



Aalto University
School of Science

PHYS-E0421 Solid State Physics

Period V, spring 2019

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Dielectric Properties of Solids

Magnetism

(Superconductivity)

Lecture 11, Monday 29.4.2019

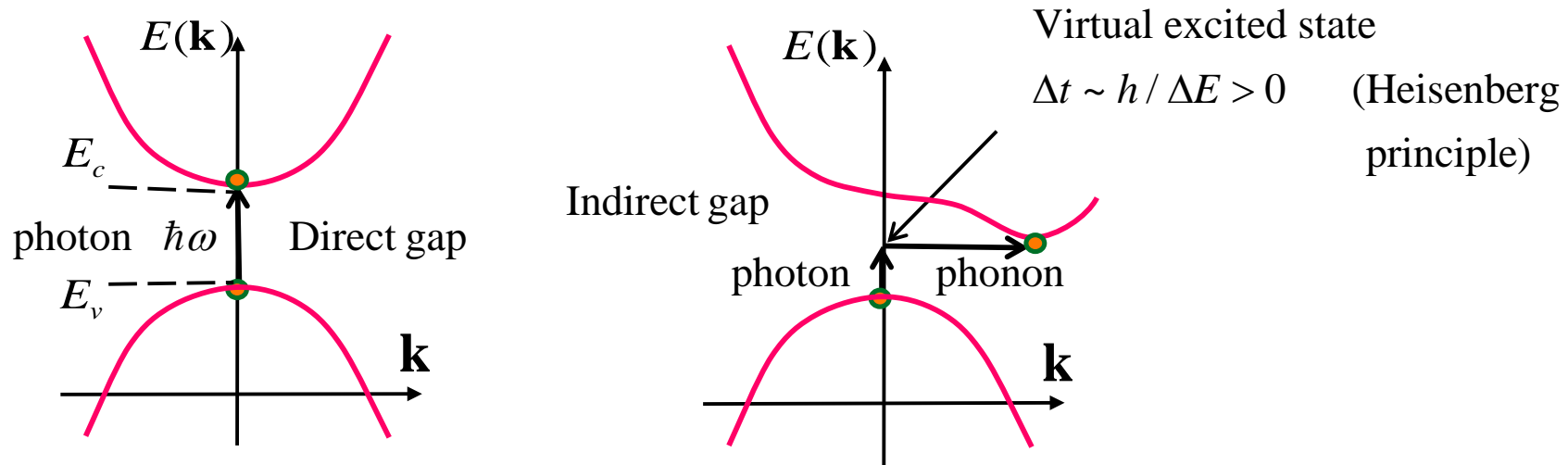
Dielectric properties of solids

- Response of materials to an external electric field
- Optical properties of ionic solids: Light – optical phonon interactions
- Measuring light – phonon interaction
- Optical properties of electrons in solids
 - Interband transitions, excitons
- Response of materials to an external electric field, Summary
- Spontaneous polarization
 - Polarization is ill-defined
 - Piezoelectric, pyroelectric, ferroelectric materials → nanoelectronics

TODAY

Optical Properties of Solids, Electron Contribution to $\epsilon(\omega)$ Interband Transitions (5.8.2)

E.g. electron - hole excitations in semiconductors



~Spherical symmetry for states around nuclei \rightarrow s-, p-, d- ... like bands

Selection rules



$\Delta l = \pm 1$ for electric - dipole - allowed transitions

$$E_f = E_i + \hbar\omega_{\text{photon}} \pm \hbar\omega_{\text{phonon}}$$

$$\mathbf{k}_f = \mathbf{k}_i + (\mathbf{k}_{\text{photon}}) \pm \mathbf{k}_{\text{phonon}}$$

~ 0

Often E_v consists of s - like states
 \Rightarrow excitations are to p - like states at E_c

