

Fundamentals of Sprays

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Motivation

- Why learn about sprays ?
- Sprays are used in many industrial applications: engines (diesel & gasoline), furnaces, gas turbines, rockets and in many other applications: aerosols, water taps...
- Therefore, understanding concepts related to sprays might be beneficial
- This lesson is about sprays: general topics of sprays in order to better understand what sprays are and how they form and interact with their surroundings





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Course related phenomena



Wehrfritz et al., Comb. Flame 2016.



What does a fuel spray look like ?

Optical Spray Investigations

- Pictures by Harri Hillamo, 2010
- Fuel injection pressure is 1400bar
- Gas density is 35 kg/m3
- Nozzle diameter d=0.34mm





















Spray penetration & opening angle

Dent (1971)
$$S = 3.07 \left(\frac{\Delta p}{\rho_g}\right)^{1/4} (tD)^{1/2} \left(\frac{294}{T_g}\right)^{1/4}$$

$$t < t_{break}: \quad S = 0.39 \left(\frac{2\Delta p}{\rho_l}\right)^{1/2} t$$

Hiroyasu & Kadota
(1980)
$$t > t_{break}: \quad S = 2.95 \left(\frac{\Delta p}{\rho_g}\right)^{1/4} (tD)^{1/2}$$

$$t_{break} = \frac{29\rho_l D}{\left(\rho_g \Delta p\right)^{1/2}}$$

- ρ_g Gas density (kg/m3)
- ρ_l Liquid density (kg/m3)
- Δp Pressure difference (Pa)
- t Time (s)
- *D* Nozzle diameter (m)

Spray penetration & opening angle

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Nozzle types

Furnaces: diameter ~ 1m

> Engines: hole diameter $D \sim 0.1-0.5$ mm

Some liquid fuel properties

| Fuel | HVO | Fischer– Tropsch diesel | FAME (RME) | EN 590 |
|-------------------------------------|---------|-------------------------------|---------------|---------|
| Density at +15C° (kg/m3) | 775-785 | 770-785 | 885 | 835 |
| Viscosity at $+40C^{\circ}$ (mm2/s) | 3.0-3.5 | 3.2-4.5 | 4.5 | 3.5 |
| Cetane number | 80-99 | 73-81 | 51 | 53 |
| Distillation range (C°) | 180-320 | 180-360 | 350-370 | 180-360 |
| Heating value (MJ/kg) | 44 | 43 | 37.5 | 43 |

Some liquid fuel properties

| Fuel | Methanol | DME |
|----------------------------|----------|------|
| Density at +15C° (kg/m3) | 722.1 | 612 |
| Viscosity at +40C° (mm2/s) | 0.45 | 0.22 |
| Cetane number | ~ 5 | 55 |
| Heating value (MJ/kg) | 22.7 | 29 |

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Some liquid fuel properties

HEP = Heptane ($C_7 H_{16}$) DF2= U.S. Diesel fuel EN590=Standard Diesel in EU HVO=Hydrotreated vegetable oil

Topics on Spray Velocity

Sprays are pressure driven

Cavitation

Fig. 5.7. Cavitation inside an acrylic glass diesel injection nozzle. The liquid phase is transparent, the gas phase is opaque. $p_{inj} = MPa$, $p_{cyl} = MPa$, $d_{noz} = mm$ [15]

Cavitation

Sprays are pressure driven

Calculating the theoretical fuel exit velocity

• Theoretical fuel exit velocity from nozzle

$$p_1 + \frac{1}{2}\rho u_1^2 = p_2 + \frac{1}{2}\rho u_2^2 \implies U_{Exit} = \sqrt{\frac{2\Delta P}{\rho_f}}$$

Velocity in practice

$$U_{Exit} = C_v \sqrt{\frac{2\Delta P}{\rho_f}}$$

Discharge coefficient Cd

Discharge coefficient

$$C_d = C_v \cdot C_a$$

- C_d used when calculating mass flow $\dot{m} = C_d A \rho_f \sqrt{\frac{2\Delta P}{\rho_f}}$
- C_{v} 'velocity coefficient', used when calculating real injection velocity
- C_a 'area contraction coefficient', used when calculating effective nozzle hole diameter

Additional material

Intact Liquid Core

Additional material Intact liquid core ?

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Smallwood, G., and Gulder, O., Atom & sprays, 2000.

Additional material Intact liquid core ?

Desjardins et al, 2010

Reynolds number compared to real high-pressure sprays. DNS done with $U_{\text{exit}}{\sim}100 \text{m/s}$

Hilamo et al. 2010, Atom. Sprays

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Chesnel et al. 2011

Additional material Intact liquid core ?

(a) Re = 5000, We = 2000

- Turbulence inside the nozzle has bigger effect than previously thought of.
- Taken the experimental and numerical evidence, it seems very likely that in modern high-pressure injection systems, the intact liquid core is either very short or even non-existent.

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(b) Re = 5000, We = 5000

Desjardins et al, 2013, Atom. Sprays

Spray Equations

Lagrangian spray modeling

- Discrete computational points are tracked
- Spray equations (Ordinary differential equation (ODE)):

- Eq. (1) can be used to calculate the particle OK/Aalto 2019 position at each time instant.

