
Neodymium

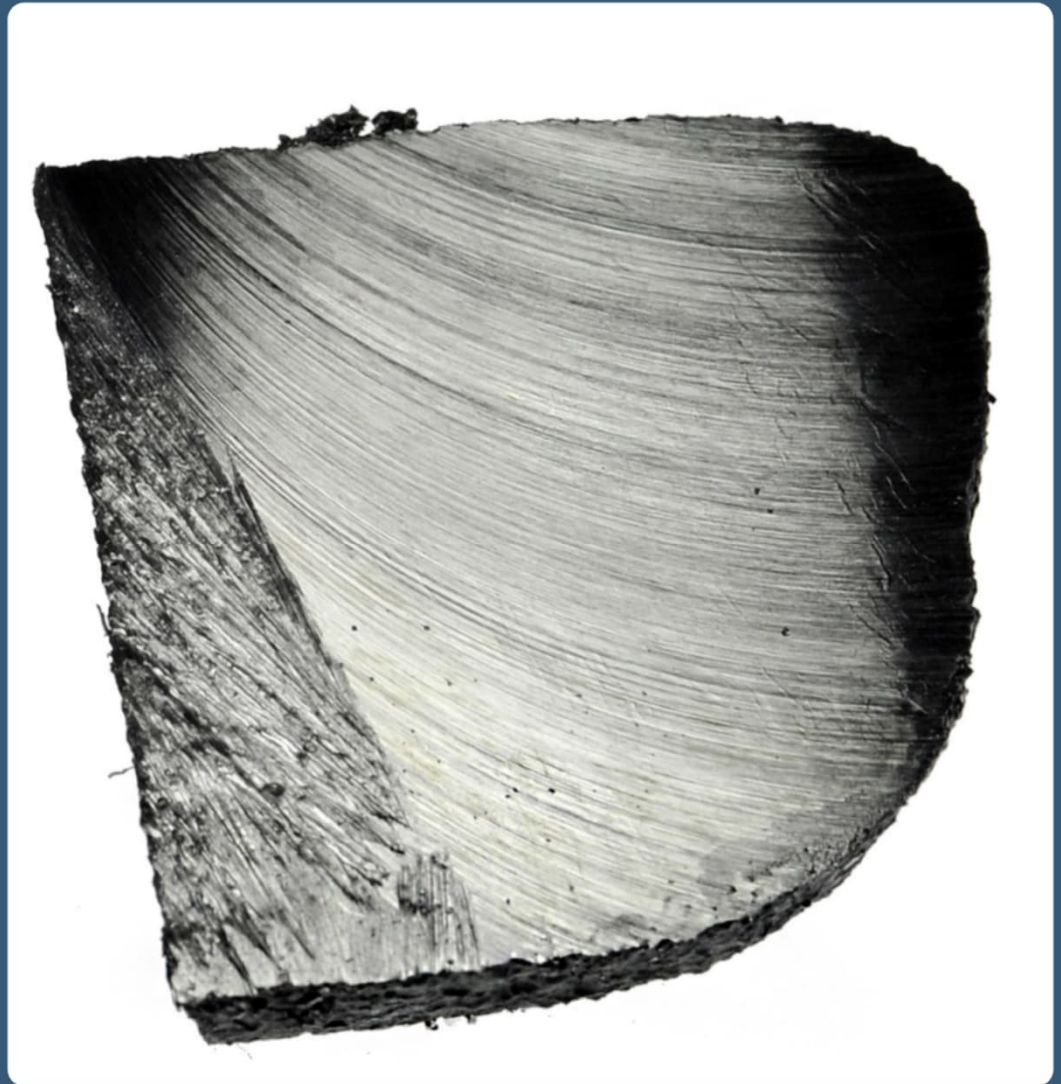
Chemistry of the Elements

04.10.2023

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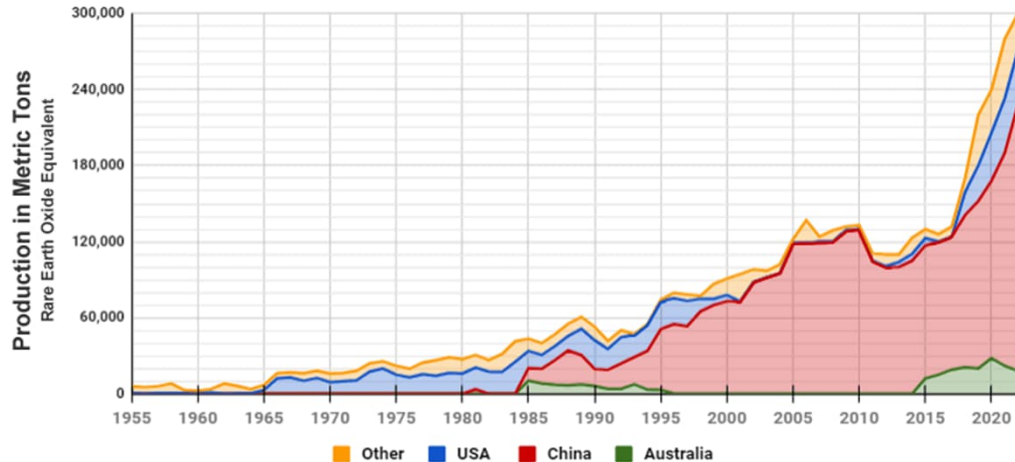


Discovery

- Nd is a rare earth metal discovered in 1885 by Baron Carl Auer von Welsbach.
- He separated a mineral called didymium into two new distinct elemental salts, one of which contained neodymium.
- Greek words, neos "new," and didymos "twin."
- The principal ores : bastnasite and monazite.
- Pure neodymium metal was not isolated until 1925.



World Production



History of rare earth element production

- Historically, a mine in California produced most of the world's rare earth minerals
- Since early 90s, China has become the world's primary source
- Today China supplies 70% of the world's REEs

Abundance

- 28th most abundant element in the earth crust.
- Very difficult to mine because it is unusual to find them in concentrations high enough for extraction.
- 2nd most abundant element in the REEs.