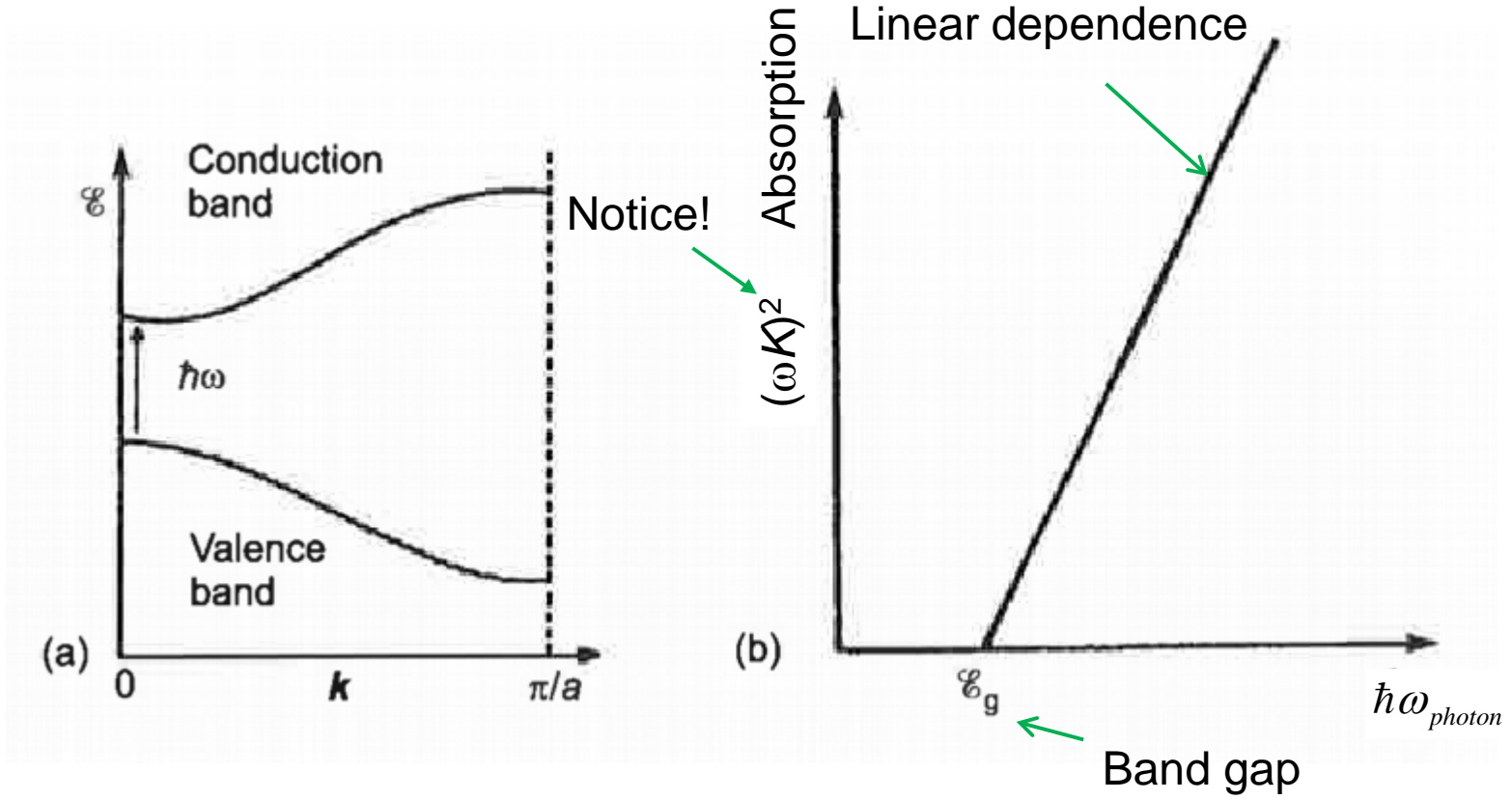
Interband Transitions

Direct-band-gap material

Vertical transitions

$$E_f = E_i + \hbar\omega_{\text{photon}}$$

$$\mathbf{k}_f = \mathbf{k}_i + \mathbf{k}_{\text{photon}}$$



Interband Transitions

Indirect-band-gap material

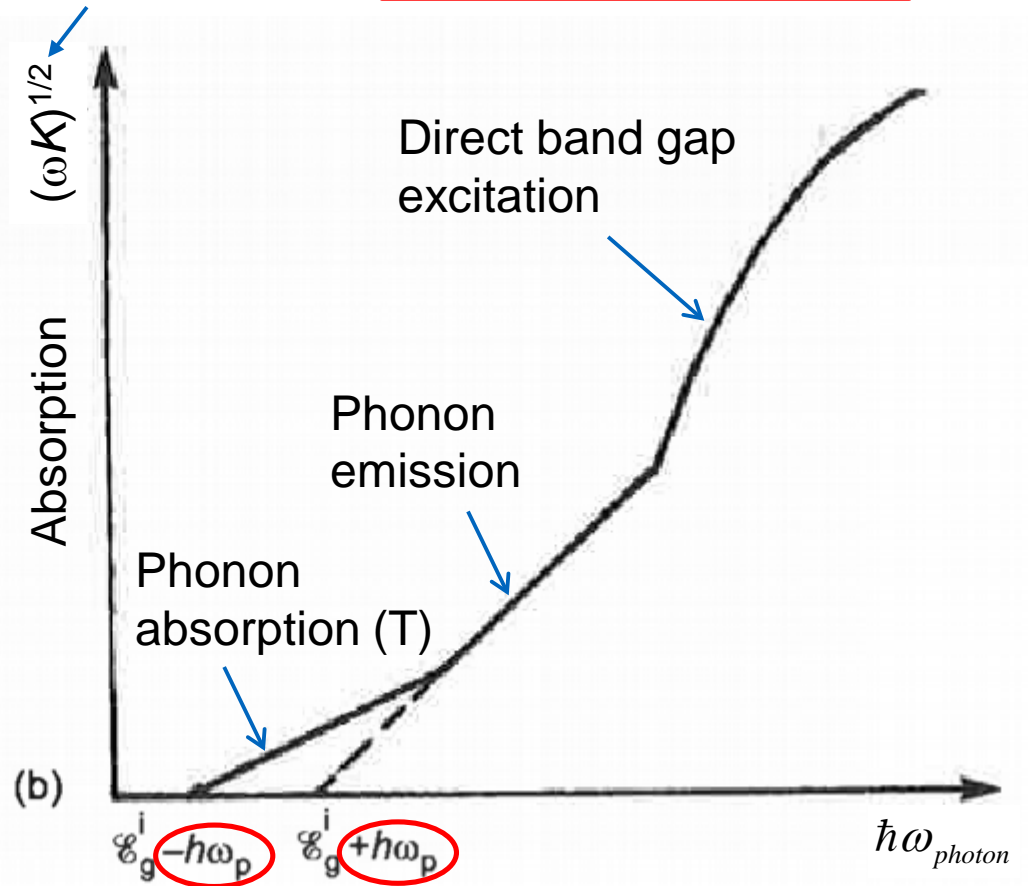
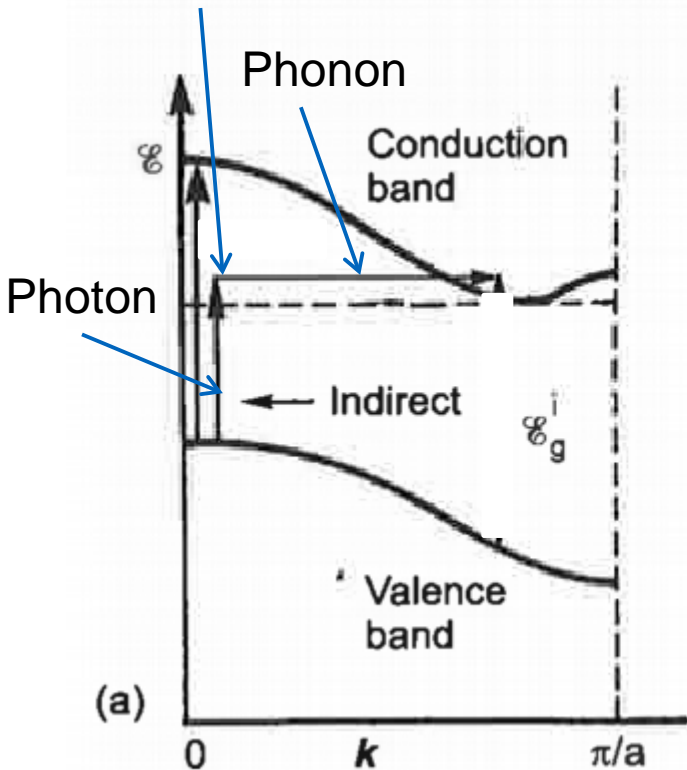
Transitions

$$E_f = E_i + \hbar\omega_{\text{photon}} \pm \hbar\omega_{\text{phonon}}$$

$$\mathbf{k}_f = \mathbf{k}_i + \mathbf{k}_{\text{photon}} \pm \mathbf{k}_{\text{phonon}}$$

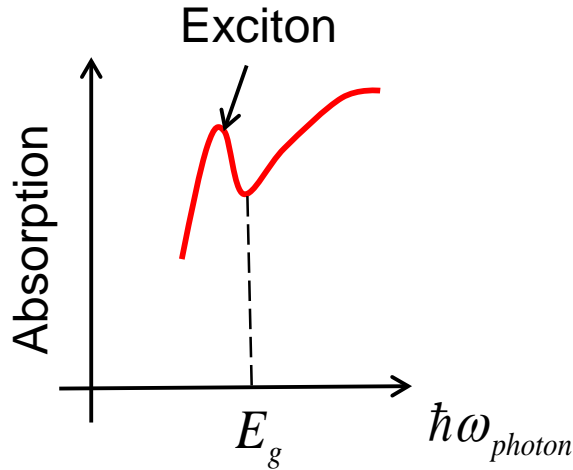
Virtual excited state

Notice!

