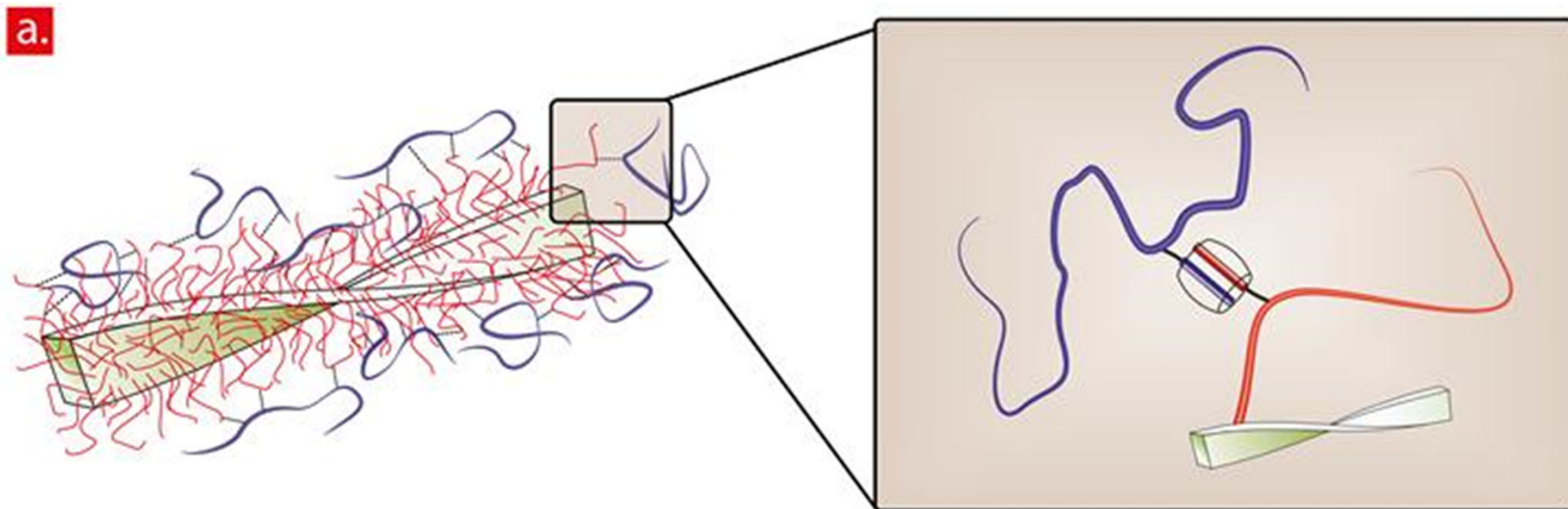
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Composites with CNCs

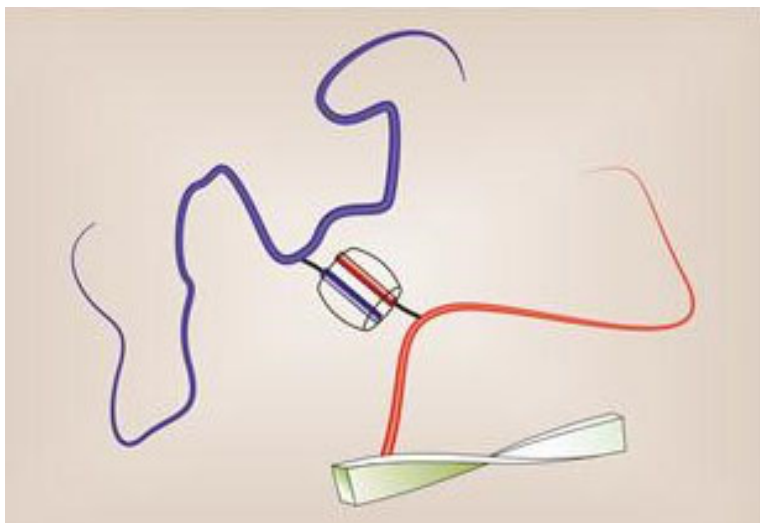
Self-healing hydrogels with CNCs

Basic idea:

- (i) Graft polymers on CNCs
- (ii) Blend grafted CNCs within a continuous polymer network
- (iii) Use strong cross linking agents that enable strength and self-healing