- In terms of reserves, China has the most, followed by Vietnam, Brazil, and Russia.

World Mine Production and Reserves (2022 Estimates)		
Country	Production (Metric Tons)	Reserves (Metric Tons)
United States	43,000	2,300,000
Australia	18,000	4,200,000
Brazil	80	21,000,000
Burma	12,000	not available
Canada	--	830,000
China	210,000	44,000,000
Greenland	--	1,500,000
India	2,900	6,900,000
Madagascar	960	not available
Russia	2,600	21,000,000
South Africa	--	790,000
Tanzania	--	890,000
Thailand	7,100	not available
Vietnam	4,300	22,000,000
Other Countries	80	280,000
World total (rounded)	300,000	130,000,000

Element	Symbol	Atomic number	Crustal abundance
Light REEs			
Lanthanum	La	57	39
Cerium	Ce	58	66.5
Praseodymium	Pr	59	9.2
Neodymium	Nd	60	41.5
Samarium	Sm	62	7.05
Europium	Eu	63	2.0
Gadolinium	Gd	64	6.2
Heavy REEs			
Terbium	Tb	65	1.2
Dysprosium	Dy	66	5.2
Holmium	Ho	67	1.3
Erbium	Er	68	3.5
Thulium	Tm	69	0.52
Ytterbium	Yb	70	3.2
Lutetium	Lu	71	0.8
Yttrium	Y	39	33