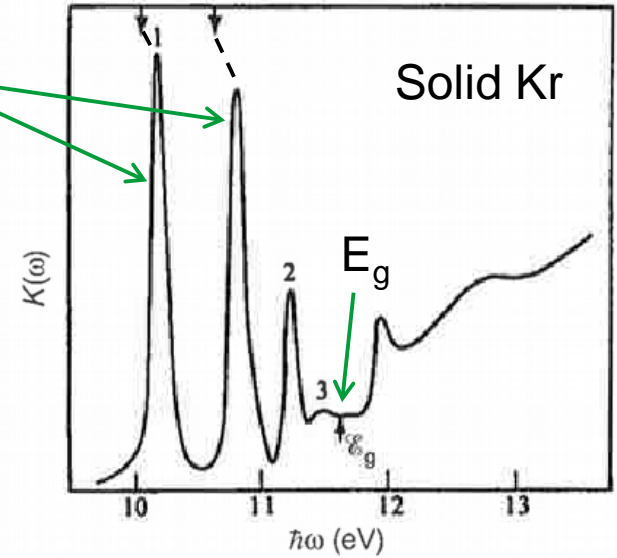


Optical Properties of Solids, Excitons (Elliott 5.8.3)

➔ Absorption of Electromagnetic Radiation in Insulators and Semiconductors:

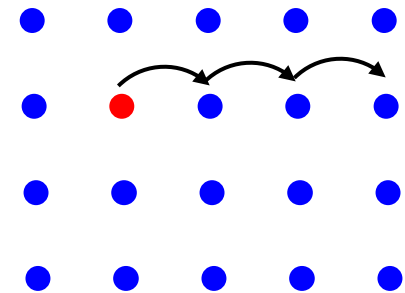


~Atomic $4p^6 \rightarrow 4p^55s$
spin-orbit splitting



A) Frenkel Exciton

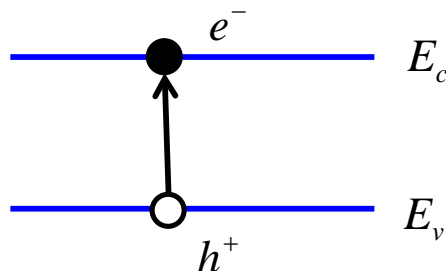
- one excited atom in lattice ((core)hole + bound electron)
- excitation hops from atom to atom
- in good insulators: rare gas solids, alkali halides



B) Mott-Wannier Exciton

2-step formation:

1) $e^- - h^+$ excitation



Excitation, e^- and h^+ ,
delocalized over several atoms

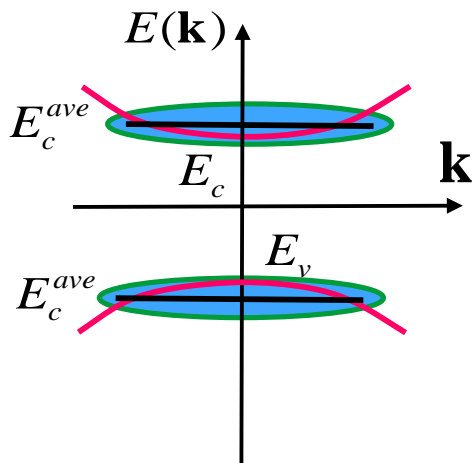
Potential

$$V^* = V^{\text{Ground state}}$$

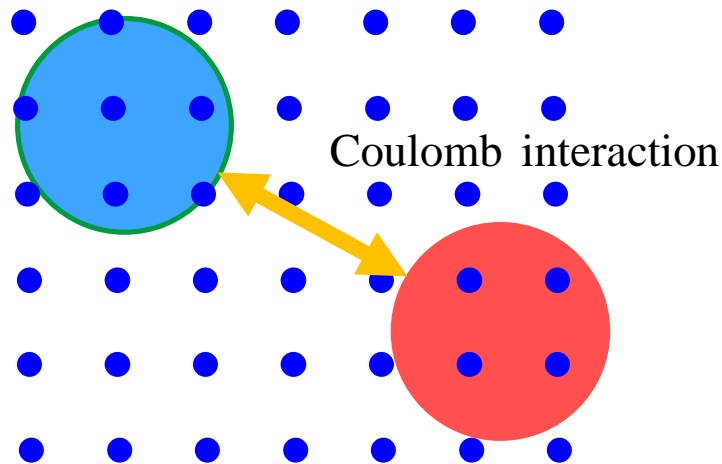
Excitation energy

$$\Delta E = E_c - E_v$$

2) Formation of wave packets



Localized e^-



Localized h^+

$$\Delta E = E_c^{ave} - E_v^{ave} - V_{Coul}(e^-, h^+)$$

Lowest excited state in lattice

B) Mott-Wannier Exciton

H-atom model

Exciton total energy

H-atom ground state

Center of mass energy
(Zero-point energy in nanostructures)

$$\Delta E = E_c^{ave} - E_v^{ave} - \frac{13.6 \text{ eV}}{(m_e/\mu^*)\epsilon^2 n^2} + \frac{\hbar^2 K^2}{2(m_e^* + m_h^*)}$$

Scaling

Screening of Coulomb int.

$$\frac{-1}{4\pi\epsilon_0 r} \rightarrow \frac{-1}{4\pi\epsilon_0 \epsilon r}$$

$$\rightarrow R^*/n^2$$

Effective Rydberg
Principal quantum number

Reduced effective
band mass

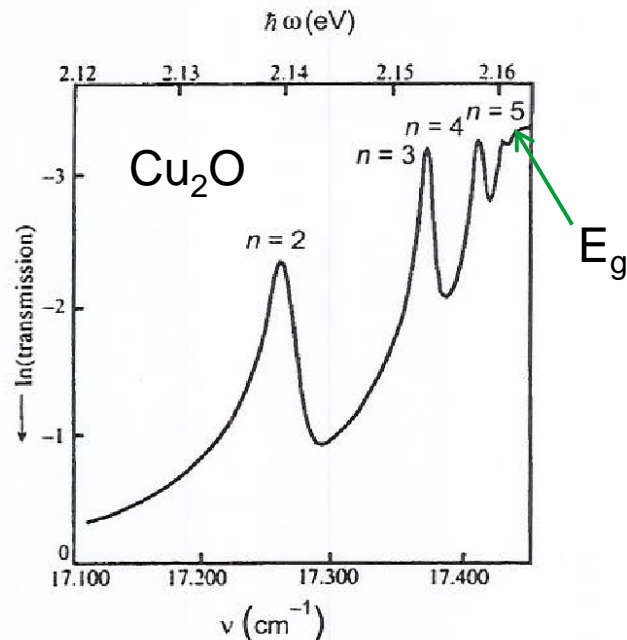
$$\frac{1}{\mu^*} = \frac{1}{m_e^*} + \frac{1}{m_h^*}$$

Model valid if

$$a_{exc} = \frac{m_e}{\mu^*} \epsilon a_0 \gg a_0$$

~Localization

~Interatomic distance



p-like Rydberg series

Frenkel vs. Mott-Wannier Excitons

Larger band gap → States are more localized

→ $\epsilon \rightarrow 1$, m_e^* and m_h^* increase

→ $a_{exc} = \frac{m_e}{\mu^*} \epsilon a_0 \rightarrow a_0$

$R^* = \frac{13.6 \text{ eV}}{(m_e/\mu^*) \epsilon^2}$ increases

→ Mott-Wannier → Frenkel excitons

Table 5.5 Exciton binding energies R^*

<i>Mott-Wannier type</i>		<i>Frenkel type</i>	
<i>Material</i>	R^* (meV)	<i>Material</i>	R^* (meV)
Si	14.7	BaO	56
Ge	4.2	KI	480
GaAs	4.9	KCl	400
GaP	3.5	KBr	400
InP	5.1	RbCl	440
CdS	29		
CdSe	15		
CdTe	11		

(After Burns (1985). Reproduced by permission of Academic Press, Inc.)

