

1. Isobaarinen prosessi ab: $W_{ab} = p_a(V_b - V_a) = 90 \text{ kJ}$. Ideaalikaasuyhtälöstä
 $T = pV/(nR)$ ja yksiatomisen kaasun $C_p = \frac{5}{2}R$, joten
 $Q_{ab} = nC_p(T_b - T_a) = nC_p \frac{p}{nR}(V_b - V_a) = \frac{5}{2}W_{ab} = 225 \text{ kJ}$.
 $\Delta U_{ab} = Q_{ab} - W_{ab} = 135 \text{ kJ}$. Isokoorinen prosessi bc: $dV = 0 \Rightarrow W_{bc} = 0 \Rightarrow$
 $\Delta U_{bc} = Q_{bc} = nC_V(T_c - T_b) = nC_V \frac{V}{nR}(p_c - p_b) = \frac{3}{2}V(p_c - p_b) = -240 \text{ kJ}$. Koko
kiertoprosessi: $W = (p_a - p_c)(V_c - V_a)/2 = 30 \text{ kJ}$ ja koska $\Delta U = 0$, niin $Q = 30 \text{ kJ}$.
Prosessi ca: $W_{ca} = (p_c + p_a)(V_a - V_c)/2 = -60 \text{ kJ}$. $\Delta U_{ca} = -\Delta U_{ab} - \Delta U_{bc} = 105 \text{ kJ}$.
 $Q_{ca} = \Delta U_{ca} + W_{ca} = 45 \text{ kJ}$.
2. Merkitään: systeemin alkutila a, isobaarinen prosessi ab, adiabaattinen bc ja isokoorinen ca.
a) Isobaarisissa prosessissa p on vakio, joten ideaalikaasuyhtälöstä myös $p = nRT/V$ on vakio $\Rightarrow T/V$ on vakio. Kun $V_b = V_a/2$, niin $T_b = T_a/2$. Adiabaattisessa prosessissa $TV^{\gamma-1}$ on vakio $\Rightarrow T_b V_b^{\gamma-1} = T_c V_c^{\gamma-1} \Rightarrow$
 $T_c = T_b (V_b/V_a)^{\gamma-1} = T_a/2 \cdot (1/2)^{\gamma-1} = T_a (1/2)^\gamma = 114 \text{ K}$ ($\gamma = 1,40$). b) Tarkastellaan isokoorista prosessia ca $\Rightarrow V$ on vakio $\Rightarrow V = nRT/p$ on vakio $\Rightarrow T/p$ on vakio $\Rightarrow p_c = p_a T_c/T_a = 0,758 \cdot 10^5 \text{ Pa}$.
3. a) Carnot-sykli: isotermiset osat $W_{ab} = \int_{V_1}^{V_2} \frac{kNT}{V} dV = kNT_1 \ln(V_2/V_1)$ ja
 $W_{cd} = kNT_2 \ln(V_4/V_3)$
Adiabaattiset: $dQ = 0 \Rightarrow W_{bc} = -\Delta U_{bc} = \frac{3}{2}kN(T_1 - T_2)$ ja $W_{da} = \frac{3}{2}kN(T_2 - T_1)$.
b) $W = kNT_1 \ln(V_2/V_1) + kNT_2 \ln(V_4/V_3)$, missä $p_1 V_1 = p_2 V_2$, $p_3 V_3 = p_4 V_4$, $p_2 V_2^\gamma = p_3 V_3^\gamma$
ja $p_4 V_4^\gamma = p_1 V_1^\gamma$. $\Rightarrow \frac{p_2}{p_1} \left(\frac{V_2}{V_1}\right)^\gamma = \frac{p_3}{p_4} \left(\frac{V_3}{V_4}\right)^\gamma \Rightarrow \frac{V_1}{V_2} \left(\frac{V_2}{V_1}\right)^\gamma = \frac{V_4}{V_3} \left(\frac{V_3}{V_4}\right)^\gamma \Rightarrow$
 $\left(\frac{V_2}{V_1}\right)^{\gamma-1} = \left(\frac{V_3}{V_4}\right)^{\gamma-1} \Rightarrow \frac{V_2}{V_1} = \frac{V_3}{V_4} \Rightarrow W = kN[T_1 - T_2] \ln\left(\frac{V_2}{V_1}\right)$. c)
 $e = \frac{W}{Q_{ab}} = \frac{W}{W_{ab}} = \frac{kN(T_1 - T_2) \ln(V_2/V_1)}{kNT_1 \ln(V_2/V_1)} = \frac{T_1 - T_2}{T_1}$.
4. a) Kappaleiden entropian muutokset $\Delta S_1 = \int_1^T \frac{dQ}{T} = m_1 c_1 \int_1^T \frac{dT}{T} = m_1 c_1 \ln \frac{T}{T_1}$, ja

$$\Delta S_2 = \int_2^{2'} \frac{dQ}{T} = m_2 c_2 \ln \frac{T'}{T_2}. \text{ Eristetty systeemi} \Rightarrow Q_1 = -Q_2 \Rightarrow$$

$$m_1 c_1 (T - T_1) = -m_2 c_2 (T' - T_2) \Rightarrow T' = T_2 - \frac{m_1 c_1}{m_2 c_2} (T - T_1) \Rightarrow$$

$$\Delta S_{tot} = m_1 c_1 \ln \frac{T}{T_1} + m_2 c_2 \ln \frac{T'}{T_2} = m_1 c_1 \ln \frac{T}{T_1} + m_2 c_2 \ln \left[1 - \frac{m_1 c_1}{m_2 c_2} \frac{T - T_1}{T_2} \right].$$

b) Lausekkeen $\Delta S_{tot}(T, T')$ derivaatan 0-kohdasta:

$$\frac{d(\Delta S_{tot})}{dT} = m_1 c_1 \cdot \frac{T_1}{T} \cdot \frac{1}{T_1} + m_2 c_2 \cdot \frac{T_2}{T'} \cdot \frac{1}{T_2} \cdot \frac{dT'}{dT} = 0, \quad \frac{dT'}{dT} = -\frac{m_1 c_1}{m_2 c_2} \Rightarrow$$

$$T = T' = \frac{m_1 c_1 T_1 + m_2 c_2 T_2}{m_2 c_2 + m_1 c_1}.$$