PERIODIC TABLE OF ELEMENTS

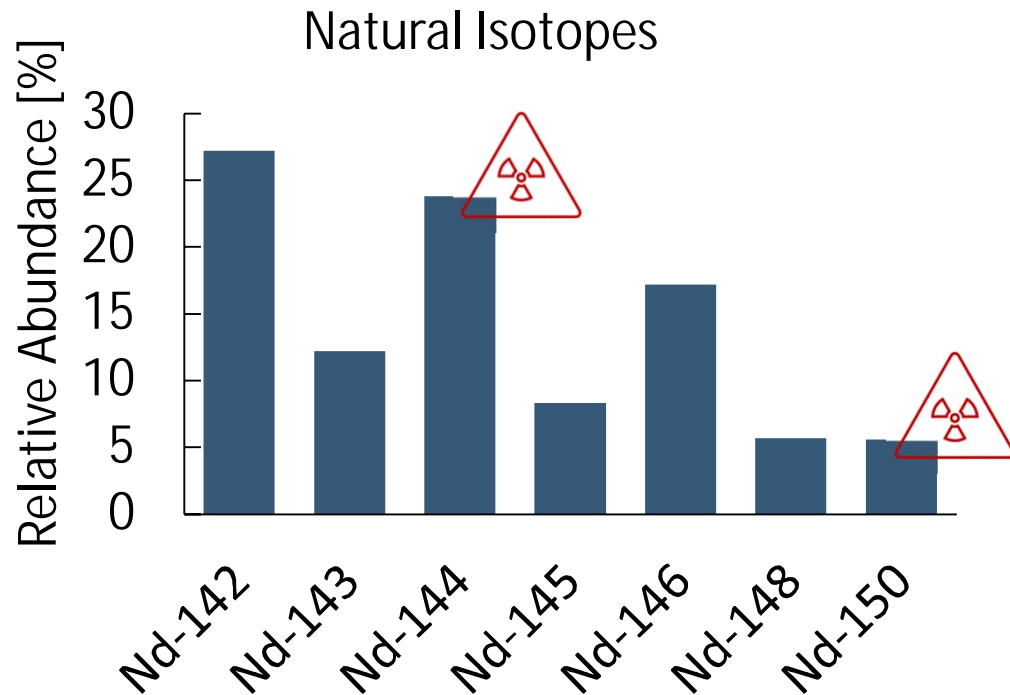
Chemical Group Block

1	Atomic Number																17	Atomic Mass, u																18	
1	Name																Cl	Symbol																2	
																	Chemical Group Block																		
1	1																	2																	18
	1																	13	14	15	16	17	18												
1	1																	5	6	7	8	9	10												
2	3	4																	B	C	N	O	F	Ne											
	Li	Be																	Boron	Carbon	Nitrogen	Oxygen	Fluorine	Neon											
3	11	12																	Al	Si	P	S	Cl	Ar											
	Na	Mg																	Aluminum	Silicon	Phosphorus	Sulfur	Chlorine	Argon											
4	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36																	
	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr																	
5	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54																	
	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe																	
6	55	56																	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86		
	Cs	Ba																	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn		
7	87	88																	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118		
	Fr	Ra																	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	Fl	Mc	Lv	Ts	Og		
																			57	58	59	60	61	62	63	64	65	66	67	68	69	70	71		
																			La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu		
																			Lanthanum	Cerium	Praseodymium	Neodymium	Promethium	Samarium	Europium	Gadolinium	Terbium	Dysprosium	Holmium	Erbium	Thulium	Ytterbium	Lutetium		
																			89	90	91	92	93	94	95	96	97	98	99	100	101	102	103		
																			Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr		
																			Actinium	Thorium	Protactinium	Uranium	Neptunium	Plutonium	Americium	Curium	Berkelium	Californium	Einsteinium	Fermium	Mendelevium	Nobelium	Lawrencium		

Physical Properties



Melting Point	1024 °C
Boiling Point	3080 °C
Density	7.01 g/cm ³



Chemical Properties

- Oxidation states: +II, +III, +IV
- Readily reacts with oxygen
- Electronegativity (Pauling): 1,14