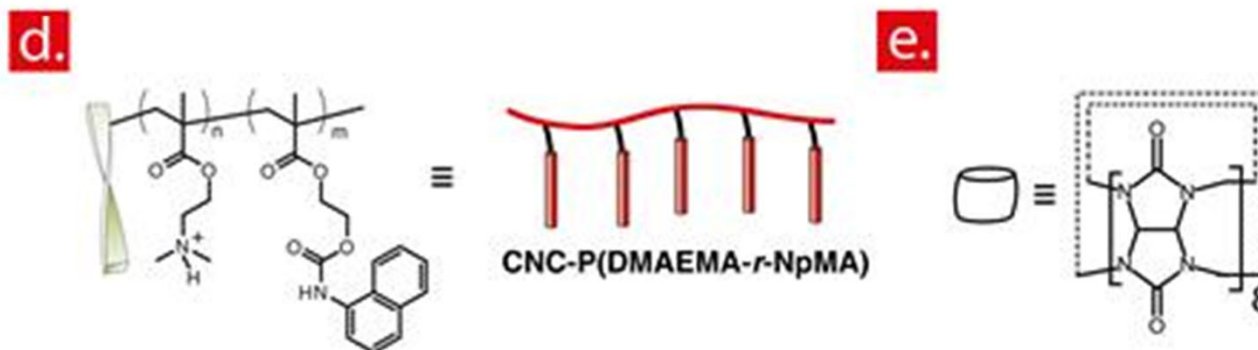
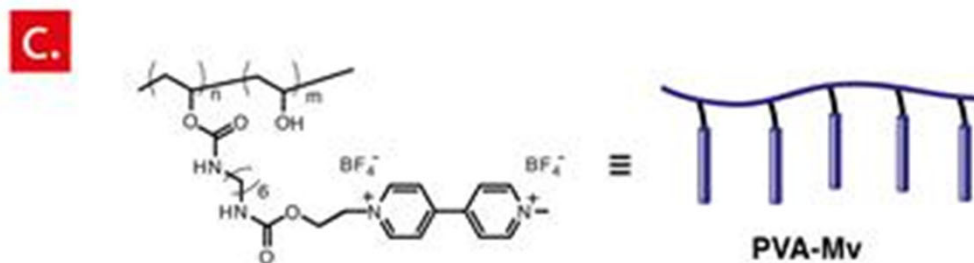
Self-healing hydrogels with CNCs



Poly(dimethylaminoethyl methacrylate) containing naphthyl methyl acrylate

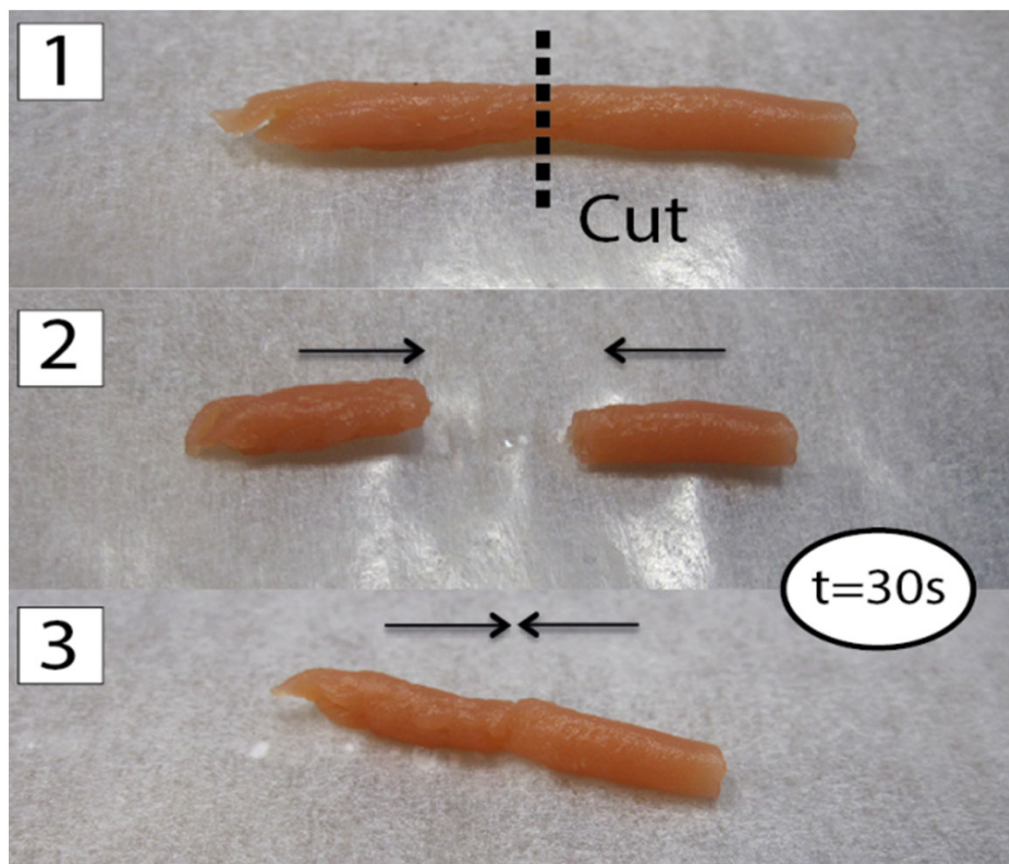
Cucurbit[8]uril: "handcuff" binder with both viologens and naphthyls as guests

Poly(vinyl alcohol) (PVA) containing methyl viologen (MV)



McKee et al. *Adv. Funct. Mater.* **2014**, 24, 2706.