Optical Properties of Solids, Summary

(Dielectric Spectroscopy, Elliott 7.1.4)

Dielectric function $\varepsilon(\omega)$:

- ← Ionic motion, large $\mu_M \rightarrow$ low- ω region
- ← Electrons, small $m_e \rightarrow$ low ... high- ω region:
atomic polarizability, collective plasmon excitations (free electron gas),
damping (ε_2) by intra- and interband transitions

Damped harmonic oscillator

$$\mu \ddot{\mathbf{r}} + \gamma \mu \dot{\mathbf{r}} + k \mathbf{r} = q \mathbf{E}_{loc}(\omega)$$



→ Frequency dependencies of different mechanisms
(Quantum mechanics may be needed, e.g., for atomic polarizability)

$$\tilde{\varepsilon}(\omega) = \varepsilon(\infty) + \sum_i \frac{f_i}{\omega_i^2 - \omega^2 - i\gamma_i \omega}$$

ω_i = resonance frequency, i th mechanism
 f_i = relative weighting (oscillator strength)
 γ_i = damping constant (dissipation)

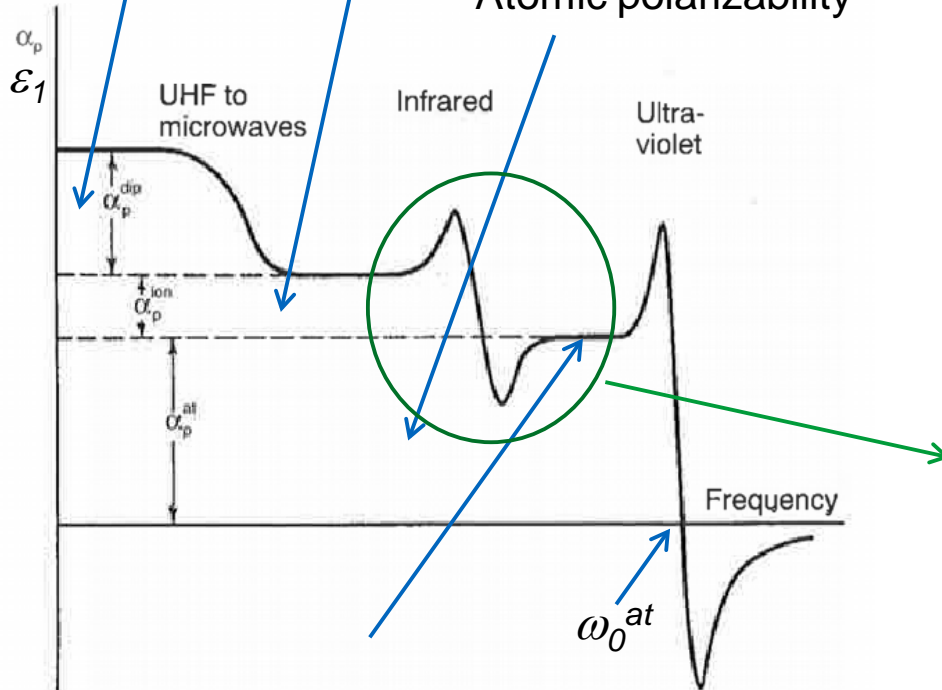
$$\varepsilon_1(\omega) = \varepsilon(\infty) + \sum_i \frac{f_i(\omega_i^2 - \omega^2)}{(\omega_i^2 - \omega^2)^2 + \gamma_i^2 \omega^2} \quad ; \quad \varepsilon_2(\omega) = \sum_i \frac{f_i \gamma_i \omega}{(\omega_i^2 - \omega^2)^2 + \gamma_i^2 \omega^2}$$

Optical Properties of Solids, Summary

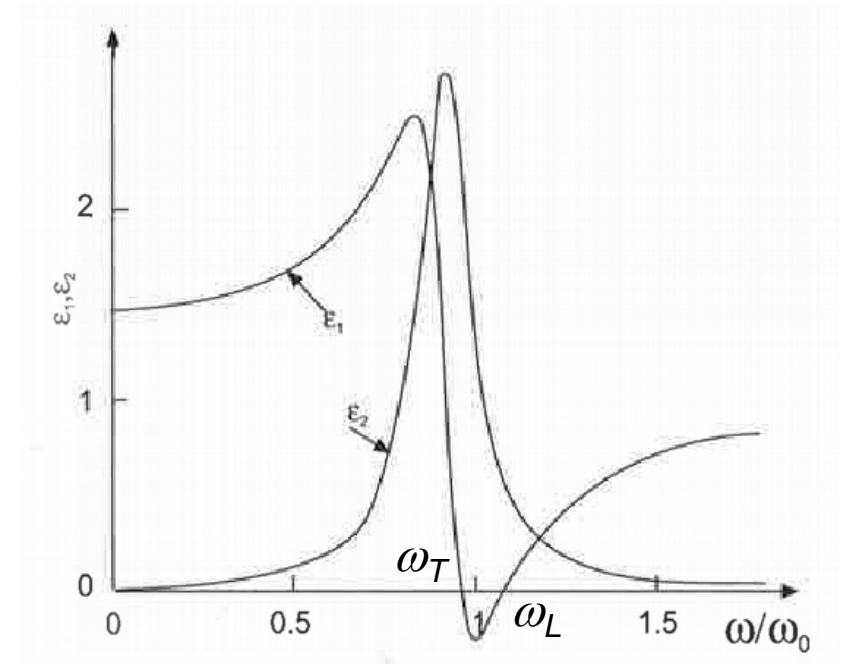
Orientation of dipoles
(no resonance)