Nd₂Fe₁₄B

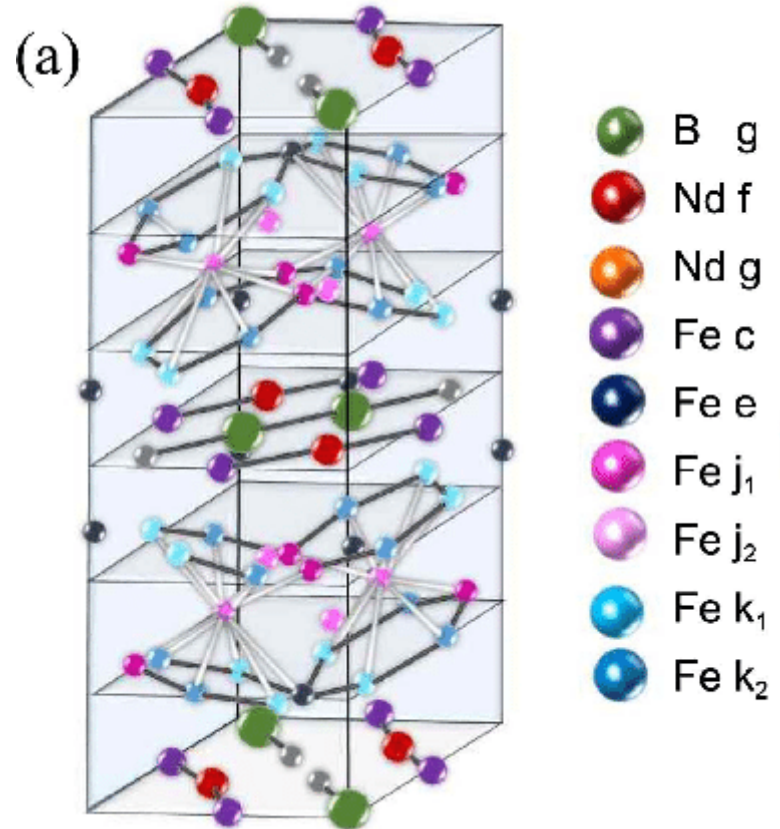
Powerful and widely used type of permanent magnet alloy

Property

- High Magnetic Strength
- Temperature Stability

Applications

- Hard Disk Drives
- Electric Vehicle Motors
- Speakers and Headphones



Schematic representation tetragonal unit cell of Nd₂Fe₁₄B (P42/mnm space group)

Neodymium-Iron-Boron Magnet-to-Magnet Recycling

Rare earth elements (REEs), which are the key materials for creating NdFeB magnets, have been subject to significant supply uncertainty in the past decade.