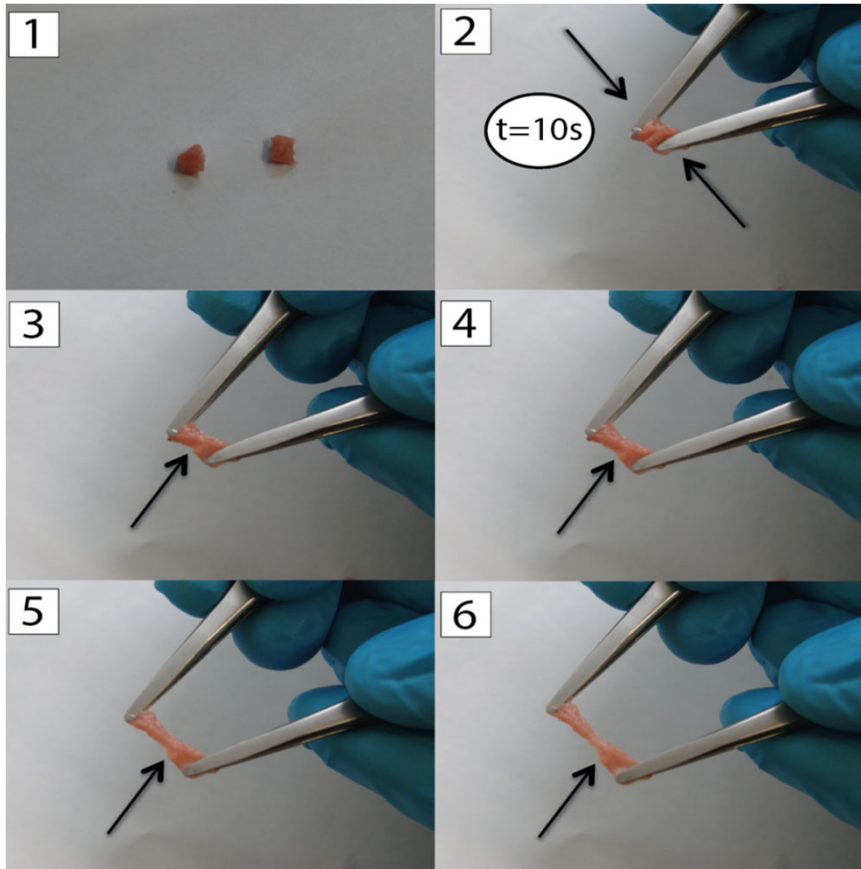
Self-healing hydrogels with CNCs



Self-healing immediately after cutting:

Supramolecular binding with cucurbit[8]uril is fast and strong.