Ion movement

Atomic polarizability



Ionic solids,
Quenching different processes with
increasing frequency



$\epsilon(\infty)$ for optical properties
of ionic solids

Light-Material Interaction, Macroscopic Maxwell's Equations

Harmonic (\mathbf{k} , ω) fields

$$\mathbf{E}, \mathbf{D}, \mathbf{B}, \mathbf{H}$$

Material parameters

$$\varepsilon(\omega), \sigma(\omega)$$

$$\mathbf{D} = \varepsilon_0 \varepsilon(\omega) \mathbf{E}$$

$$\mathbf{j} = \sigma(\omega) \mathbf{E}$$

Faraday

$$\nabla \times \mathbf{E} = -\frac{\partial \mathbf{B}}{\partial t}$$

Ampere-Maxwell

$$\nabla \times \mathbf{B} = \mu_0 \mathbf{j} + \mu_0 \frac{\partial \mathbf{D}}{\partial t}$$

Dispersion relation

$$k^2 = \frac{\omega^2 \varepsilon(\omega)}{c^2}$$

- Propagation or reflection of EM radiation at the vacuum-matter interface
- Coupled propagating radiation-matter modes, polaritons in ionic solids

Refractive index

$$\tilde{n} = \sqrt{\varepsilon}$$

Intensity decay

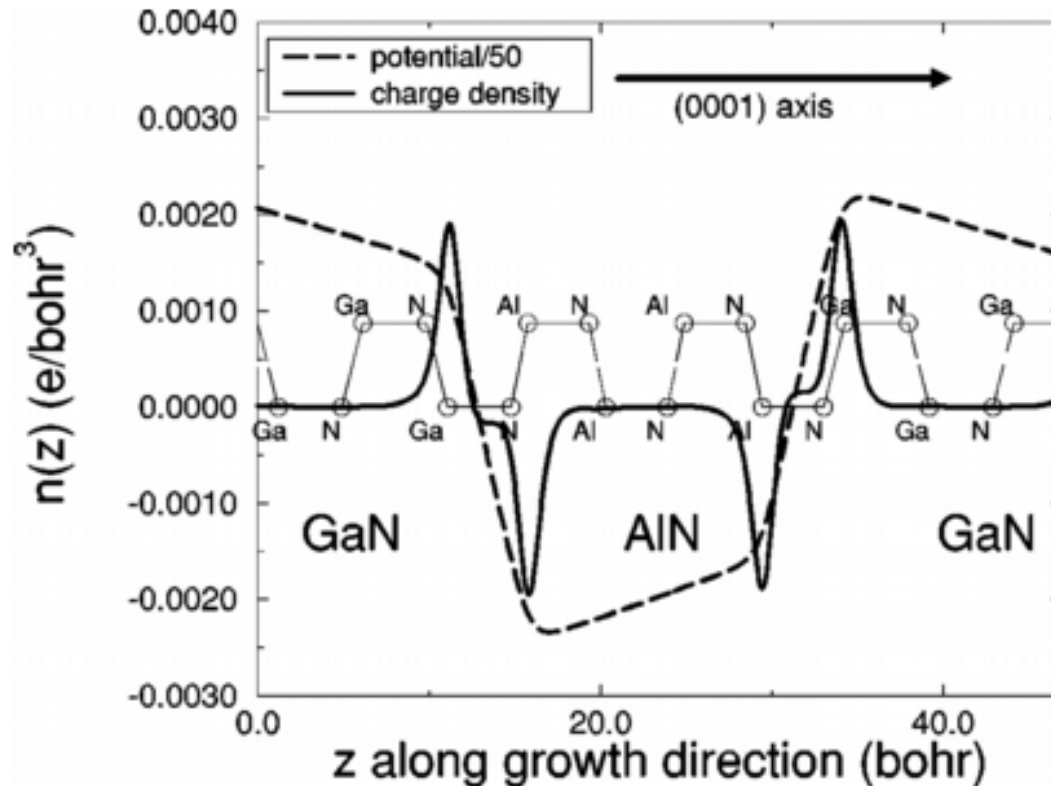
$$I \propto E_0^2 \exp\left[-\frac{2\omega\kappa_i}{c} x\right]$$

Reflectivity

$$R = \frac{(n_r - 1)^2 + \kappa_i^2}{(n_r + 1)^2 + \kappa_i^2}$$

Lecture Assignment Heterostructures of polarized materials

GaN and AlN have (non-cubic) wurtzite structure → spontaneous polarization in c (or (0001)) –direction → internal electric fields



Charge density and potential averaged over planes parallel to the interface.
F. Bernardini and V. Fiorentini, PRB, **57**, R9427 (1998).

Spontaneous Polarization

Polarization is a ill-defined bulk property

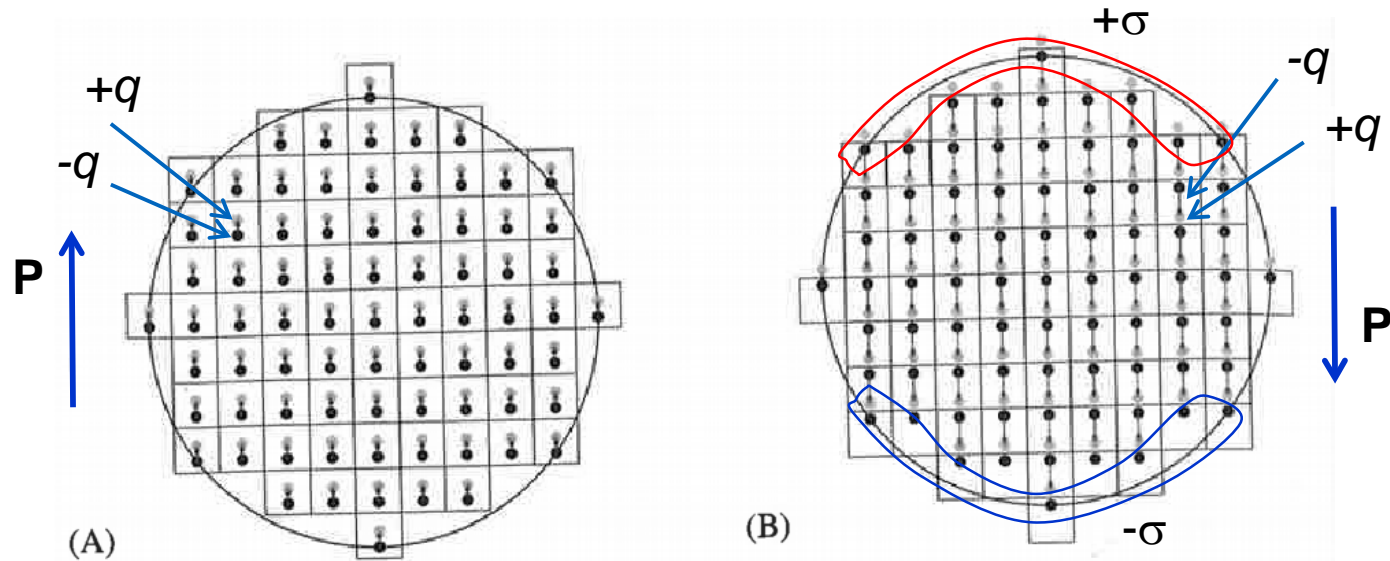


Figure 22.1. When charges are distributed throughout a solid, the polarization density depends upon how they are grouped. In (A), positive and negative charges have been grouped with their nearest neighbors. In (B), they have been grouped so that the polarization now points in the opposite direction. In compensation, there is now a substantial surface charge.