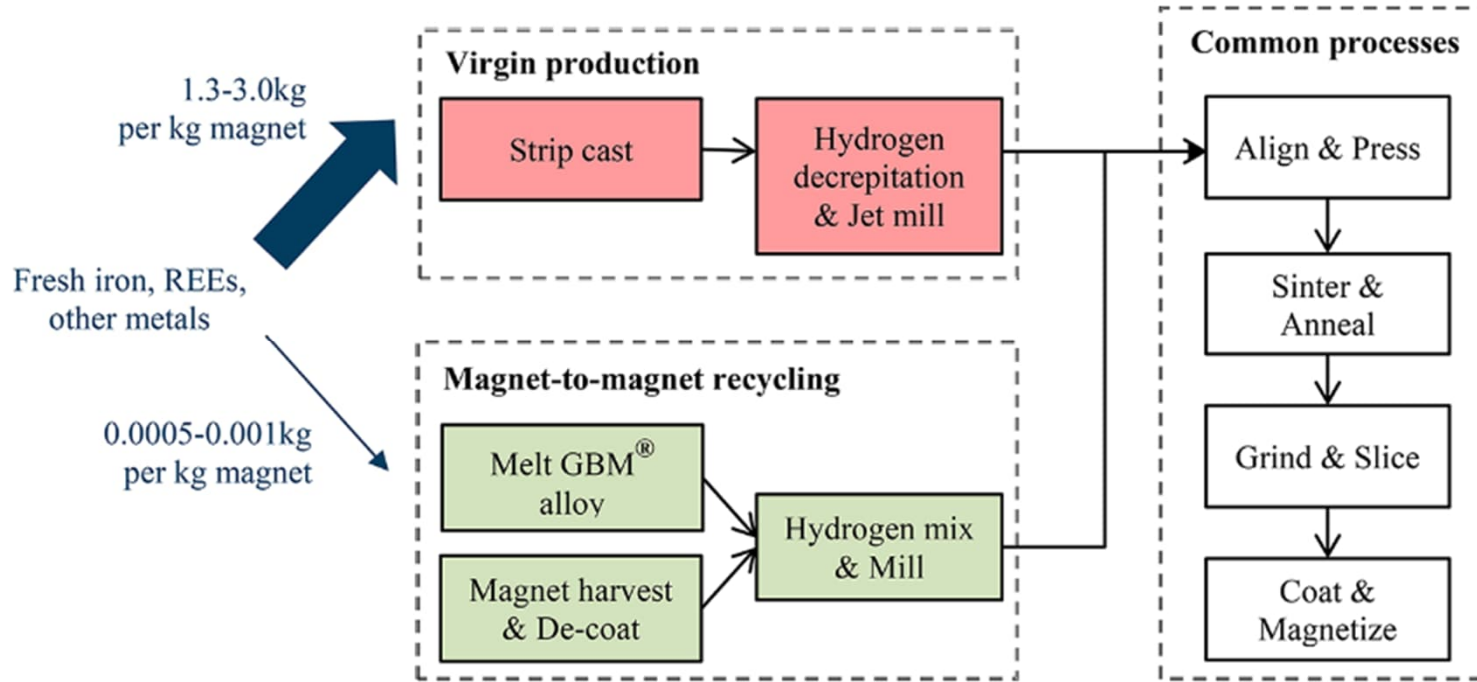


Magnet-to-Magnet Recycling



Advantage

- 1) Minimize waste and resource exhaustion
- 2) The use of mechanical processes reduces the use of chemicals and harmful emissions



Process flows for NdFeB magnet virgin production and magnet-to-magnet recycling

Properties of Virgin and Recycled NdFeB Magnets

parameters (unit)	virgin magnet	recycled magnet
B_r (T)	1.2	1.3
BHc (kOe)	11.5	12.6
IHc (kOe)	19.0	>20.0
BH_{max} (MGOe)	34.0	40.7
operating temperature (°C)	180	180

Both are suitable for high temperature applications such as electric vehicles (EVs), offer similar performance, and thus can be used interchangeably.

Life Cycle Impacts of Producing 1 kg of NdFeB Magnet through Magnet-to-Magnet Recycling

impact category	unit	virgin, China	virgin, U.S.	recycled, China	recycled, U.S.
ozone depletion	mg CFC-11 eq	12–29	13–30	0.4–0.9	0.6–1.3
global warming	kg CO ₂ eq	94–222	88–207	25–56	18–41
smog	kg O ₃ eq	9–21	8–19	2–4	1–2
acidification	kg SO ₂ eq	0.8–1.7	0.8–1.6	0.4–0.6	0.3–0.5
eutrophication	kg N eq	1.0–2.4	1.0–2.4	0.1	0.1–0.2
carcinogenics	CTUh	4.5×10^{-06} – 1.0×10^{-05}	4.8×10^{-06} – 1.1×10^{-05}	6.4×10^{-07} – 1.3×10^{-06}	9.7×10^{-7} – 2.0×10^{-6}
non carcinogenics	CTUh	2.6×10^{-05} – 5.7×10^{-05}	2.6×10^{-05} – 5.8×10^{-05}	6.7×10^{-06} – 1.1×10^{-05}	7.2×10^{-6} – 1.2×10^{-5}
respiratory effects	kg PM _{2.5} eq	0.2–0.4	0.2–0.3	0.04–0.06	0.02–0.03
ecotoxicity	CTUe	646–1,430	643–1,422	188–327	184–320
fossil fuel depletion	MJ surplus	101–240	113–267	5–10	18–39

Magnet-to-magnet recycling significantly reduces the environmental footprint of NdFeB magnet production, because it minimizes the use of fresh rare earths.