Self-healing hydrogels with CNCs

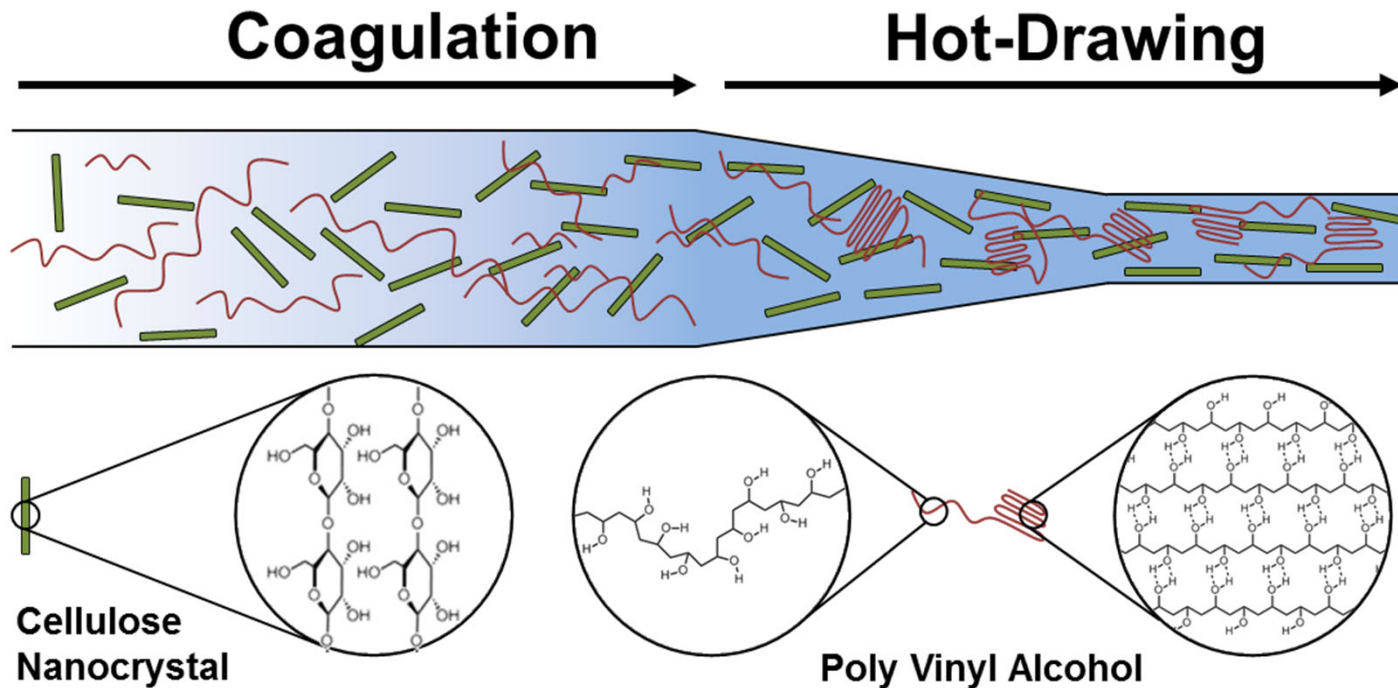


Self-healing after 4 months storage:

The mechanism with cucurbit[8]uril still works and it is still very fast.

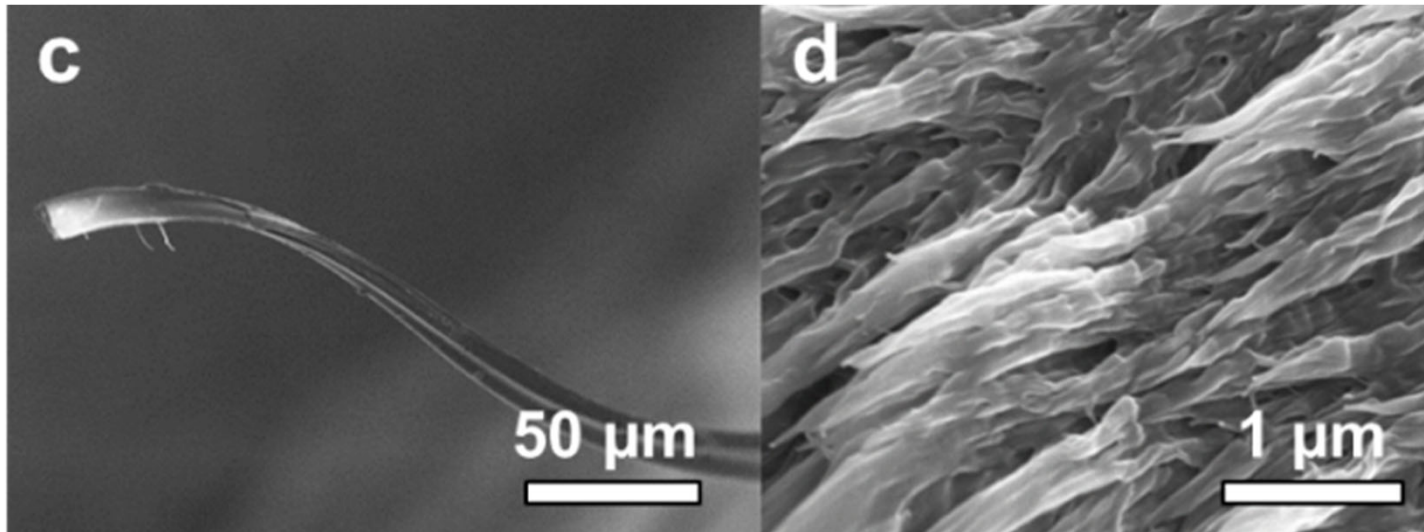
Composite fibres with CNCs

- Polyvinyl alcohol (PVA) with loads of hydroxyl groups is a compatible matrix with CNCs
→ No compatibilizing modifications on CNCs required