M.P. Marder, Condensed Matter Physics

Surface charge is important - But polarization should be a bulk property?

Changes in polarization can be measured and are sufficient

→ Solution of the dilemma

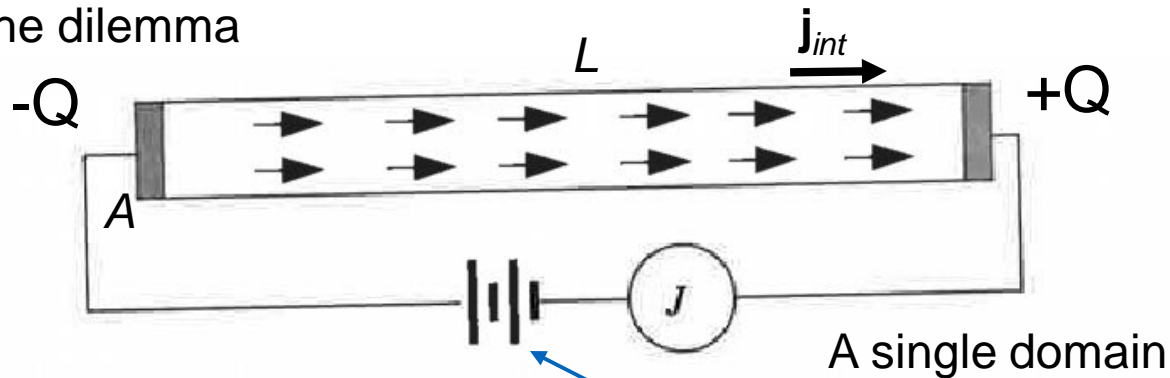


Figure 22.2. The spontaneous electric polarization of a sample is measured by cooling it below the ferroelectric transition temperature in an electric field, and then measuring the total charge that flows between the ends.

M.P. Marder, Condensed Matter Physics

Polarization

$$|\mathbf{P}| = QL/V = Q/A = \left| \int dt' \mathbf{j}_{\text{int}}(t') \right|$$

Absolute values of \mathbf{P} are actually not needed!

Berry's phase formalism (1984)
(based on electron system wavefunction)

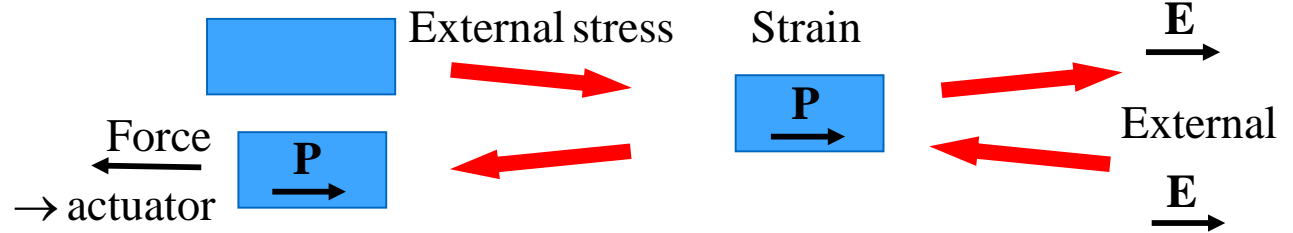
\mathbf{P} , e.g., from first – principles calculations

Distinguishing between metals and insulators

Spontaneous Polarization

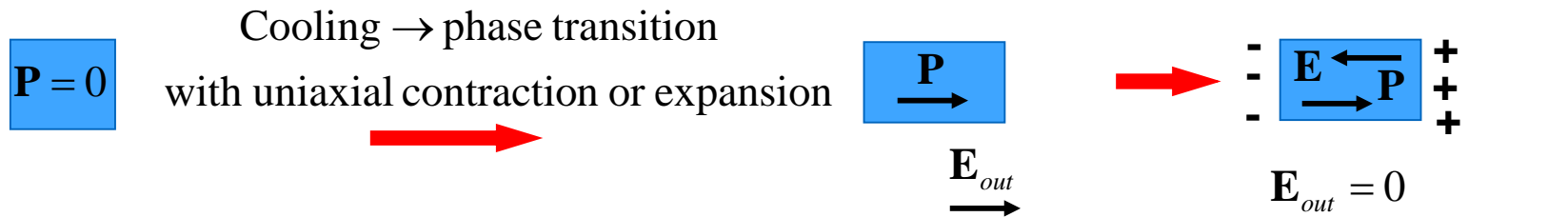
(Elliott 7.1.5)

Piezoelectric materials



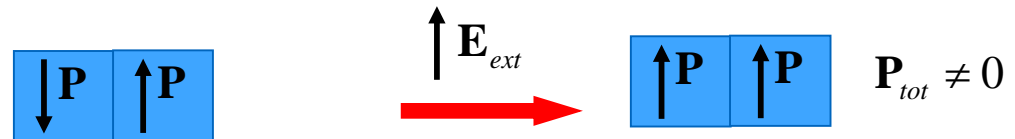
Subset: Pyroelectric materials

High temperature ("burning")



Subsubset: Ferroelectric materials

Domain structure with $P_{tot} = 0$



Ferroelectricity, example BaTiO₃

Elliott 7.1.5

Perovskite structure

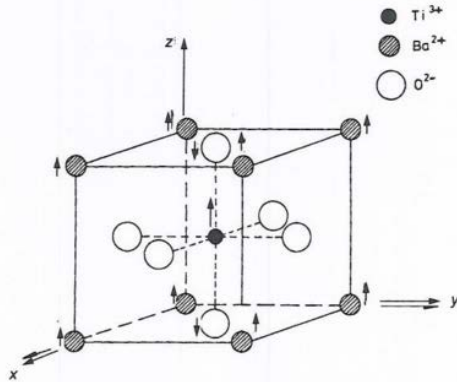
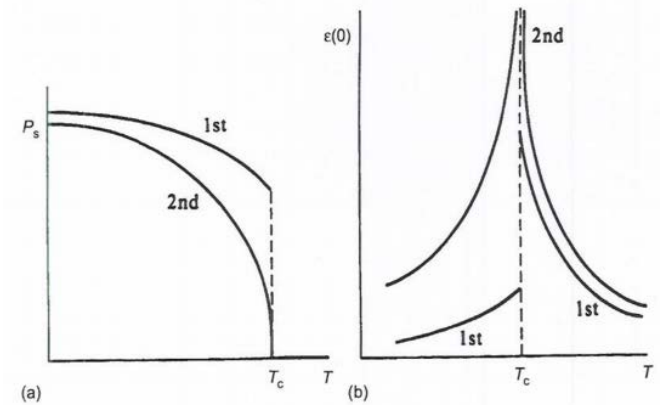


Fig. 7.13 The cubic perovskite (non-pyroelectric) structure of BaTiO₃ stable above 134°C. The arrows show the relative magnitude and direction of the atomic displacements in the transition to the pyroelectric tetragonal phase stable between -5 and 134°C. The oxygen atoms have been taken to be fixed (see also Fig. 7.19a).