Recovery of Neodymium

Hydrometallurgy

Electrochemistry

Gas-phase extraction

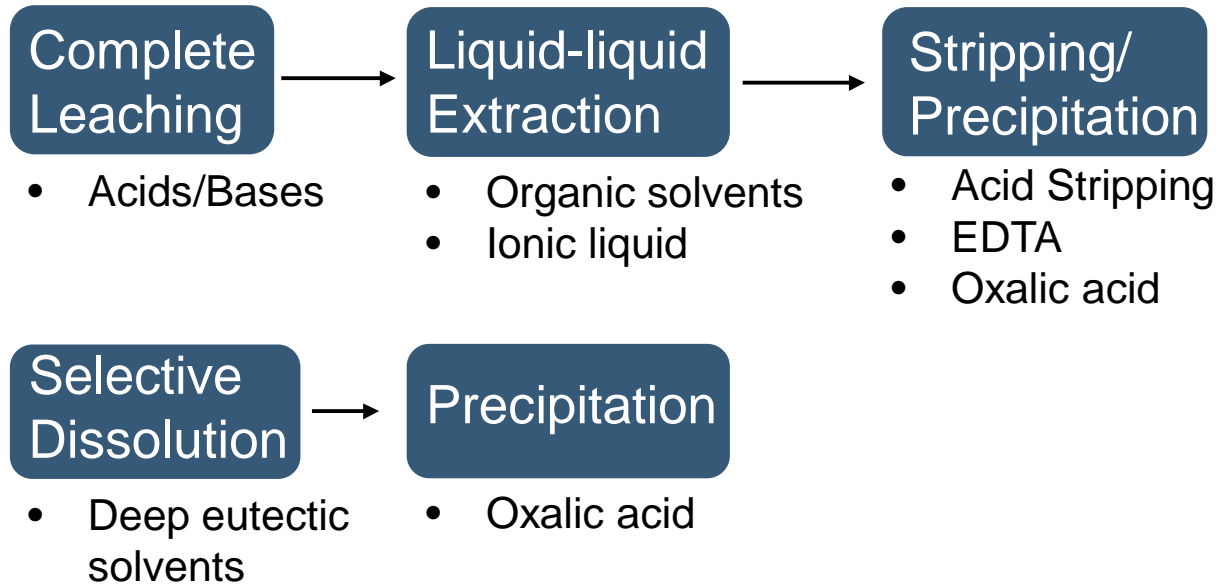
Membrane Separation

Biological Extraction

Pyrometallurgy

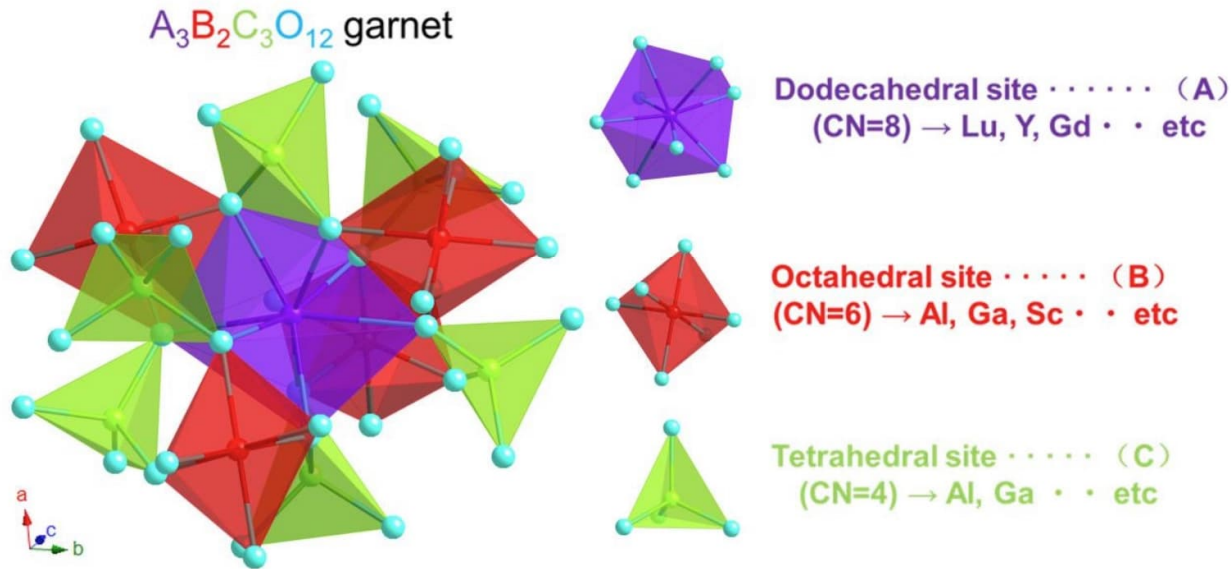
- High processing capacity
- Low cost
- Big environmental impact

Recovery of Neodymium



Nd:YAG Laser

- Neodymium-doped yttrium aluminum garnet ($\text{Nd:Y}_3\text{Al}_5\text{O}_{12}$)



Nd:YAG Laser



- **Cutting**
- **Drilling**
- **Many more**



- **Skin treatment**
- **Ophthalmology**
- **Surgery**



- **Rangefinders**

Thank you !

Any questions or comments?