Composite fibres with CNCs

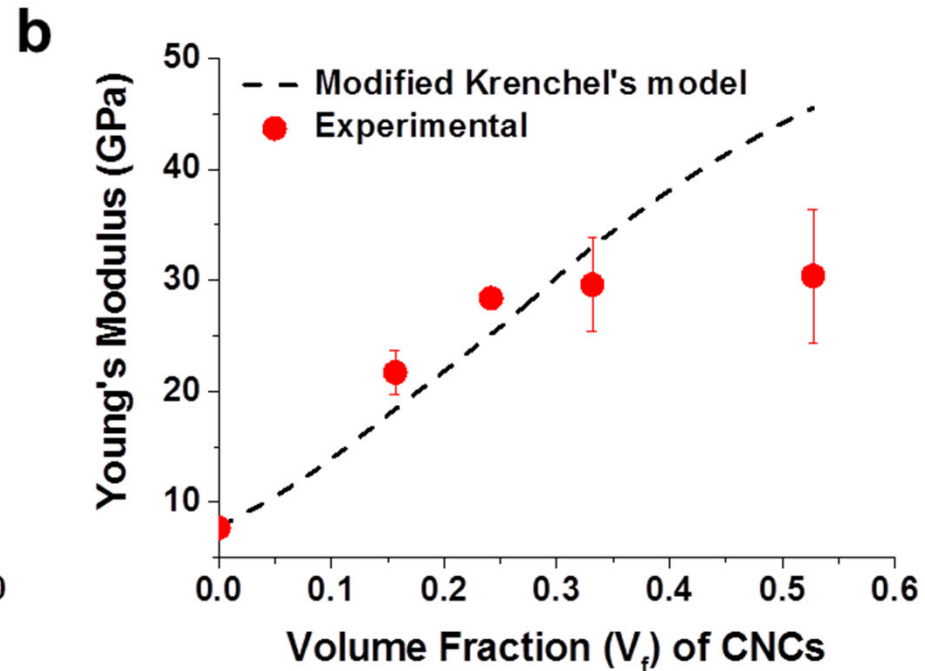
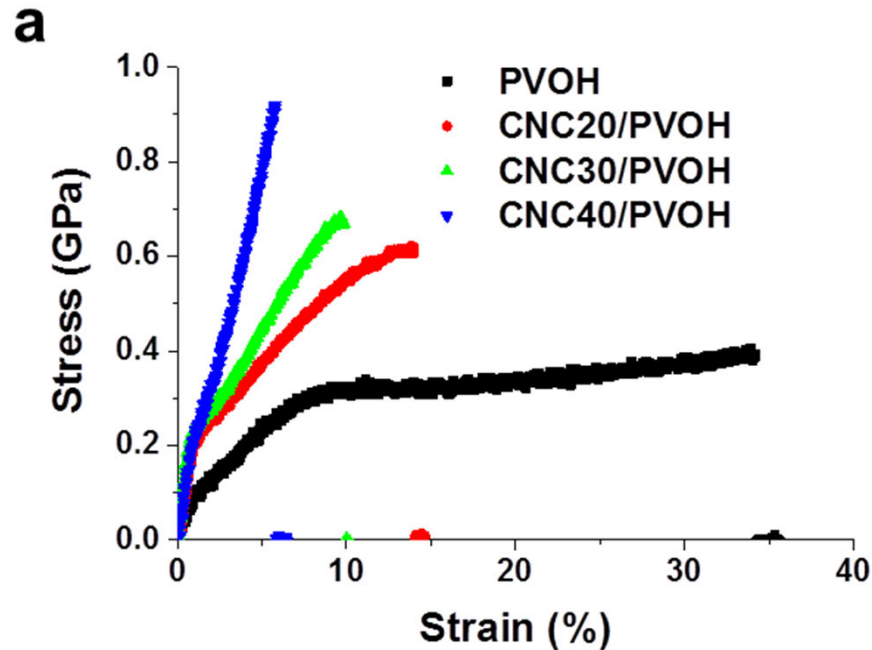
- Result: CNC/PVA composite fibers with tens of μm diameter



- CNC concentration upon fibre spinning is fairly high
 - Above the concentration for liquid crystal formation
 - Liquid crystal phase induces alignment of CNCs in resulting fibres

Composite fibres with CNCs

- Very high mechanical properties for the resulting PVA/CNC fibres: tensile strength of close to 1 MPa with 40% CNC loading



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Applications of nanocellulose hydrogels

Nanocellulose hydrogels

- **Particularly CNFs form gels at low concentrations in water**
- **Gel formation can be an advantage, utilized in specific applications**
- **Most biomedical applications utilise the gel formation properties of CNF**