Phase transitions

- First - order : - Discontinuous **P**
 - Discontinuous $\epsilon(0)$
 Second - order : - Continuous **P**
 - Divergent $\epsilon(0)$



$T > 134^\circ\text{C}$

Paraelectric (non-ferroelectric)
 cubic structure



$-5^\circ\text{C} < T < 134^\circ\text{C}$

Tetragonal structure with
 + ions moving \uparrow , - ions \downarrow



P
 Ferroelectric

$T = 134^\circ\text{C} = \text{Ferroelectric Curie - temperature}$

First - order phase transition

(Cf. Metal-organic perovskite solar cells)

Spontaneous Polarization

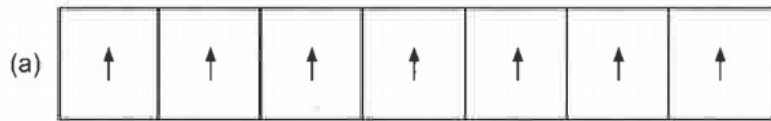
Elliott 7.1.5

Domain Structures and Hysteresis of Polarization

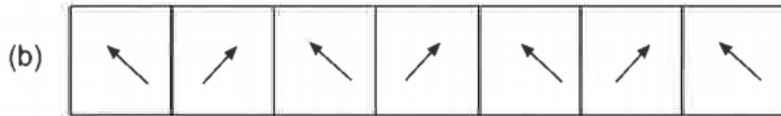
Movement of domain walls

→ Hysteresis

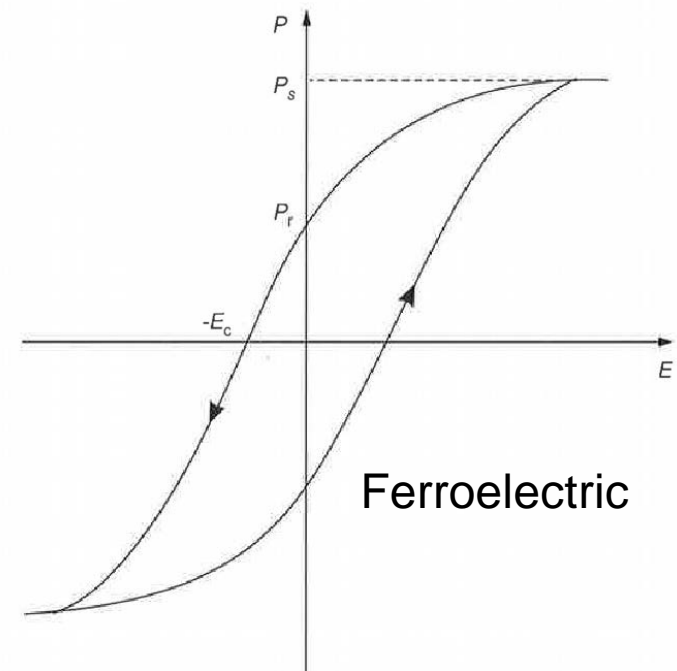
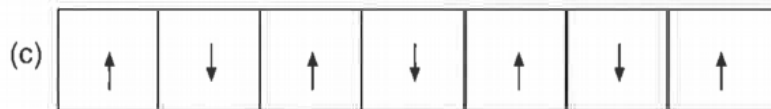
Ferroelectric



Ferrielectric



Anti-ferrielectric



Magnetic properties of materials

- Response of materials to an external magnetic field
 - Magnetic quantities, magnetism is quantum mechanics (home work)
 - Quantum mechanical description **TODAY**
 - Atomic diamagnetism, paramagnetism
 - Response of free electron gas
- Spontaneous magnetism
 - Exchange interaction, H₂ molecule, Heisenberg spin Hamiltonian
 - Mean-field approximation for ferromagnetism of magnetic moments
 - Spin waves (low-energy excitations)
 - Free electron gas
 - Stoner model for ferromagnetism of itinerant electrons
 - Antiferromagnetism
 - Domain structure

Models for materials' magnetic properties and phenomena

Magnetic properties of materials, Literature

- Griffiths Ch. 6 (Classical field theory) (Mycourses)
- Bolton-Freake (Eds.) Quantum mechanics of matter Chs. 5, 2, 1 (ebook)
- Elliott Ch. 7.2
- Ashcroft-Mermin, Chs. 31, (32), 33, App. P (Mycourses)
- Ibach-Lüth, Ch. 8
- Simon, Oxford solid state basics, Chs. 19-23: a short presentation (ebook)

Magnetic properties of materials

- Technology (engines, transformers, actuators, levitation, memory devices, antennas, electronics ...)
- ← Novel materials: (neo magnets), dilute magnetic semiconductors, multiferroic materials for spintronics, magnetic nanoparticles
- Theory: Electrons in solids, (nuclei)
 - single-electron band theory ↔ many interacting electrons
 - local effects ↔ collective effects

Magnetic quantities

(Elliott 7.2.1)

Magnetic dipole moment of a current on a loop

$$\boldsymbol{\mu} = iA\hat{\mathbf{n}} \quad [\boldsymbol{\mu}] = 1 \text{ Am}^2$$

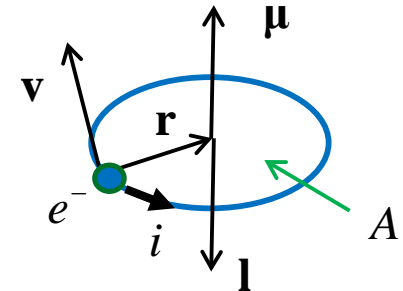
[E(7.78)]

$$|\boldsymbol{\mu}| = \frac{|-e|v}{2\pi r} \pi r^2 = \frac{|-e|}{2m_e} |\mathbf{r} \times \mathbf{p}| = \frac{|-e|}{2m_e} |\mathbf{l}|$$

Orbital angular momentum

$$\boldsymbol{\mu} = \frac{-e}{2m_e} \mathbf{l}$$

Electron on a circular path around a nucleus



Quantum Mechanics

$$\mu_z = -\mu_B \langle \hat{\mathbf{l}}_z \rangle$$

Expectation value in \hbar

$$\mu_B = \frac{e\hbar}{2m_e}$$

Bohr magneton

Spin

$$\mu_z = -\mu_B g_0 \langle \hat{\mathbf{s}}_z \rangle$$

$$\langle \hat{\mathbf{s}}_z \rangle = \pm 1/2$$

$g_0 \approx 2$ Gyromagnetic ratio

Many electrons in atom

$$\mu_z = -\mu_B \langle \hat{\mathbf{L}}_z + 2\hat{\mathbf{S}}_z \rangle = -\mu_B g \langle \hat{\mathbf{J}}_z \rangle$$