Lagrangian spray modeling

• Parcel equation of motion F = ma

$$ma = F_{drag} = \frac{1}{2}\rho_g C_d A U^2$$

$$\frac{1}{6}\rho_p \pi d^3 \frac{du_p}{dt} = \frac{1}{2} (u_g - u_p) |u_g - u_p| \rho_g C_d \frac{\pi d^2}{4}$$

$$\rightarrow \quad \frac{du_p}{dt} = \frac{C_d}{\tau_p} \frac{Re_p}{24} (u_g - u_p)$$

$$\tau_p = \frac{\rho_l d^2}{18 \rho_g v_g}$$

Lagrangian spray modeling

or

$$u_{p,2} = u_{p,1} + \frac{C_d}{\tau_p} \frac{Re_p}{24} \left(u_g - u_{p,1} \right) dt$$
(2)

Eq. (2) can be used to update the particle velocity for each time instant, and then use Eq. (1) to update the position. OK

Droplet Drag

- The resistance a droplet/particle encounters, is due to shear and form drag
- Droplet drag coefficient Cd is calculated from

Droplet drag coefficient

Lagrangian spray modeling (Matlab)

Constant Cd=0.424 and constant gas phase velocity (0 m/s)

Droplet size is constant d=100micro-m

Gas density is 2 kg/m3 and liquid density is 800kg/m3.

Constant gas phase velocity may be a strong assumption.

Dispersed and continuous phase coupling

Stokes number

• Stokes number
$$St = \frac{\tau_p}{\tau_f}$$

• Ratio of the characteristic particle time scale (or the momentum relaxation time scale)

$$\tau_p = \frac{\rho_l d^2}{18 \rho_g v_g}$$

• To the characteristic flow time scale

$$\tau_f = \frac{D}{U_0}$$

 $\begin{array}{l} d \\ d \\ roplet \\ diameter \\ D \\ nozzle \\ diameter \\ U_0 \\ flow \\ velocity \end{array}$

Figure by G. Stiesch

Coupling regimes

Stokes number

- If $St \ll 1$ Particles will have plenty of time to adjust to changes in flow field velocity. Particle and fluid velocities nearly equal.
- If $St \gg 1$ Particle velocity will be little affected by flow velocity.

Source term in N-S Eqs.

 The flow field 'sees' the droplets via source terms in the N-S eqs. In the 2-way coupling regime, the source term M_d accounts for the coupling. Droplets 'see' the flow field effect from the velocity difference.

$$\frac{\partial \rho}{\partial t} + \frac{\partial \rho u_j}{\partial x_j} = 0$$

$$\frac{\partial \rho u_i}{\partial t} + \frac{\partial \rho u_i u_j}{\partial x_j} = -\frac{\partial}{\partial x_j} \left(p \delta_{ij} - \tau_{ij} \right) + \underline{M_d}$$

$$\underline{M_d} = \frac{1}{2} \rho_g C_d A |u_g - u_p| (u_g - u_p)$$

Group Work

Discuss in pairs for about 10 min:

- **1.** What information is needed to calculate droplet/particle Stokes number ?
- **2.** What information is not needed ?
- **3.** What does a particle St number mean ?
- 4. If you would be given a task to design a spray configuration that would give a homogeneous mixture (of droplets and air / vapor and air), what kind of a St number would you aim for and why ?

What is dense ?

What is dense ?

Fig. 5.2. Schematic illustration of different flow regimes

What is dense ?

• Typically assumed dense if $V_p/V_f > 10^{-3}$

- Assuming spray mass is divided homogenously in a cone –like volume
- From the schematic spray opening angle picture, the area ratio:

S < 90D assuming $\alpha = 20^{\circ}$

If *D*=90µm, S<8.1mm

D nozzle diameterS spray penetrationα spray opening angle

Additional material Nozzle Hole Size vs Average Fuel Concentration

- Average concentration $c_{ave} = \frac{m_s}{V_c}$
- Assumed constant injection rate $m_s \sim t D^2$
- The spray volume evolves as a cone
- Bottom area of the cone

a cone
$$V = \frac{1}{3}AS$$

 $A(t) \sim \left(\tan\left(\frac{\alpha}{2}\right)S\right)^2$

1

• Spray penetration assumes well known correlation $S(t) \sim (tD)^{1/2}$

• We obtain
$$c_{ave} \sim \frac{tD^2}{\frac{1}{3}\pi \left(\tan\left(\frac{\alpha}{2}\right)S\right)^2 S} \sim \left(\frac{D}{t}\right)^{\frac{1}{2}}$$

Vuorinen, V., thesis 2010

Additional material Nozzle Hole Size vs Average Fuel Concentration

Calculated result from the previous slide final result

D=0.12mm, α=15° or 20°

D=0.12mm or 0.36mm, α=20°

Turbulence in flow

- Turbulent flow consists of vortices (eddies) of different size
- The biggest vortices are of the size of the flow geometry
- The eddies are breaking up into smaller eddies. At the same time kinetic energy is transported from the bigger eddies to smaller ones
- At the smallest turbulent scale (Kolmogorov scale) kinetic energy is dissipated into heat
- Fluctuations
- Convection

Fig. 12.1. Formation of a turbulent shear layer (Roshko 1975)

Kuva 12.4 Turbulenssin kineettisen energian jakauma aaltoluvun k funktiona. Aaltoluku on pyörteen halkaisijan käänteisluku.

Turbulence in flow

- Turbulence \leftrightarrow Shear
- Turbulence is characterized by the following:
 - Three dimensional
 - Unsteady
 - Random
 - Strong vorticity
 - **Dissipative**
 - Strongly diffusive

Turbulence = increased mixing !

Jet development: Kelvin Helmholtz Instability and Spray Mixing

- Spray and jet turbulence is initiated by a Kelvin Helmholtz instability at the near nozzle region
- Growth rate of KH instability is related to the velocity difference of the fluids and to the wavelength.

KH Instability

KH Instability and Mixing

Figure V. Vuorinen

Additional Reading

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