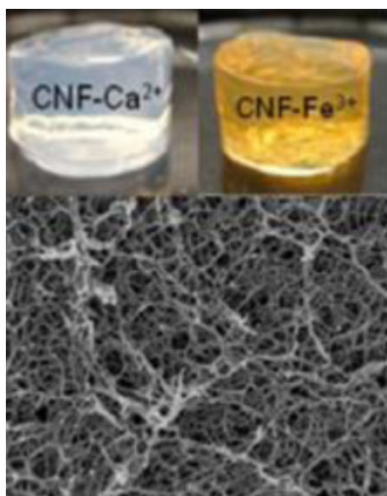
Hydrogels for tissue engineering

- Tissue engineering scaffolds provide 3D foundation to direct cellular attachment, proliferation, and differentiation – ultimately tissue formation
- Requirements: biocompatibility, non-toxicity, biodegradability, sufficient (tunable) porosity and mechanical properties

Bottlenecks to CNF hydrogels as tissue scaffolds:

- Cellulose lacks the adhesive sites necessary for cell signalling and migration
- Tuning the porosity and mechanical properties is not always straightforward

CNF hydrogels for tissue growth

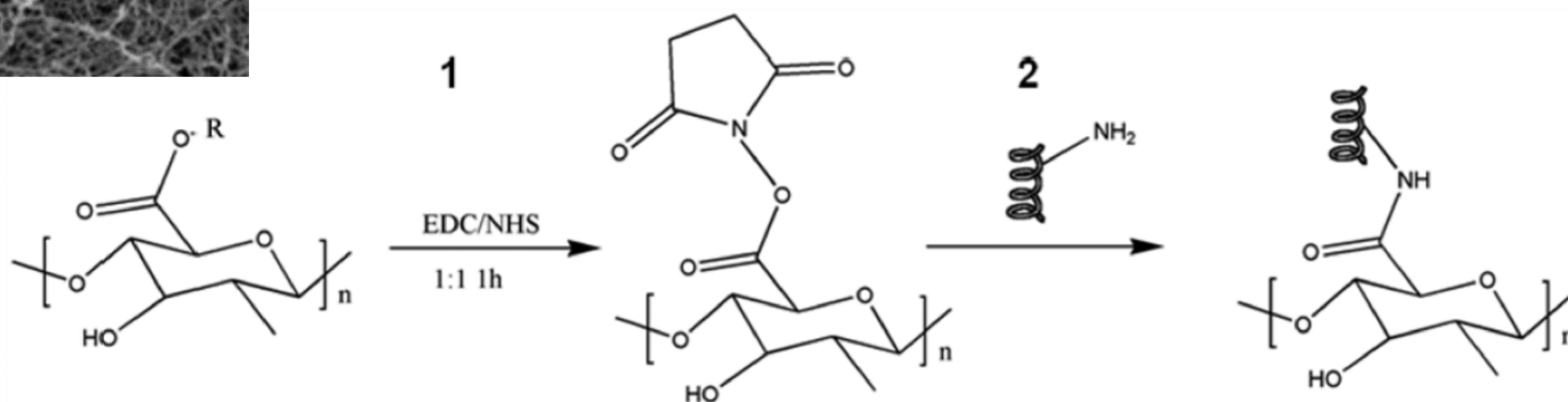


Cross-linking TEMPO-oxidized CNFs by multivalent metal ions

→ Stronger gels for tissue growth scaffolds

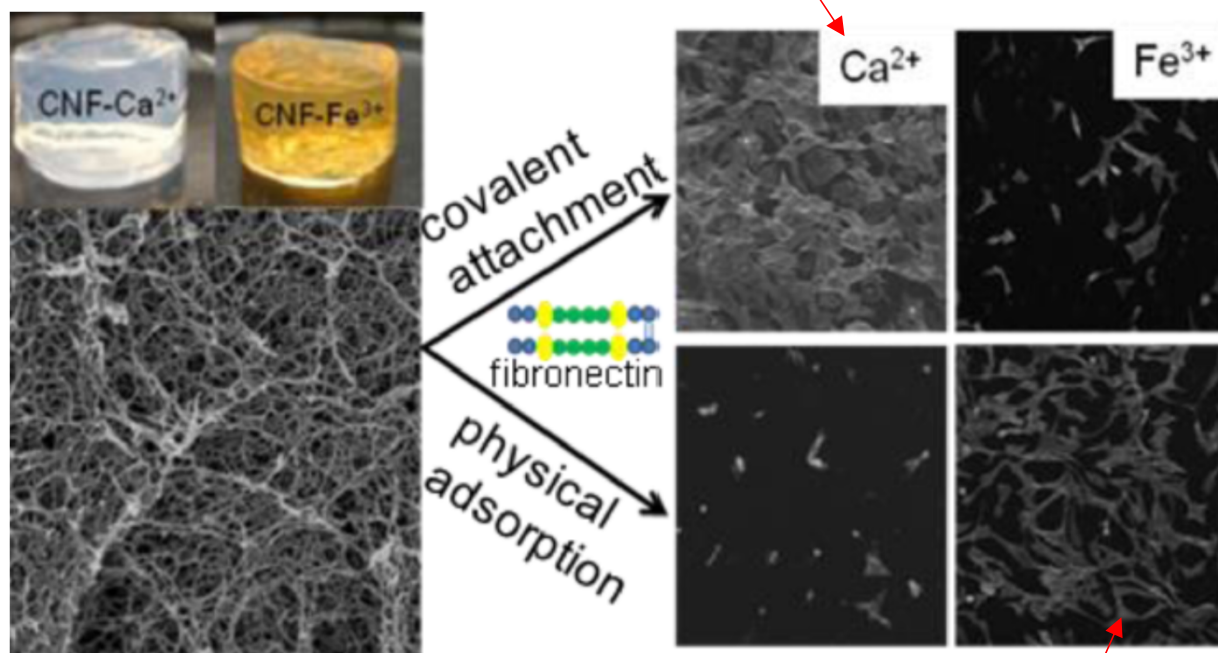
Physical adsorption or covalent attachment of fibronectin protein