Landé factor

Total orbital and total spin angular momenta of all electrons

Total angular momentum

$$\hat{\mathbf{J}} = \hat{\mathbf{L}} + \hat{\mathbf{S}}$$

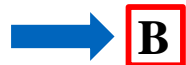
Magnetic quantities

(Elliott 7.2.1)

Torque on
magn. moment

$$\mathbf{T} = \boldsymbol{\mu} \times \mathbf{B}$$

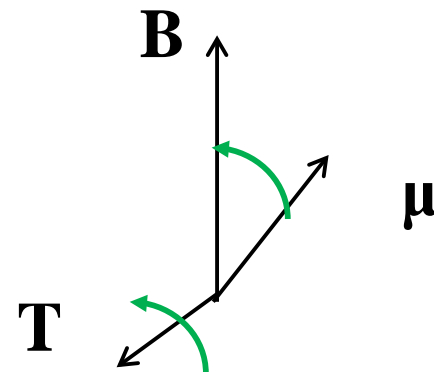
[E(7.80)]



Magnetic flux density

$$[\mathbf{B}] = 1 \text{ T}$$

(Classical physics)



Magnetic field strength

External
applied field

$$\mathbf{B}_0 = \mu_0 \mathbf{H}$$

$$[\mathbf{H}] = 1 \text{ A/m}$$

Vacuum permeability (SI system)

Magnetization = Magnetic moment / volume

[E(7.79)]

$$\mathbf{M} = \sum_i \boldsymbol{\mu}_i / V$$

$$[\mathbf{M}] = 1 \text{ A/m}$$

Total magnetic flux density

$$\mathbf{B} = \mathbf{B}_0 + \mu_0 \mathbf{M}$$



$$\mathbf{B} = \mu_0 (\mathbf{H} + \mathbf{M})$$

(Cf. homework)

[E(7.81)]

Magnetic quantities

(Elliott 7.2.1)

Magnetic susceptibility (Material property)

$$\mathbf{M} = \chi_m \mathbf{H}$$



$$\mu_0 \mathbf{M} = \chi_m \mathbf{B}_0$$



[E(7.83)]

$$\chi_m = \frac{\partial M}{\partial H} = \mu_0 \frac{\partial M}{\partial B_0}$$

Paramagnetic material



$$\chi_{m,p} > 0$$
$$\chi_{m,d} < 0$$

Diamagnetic material

Relative permeability

[E(7.82)]

$$\mathbf{B} = \mu_0 (\mathbf{H} + \mathbf{M}) = \mu_r \mu_0 \mathbf{H}$$

(cf. $\mathbf{D} = \epsilon_0 \mathbf{E} + \mathbf{P} = \epsilon_0 \epsilon \mathbf{E}$)



$$\mu_r = 1 + \chi_m$$

[E(7.84)]

For dia- and paramagnets

$$\chi_m \ll 1$$



$$\mu_r \approx 1$$
$$\mathbf{B}_{loc} \approx \mathbf{B}_0$$

Magnetization vs. thermodynamics

(AM Eqs. (31.1)-(31.8))

Energy of moment i in field \mathbf{B}_0

$$\mathbf{M} = \sum_i \boldsymbol{\mu}_i / V$$

$$\Delta E_i = -\boldsymbol{\mu}_i \cdot \mathbf{B}_0 \quad (\text{Classical } \boldsymbol{\mu}_i)$$

Homogeneous system of moments at $T=0$ in field \mathbf{B}_0

Total energy

$$E_0 = \sum_i -\boldsymbol{\mu}_i \cdot \mathbf{B}_0 = -V \mathbf{M}_0 \cdot \mathbf{B}_0$$



$$M_0(B_0) = -\frac{1}{V} \frac{\partial E_0}{\partial B_0}$$

$\mathbf{M}_0 \parallel \mathbf{B}_0$

$T > 0$, Several excited states n

$$E_n = \sum_i -\boldsymbol{\mu}_i^n \cdot \mathbf{B}_0 = -V \mathbf{M}_n \cdot \mathbf{B}_0$$

Maxwell-Boltzmann

statistics

$$M(B_0) = \frac{\sum_n M_n \exp(-E_n / k_B T)}{\sum_n \exp(-E_n / k_B T)}$$

$\mathbf{M}_n \parallel \mathbf{B}_0$

Partition function

$$Z(T) = \sum_n \exp[-E_n(\mathbf{B}_0) / k_B T]$$

Helmholtz free energy

$$F_H (= U - TS) = -k_B T \ln Z$$



$$M = -\frac{1}{V} \frac{\partial F_H}{\partial B_0} \quad [\text{E(7.74)}]$$

$$\mathbf{M} = \chi_m \mathbf{H} = \chi_m \mathbf{B}_0 / \mu_0$$



$$\chi_m = \mu_0 \frac{\partial M}{\partial B_0} = -\frac{\mu_0}{V} \frac{\partial^2 F_H}{\partial B_0^2}$$

Magnetism is a quantum phenomenon

Bohr – van Leeuwen theorem: Classical interacting system is insensitive to the magnetic field \mathbf{B} .

Proof.

$$M = -\frac{1}{V} \frac{\partial F_H}{\partial B_0}$$

$$F_H = -k_B T \ln Z$$

$$Z(T) = \sum_n \exp[-E_n(\mathbf{B}_0) / k_B T]$$

Classical mechanics

$$Z(T) \propto \iint \exp[-H(\mathbf{p}, \mathbf{r}) / k_B T] d\mathbf{p} d\mathbf{r}$$

Minimal substitution

$$\mathbf{p} = m_e \mathbf{v} - e\mathbf{A}$$

[E(7.76)]

Vector potential

$$\mathbf{B} = \nabla \times \mathbf{A}$$

Hamiltonian function

$$\rightarrow H(\mathbf{p}, \mathbf{r}) = \frac{1}{2m_e} (\mathbf{p} + e\mathbf{A})^2 + V(\mathbf{r}) \quad [\text{E}(7.77)]$$

\mathbf{p} - integral over an ∞ p - volume
 $e\mathbf{A}$ finite shift of origin in integral



$$Z \neq Z(\mathbf{A})$$

or $Z(\mathbf{B}_0)$



$$\mathbf{M} = 0$$

Magnetic properties

- Response of materials to an external magnetic field
 - Magnetic quantities, magnetism is quantum mechanics (home work)
 - Quantum mechanical description
 - Atomic diamagnetism, paramagnetism (lecture work) **The next lecture**
 - Response of free electron gas
- Spontaneous magnetism Ferromagnetism and antiferromagnetism
 - Exchange interaction, H₂ molecule, Heisenberg spin Hamiltonian
 - Mean-field approximation for ferromagnetism of magnetic moments
 - Spin waves (low-energy excitations)
 - Free electron gas
 - Stoner model for ferromagnetism of itinerant electrons
 - Antiferromagnetism
 - Domain structure