→ Supports biorecognition



CNF hydrogels: fibroblast cells

Good cell proliferation:
Ca cross linking + covalent fibronectin attachment

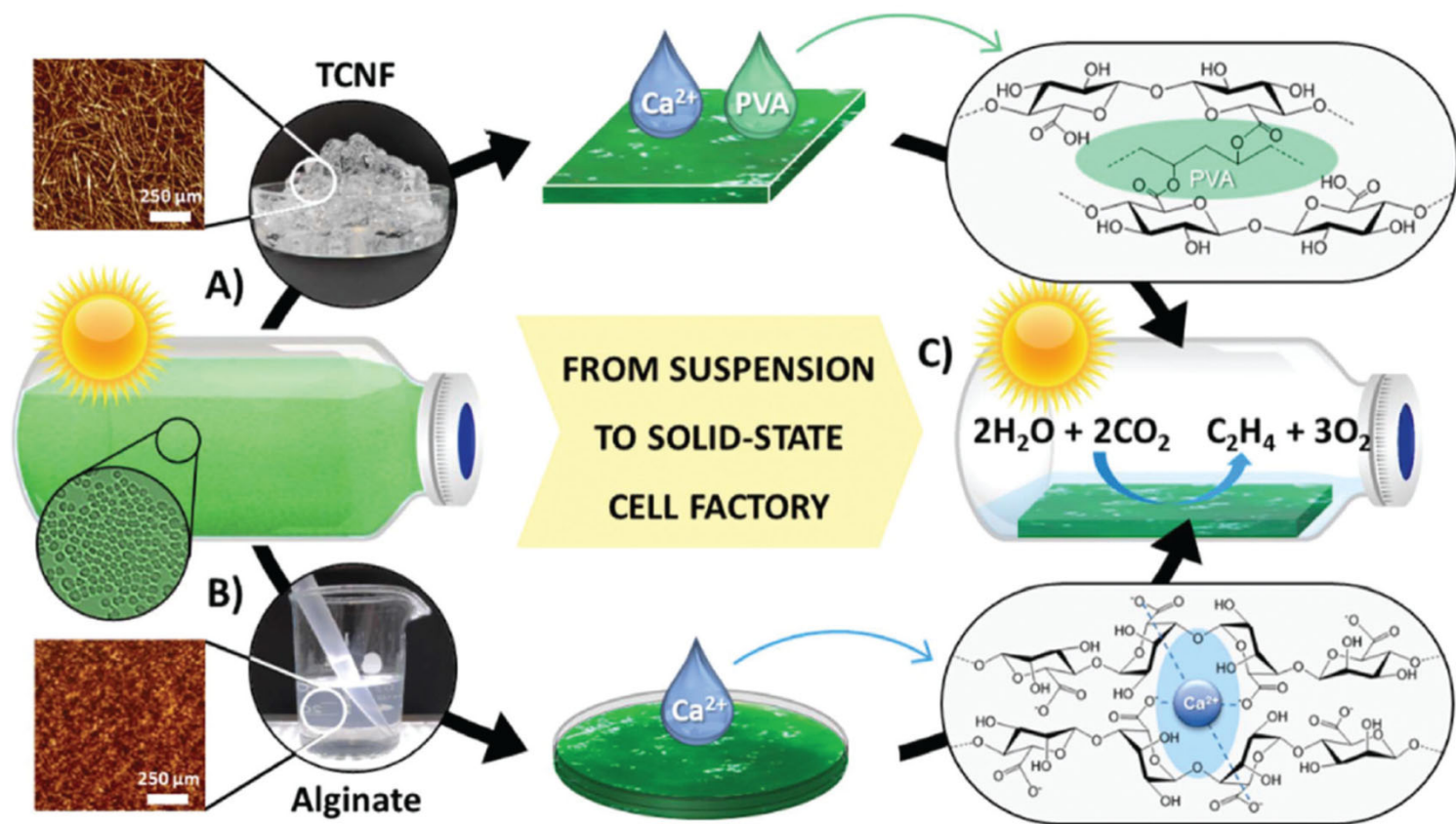


Good cell proliferation:
Fe cross linking + physical fibronectin attachment

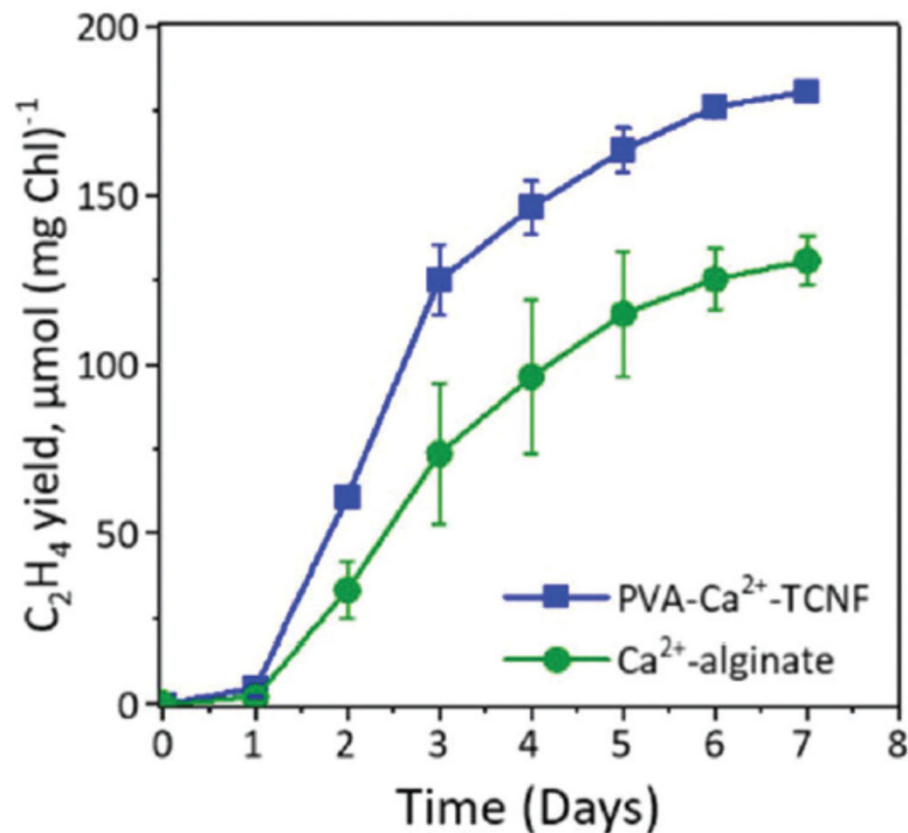
Solid state cell factories

- **Photosynthetic cell factories are platforms where microbial cells of cyanobacteria or green algae are used as biocatalysts to produce biofuels or platform chemicals**
- **They generally consist of pools of water where the cells are located (suspension culture)**
- **Application for nanocellulose hydrogels: “solid state” cell factories where the photosynthetic cells are trapped in a nanocellulose gel matrix**
 - Enables better control over the chemicals production and collection
 - Harmful proliferation of cells (instead of chemical production) can be better controlled
 - Enables better control over light-to-production efficiency

Solid state cell factories with CNFs



Solid state cell factories with CNFs



Production of ethylene by photosynthetic cells:

- Higher conversion by cross linked CNF matrix than by alginate matrix

(Note: alginate is currently the only other option for “solid state” cell factories)

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Conclusions

- **Materials incorporating nanocellulosics is a highly visible research area**
- **Some applications require just film formation (e.g., gas barriers, transistor supports), some require complex chemistry and recognition mechanisms (e.g., self-healing composites, tissue growth scaffolds)**

Examples of versatility in research:

- Transparent films
- Plastic surrogates
- Supports for electronics
- Tissue growth scaffolds
- Solid state cell factories

Conclusions

Nanocellulose:

- **High potential: high strength, bio-based origin, biodegradability, non-toxicity**
- **Difficulties: compatibility with composite matrix, realisation of potential, susceptibility to water**
- **Native properties of cellulose in the plant cell wall are underutilised**

Feedback for the course

Please take a couple of minutes to fill in a feedback form for the whole lecture bit